## RAPID AND CONVERGENT SYNTHESIS OF A 2,4'-LINKED TRI-OXAZOLE IN AN APPROACH TO POLY-OXAZOLES

# Author's name(s) : Daniel D. Caspi, Haiming Zhang, Scott C. Virgil, Fabian M. Piller, and Brian M. Stoltz\*

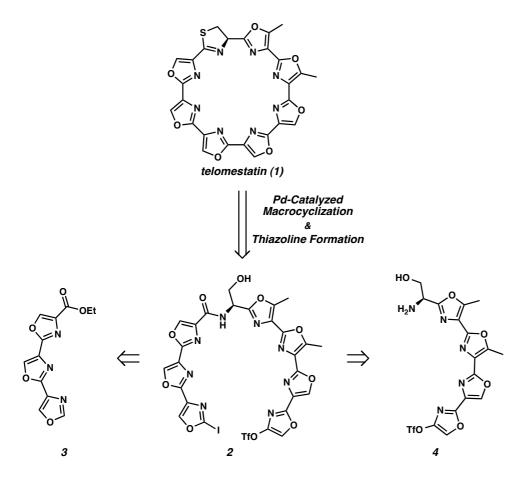
The Warren and Katharine Schlinger Laboratory of Chemistry and Chemical Engineering, California Institute of Technology, Pasadena, California 91125, United States stoltz@caltech.edu

**Abstract** – A rapid and convergent synthesis of a 2,4'-linked tri-oxazole using a Negishi coupling is described.

Due to the biological importance of polyoxazole-containing compounds, a multitude of synthetic approaches have been described.<sup>1</sup> While preparations of mono- and di-substituted oxazoles are common, reports describing the synthesis of longer oxazole chains are much more scarce.<sup>2,3</sup> We became interested in efficient routes to 2,4'-polyoxazoles during studies on the total synthesis of telomestatin (**1**, Scheme 1), a potent telomerase inhibitor.<sup>4,5,6,7</sup> Herein, we present a rapid and convergent synthesis of a 2,4'-linked tri-oxazole using a Negishi coupling.

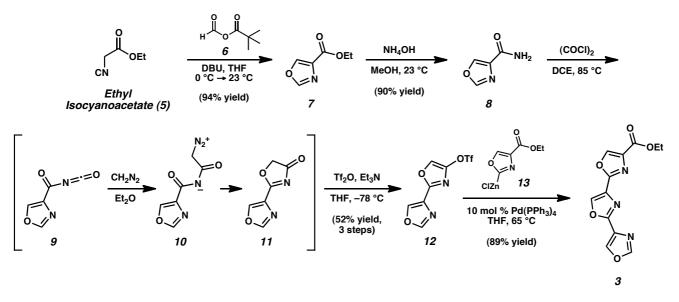
Our retrosynthetic analysis for telomestatin (1) is featured in Scheme 1. On the basis of the reported telomestatin (1) syntheses,<sup>5</sup> we envisioned a late-stage installation of the sulfur moiety and the thiazoline ring. Additionally, in order to maximize synthetic efficiency, we sought to complete the final aryl–aryl linkage of **2** and induce macrocyclization using a palladium-catalyzed cross-coupling.<sup>8,9</sup> This maneuver would also allow for a high degree of convergency by dividing the molecule into two roughly equal halves. Disconnection across the amide bond in **2** then reveals tri-oxazole fragment **3** and tetrakis-oxazole amino alcohol **4**.

Scheme 1



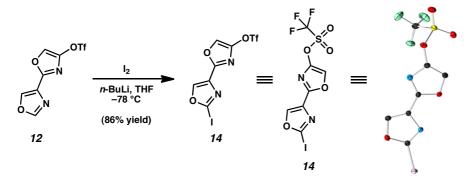
Our synthesis of the left-hand tri-oxazole portion (3) of telomestatin (1) began with the preparation of known oxazole ester 7. Exposure of ethyl isocyanoacetate (5) to mixed anhydride  $6^{10}$  and DBU led to a high yield of oxazole ester 7,<sup>11</sup> which was smoothly converted to amide 8 by the action of aqueous ammonia in methanol (Scheme 2). Conversion to bis-oxazole triflate 12 was achieved by means of a three-step sequence, which commenced by heating amide 8 in the presence of oxalyl chloride to give rise to acyl isocyanate 9.

#### Scheme 2



Subjection of acyl isocyanate 9 to anhydrous, alcohol-free diazomethane dried over sodium metal<sup>12</sup> led to in situ production of 10, which rapidly cyclized with loss of nitrogen to form oxazolone 11.<sup>13</sup> Treatment of this intermediate with Tf<sub>2</sub>O and amine base produced bis-oxazole triflate 12 in 52% yield over 3 steps.<sup>14</sup> Further confirmation of the structural identity was achieved by single crystal X-ray diffraction upon conversion to the iodo derivative (12  $\rightarrow$  14, Scheme 3).<sup>15</sup>

Scheme 3



A wide range of cross-couplings of appropriate mono- and bis-oxazole subunits were investigated to prepare the desired tri-oxazole fragment (3), including Stille, Suzuki, and Negishi protocols.<sup>16,17</sup> Ultimately, the Negishi approach proved to be the most robust to accomplish this union.<sup>16</sup> The necessary zinc reagent (13) for this reaction could be prepared from 7 via a deprotonation/quenching event with LiHMDS and ZnCl<sub>2</sub>, furnishing tri-oxazole 3 after successful aryl fusion with bis-oxazole 12.<sup>18</sup> This approach was also used in a similar fashion to prepare a tetrakis-oxazole.<sup>19</sup>

In conclusion, we have presented an efficient and convergent synthesis of a tri-oxazole fragment. Existing studies to utilize this methodology toward a total synthesis of telomestatin are ongoing in our laboratory.

### ACKNOWLEDGEMENTS

The authors gratefully acknowledge the NIH-NIGMS (R01 GM65961-01), Eli Lilly (predoctoral fellowship to D.D.C.), AstraZeneca, Boehringer Ingelheim, Johnson & Johnson, Pfizer, Merck, Amgen, Research Corporation, Roche, and GlaxoSmithKline for generous funding. We also thank Tim Dong and J. T. Mohr for experimental assistance.

#### **REFERENCES (AND NOTES)**

<sup>1</sup> For a review on oxazole-containing natural products, see: V. S. C. Yeh, *Tetrahedron*, 2004, **60**, 11995–12042.

<sup>2</sup> For a review on directly linked polyazoles, see: (a) E. Riego, D. Hernández, F. Albericio, and M. Álvarez, *Synthesis*, 2005, 1907–1922; (b) C. D. Turner and H. S. Liang, *Curr. Org. Chem.*, 2011, **15**, 2846-2870.

<sup>3</sup> For excellent and comprehensive discussions on oxazole methodologies, see: (a) E. F. Flegeau, "Palladium Cross-Couplings of Oxazoles". Ph.D. dissertation, Edinburgh University, 2008; (b) E. Charalambidou, "Synthesis of oxazole ring systems for use as inhibitors of telomerase function". Masters thesis, University College London, 2008.

<sup>4</sup> K. Shin-ya, K. Wierzba, K. Matsuo, T. Ohtani, Y. Yamada, K. Furihata, Y. Hayakawa, and H. Seto, *J. Am. Chem. Soc.*, 2001, **123**, 1262–1263.

<sup>5</sup> For the total synthesis of telomestatin, see: (a) S. Yamada, K. Shigeno, K. Kitagawa, S. Okajima, and T. Asao (Taiho Pharmaceutical Co. Ltd., Sosei Co. Ltd.). WO2002048153, *Chem. Abstr.* 2002, **137**, 47050; (b) T. Doi, M. Yoshida, K. Shin-ya, and T. Takahashi *Org. Lett.*, 2006, **8**, 4165–4167; (c) For the synthesis of *(S)*-telomestatin, see: T. Doi, K. Shibata, M. Yoshida, M. Takagi, M. Tera, K. Nagasawa, K. Shin-ya, and T. Takahashi, *Org. Biomol. Chem.*, 2011, **9**, 387–393; (d) For a formal total synthesis, see: J. Linder, T. P. Garner, H. E. L. Williams, M. S. Searle, and C. J. Moody, *J. Am. Chem. Soc.*, 2011, **133**, 1044–1051.

<sup>6</sup> For approaches toward telomestatin, see: (a) N. Endoh, K. Tsuboi, R. Kim, Y. Yonezawa, and C. Shin, *Heterocycles*, 2003, **60**, 1567–1572; (b) S. K. Chattopadhyay and S. Biswas, *Tetrahedron Lett.*, 2006, **47**, 7897–7900; (c) S. K. Chattopadhyay, S. Biswas, and B. K. Pal, *Synthesis*, 2006, 1289–1294; (d) C. M. Marson and M. Saadi, *Org. Biomol. Chem.*, 2006, **4**, 3892–3893.

<sup>7</sup> D. D. Caspi, "The Adaptive Nature of Palladium Reactivity in Synthesis", Ph.D. dissertation, California Institute of Technology, 2008.

<sup>8</sup> For examples of palladium-catalyzed biaryl formation for macrocyclization, see: (a) H. K. Patel, J. D. Kilburn, G. J. Langley, P. D. Edwards, T. Mitchell, and R. Southgate, *Tetrahedron Lett.*, 1994, 35, 481–484; (b) A.-C. Carbonnelle and J. Zhu, *Org. Lett.*, 2000, 2, 3477–3480; (c) K. C. Nicolaou, P. G. Bulger, and D. Sarlah, *Angew. Chem., Int. Ed.*, 2005, 44, 4442–4489; (d) J. Blankenstein and J. Zhu, *Eur. J. Org. Chem.*, 2005, 1949–1964.

<sup>9</sup> Heteroaryl-heteroaryl cross-couplings, especially those containing nitrogen heterocycles, often exhibit poor reactivity. For relevant discussions, see: (a) N. Kudo, M. Perseghini, and G. C. Fu, *Angew*. *Chem.*, *Int. Ed.*, 2006, **45**, 1282–1284 and references therein. (b) K. L. Billingsley, K. W. Anderson, and S. L. Buchwald, *Angew. Chem.*, *Int. Ed.*, 2006, **45**, 3484–3488 and references therein.

(a) M. Behforouz, J. Haddad, W. Cai, M. B. Arnold, M. Farahnaz, A. C. Sousa, and M. A. Horn, J. Org. Chem., 1996, 61, 6552–6555; (b) W. K. Fife and Z.-d. Zhang, J. Org. Chem., 1986, 51, 3744–3746;
(c) C. R. Hutchinson and A. D. Harmon, J. Org. Chem., 1975, 40, 3474–3480.

<sup>11</sup> C. M. Shafer and T. F. Molinski, *Heterocycles*, 2000, **53**, 1167–1170.

<sup>12</sup> T. J. de Boer and H. J. Backer, *Org. Synth.*, 1956, **36**, 16–19.

For the synthesis of oxazolones and oxazole triflates, see reference 3 and see: (a) J. C. Sheehan and P. T. Izzo, J. Am. Chem. Soc., 1949, **71**, 4059–4062; (b) Y. S. Rao and R. Filler, Chem. Commun., 1970, 1622; (c) R. M. Rodehorst and T. H. Koch, J. Am. Chem. Soc., 1975, **97**, 7298–7304; (d) T. R. Kelly and F. Lang, *Tetrahedron Lett.*, 1995, **36**, 5319–5322; (e) J. V. Schaus and J. S. Panek, Org. Lett., 2000, **2**, 469–471; (f) A. B. Smith, III, K. P. Minbiole, and S. Freeze, Synlett, 2001, 1739–1742; (g) A. B. Smith, III, K. P. Minbiole, P. R. Verhoest, and M. Schelhaas, J. Am. Chem. Soc., 2001, **123**, 4834–4836; (h) A. B. Smith, III, K. P. Minbiole, P. R. Verhoest, and M. Schelhaas, J. Am. Chem. Soc., 2001, **123**, 4834–4836; (h) A. B. Smith, III, K. P. Minbiole, P. R. Verhoest, and M. Schelhaas, J. Am. Chem. Soc., 2001, **123**, 10942–10953; (i) T. R. Kelly and F. Lang, J. Org. Chem., 1996, **61**, 4623–4633; (j) N. F. Langille, L. A. Dakin and J. S. Panek, Org. Lett., 2002, **4**, 2485–2488; (k) B. M. Trost, K. Dogra, and M. Franzini, J. Am. Chem. Soc., 2004, **126**, 1944–1945; (l) A. B. Smith, III, T. M. Razler, G. R. Pettit, and J.-C. Chapuis, Org. Lett., 2005, **7**, 4403–4406; (m) A. B. Smith, III, T. M. Razler, R. M. Meis, and G. R. Pettit, Org. Lett., 2006, **8**, 797–799; (n) A. G. M. Barrett and J. T. Kohrt, Synlett, 1995, 415-416.

<sup>14</sup> Although no characterization data or experimental details were provided, the synthesis of compound **12** via a related route was reported. See H. Araki, T. Katoh, and M. Inoue, *Tetrahedron Lett.*, 2007, **48**, 3713–3717. An alternate route to **12** from **8** is also described in the supporting information (23% overall yield).

<sup>15</sup> See supporting information; iodobis-oxazole triflate **14** is shown with 50% probability ellipsoids (Note: Only *Molecule A* is depicted). Crystallographic data have been deposited at the CCDC, 12 Union

Road, Cambridge CB2 1EZ, UK, and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 282586.

<sup>16</sup> For useful reports of oxazole cross-couplings, see reference 3 and references therein.

<sup>17</sup> For Negishi cross-couplings of oxazoles, see: (a) N. K. Harn, C. J. Gramer, and B. A. Anderson, *Tetrahedron Lett.*, 1995, **36**, 9453–9456; (b) B. A. Anderson and N. K. Harn *Synthesis*, 1996, 583–585;
(c) B. A. Anderson, L. M. Becke, R. N. Booher, M. E. Flaugh, N. K. Harn, T. J. Kress, D. L. Varie, and J. P. Wepsiec, *J. Org. Chem.*, 1997, **62**, 8634–8639; (d) E. Vedejs and L. M. Luchetta, *J. Org. Chem.*, 1999, **64**, 1011–1014; (e) M. R. Reeder, H. E. Gleaves, S. A. Hoover, H. R. Imbordino, and J. Pangborn, *J. Org. Process Res. Dev.*, 2003, **7**, 696–699.

<sup>18</sup> Although no characterization data or experimental details were provided, the synthesis of **3** was reported in reference 13. The synthesis of **3** has also been reported by: E. F. Flegeau, M. E. Popkin, and M. F. Greaney, *Org. Lett.*, 2008, **10**, 2717–2720.

<sup>19</sup> This approach was also successful in producing small quantities of tetrakis-oxazole **ii**, which was prepared from triflate **12** and Negishi reagent **i**. Negishi reagent **i** was made in a two-step process: ethoxycarbonylation of **12**, followed by metalation in a similar fashion as **13**.

