

Supporting Information

Palladium-Catalyzed Decarbonylative Dehydration for the Synthesis of α -Vinyl Carbonyl Compounds and Total Synthesis of (–)-Aspewentins A, B, and C^{**}

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Materials and Methods

Unless otherwise stated, reactions were performed in flame-dried glassware under a nitrogen atmosphere using dry, deoxygenated solvents, or under vacuum without the use of solvents. Solvents were dried by passage through an activated alumina column under argon.¹ Reaction progress was monitored by thin-layer chromatography (TLC) or ¹H NMR analysis of the crude reaction mixture. TLC was performed using E. Merck silica gel 60 F254 precoated glass plates (0.25 mm) and visualized by UV fluorescence quenching, p-anisaldehyde, phosphomolybdic acid, or KMnO₄ staining. Silicycle SiliaFlash® P60 Academic Silica gel (particle size 40-63 nm) was used for flash chromatography. ¹H NMR and ¹³C NMR spectra were recorded on a Varian Inova 500 (500 MHz and 126 MHz, respectively) or a Bruker CryoProbe Prodigy 400 spectrometer (400 MHz and 101 MHz, respectively) and are reported relative to residual CHCl₃ (δ 7.26 ppm and δ 77.16 ppm, respectively). Data for ¹H NMR are reported as follows: chemical shift (δ ppm) (multiplicity, coupling constant (Hz), integration). Multiplicities are reported as follows: s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, sept = septuplet, m =multiplet, br s = broad singlet, br d = broad doublet, app = apparent. Data for ${}^{13}C$ NMR are reported in terms of chemical shifts (δ ppm). IR spectra were obtained by use of a Perkin Elmer Spectrum BXII spectrometer using thin films deposited on NaCl plates and reported in frequency of absorption (cm⁻¹). Optical rotations were measured with a Jasco P-2000 polarimeter operating on the sodium D-line (589 nm), using a 100 mm path-length cell and are reported as: $\left[\alpha\right]_{D}^{T}$ (concentration in g/100 mL, solvent). Analytical chiral GC was performed with an Agilent 6850 GC utilizing a G-TA (30 m x 0.25 mm) column (1.0 mL/min carrier gas flow). Analytical chiral HPLC was performed with an Agilent 1100 Series HPLC utilizing a Chiralpak (AD-H or AS) or Chiralcel (OD-H, OJ-H, or OB-H) columns (4.6 mm x 25 cm) obtained from Daicel Chemical Industries, Ltd. Analytical chiral SFC was performed with a Mettler SFC supercritical CO₂ analytical chromatography system utilizing Chiralpak (AD-H, AS-H or IC) or Chiralcel (OD-H, OJ-H, or OB-H) columns (4.6 mm x 25 cm) obtained from Daicel Chemical Industries, Ltd. High resolution mass spectra (HRMS) were provided by the California Institute of Technology Mass Spectrometry Facility using a JEOL JMS-600H High Resolution Mass Spectrometer (EI+ or FAB+), or obtained with an Agilent 6200 Series TOF using Agilent G1978A Multimode source in electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), or mixed ionization mode (MM: ESI-APCI).

Reagents were purchased from Sigma-Aldrich, Acros Organics, Strem, or Alfa Aesar and used as received unless otherwise stated. i-Pr₂NH was distilled from calcium hydride prior to use. (*S*)-*t*-BuPHOX was prepared by a known method.²

List of Abbreviations:

9-BBN – 9-borabicyclo(3.3.1)nonane, Ac – acetyl, Cy – cyclohexyl, Bz – benzoyl, DBU – 1,8-Diazabicycloundec-7-ene, DIBAL – diisobutylaluminium hydride, ee – enantiomeric excess, HPLC – high-performance liquid chromatography, LDA – lithium diisopropylamide, MTBE – methyl *tert*-butyl ether, LiHMDS – lithium hexamethyldisilazide, NMP – *N*-methylpyrrolidone, SFC – supercritical fluid chromatography, TLC – thin-layer chromatography, TEMPO – 2,2,6,6-Tetramethylpiperidin-1-yloxy, TFA – trifluoroacetic acid, THF – tetrahydrofuran, Ts – *p*toluenesulfonyl

Preparation of Carboxylic Acid Substrates



3-(1-(Ethoxycarbonyl)-2-oxocyclopentyl)propanoic acid (8a). A flame-dried 100 mL roundbottom flask was charged with a magnetic stir bar, anhydrous MeCN (30 mL), β -keto ester SI-1 (2.9 mL, 20 mmol, 1 equiv), tert-butyl acrylate (3.0 mL, 20.6 mmol, 1.02 equiv), and DBU (0.15 mL, 1 mmol, 0.05 equiv). The light yellow reaction mixture was stirred at 23 °C. After 12 h, TLC analysis indicated complete consumption of starting material. Solvents were evaporated, and the crude residue was purified by flash column chromatography on silica gel $(10 \rightarrow 16 \rightarrow 25\%)$ EtOAc in hexanes) to afford a colorless oil (5.22 g). To a solution of this oil (1.42 g) in CH₂Cl₂ (4 mL) was added trifluoroacetic acid (4 mL). The reaction mixture was stirred at 23 °C for 30 min and concentrated under reduced pressure. Removal of remaining trifluoroacetic acid by azeotropic evaporation from toluene (5 mL x 5) afforded carboxylic acid 8a (1.14 g, 92% yield over 2 steps) as a viscous colorless oil. $R_f = 0.1$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.86 (br s, 1H), 4.17 (q, J = 7.1 Hz, 2H), 2.59 (ddd, J = 16.3, 10.2, 5.7 Hz, 1H), 2.53– 2.36 (m, 3H), 2.30 (dt, J = 19.0, 8.0 Hz, 1H), 2.19 (ddd, J = 14.2, 10.2, 5.7 Hz, 1H), 2.09–1.92 (m, 3H), 1.89 (dt, J = 13.1, 7.4 Hz, 1H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 214.7, 178.6, 171.1, 61.8, 59.2, 38.0, 34.0, 29.6, 28.2, 19.7, 14.2; IR (Neat Film, NaCl) 2979, 1713, 1408, 1158, 1028 cm⁻¹; HRMS (MM: ESI-APCI-) *m/z* calc'd for C₁₁H₁₅O₅ [M-H]⁻: 227.0925, found 227.0925.



Methyl (R)-3-(1-methyl-2-oxocyclohexyl)propanoate (SI-3). Synthesis of SI-3 was based on a literature procedure.³ A 100 mL round-bottom flask was charged with a magnetic stir bar, 2methylcyclohexanone (SI-2, 3.7 mL, 30.6 mmol, 1 equiv), (S)-phenylethylamine (3.71 g, 30.6 mmol, 1 equiv), p-toluenesulfonic acid hydrate (58 mg, 0.306 mmol, 0.01 equiv), and toluene (30 mL). The flask was equipped with a Dean-Stark trap filled with toluene and a reflux condenser. The reaction mixture was heated at reflux for 3.5 h. The Dean-Stark trap was replaced with a distillation head, and the toluene was distilled off under reduced pressure. The residue was cooled to 60 °C under nitrogen and methyl acrylate (3.4 mL, 36.7 mmol, 1.2 equiv) was added. The reaction mixture was stirred at 60 °C for 12 h. After cooling to ambient temperature, the reaction mixture was quantitatively transferred to a 250 mL round-bottom flask by rinsing with THF (50 mL total). Aqueous 20% acetic acid (30 mL) was added and the solution was stirred at 23 °C for 5 h. THF was evaporated under reduced pressure and 1N HCl (11 mL) was added. The biphasic mixture was extracted with Et₂O (25 mL x 3). The combined organic layers were washed with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel $(5 \rightarrow 6 \rightarrow 10 \rightarrow 12\%$ EtOAc in hexanes) to afford δ -keto ester SI-3 (4.96 g, 81% yield over 2 steps) as a light brown oil. $R_f = 0.4$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 3.66 (s, 3H), 2.46–2.26 (m, 3H), 2.22–2.10 (m, 1H), 2.11–1.98 (m, 1H), 1.93–1.67 (m, 6H), 1.67–1.54 (m, 1H), 1.07 (d, J = 2.7 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 215.4, 174.2, 51.8, 48.0, 39.4,

38.8, 32.6, 29.1, 27.6, 22.5, 21.1; IR (Neat Film, NaCl) 2936, 2865, 1740, 1705, 1437, 1378, 1304, 1197, 1172, 1123, 988 cm⁻¹; HRMS (ESI-APCI+) *m/z* calc'd for C₁₁H₁₉O₃ [M+H]⁺: 199.1329, found 199.1325; [α]_D²⁵ +31.5 (*c* 3.00, EtOH, 91% ee).



(*R*)-3-(1-Methyl-2-oxocyclohexyl)propanoic acid (8b). To a solution of SI-3 (2.37 g, 11.9 mmol, 1.0 equiv) in MeOH (11 mL) was added aqueous 2N NaOH (7.8 mL, 15.5 mmol, 1.3 equiv). The reaction mixture was stirred at 23 °C for 2 h, then MeOH was evaporated under reduced pressure. The aqueous layer was washed with Et₂O (10 mL x 1), acidified with 1N HCl (25 mL), and extracted with Et₂O (25 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure to a wet residue. The residue was redissolved in CH₂Cl₂, dried over Na₂SO₄, filtered, and concentrated under reduced pressure to a wet residue. The residue was redissolved in CH₂Cl₂, dried over Na₂SO₄, filtered, and concentrated under reduced pressure to afford carboxylic acid **8b** (2.14 g, 95% yield) as a light yellow viscous oil. R_f = 0.1 (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 2.43–2.31 (m, 3H), 2.22 (ddd, *J* = 16.8, 13.2, 5.2 Hz, 1H), 2.06–1.96 (m, 1H), 1.89–1.68 (m, 6H), 1.66–1.56 (m, 1H), 1.08 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 215.5, 179.7, 48.0, 39.3, 38.8, 32.4, 29.1, 27.5, 22.6, 21.1; IR (Neat Film, NaCl) 2936, 1706, 1455, 1312, 1224, 1124, 1097, 902, 856 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₀H₁₅O₃ [M–H]⁻⁻: 183.1027, found 183.1034; [α]p²⁵+36.0 (*c* 4.77, EtOH).



3-(1-Ethyl-2-oxocyclohexyl)propanoic acid (8c). Using 2-ethylcyclohexanone (**SI-4**) as staring material, the procedure for the synthesis of **8b** was followed to provide carboxylic acid **8c** (3.85 g, 85% yield over 3 steps) as a white solid. m.p. 60–62 °C; $R_f = 0.1$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 2.45–2.26 (m, 3H), 2.23–2.12 (m, 1H), 1.93–1.58 (m, 9H), 1.54–1.43 (m, 1H), 0.77 (t, J = 7.8 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 215.1, 179.9, 51.1, 39.2, 35.8, 28.9, 28.8, 27.3, 27.2, 20.8, 7.8; IR (Neat Film, NaCl) 2939, 2868, 1704, 1455, 1423, 1312, 1229, 1127, 1091 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₁H₁₇O₃ [M–H][–]: 197.1183, found 197.1187.



3-(1-Allyl-2-oxocyclohexyl)propanoic acid (8d). Basic hydrolysis of known δ -keto ester **SI-5**⁴ (2.24 g, 10.0 mmol, 1.0 equiv) afforded carboxylic acid **8d** (2.04 g, 97% yield) as a viscous colorless oil. $R_f = 0.1$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 10.52 (br s, 1H), 5.64 (dq, J = 17.0, 7.7 Hz, 1H), 5.11–5.02 (m, 2H), 2.44–2.28 (m, 4H), 2.28–2.20 (m, 1H), 2.16 (ddd, J = 16.4, 11.2, 5.1 Hz, 1H), 1.96 (ddd, J = 15.8, 11.2, 5.0 Hz, 1H), 1.88–1.65 (m, 7H); ¹³C NMR (126 MHz, CDCl₃) δ 214.4, 179.7, 133.1, 118.7, 50.9, 39.2, 39.1, 36.2, 29.5, 28.7, 27.1, 20.8; IR (Neat Film, NaCl) 2937, 2866, 1704, 1419, 1312, 1221, 1126, 995, 917 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₂H₁₇O₃ [M–H]⁻: 209.1183, found 209.1190.



2-Methylallyl 1-(3-methoxy-3-oxopropyl)-2-oxocyclohexane-1-carboxylate (SI-7). A flamedried 100 mL round-bottom flask was charged with a magnetic stir bar, MeCN (30 mL), β-keto ester **SI-6** (3.32 g, 16.9 mmol, 1.0 equiv), methyl acrylate (1.6 mL, 17.3 mmol, 1.02 equiv), and DBU (0.25 mL, 1.69 mmol, 0.1 equiv). The light yellow reaction mixture was stirred at 23 °C. After 14 h, TLC analysis indicated complete consumption of starting material. Solvents were evaporated, and the crude residue was purified by flash column chromatography on silica gel (8→16% EtOAc in hexanes) to afford ester **SI-7** (4.69 g, 98% yield) as a colorless oil. R_{*f*} = 0.4 (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 4.96 (d, *J* = 16.1 Hz, 2H), 4.54 (s, 2H), 3.65 (s, 3H), 2.54–2.34 (m, 4H), 2.30–2.15 (m, 2H), 2.06–1.89 (m, 2H), 1.82–1.74 (m, 1H), 1.73 (s, 3H), 1.71–1.57 (m, 2H), 1.54–1.43 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 207.4, 173.6, 171.5, 139.2, 114.2, 68.8, 60.2, 51.8, 41.1, 36.4, 29.8, 29.5, 27.6, 22.6, 19.7; IR (Neat Film, NaCl) 2948, 2867, 1738, 1715, 1436, 1307, 1176, 1135, 990, 907 cm⁻¹; HRMS (ESI-APCI+) *m/z* calc'd for C₁₅H₂₃O₅ [M+H]⁺: 283.1540, found 283.1533.



Methyl (*R*)-3-(1-(2-methylallyl)-2-oxocyclohexyl)propanoate (SI-8). In a nitrogen-filled glove box, a 250 mL Schlenk flask was charged with a magnetic stir bar, $Pd_2(dba)_3$ (32 mg,

0.035 mmol, 0.005 equiv), (*S*)-*t*-BuPHOX (34 mg, 0.0875 mmol, 0.0125 equiv), and MTBE (40 mL). The solution was stirred at ambient temperature for 30 min. Then additional MTBE (21 mL) and **SI-7** (1.98 g, 7.0 mmol, 1.0 equiv) were added via syringe. The syringe was rinsed with MTBE (3 mL x 3) to ensure complete transfer of **SI-7**. The Schlenk flask was sealed and taken out of the glove box. The reaction mixture was stirred at 40 °C for 28 h, then concentrated under reduced pressure. The crude residue was purified by flash column chromatography on silica gel (6→10% EtOAc in hexanes) to afford δ-keto ester **SI-8** (1.49 g, 89% yield) as a colorless oil. $R_f = 0.5$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 4.83 (s, 1H), 4.65 (s, 1H), 3.64 (s, 3H), 2.47 (dt, J = 14.0, 6.5 Hz, 1H), 2.42–2.27 (m, 4H), 2.15–2.06 (m, 1H), 2.02–1.92 (m, 1H), 1.91–1.65 (m, 7H), 1.64 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 214.4, 174.2, 141.9, 115.4, 51.8, 51.0, 42.9, 39.4, 36.7, 30.5, 28.9, 27.2, 24.5, 20.9; IR (Neat Film, NaCl) 2943, 2865, 1738, 1699, 1436, 1374, 1173, 1128, 1080, 895 cm⁻¹; HRMS (ESI-APCI+) *m/z* calc'd for C₁₄H₂₃O₃ [M+H]⁺: 239.1642, found 239.1633; [α]p²⁵ +9.0 (*c* 1.00, CHCl₃, 91% ee).



(*R*)-3-(1-(2-Methylallyl)-2-oxocyclohexyl)propanoic acid (8e). Basic hydrolysis of δ -keto ester SI-8 (1.49 g, 6.25 mmol, 1 equiv) afforded carboxylic acid 8e (1.39 g, 99% yield) as a white solid. m.p. 81–82 °C; $R_f = 0.1$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 4.82 (s, 1H), 4.64 (s, 1H), 2.54–2.43 (m, 1H), 2.43–2.27 (m, 4H), 2.15 (ddd, J = 16.4, 10.8, 5.5 Hz, 1H), 1.96–1.69 (m, 7H), 1.69–1.63 (m, 1H), 1.62 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 214.7, 179.9, 141.7, 115.5, 51.0, 43.0, 39.3, 36.7, 30.1, 28.9, 27.1, 24.4, 20.9; IR (Neat Film,

NaCl) 3074, 2940, 2866, 1704, 1455, 1312, 1219, 1128, 896 cm⁻¹; HRMS (ESI-APCI–) m/z calc'd for C₁₃H₁₉O₃ [M–H]⁻: 223.1340, found 223.1345; $[\alpha]_D^{25}$ +5.4 (*c* 1.00, CHCl₃).



4,4-Dimethyl-5-oxo-5-phenylpentanoic acid (8f). A flame-dried 100 mL round-bottom flask was charged with a magnetic stir bar, THF (26 mL), and cooled to 0 $^{\circ}$ C. A solution of *n*butyllithium in hexanes (2.5 M, 5.7 mL, 14.3 mmol, 1.1 equiv) was added, followed by dropwise addition of diisopropylamine (2.0 mL, 14.3 mmol, 1.1 equiv). The light yellow solution was stirred at 0 °C for 10 min, then cooled to -78 °C in a dry ice-acetone bath. Isobutyrophenone (SI-9, 2.0 mL, 13 mmol, 1.0 equiv) was added dropwise. The orange-red solution was stirred at the same temperature for 30 min, and methyl 3-bromopropionate (1.7 mL, 15.6 mmol, 1.2 equiv) was added dropwise. The dry ice-acetone bath was removed, and the yellow reaction mixture was warmed to 0 °C and stirred for an additional 2 h. The reaction was quenched with half saturated aqueous NH₄Cl solution (30 mL) and extracted with Et₂O (30 mL x 2). The combined organic layers were washed with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by flash column chromatography on silica gel (5% EtOAc in hexanes) to afford a colorless oil (1.20 g), which was subjected to basic hydrolysis to provide carboxylic acid 8f (0.93 g, 32% yield over 2 steps) as a viscous colorless oil. $R_f = 0.1$ (25% EtOAc in hexanes); ¹H NMR (400 MHz, CDCl₃) δ 7.69–7.65 (m, 2H), 7.50– 7.45 (m, 1H), 7.43–7.37 (m, 2H), 2.36–2.29 (m, 2H), 2.13–2.07 (m, 2H), 1.34 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) & 208.2, 179.7, 138.6, 131.3, 128.4, 127.8, 47.2, 35.3, 29.9, 26.0; IR (Neat Film, NaCl) 2972, 1709, 1597, 1444, 1303, 1200, 962, 719, 700 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₃H₁₅O₃ [M–H]⁻: 219.1027, found 219.1035.



4-(Butoxycarbonyl)-4-ethyloctanoic acid (8g). A flame-dried 50 mL round-bottom flask was charged with a magnetic stir bar, THF (10 mL), and cooled to 0 °C. A solution of *n*-butyllithium in hexanes (2.5 M, 2.4 mL, 6.09 mmol, 1.1 equiv) was added, followed by dropwise addition of diisopropylamine (0.93 mL, 6.65 mmol, 1.2 equiv). The light yellow solution was stirred at 0 °C for 10 min, then cooled to -78 °C in a dry ice-acetone bath. Butyl ester **SI-10** (1.11 g, 5.54 mmol, 1.0 equiv) was added dropwise. The solution was stirred at the same temperature for 40 min, and allyl bromide (0.58 mL, 6.65 mmol, 1.2 equiv) was added dropwise. After stirring at -78 °C for 1 h, the dry ice-acetone bath was removed, and the yellow reaction mixture was warmed to 23 °C and stirred for an additional 1 h. The reaction was quenched with half saturated aqueous NH₄Cl solution (10 mL) and extracted with hexanes (30 mL x 2). The combined organic layers were washed with H₂O, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by flash column chromatography on silica gel (3% Et₂O in hexanes) to afford the allylated ester as a colorless oil (1.24 g, R_f = 0.8 (10% Et₂O in hexanes)), which was used directly in the next reaction.

A solution of 9-BBN in THF (0.5 M, 10 mL, 5.0 mmol, 1.1 equiv) was added to a 50 mL round-bottom flask containing the allylated ester (1.24 g, 4.5 mmol, 1.0 equiv) at 0 °C. After 10 min, the ice bath was removed, and the reaction mixture was stirred for another 3 h. Water (10

mL) was added to the reaction mixture, followed by careful portionwise addition of sodium perborate hydrate (1.78 g, 17.8 mmol, 4.0 equiv) to oxidize the organoborane intermediate to the corresponding alcohol. The biphasic mixture was stirred for 40 min, and extracted with EtOAc (25 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by flash column chromatography on silica gel (16 \rightarrow 25% EtOAc in hexanes) to afford the desired primary alcohol as a colorless oil (1.14 g, R_f = 0.4 (25% EtOAc in hexanes)), which was used directly in the next reaction.

Oxidation of the primary alcohol to the corresponding carboxylic acid was carried out following a literature procedure.⁵ A 100 mL round-bottom flask was charged with a magnetic stir bar, the primary alcohol (517 mg, 2.0 mmol, 1.0 equiv), TEMPO (22 mg, 0.14 mmol, 0.07 equiv), MeCN (10 mL), H₂O (2 mL), and aqueous phosphate buffer (0.33 M in NaH₂PO₄ and 0.33 M in Na₂HPO₄, 7.5 mL), and stirred at 20 °C for 5 min. Solid NaClO₂ (452 mg, 4.0 mmol, 2.0 equiv) was added to the flask and the reaction mixture was stirred for 2 min when the solid dissolved. A solution of NaClO (0.26 wt%, 1.1 mL, 0.04 mmol, 0.02 equiv) was added, and the reaction mixture immediately turned dark red. The flask was placed in a pre-heated 35 °C oil bath and stirred for 14 h. TLC analysis showed complete consumption of the alcohol. To the reaction mixture was added H₂O (15 mL) and 2N NaOH (4 mL) to bring the solution's pH to 10. The biphasic mixture was poured into an ice-cold solution of Na₂SO₃ (610 mg) in H₂O (10 mL). stirred for 30 min, and extracted with Et₂O (30 mL x 1). The ethereal layer was back-extracted with 0.7N NaOH (8 mL x 2). The alkaline aqueous layers were combined and acidified with 6N HCl (12 mL), and extracted with EtOAc (25 mL x 3). The combined EtOAc extracts were washed with H₂O, dried over Na₂SO₄, filtered, and concentrated under reduced pressure to a liquid/solid mixture. The mixture was dissolved in Et₂O and solid impurities were filtered off.

The filtrate was concentrated and purified by flash column chromatography on silica gel (5% MeOH in CH₂Cl₂) to afford carboxylic acid **8g** (485 mg, 71% yield over 3 steps) as a viscous colorless oil. $R_f = 0.3$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 10.27 (br s, 1H), 4.07 (t, J = 6.6 Hz, 2H), 2.29–2.19 (m, 2H), 1.95–1.85 (m, 2H), 1.64–1.55 (m, 4H), 1.55–1.49 (m, 2H), 1.43–1.33 (m, 2H), 1.33–1.23 (m, 2H), 1.20–1.05 (m, 2H), 0.93 (t, J = 7.4 Hz, 3H), 0.89 (t, J = 7.4 Hz, 3H), 0.79 (t, J = 7.5 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 179.9, 176.7, 64.4, 48.8, 34.0, 30.8, 29.3, 28.6, 27.2, 26.2, 23.3, 19.3, 14.1, 13.8, 8.4; IR (Neat Film, NaCl) 2961, 2875, 1716, 1458, 1207, 1133, 947 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₅H₂₇O₄ [M–H]⁻: 271.1915, found 271.1923.



3-Allyl-1-benzyl-3-ethylpiperidin-2-one (SI-12). A flame-dried 100 mL round-bottom flask was charged with a magnetic stir bar, THF (15 mL), diisopropylamine (1.0 mL, 7.08 mmol, 1.1 equiv), and cooled to 0 °C. A solution of *n*-butyllithium in hexanes (2.5 M, 2.8 mL, 7.08 mmol, 1.1 equiv) was added dropwise. The light yellow solution was stirred at 0 °C for 10 min, then cooled to -78 °C in a dry ice-acetone bath. A solution of benzyl lactam **SI-11** (1.40 g, 6.44 mmol, 1.0 equiv) in THF (5 mL) was added dropwise. The dark greenish yellow solution was stirred at -78 °C for 20 min, and allyl bromide (0.67 mL, 7.73 mmol, 1.2 equiv) was added dropwise. The color of the reaction mixture immediately turned into light yellow. The dry ice-acetone bath was removed, and the yellow reaction mixture was warmed to 23 °C and stirred for an additional 1 h. The reaction was quenched with half saturated aqueous NH₄Cl solution (20 mL) and extracted with EtOAc (25 mL x 3). The combined organic layers were washed with

H₂O, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by flash column chromatography on silica gel (hexanes:EtOAc:CH₂Cl₂ = 30:3:2) to afford allyl lactam **SI-12** (1.41 g, 85% yield) as a colorless oil. R_f = 0.5 (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.37–7.31 (m, 2H), 7.31–7.24 (m, 3H), 5.88–5.76 (m, 1H), 5.15–5.07 (m, 2H), 4.68–4.56 (m, 2H), 3.24–3.16 (m, 2H), 2.59 (ddt, *J* = 13.5, 6.7, 1.4 Hz, 1H), 2.25 (ddt, *J* = 13.5, 8.1, 1.1 Hz, 1H), 1.87 (dq, *J* = 13.5, 7.5 Hz, 1H), 1.82–1.70 (m, 4H), 1.57 (dq, *J* = 13.5, 7.4 Hz, 1H), 0.93 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 174.5, 137.8, 135.0, 128.6, 128.2, 127.3, 117.9, 50.6, 47.8, 45.4, 43.4, 31.6, 28.9, 19.8, 8.9; IR (Neat Film, NaCl) 2938, 2876, 1635, 1490, 1453, 1352, 1196, 913, 737, 701 cm⁻¹; HRMS (ESI-APCI+) *m/z* calc'd for C₁₇H₂₄NO [M+H]⁺: 258.1852, found 258.1862.



3-(1-Benzyl-3-ethyl-2-oxopiperidin-3-yl)propanoic acid (8h). A flame-dried 50 mL roundbottom flask was charged with a magnetic stir bar, $BH_3 \cdot SMe_2$ (3.8 mL, 2.0 M in THF, 7.5 mmol, 2.5 equiv), and THF (4.0 mL). The solution was cooled to 0 °C in an ice bath, and cyclohexene (1.52 mL, 15.0 mmol, 5.0 equiv) was added. After stirring at 0 °C for 1 h, a solution of allyl lactam **SI-12** (772 mg, 3.0 mmol, 1.0 equiv) in THF (1.0 mL) was added. The vial containing **SI-12** was washed with THF (1.2 mL x 3) and the washings were also added. The ice bath was removed, and the white slurry was stirred at 20 °C for 17 h. TLC analysis then indicated complete consumption of starting material. The reaction was quenched by careful addition of H_2O (10 mL) and NaBO₃·H₂O (2.25 g). The aqueous phase was extracted with EtOAc (25 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under

reduced pressure. The residue was purified by flash column chromatography on silica gel (hexanes:EtOAc:MeOH 20:10:1 \rightarrow 15:10:1) to afford the desired primary alcohol (591 mg, R_f = 0.2 (hexanes:EtOAc:MeOH = 10:10:1)) as a colorless oil, which was used directly in the next reaction.

The primary alcohol (591 mg, 2.15 mmol, 1.0 equiv) was dissolved in acetone (20 mL) and cooled to 0 °C. Jones' reagent (1.2 mL, 3.1 M in CrO₃, 1.7 equiv) was added dropwise until the reaction mixture became persistently orange. Excess Jones' reagent was quenched with *i*PrOH (5 mL), and green solids precipitated out of the solution. The solids were filtered and washed with additional acetone (10 mL x 2). The filtrate was concentrated and partitioned between EtOAc and H₂O to remove residual chromium salts. The aqueous layer was extracted with EtOAc (20 mL x 3). The organic extracts were combined, concentrated, redissolved in Et₂O (20 mL), and extracted with aqueous 2N NaOH (20 mL x 1). The organic layer was washed with H₂O (10 mL x 1). The combined aqueous layers were acidified with 6N HCl to pH 1 and extracted with CH₂Cl₂ (10 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated. The residue was purified by flash column chromatography on silica gel (hexanes:EtOAc:MeOH 100:50:7.5 \rightarrow 60:30:6) to afford carboxylic acid **8h** (425 mg, 49%) yield over 2 steps) as a viscous colorless oil that slowly solidified during storage in a freezer. R_f = 0.3 (67% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.38–7.33 (m, 2H), 7.33–7.24 (m, 3H), 4.72 (d, J = 14.5 Hz, 1H), 4.52 (d, J = 14.5 Hz, 1H), 3.24 (td, J = 5.5, 1.8 Hz, 2H), 2.54 (ddd, J = 16.2, 10.4, 5.8 Hz, 1H), 2.42 (ddd, J = 16.0, 10.3, 5.5 Hz, 1H), 2.07 (ddd, J = 13.9),10.4, 5.5 Hz, 1H), 1.97–1.77 (m, 5H), 1.74–1.58 (m, 2H), 0.94 (t, J = 7.5 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) & 178.2, 174.6, 137.5, 128.8, 128.2, 127.5, 50.8, 47.8, 44.7, 32.8, 31.0, 30.0,

29.6, 19.5, 8.6; IR (Neat Film, NaCl) 2943, 1732, 1599, 1495, 1454, 1354, 1261, 1196, 910, 735, 701 cm⁻¹; HRMS (ESI-APCI+) *m/z* calc'd for C₁₇H₂₄NO₃ [M+H]⁺: 290.1751, found 290.1754.



4-Formyl-4-methyltridecanoic acid (8i). Using 2-methylundecanal (**SI-13**) as staring material, the procedure for the synthesis of **8b** was followed (Note: the conjugate addition reaction was allowed to proceed for 20 h at 70 °C and then 28 h at 80 °C) to provide carboxylic acid **8i** (481 mg, 32% yield over 3 steps) as a colorless oil. $R_f = 0.2$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 11.35 (br s, 1H), 9.41 (s, 1H), 2.34–2.19 (m, 2H), 1.93–1.83 (m, 1H), 1.82–1.72 (m, 1H), 1.53–1.37 (m, 2H), 1.33–1.09 (m, 14H), 1.02 (s, 3H), 0.86 (t, *J* = 6.5 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 205.9, 179.7, 48.5, 35.6, 32.0, 30.3, 29.6, 29.5, 29.4, 29.4, 29.1, 23.9, 22.8, 18.2, 14.2; IR (Neat Film, NaCl) 2924, 2853, 1711, 1458, 1300, 1225, 913 cm⁻¹; HRMS (ESI-APCI–) *m/z* calc'd for C₁₅H₂₇O₃ [M–H]⁻: 255.1966, found 255.1976.



3-((3*S***,3a***S***,5a***S***,6***R***,9a***S***,9b***S***)-3-Acetoxy-3a,6-dimethyl-7-oxododecahydro-1***H***-cyclopenta**[*a*]**naphthalen-6-yl)propanoic acid (8j).** Carboxylic acid **8j** was prepared based on a literature procedure.⁶ A 250 mL round-bottom flask was charged with a magnetic stir bar, testosterone acetate **SI-14** (661 mg, 2.0 mmol, 1.0 equiv), Na₂CO₃ (312 mg, 2.94 mmol, 1.47

equiv), t-BuOH (21 mL) and H₂O (1 mL). The flask was placed in a pre-heated 70 °C oil bath, and a hot solution of NaIO₄ (2.14 g, 10.0 mmol, 5.0 equiv) and KMnO₄ (25 mg, 0.16 mmol, 0.08 equiv) in H₂O (21 mL) was added. The reaction mixture was stirred at 70 °C for 20 min when TLC analysis indicated complete consumption of starting material. The flask was cooled to ambient temperature, and 1N HCl (100 mL) was added. After stirring 2 min, the reaction mixture was extracted with EtOAc (30 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (50% EtOAc in hexanes) to afford carboxylic acid **8**i (664 mg, 88% yield) as a solid/liquid mixture, which was recrystallized from hexanes: EtOAc 2:1 to give a white solid. m.p. 135–136 °C; $R_f = 0.4$ (50% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 4.58 (t, J = 8.5 Hz, 1H), 2.63–2.44 (m, 1H), 2.40–2.14 (m, 4H), 2.14–2.06 (m, 1H), 2.04 (s, 3H), 1.99–1.89 (m, 1H), 1.85–1.30 (m, 8H), 1.29–1.14 (m, 3H), 1.12 (s, 3H), 1.10–1.02 (m, 1H), 0.84 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) & 214.6, 179.6, 171.3, 82.5, 50.5, 50.2, 48.0, 42.7, 38.0, 36.4, 34.8, 30.9, 29.3, 29.2, 27.5, 23.7, 21.3, 21.1, 20.5, 12.2; IR (Neat Film, NaCl) 2941, 1732, 1705, 1448, 1375, 1248, 1043, 952, 735 cm⁻¹; HRMS (ESI-APCI-) *m/z* calc'd for $C_{20}H_{29}O_5 [M-H]^-: 349.2020$, found 349.2037; $[\alpha]_D^{25} + 26.1$ (*c* 0.50, CHCl₃).

Palladium-Catalyzed Decarbonylative Dehydration of Carboxylic Acids



General Procedure A: Large Scale Distillation Process

Ethyl 2-oxo-1-vinylcyclopentane-1-carboxylate (9a). A flame-dried 15 mL round-bottom flask was charged with a magnetic stir bar, PdCl₂(PPh₃)₂ (7.0 mg, 0.01 mmol, 0.002 equiv), Xantphos (6.9 mg, 0.012 mmol, 0.0024 equiv), $(t-Bu)_4$ biphenol⁷ (20.5 mg, 0.05 mmol, 0.01 equiv), and carboxylic acid 8a (1.14 g, 5.0 mmol, 1.0 equiv). The flask was equipped with a distillation head and a 25 mL round-bottom receiving flask. The closed system was connected to a vacuum manifold and equipped with a needle valve. The system was evacuated and backfilled with N₂ (x 3), and the first portion of acetic anhydride (6.0 mmol, 1.2 equiv) was added via syringe through the septum that seals the top of the distillation head. The flask was lowered into a pre-heated 60 °C oil bath and gradually heated to 132 °C in 18 min. When the oil bath temperature reached 122 °C, the needle valve was closed, switched to vacuum, and the needle valve carefully and slowly opened to allow distillation of acetic acid into a receiving flask, which was cooled to -78 °C. When the oil bath temperature reached 130 °C, time was recorded as t = 0. After distillation ceased (about t = 3 min), the needle valve was opened fully and a vacuum of 1–5 mmHg was drawn. At t = 30 min, the system was backfilled with N_2 , and the second portion of acetic anhydride (2.5 mmol, 0.5 equiv) was added via syringe. The system was then gradually (t = 35 min) resubjected to a vacuum of 1–5 mmHg. Acetic anhydride was added as follows (0.3, 0.2 equiv) in the same manner every 30 min. The reaction was stopped at t = 2 h and allowed to cool under N2 to ambient temperature. The distillate was added to a saturated

aqueous solution of NaHCO₃, stirred for 20 min, and the biphasic mixture was extracted with CH₂Cl₂ (20 mL x 3). The combined extracts were dried over Na₂SO₄ and filtered into a 250 round-bottom flask. To this filtrate were added the residual dark red reaction mixture and the washings of the distillation head's inside (with ~5 mL CH₂Cl₂). The solvents were evaporated and the residue was purified by flash column chromatography on silica gel (5→8% EtOAc in hexanes for **9a** and then 16% EtOAc in hexanes for **10**) to afford vinyl ketone **9a** (613 mg, 67% yield) as a colorless, fragrant oil. $R_f = 0.5$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.04 (dd, J = 17.6, 10.7 Hz, 1H), 5.28 (d, J = 10.7 Hz, 1H), 5.19 (d, J = 17.6 Hz, 1H), 4.18 (q, J = 7.2 Hz, 2H), 2.63–2.55 (m, 1H), 2.45–2.27 (m, 2H), 2.23–2.15 (m, 1H), 2.07–1.87 (m, 2H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 212.5, 170.4, 134.5, 117.0, 63.5, 61.9, 37.7, 32.9, 19.6, 14.2; IR (Neat Film, NaCl) 2980, 1753, 1729, 1635, 1456, 1406, 1255, 1142, 1034, 927 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₀H₁₄O₃ [M]⁺: 182.0943, found 182.0939.

Ethyl 2-oxo-3,4,5,6-tetrahydrocyclopenta[*b*]pyran-4a(2*H*)-carboxylate (10). The above reaction also furnished enol lactone 10 (206 mg, 20% yield) as a colorless oil. $R_f = 0.2$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.30 (app. s, 1H), 4.20 (q, J = 7.1 Hz, 2H), 2.76–2.62 (m, 2H), 2.55–2.42 (m, 3H), 2.35–2.26 (m, 1H), 1.96–1.84 (m, 1H), 1.84–1.73 (m, 1H), 1.26 (t, J = 7.1 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 173.3, 167.4, 151.6, 108.0, 61.8, 50.8, 35.9, 29.5, 28.1, 26.7, 14.2; IR (Neat Film, NaCl) 2940, 2862, 1768, 1726, 1668, 1456, 1248, 1159, 1124, 1020, 891, 805 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₁H₁₄O₄ [M]⁺: 210.0892, found 210.0910.



Butyl 2-ethyl-2-vinylhexanoate (9g). A flame-dried 20 x 150 mm Kimax culture tube was charged with a magnetic stir bar, PdCl₂(nbd) (1.3 mg, 0.005 mmol, 0.01 equiv), Xantphos (3.5 mg, 0.006 mmol, 0.012 equiv), carboxylic acid **8g** (136 mg, 0.5 mmol, 1.0 equiv), and benzoic anhydride (136 mg, 0.6 mmol, 1.2 equiv). The tube was sealed with a rubber septum, and the system was evacuated and backfilled with N₂ (x 3). NMP (0.25 mL) was added via syringe. The reaction mixture was stirred at 20 °C for 2 min, then placed in a pre-heated 132 °C oil bath and stirred for 3 h. After cooling to ambient temperature, Et₃N (0.3 mL) was added, and the mixture was purified by flash column chromatography on silica gel (2% Et₂O in hexanes) to afford vinyl ester **9g** (58 mg, 51% yield) as a colorless oil. $R_f = 0.5$ (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.98 (dd, J = 17.8, 10.9 Hz, 1H), 5.17 (d, J = 11.0 Hz, 1H), 5.07 (d, J = 17.8 Hz, 1H), 4.08 (t, J = 6.6 Hz, 2H), 1.77–1.69 (m, 2H), 1.69–1.63 (m, 2H), 1.63–1.55 (m, 2H), 1.45–1.23 (m, 4H), 1.23–1.08 (m, 2H), 0.93 (t, J = 7.4 Hz, 3H), 0.88 (t, J = 7.4 Hz, 3H), 0.80 (t, J = 7.4 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 175.7, 140.3, 114.3, 64.5, 52.9, 35.7, 30.9, 29.0, 26.7, 23.4, 19.4, 14.1, 13.8, 8.9; IR (Neat Film, NaCl) 2960, 2874, 1731, 1459, 1380, 1240,

1206, 1136, 1001, 915 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₄H₂₆O₂ [M]⁺: 226.1933, found 226.1933.

General Procedure B: Small Scale Nondistillation Process

Spectroscopic Data for Pd-Catalyzed Decarbonylative Dehydration Products



(*R*)-2-Methyl-2-vinylcyclohexan-1-one (*ent-7*). Ketone *ent-7* was prepared according to General Procedure A and isolated by silica gel chromatography (3% Et₂O in hexanes) as a colorless, fragrant oil. 60% yield. $R_f = 0.6$ (20% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.97 (dd, J = 17.8, 10.8 Hz, 1H), 5.13 (d, J = 10.8 Hz, 1H), 4.98 (d, J = 17.7 Hz, 1H), 2.56–2.45 (m, 1H), 2.37–2.28 (m, 1H), 2.01–1.90 (m, 2H), 1.84–1.65 (m, 3H), 1.65–1.56 (m, 1H), 1.15 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 213.6, 142.7, 114.9, 52.2, 39.9, 39.3, 27.8, 24.0, 21.8; IR (Neat Film, NaCl) 2932, 2864, 1709, 1635, 1450, 1371, 1313, 1123, 1093, 986, 918 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₉H₁₄O [M]⁺: 138.1045, found 138.1026; [α]_D²⁵ +113.2 (*c* 1.03, CHCl₃, 92% ee).



2-Ethyl-2-vinylcyclohexan-1-one (9c). Ketone **9c** was prepared according to General Procedure A and isolated by silica gel chromatography (2% Et₂O in hexanes) as a colorless, fragrant oil. 66% yield. $R_f = 0.4$ (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.90 (dd, J = 17.8, 10.9 Hz, 1H), 5.19 (d, J = 10.9 Hz, 1H), 4.96 (d, J = 17.7 Hz, 1H), 2.52–2.41 (m, 1H), 2.36–2.27 (m, 1H), 1.98–1.83 (m, 2H), 1.80–1.53 (m, 6H), 0.79 (t, J = 7.5 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 213.3, 141.5, 115.5, 55.5, 39.6, 35.6, 29.8, 27.4, 21.6, 8.2; IR (Neat

Film, NaCl) 2938, 2863, 1707, 1448, 1313, 1230, 1124, 993, 918 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₀H₁₆O [M]⁺: 152.1201, found 152.1176.



2-Allyl-2-vinylcyclohexan-1-one (9d). Ketone **9d** was prepared according to General Procedure A and isolated by silica gel chromatography (3% Et₂O in hexanes) as a colorless, fragrant oil. 54% yield. $R_f = 0.4$ (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.86 (dd, J = 17.7, 10.8 Hz, 1H), 5.68 (ddt, J = 18.0, 11.1, 7.3 Hz, 1H), 5.22 (d, J = 10.8 Hz, 1H), 5.05–4.95 (m, 3H), 2.54–2.45 (m, 1H), 2.42–2.27 (m, 3H), 1.99–1.88 (m, 2H), 1.81–1.60 (m, 4H); ¹³C NMR (126 MHz, CDCl₃) δ 212.7, 141.4, 134.3, 117.9, 116.2, 54.9, 42.0, 39.6, 36.0, 27.3, 21.6; IR (Neat Film, NaCl) 3077, 2936, 2864, 1708, 1636, 1448, 1314, 1222, 1124, 998, 916 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₁H₁₆O [M]⁺: 164.1201, found 164.1173.



(*S*)-2-(2-Methylallyl)-2-vinylcyclohexan-1-one (9e). Ketone 9e was prepared according to General Procedure A and isolated by silica gel chromatography (2 \rightarrow 3% Et₂O in hexanes) as a colorless, fragrant oil. 69% yield. R_f = 0.4 (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.94 (dd, *J* = 18.1, 11.1 Hz, 1H), 5.19 (d, *J* = 10.8 Hz, 1H), 4.98 (d, *J* = 17.8 Hz, 1H), 4.80 (s, 1H), 4.65 (s, 1H), 2.55–2.43 (m, 2H), 2.43–2.32 (m, 2H), 2.01–1.87 (m, 2H), 1.79–1.66 (m, 4H), 1.65 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 212.3, 142.5, 142.4, 115.8, 115.0, 54.8, 45.5, 39.5,

36.0, 27.1, 25.0, 21.7; IR (Neat Film, NaCl) 3075, 2938, 2863, 1707, 1641, 1448, 1125, 919, 892 cm⁻¹; HRMS (EI+) *m/z* calc'd for $C_{12}H_{18}O$ [M]⁺: 178.1358, found 178.1350; $[\alpha]_D^{25}$ +129.5 (*c* 1.00, CHCl₃, 92% ee).



2,2-Dimethyl-1-phenylbut-3-en-1-one (9f). Ketone **9f** was prepared according to General Procedure A and isolated by silica gel chromatography (2.5% Et₂O in hexanes) as a colorless oil. 75% yield. $R_f = 0.4$ (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.89–7.85 (m, 2H), 7.49–7.43 (m, 1H), 7.40–7.35 (m, 2H), 6.19 (dd, J = 17.6, 10.6 Hz, 1H), 5.26–5.18 (m, 2H), 1.40 (s, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 204.8, 144.0, 137.2, 131.8, 129.4, 128.1, 114.2, 50.3, 26.2; IR (Neat Film, NaCl) 3084, 2975, 2933, 1679, 1634, 1446, 1258, 971, 918, 719, 694 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₂H₁₄O [M]⁺: 174.1045, found 174.1069.



1-Benzyl-3-ethyl-3-vinylpiperidin-2-one (9h). Lactam **9h** was prepared according to General Procedure B and isolated by silica gel chromatography (9% EtOAc in hexanes) as a colorless oil. 57% yield. $R_f = 0.4$ (25% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 7.36–7.31 (m, 2H), 7.31–7.24 (m, 3H), 5.96 (dd, J = 17.6, 10.8 Hz, 1H), 5.19 (dd, J = 10.8, 1.0 Hz, 1H), 5.11 (dd, J = 17.6, 1.0 Hz, 1H), 4.68–4.57 (m, 2H), 3.29–3.16 (m, 2H), 1.98–1.80 (m, 4H), 1.80–1.66 (m, 2H), 0.90 (t, J = 7.4 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 173.0, 142.6, 137.7, 128.6, 128.1,

127.3, 114.1, 50.6, 49.4, 47.8, 31.8, 28.8, 19.3, 8.6; IR (Neat Film, NaCl) 2938, 2876, 1634, 1494, 1453, 1353, 1261, 1196, 915, 742, 700 cm⁻¹; HRMS (MM: ESI-APCI+) m/z calc'd for C₁₆H₂₂NO [M+H]⁺: 244.1696, found 244.1702.



2-Methyl-2-vinylundecanal (9i). Aldehyde **9i** was prepared according to General Procedure B and isolated by silica gel chromatography (2% Et₂O in hexanes) as a colorless, fragrant oil. 77% yield. $R_f = 0.5$ (10% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 9.38 (s, 1H), 5.79 (dd, J = 17.6, 10.8 Hz, 1H), 5.25 (d, J = 10.8 Hz, 1H), 5.11 (d, J = 17.5 Hz, 1H), 1.57 (td, J = 9.6, 5.9 Hz, 2H), 1.25 (br s, 14H), 1.15 (s, 3H), 0.87 (t, J = 6.7 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 203.2, 139.1, 116.6, 52.9, 35.7, 32.0, 30.3, 29.7, 29.6, 29.4, 24.0, 22.8, 17.8, 14.3; IR (Neat Film, NaCl) 2927, 2854, 1732, 1463, 998, 920 cm⁻¹; HRMS (EI+) *m/z* calc'd for C₁₄H₂₆O [M]⁺: 210.1984, found 210.2009.



(3S,3aS,5aS,6R,9aS,9bS)-3a,6-dimethyl-7-oxo-6-vinyldodecahydro-1H-

cyclopenta[*a*]naphthalen-3-yl acetate (9j). Ketone 9j was prepared according to General Procedure B and isolated by silica gel chromatography (10 \rightarrow 14% EtOAc in hexanes) as a white solid. 41% yield. m.p. 138–139 °C; $R_f = 0.3$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 5.78 (dd, J = 17.6, 10.9 Hz, 1H), 5.24 (d, J = 10.9 Hz, 1H), 5.02 (d, J = 17.6 Hz, 1H),

4.59 (t, J = 8.5 Hz, 1H), 2.61 (td, J = 14.3, 6.3 Hz, 1H), 2.37–2.30 (m, 1H), 2.24–2.14 (m, 1H), 2.03 (s, 3H), 2.02–1.95 (m, 1H), 1.83–1.62 (m, 3H), 1.58–1.47 (m, 1H), 1.45–1.33 (m, 3H), 1.32–1.25 (m, 2H), 1.24 (s, 3H), 1.18–1.05 (m, 2H), 0.85 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 214.3, 171.2, 141.7, 114.7, 82.4, 54.6, 51.0, 50.4, 43.0, 38.0, 36.4, 34.8, 31.2, 27.6, 23.6, 21.9, 21.3, 15.2, 12.3; IR (Neat Film, NaCl) 2946, 2845, 1735, 1696, 1373, 1250, 1041, 922 cm⁻¹; HRMS (EI+) m/z calc'd for C₁₉H₂₈O₃ [M]⁺: 304.2039, found 304.2044; [α]_D²⁵ +9.0 (*c* 0.50, CHCl₃). Total Synthesis of (-)-Aspewentin A, B, C, and Related Compounds



Ester SI-15. A 100 mL 2-necked round-bottom flask was charged with a magnetic stir bar and magnesium turnings (560 mg, 23.0 mmol, 1.1 equiv). The flask was equipped with a reflux condenser, flame-dried under vacuum, and allowed to cool to ambient temperature under N_2 . THF (3 mL) was added, followed by a small portion (2 mL) of a solution of bromoarene 14^8 (5.63 g, 20.9 mmol, 1.0 equiv) in THF (8.5 mL). DIBAL-H (0.4 mL, 1.0 M in hexanes, 0.4 mmol, 0.02 equiv) was added, and the reaction mixture was gently heated to reflux using a heat gun. Grignard reagent formation initiated as the reaction mixture turned dark with a strong exotherm. The remainder of the bromoarene solution in THF (ca. 6.5 mL) was added dropwise to maintain a gentle reflux. After addition was finished, the reaction mixture was further refluxed for 1 h, and allowed to cool to ambient temperature under N₂. The dark gray solution of Grignard reagent was taken up in a syringe and added dropwise to a separate 100 mL roundbottom flask containing a stirred suspension of CuI (400 mg, 2.09 mmol, 0.1 equiv) and ethyl 4iodobutyrate (5.06 g, 20.9 mmol, 1.0 equiv) in THF (20 mL) at 0 °C. After 15 min, the reaction mixture was warmed to 20 °C and guenched with aqueous NH₄Cl and NaHSO₄ solution. The layers were separated, and the aqueous layer was extracted with EtOAc (30 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel $(3\rightarrow 4\rightarrow 10\% \text{ EtOAc in hexanes})$ furnished

ester SI-15 (4.16 g, 65% yield) as a colorless oil. $R_f = 0.5$ (10% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.78 (s, 1H), 6.56 (s, 1H), 4.15 (g, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.79 (s, 3H), 3.79 (s

6.4 Hz, 2H), 2.60–2.55 (m, 2H), 2.39 (t, J = 7.4 Hz, 2H), 1.95–1.85 (m, 2H), 1.84–1.76 (m, 2H), 1.66–1.60 (m, 2H), 1.30–1.25 (m, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 173.7, 157.3, 147.6, 140.7, 126.6, 112.1, 110.2, 60.4, 55.3, 39.0, 34.5, 34.3, 32.9, 32.2, 26.6, 25.3, 19.8, 14.4; IR (Neat Film, NaCl) 2930, 1734, 1603, 1465, 1304, 1254, 1188, 1116, 1065, 847 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₁₉H₂₉O₃ [M+H]⁺: 305.2111, found 305.2105.



Ester 15. To a stirred solution of *n*-BuLi (3.2 mL, 2.5 M in hexanes, 8.10 mmol, 1.1 equiv) in THF (12 mL) was added *i*-Pr₂NH (1.2 mL, 8.83 mmol, 1.2 equiv) dropwise at 0 °C. The light yellow solution was stirred for 5 min, then cooled to -78 °C in a dry ice-acetone bath. A solution of ester SI-15 (2.24 g, 7.36 mmol, 1.0 equiv) in THF (6 mL) was added dropwise, followed by washings of the syringe (THF, 1 mL x 2). The reaction mixture was stirred at -78 °C for 15 min, and iodomethane (0.60 mL, 9.57 mmol, 1.3 equiv) was added dropwise. After stirring at -78 °C for another 45 min, the reaction mixture was allowed to warm to 20 °C, and quenched with aqueous 1N HCl (ca. 30 mL). The layers were separated, and the aqueous layer was extracted with EtOAc (30 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (2.5 \rightarrow 3% EtOAc in hexanes) furnished ester **15** (1.99 g, 85% yield) as a colorless oil. R_{*J*} = 0.6 (10% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.77 (s, 1H), 6.56 (s, 1H), 4.16 (q, *J* = 7.1 Hz, 2H), 3.79 (s, 3H), 2.61 (t, *J* = 6.5 Hz, 2H), 2.58–2.48 (m, 3H), 1.98–1.88 (m, 1H), 1.84–1.75 (m, 2H), 1.71–1.58 (m, 3H), 1.30–1.26 (m, 9H), 1.22 (d, *J* = 7.0 Hz, 3H); ¹³C NMR (126

MHz, CDCl₃) δ 176.7, 157.4, 147.6, 141.0, 126.5, 112.0, 110.2, 60.4, 55.3, 39.8, 39.0, 34.6, 34.3, 32.2, 32.2, 31.3, 26.6, 19.8, 17.4, 14.5; IR (Neat Film, NaCl) 2935, 1732, 1605, 1467, 1304, 1255, 1156, 1126, 1053, 864 cm⁻¹; HRMS (FAB+) *m/z* calc'd for C₂₀H₃₀O₃ [M]⁺: 318.2195, found 318.2202.



Ketone 13. To a 50 mL round-bottom flask containing ester **15** (1.96 g, 6.16 mmol, 1.0 equiv) was added a magnetic stir bar, MeOH (9 mL), and aqueous 2N NaOH (4.6 mL, 9.24 mmol, 1.5 equiv). The reaction mixture was stirred at 60 °C for 4 h and cooled to ambient temperature. Most of the MeOH was evaporated under reduced pressure, and the alkaline solution was acidified with 1N NCl (15 mL) and extracted with CH_2Cl_2 (20 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure to the crude carboxylic acid as a colorless oil, which was used immediately for the next reaction.

The crude carboxylic acid was added dropwise to a mixture of conc. H₂SO₄ and H₂O (28 mL total, 3:1 v/v) at 0 °C via pipette. Washings of the pipette (Et₂O, 2 mL x 3) were also added. The cooling bath was removed, and the reaction mixture was stirred for 15 min, and then placed in a pre-heated 80 °C oil bath. After 20 min, the reaction mixture was allowed to cool to ambient temperature, diluted with ice water (ca. 90 mL), and extracted with EtOAc (30 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (2.5 \rightarrow 3% Et₂O in CH₂Cl₂) furnished ketone **13** (1.64 g, 97% yield over 2 steps) as a white solid. m.p. 85–86 °C; R_f = 0.7 (10% Et₂O in CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 6.83 (s, 1H), 3.87 (s, 3H), 2.85 (dt, J = 17.6, 4.0 Hz, 1H), 2.72 (ddd, J = 17.0, 11.0, 5.2 Hz, 1H), 2.60–2.47 (m, 3H), 2.20–2.10 (m, 1H), 1.91–1.71 (m, 3H), 1.69–1.57 (m, 2H), 1.30 (s, 6H), 1.19 (d, J = 6.7 Hz, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 200.7, 158.0, 152.2, 144.3, 126.3, 120.9, 108.4, 56.1, 56.1, 43.1, 38.4, 31.7, 31.7, 30.7, 27.1, 26.8, 19.4, 15.5; IR (Neat Film, NaCl) 2928, 1685, 1591, 1559, 1457, 1317, 1246, 1102, 1012 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₁₈H₂₅O₂ [M+H]⁺: 273.1849, found 273.1855.



Allyl enol carbonate 16. To a stirred solution of LiHMDS (1.11 g, 6.61 mmol, 1.1 equiv) in THF (10 mL) at 0 °C was added a solution of ketone 13 (1.64 g, 6.01 mmol, 1.0 equiv) in THF (5 mL) via syringe. Washings of the syringe (THF, 1.5 mL x 2) were also added. The deep red solution was stirred at 0 °C for 1 h. In a separate, flame-dried 200 mL round-bottom flask, THF (34 mL) and allyl chloroformate (0.77 mL, 7.21 mmol, 1.2 equiv) were added, and the solution was cooled to -78 °C in a dry ice-acetone bath. To this cooled solution was added the deep red enolate solution dropwise via cannula (ca. 15 min addition time). The reaction mixture was stirred at -78 °C for another 15 min, and the cold bath was removed. After warming to room temperature, the reaction was quenched with half saturated NH₄Cl solution (ca. 50 mL). The layers were separated, and the aqueous layer was extracted with EtOAc (30 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (3 \rightarrow 5% EtOAc in hexanes) furnished allyl enol carbonate 16 (1.91 g, 89% yield) as a white solid. R_f = 0.6 (16% EtOAc in hexanes); ¹H

NMR (500 MHz, CDCl₃) δ 6.74 (s, 1H), 6.06–5.96 (m, 1H), 5.41 (d, J = 17.1 Hz, 1H), 5.29 (d, J = 10.4 Hz, 1H), 4.70 (d, J = 5.8 Hz, 2H), 3.75 (s, 3H), 2.67 (t, J = 8.0 Hz, 2H), 2.56 (t, J = 6.5 Hz, 2H), 2.27 (t, J = 8.0 Hz, 2H), 1.85 (s, 3H), 1.83–1.76 (m, 2H), 1.64–1.58 (m, 2H), 1.27 (s, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 153.4, 152.8, 146.3, 140.3, 136.1, 132.0, 126.0, 123.5, 118.8, 117.2, 109.1, 68.7, 56.4, 38.7, 34.4, 31.9, 28.7, 27.5, 24.2, 19.7, 16.4; IR (Neat Film, NaCl) 2929, 1762, 1670, 1592, 1465, 1363, 1247, 1107, 1046 cm⁻¹; HRMS (FAB+) *m/z* calc'd for C₂₂H₂₈O₄ [M]⁺: 356.1988, found 356.1979.



α-Allyl ketone 12. In a nitrogen-filled glove box, a 100 mL Schlenk flask was charged with $Pd(OAc)_2$ (2.7 mg, 0.012 mmol, 0.003 equiv), (*S*)-*t*-BuPHOX (15.5 mg, 0.0399 mmol, 0.01 equiv), and MTBE (15 mL). The solution was stirred at ambient temperature for 30 min. Another portion of MTBE (15 mL) was added, and then allyl enol carbonate 16 (1.42 g, 3.99 mmol, 1.0 equiv) was added as a solid to the reaction mixture. Washings of the vial containing 16 (MTBE, 2.5 mL x 4) were also added. The Schlenk flask was sealed with a Kontes valve, brought out of the glove box, and placed in a pre-heated 40 °C oil bath. The reaction mixture was stirred at this temperature for 17 h, at which time TLC analysis indicated complete conversion of starting material. Evaporation of solvent and flash column chromatography on silica gel (10% EtOAc in hexanes) afforded α-allyl ketone 12 (1.24 g, >99% yield) as a viscous colorless oil. $R_f = 0.4$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.83 (s, 1H), 5.86–5.75 (m, 1H), 5.08–5.00 (m, 2H), 3.87 (s, 3H), 2.80–2.66 (m, 2H), 2.52 (t, *J* = 6.5 Hz, 2H),

2.37 (dd, J = 13.8, 7.5 Hz, 1H), 2.28 (dd, J = 13.9, 7.3 Hz, 1H), 1.98 (dt, J = 13.2, 6.4 Hz, 1H), 1.88–1.78 (m, 3H), 1.69–1.60 (m, 2H), 1.30 (s, 3H), 1.29 (s, 3H), 1.14 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 202.1, 158.7, 152.2, 143.7, 134.7, 126.1, 119.8, 117.9, 108.6, 56.1, 45.1, 41.4, 38.4, 34.9, 32.9, 31.7, 31.7, 27.0, 23.6, 21.9, 19.5; IR (Neat Film, NaCl) 2928, 1684, 1591, 1559, 1457, 1318, 1245, 1104, 1016, 913 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₂₁H₂₉O₂ [M+H]⁺: 313.2162, found 313.2158; [α]_D²⁵–13.1 (*c* 1.00, CHCl₃, 94% ee).



Carboxylic acid 11. A flame-dried 25 mL round-bottom flask was charged with a magnetic stir bar, BH₃·SMe₂ (0.6 mL, 2.0 M in THF, 1.2 mmol, 1.2 equiv), and THF (0.5 mL). The solution was cooled to 0 °C in an ice bath, and cyclohexene (0.23 mL, 2.3 mmol, 2.3 equiv) was added. After stirring at 0 °C for 1 h, a solution of α -allyl ketone **12** (312 mg, 1.0 mmol, 1.0 equiv) in THF (0.6 mL) was added. The vial containing **12** was washed with THF (0.6 mL x 2) and the washings were also added. The white slurry soon turned into a light yellow and clear solution (ca. 5 min). The ice bath was removed, and the solution was stirred at 20 °C for 12 h. TLC analysis then indicated almost complete consumption of starting material. To the solution was added H₂O (1 mL) and NaBO₃·H₂O (350 mg). Additional H₂O (10 mL) was added to break up the white solid formed in the mixture. The aqueous phase was extracted with EtOAc (15 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, concentrated under reduced pressure, and quantitatively transferred to a 50 mL round-bottom flask. To this flask was added

a magnetic stir bar, TEMPO (22 mg, 0.14 mmol, 0.14 equiv), MeCN (5 mL), H₂O (2.6 mL), and aqueous phosphate buffer (0.33 M in NaH₂PO₄ and 0.33 M in Na₂HPO₄, 4 mL), and stirred at 20 °C for 5 min. Solid NaClO₂ (622 mg, 5.5 equiv) was added to the flask and the reaction mixture was stirred for 2 min when the solid dissolved. A solution of NaClO (0.26 wt%, 1.2 mL, 0.04 mmol, 0.04 equiv) was added, and the reaction mixture immediately turned dark red. The flask was placed in a pre-heated 35 °C oil bath and stirred for 53 h. TLC analysis showed complete consumption of the alcohol. After cooling to 23 °C, 2N NaOH (5 mL) was added, followed by a solution of Na_2SO_3 (0.88 g) in H₂O (6 mL). The mixture was stirred vigorously for 30 min, and extracted with Et₂O (30 mL x 1). The ethereal layer was discarded. The alkaline aqueous layer was acidified with 8N HCl (ca. 3 mL), and extracted with EtOAc (30 mL x 1). The EtOAc layer was washed with 1N HCl (10 mL) and H₂O (10 mL), dried over Na₂SO₄, filtered, and concentrated under reduced pressure to afford carboxylic acid 11 (252 mg, 73% yield) as a viscous light brown oil. Recrystallization from hexanes/EtOAc gave pure 11 as white crystals. m.p. 125–127 °C; $R_f = 0.4$ (67% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.82 (s, 1H), 3.85 (s, 3H), 2.82-2.68 (m, 2H), 2.51 (t, J = 6.5 Hz, 2H), 2.44 (dt, J = 16.5, 8.1 Hz, 1H), 2.34 (dt, J = 16.5, 8.2 Hz, 1H), 1.98 (dt, J = 13.1, 6.3 Hz, 1H), 1.94–1.79 (m, 5H), 1.66–1.60 (m, 2H), 1.30 (s, 3H), 1.29 (s, 3H), 1.15 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 201.8, 179.5, 158.6. 152.5, 143.4, 126.2, 119.5, 108.6, 56.0, 44.5, 38.4, 34.9, 33.5, 31.7, 31.7, 31.6, 29.4, 27.0, 23.6, 21.9, 19.4; IR (Neat Film, NaCl) 2931, 1708, 1674, 1591, 1558, 1460, 1318, 1245, 1227, 1103, 913, 731 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₂₁H₂₉O₄ [M+H]⁺: 345.2060, found 345.2047; $[\alpha]_D^{25}$ -1.5 (*c* 1.00, CHCl₃).



α-Vinyl ketone 17. Following General Procedure B, palladium-catalyzed decarbonylative dehydration of carboxylic acid 11 (172 mg, 0.5 mmol, 1.0 equiv) and flash column chromatography on silica gel (7→10% EtOAc in hexanes) furnished α-vinyl ketone 17 (140 mg, 93% yield) as a colorless oil. $R_f = 0.3$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.82 (s, 1H), 6.01 (dd, J = 17.7, 10.9 Hz, 1H), 5.08–4.98 (m, 2H), 3.86 (s, 3H), 2.81–2.68 (m, 2H), 2.58–2.43 (m, 2H), 2.08 (dt, J = 13.8, 5.1 Hz, 1H), 2.00–1.90 (m, 1H), 1.89–1.73 (m, 2H), 1.68–1.55 (m, 2H), 1.28 (s, 6H), 1.27 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 199.9, 158.5, 152.3, 143.7, 141.1, 126.0, 120.2, 114.4, 108.5, 56.1, 49.0, 38.4, 34.9, 34.5, 31.7, 31.6, 27.0, 24.0, 23.4, 19.4; IR (Neat Film, NaCl) 2928, 1684, 1591, 1559, 1458, 1318, 1245, 1105, 1020, 915 cm⁻¹; HRMS (MM: ESI-APCI+) *m*/*z* calc'd for C₂₀H₂₇O₂ [M+H]⁺: 299.2006, found 299.1992; [α]_p²⁵–43.2 (*c* 1.00, CHCl₃).



(-)-Aspewentin B (1). A 50 mL round-bottom flask was charged with a magnetic stir bar, NaI (329 mg, 2.20 mmol, 6.0 equiv), AlCl₃ (146 mg, 1.10 mmol, 3.0 equiv), and MeCN (4 mL) in air. To this slurry was added a solution of α -vinyl ketone 17 (109 mg, 0.37 mmol, 1.0 equiv) in MeCN (6 mL). The reaction mixture immediately turned yellow-orange. After stirring for 5

min, the reaction was quenched with 1N HCl (30 mL) and extracted with CH₂Cl₂ (10 mL x 4). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (2.5% Et₂O in hexanes) furnished (–)-aspewentin B (1, 82 mg, 78% yield) as a light yellow oil. $R_f = 0.7$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 12.40 (s, 1H), 6.84 (s, 1H), 5.98 (dd, J = 17.7, 10.8 Hz, 1H), 5.13 (d, J = 10.7 Hz, 1H), 5.03 (d, J = 17.6 Hz, 1H), 2.81–2.69 (m, 2H), 2.59–2.44 (m, 2H), 2.14–2.06 (m, 1H), 2.04–1.94 (m, 1H), 1.88–1.73 (m, 2H), 1.68–1.55 (m, 2H), 1.33 (s, 3H), 1.27 (s, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 207.2, 160.9, 156.1, 142.3, 140.5, 124.7, 115.1, 114.6, 113.3, 47.9, 38.3, 35.0, 34.4, 31.7, 31.6, 26.9, 23.4, 23.3, 19.4; IR (Neat Film, NaCl) 2929, 1635, 1464, 1358, 1222, 1083, 920, 809 cm⁻¹; HRMS (FAB+) *m/z* calc'd for C₁₉H₂₅O₂ [M+H]⁺: 285.1855, found 285.1842; [α]_D²⁵–90.5 (*c* 0.20, MeOH, 98% ee).⁹



(–)-Aspewentin A (18). A 20 mL scintillation vial was charged with a magnetic stir bar and NaBH₄ (38 mg, 1.0 mmol, 10 equiv). Trifluoroacetic acid (1 mL) was added carefully with stirring. Solids soon dissolved, with a slight exotherm. A solution of (–)-aspewentin B (1, 29 mg, 0.10 mmol, 1.0 equiv) in CH₂Cl₂ (2 mL) was added, and the reaction mixture immediately turned yellow. After stirring 1 h, additional NaBH₄ (7.5 mg, 0.20 mmol, 2.0 equiv) was added. After 45 min, a third portion of NaBH₄ (11 mg, 0.30 mmol, 3.0 equiv, 15 equiv total) was added. After stirring another 2 h, the reaction was quenched with saturated aq. NaHCO₃ solution (ca. 15 mL). Care should be taken during addition of NaHCO₃ due to rapid gas evolution. The biphasic

mixture was extracted with EtOAc (15 mL x 2). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (9% Et₂O in hexanes) furnished (–)-aspewentin A (**18**, 23 mg, 85% yield) as a colorless oil. $R_f = 0.4$ (20% Et₂O in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.67 (s, 1H), 5.90 (dd, J = 17.5, 10.7 Hz, 1H), 4.99 (dd, J = 17.5, 1.3 Hz, 1H), 4.96 (dd, J = 10.8, 1.2 Hz, 1H), 4.69 (br s, 1H), 2.66 (d, J = 16.3 Hz, 1H), 2.57 (t, J = 6.7 Hz, 2H), 2.50 (t, J = 6.5 Hz, 2H), 2.46 (d, J = 16.3 Hz, 1H), 1.86–1.79 (m, 2H), 1.76–1.70 (m, 1H), 1.69–1.64 (m, 1H), 1.64–1.60 (m, 2H), 1.27 (s, 3H), 1.27 (s, 3H), 1.11 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 151.3, 147.2, 143.9, 135.2, 126.5, 119.7, 111.2, 110.0, 38.8, 34.5, 34.5, 34.1, 33.9, 32.1, 32.0, 26.8, 26.0, 24.4, 19.7; IR (Neat Film, NaCl) 3307, 2923, 1606, 1456, 1418, 1324, 1252, 1019, 910, 858 cm⁻¹; HRMS (FAB+) *m/z* calc'd for C₁₉H₂₆O [M]⁺: 270.1984, found 270.1972; [α]_D²⁵–38.6 (*c* 0.20, MeOH).



(-)-Aspewentin C (19) and (+)-10-*epi*-aspewentin C (20). The reaction was performed according to a literature procedure.¹⁰ To a 20 mL scintillation vial containing (-)-aspewentin A (18, 23 mg, 0.085 mmol, 1.0 equiv) was added a magnetic stir bar and $Rh_2(cap)_4$ (0.6 mg, 0.00085 mmol, 0.01 equiv). The vial was sealed with a hollow plastic cap containing a PTFE septum, and evacuated and refilled with N₂ (x 3). 1,2-DCE (2 mL) was added via syringe, and the solution was stirred at 60 °C in an oil bath for 2 min to ensure complete dissolution of the catalyst. The reaction mixture was cooled to 0 °C, and *tert*-butyl hydroperoxide (118 μ L, 0.85 mmol, 10 equiv) was added dropwise. The reaction mixture immediately turned red. After
stirring at 0 °C for 40 min, the reaction was quenched with H₂O and extracted with CH₂Cl₂ (10 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (6% Et₂O in hexanes) afforded the peroxide intermediate (12.5 mg, $R_f = 0.6$ (20% Et₂O in hexanes)), which was used directly in the next reaction.

To a 20 mL scintillation vial containing the peroxide intermediate (11 mg, 0.031 mmol, 1.0 equiv) was added THF (0.65 mL) and ag 1M NH₄OAc (0.65 mL) under N₂. 10% Cd/Pb couple (150 mg, 1.24 mmol, 40 equiv) was added, and the reaction mixture was stirred vigorously at 20 °C. After 4 h, another portion of 10% Cd/Pb couple (37 mg, 0.31 mmol, 10 equiv) was added to increase conversion. After 2 h, a third portion of 10% Cd/Pb couple (37 mg, 0.31 mmol, 10 equiv) was added. After stirring for 4.5 h, the reaction mixture was filtered through a cotton plug and eluted with H₂O and CH₂Cl₂. The biphasic mixture was extracted with CH₂Cl₂ (10 mL x 3). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Flash column chromatography on silica gel (9% EtOAc in hexanes) furnished (-)-aspewentin C (19, 2.8 mg, 13% yield over 2 steps) as a white solid. $R_f =$ 0.3 (16% EtOAc in hexanes); ¹H NMR (500 MHz, Acetone- d_6) δ 5.98 (s, 1H), 5.75 (dd, J = 17.5, 10.7 Hz, 1H), 4.90 (dd, J = 10.7, 1.4 Hz, 1H), 4.81 (dd, J = 17.5, 1.4 Hz, 1H), 4.23 (d, J = 2.3Hz, 1H), 2.72-2.60 (m, 1H), 2.44 (br d, J = 17.4 Hz, 1H), 2.27-2.19 (m, 2H), 2.18-2.08 (m, 1H), 1.95 (br d, J = 17.4 Hz, 1H), 1.72–1.60 (m, 2H), 1.56–1.47 (m, 1H), 1.42 (s, 3H), 1.41–1.30 (m, 2H), 1.16 (s, 3H), 1.15–1.07 (m, 1H), 1.03 (s, 3H); ¹³C NMR (101 MHz, Acetone-d₆) δ 185.1, 167.7, 157.2, 145.6, 128.5, 122.4, 110.8, 70.2, 41.9, 37.9, 37.5, 34.4, 33.4, 32.4, 30.4, 27.4, 26.9, 21.5, 17.6; IR (Neat Film, NaCl) 3419, 2956, 2919, 1659, 1623, 1603, 1398, 1302, 1137, 1073, 1040, 1016, 959, 909, 878, 734 cm⁻¹; HRMS (MM: ESI-APCI+) m/z calc'd for C₁₉H₂₇O₂ $[M+H]^+$: 287.2006, found 287.2016; $[\alpha]_D^{25}$ –123.2 (*c* 0.10, MeOH). [Note: ¹³C NMR spectrum also contains smaller peaks at 70.1 ppm and 122.4 ppm. We think that these additional peaks belong to an unknown impurity, which may also be responsible for the large difference in optical rotation of the synthetic compound and the natural product.]

The same reaction also formed (+)-10-*epi*-aspewentin C (**20**, 1.6 mg, 7% yield over 2 steps) as a white solid. $R_f = 0.4$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, Acetone-*d*₆) δ 5.99 (s, 1H), 5.86 (dd, J = 17.5, 10.8 Hz, 1H), 4.95 (dd, J = 17.5, 1.4 Hz, 1H), 4.90 (dd, J = 10.8, 1.3 Hz, 1H), 4.24 (d, J = 2.3 Hz, 1H), 2.75–2.64 (m, 1H), 2.41–2.33 (m, 1H), 2.33–2.27 (m, 1H), 2.21–2.10 (m, 3H), 1.69 (dtd, J = 13.1, 3.4, 2.0 Hz, 1H), 1.60–1.50 (m, 2H), 1.50–1.44 (m, 1H), 1.43 (s, 3H), 1.36 (td, J = 13.4, 4.1 Hz, 1H), 1.22 (tdd, J = 13.2, 4.3, 2.1 Hz, 1H), 1.16 (s, 3H), 0.96 (s, 3H); ¹³C NMR (126 MHz, Acetone-*d*₆) δ 185.2, 167.8, 157.0, 148.1, 128.3, 122.4, 109.9, 70.2, 42.0, 38.0, 37.5, 33.9, 33.2, 32.7, 30.5, 26.9, 23.3, 21.1, 17.6; IR (Neat Film, NaCl) 3400, 2958, 2923, 1660, 1626, 1604, 1397, 1302, 1128, 1075, 1039, 1017, 958, 939, 906, 877, 825 cm⁻¹; HRMS (MM: ESI-APCI+) *m*/*z* calc'd for C₁₉H₂₇O₂ [M+H]⁺: 287.2006, found 287.2016; [α] $_{0}^{25}$ +39.4 (*c* 0.10, MeOH). [Note: the two big peaks not having integration values (around 2.8 ppm) in the ¹H NMR spectra of compounds **19** and **20** belong to water; see spectra on pages 114 and 116, respectively.]



Ketone SI-16. Using α -allyl ketone **12** as starting material (20 mg, 0.064 mmol, 1.0 equiv), and following the same procedure as that for (–)-aspewentin B (1), ketone **SI-16** was obtained (17

mg, 89% yield) as a colorless oil. $R_f = 0.7$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 12.47 (s, 1H), 6.84 (s, 1H), 5.83–5.72 (m, 1H), 5.13–5.05 (m, 2H), 2.82–2.66 (m, 2H), 2.52 (t, J = 6.5 Hz, 2H), 2.46 (dd, J = 13.9, 7.3 Hz, 1H), 2.26 (dd, J = 13.8, 7.5 Hz, 1H), 2.08–1.98 (m, 1H), 1.90–1.77 (m, 3H), 1.65–1.59 (m, 2H), 1.28 (s, 6H), 1.19 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 209.1, 161.0, 156.0, 142.2, 133.8, 124.7, 118.6, 114.2, 113.4, 44.1, 41.2, 38.3, 35.0, 32.6, 31.6, 31.6, 26.9, 23.0, 22.2, 19.4; IR (Neat Film, NaCl) 2930, 1634, 1464, 1360, 1220, 1190, 919, 811 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₂₀H₂₇O₂ [M+H]⁺: 299.2006, found 299.1999; [α]_D²⁵–23.5 (*c* 0.20, MeOH).



Ketone SI-17. To a solution of α-allyl ketone 12 (109 mg, 0.349 mmol, 1.0 equiv) and vinyloxytrimethylsilane (0.52 mL, 3.49 mmol, 10.0 equiv) in toluene (19 mL) was added Grubbs 2^{nd} generation catalyst (14.8 mg, 0.01745 mmol, 0.05 equiv) at 20 °C. The purple reaction mixture was immersed in a pre-heated 128°C oil bath (color changed to yellow) and refluxed for 16 h. The reaction mixture was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (10% EtOAc in hexanes) to afford ketone SI-17 (103 mg, 92% conv., 94% yield) as a colorless oil. $R_f = 0.4$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 6.82 (s, 1H), 5.60 (d, *J* = 15.9 Hz, 1H), 5.48–5.38 (m, 1H), 3.86 (s, 3H), 2.79–2.66 (m, 2H), 2.58–2.45 (m, 2H), 2.08–1.99 (m, 1H), 1.96–1.88 (m, 1H), 1.87–1.74 (m, 2H), 1.68–1.56 (m, 5H), 1.30 (s, 3H), 1.28 (s, 3H), 1.24 (s, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 200.4, 158.6, 152.1, 143.9, 133.6, 126.0, 125.0, 120.2, 108.5, 56.1, 48.2, 38.4, 35.1, 34.9, 31.8,

31.6, 27.0, 24.1, 24.0, 19.5, 18.5; IR (Neat Film, NaCl) 2928, 1683, 1591, 1558, 1456, 1318, 1245, 1106, 1020, 966 cm⁻¹; HRMS (MM: ESI-APCI+) m/z calc'd for C₂₁H₂₉O₂ [M+H]⁺: 313.2162, found 313.2154; $[\alpha]_D^{25}$ –49.0 (*c* 1.00, CHCl₃).



Ketone SI-18. Using ketone **SI-17** as starting material (35 mg, 0.112 mmol, 1.0 equiv), and following the same procedure as that for (–)-aspewentin B (1), ketone **SI-18** was obtained (30 mg, 90% yield) as a colorless oil. $R_f = 0.7$ (16% EtOAc in hexanes); ¹H NMR (500 MHz, CDCl₃) δ 12.46 (s, 1H), 6.84 (s, 1H), 5.57 (d, J = 15.9 Hz, 1H), 5.49–5.39 (m, 1H), 2.79–2.67 (m, 2H), 2.59–2.44 (m, 2H), 2.09–2.00 (m, 1H), 2.00–1.91 (m, 1H), 1.89–1.74 (m, 2H), 1.66 (d, J = 5.9 Hz, 3H), 1.64–1.55 (m, 2H), 1.30 (s, 3H), 1.28 (s, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 207.7, 160.9, 155.9, 142.5, 133.1, 125.9, 124.6, 114.6, 113.3, 47.1, 38.3, 35.1, 35.0, 31.7, 31.6, 26.9, 23.9, 23.5, 19.4, 18.4; IR (Neat Film, NaCl) 2929, 1634, 1464, 1359, 1267, 1221, 1191, 964 cm⁻¹; HRMS (MM: ESI-APCI+) *m/z* calc'd for C₂₀H₂₇O₂ [M+H]⁺: 299.2006, found 299.2002; [α]_D²⁵–65.1 (*c* 0.20, MeOH).



Ethenolysis of SI-17. In a nitrogen-filled glove box, a Fisher-Porter bottle was charged with a magnetic stir bar, toluene (10 mL), and Grubbs catalyst (1.7 mg, 0.0028 mmol, 0.04 equiv). A solution of **SI-17** (22 mg, 0.07 mmol, 1.0 equiv) in toluene (0.2 mL) was added. The head of the Fisher-Porter bottle was equipped with a pressure gauge, and a dip-tube was adapted on the bottle. The system was sealed and taken out of the glove box and connected to the ethylene line. The vessel was then purged with ethylene (polymer purity 99.9% from Matheson Tri Gas) for 5 min, pressurized to 150 psi, and placed in an oil bath at 40 °C. After stirring for 1.5 h, the solvent was evaporated, and the residue was diluted in EtOAc and passed through a short plug of silica gel to remove the ruthenium catalyst. The filtrate was concentrated under reduced pressure, and ¹H NMR analysis of the residue showed starting material **SI-17** remaining, and no peaks corresponding to the desired product **17**.

Synthetic (–)-Aspewentin A	Natural (+)-Aspewentin A ¹¹			
¹ H NMR (500 MHz, CDCl ₃)	¹ H NMR (500 MHz, CDCl ₃)			
6.67 (s, 1H)	6.66 (s, 1H)			
5.90 (dd, <i>J</i> = 17.5, 10.7 Hz, 1H)	5.89 (dd, J = 17.6, 10.8 Hz, 1H)			
4.99 (dd, J = 17.5, 1.3 Hz, 1H)	4.98 (br d, $J = 17.6$ Hz, 1H)			
4.96 (dd, J = 10.8, 1.2 Hz, 1H)	4.94 (br d, $J = 10.8$ Hz, 1H)			
4.69 (br s, 1H)	4.44 (br s, 1H)			
2.66 (d, <i>J</i> = 16.3 Hz, 1H)	2.64 (d, 16.4 Hz, 1H)			
2.57 (t, J = 6.7 Hz, 2H)	2.55 (t, 6.6 Hz, 2H)			
2.50 (t, J = 6.5 Hz, 2H)	2.48 (t, 6.5 Hz, 2H)			
2.46 (d, <i>J</i> = 16.3 Hz, 1H)	2.44 (d, 16.4 Hz, 1H)			
1.86–1.79 (m, 2H)	1.80 (m, 2H)			
1.76–1.70 (m, 1H)	1.71 (m, 1H)			
1.69–1.64 (m, 1H)	1.64 (m, 1H)			
1.64–1.60 (m, 2H)	1.60 (m, 2H)			
1.27 (s, 3H), 1.27 (s, 3H)	1.25 (s, 3H), 1.25 (s, 3H)			
1.11 (s, 3H)	1.08 (s, 3H)			
¹³ C NMR (101 MHz, CDCl ₃)	^{13}C NMR (125 MHz, CDCl ₃)			
151.3	151.2			
147.2	147.1			
143.9	143.8			
135.2	135.1			
126.5	126.4			
119.7	119.6			
111.2	111.1			
110.0	109.9			
38.8	38.7			
34.5	34.4			
34.5	34.4			
34.1	34.0			
33.9	33.8			
32.1	32.0			
32.0	31.9			
26.8	26.7			
26.0	25.8			
24.4	24.2			
19.7	19.6			
Optical Rotation	Optical Rotation			
$[\alpha]_{D}^{25}$ –38.6 (<i>c</i> 0.20, MeOH)	$[\alpha]_{D}^{14}$ +41.8 (<i>c</i> 0.19, MeOH)			

Comparison of Synthetic and Natural Aspewentin A (Table S1)

Synthetic (–)-Aspewentin B	Natural (+)-Aspewentin B ¹¹			
^{$IH NMR (500 MHz, CDCl3)$}	¹ H NMR (500 MHz, CDCl ₃)			
12.40 (s, 1H)	12.40 (s, 1H)			
6.84 (s, 1H)	6.84 (s, 1H)			
5.98 (dd, <i>J</i> = 17.7, 10.8 Hz, 1H)	5.98 (dd, $J = 17.6$, 10.8 Hz, 1H)			
5.13 (d, J = 10.7 Hz, 1H)	5.14 (d, J = 10.8 Hz, 1H)			
5.03 (d, <i>J</i> = 17.6 Hz, 1H)	5.03 (d, J = 17.6 Hz, 1H)			
2.81–2.69 (m, 2H)	2.75 (m, 2H)			
2.59–2.44 (m, 2H)	2.51 (m, 2H)			
2.14–2.06 (m, 1H)	2.10 (m, 1H)			
2.04–1.94 (m, 1H)	2.00 (m, 1H)			
1.88–1.73 (m, 2H)	1.81 (m, 2H)			
1.68–1.55 (m, 2H)	1.61 (m, 2H)			
1.33 (s, 3H)	1.33 (s, 3H)			
1.27 (s, 6H)	1.28 (s, 3H), 1.27 (s, 3H)			
$^{13}CNMR$ (126 MHz, CDCl ₃)	$^{13}C NMR (125 MHz, CDCl_3)$			
207.2	207.1			
160.9	160.8			
156.1	156.2			
142.3	142.2			
140.5	140.4			
124.7	124.6			
115.1	115.0			
114.6	114.4			
113.3	113.3			
47.9	47.8			
38.3	38.2			
35.0	34.9			
34.4	34.3			
31.7	31.6			
31.6	31.5			
26.9	26.8			
23.4	23.3			
23.3	23.2			
19.4	19.3			
Optical Rotation	Optical Rotation			
$[\alpha]_{D}^{25}$ –90.5 (<i>c</i> 0.20, MeOH, 98% ee)	$[\alpha]_{D}^{20}$ +23.3 (c 0.20, MeOH)			

Comparison of Synthetic and Natural Aspewentin B (Table S2)

Synthetic (–)-Aspewentin C	Natural (+)-Aspewentin C ¹¹			
¹ H NMR (500 MHz, Acetone-d ₆)	¹ H NMR (500 MHz, Acetone-d ₆)			
5.98 (s, 1H)	5.99 (s, 1H)			
5.75 (dd, <i>J</i> = 17.5, 10.7 Hz, 1H)	5.76 (dd, J = 17.5, 10.8 Hz, 1H)			
4.90 (dd, <i>J</i> = 10.7, 1.4 Hz, 1H)	4.91 (br d, $J = 10.8$ Hz, 1H)			
4.81 (dd, <i>J</i> = 17.5, 1.4 Hz, 1H)	4.82 (br d, $J = 17.5$ Hz, 1H)			
4.23 (d, $J = 2.3$ Hz, 1H)	4.22 (br s, 1H)			
2.72–2.60 (m, 1H)	2.68 (m, 1H)			
2.44 (br d, $J = 17.4$ Hz, 1H)	2.44 (br d, 17.8 Hz, 1H)			
2.27–2.19 (m, 2H)	2.24 (m, 1H), 2.24 (m, 1H)			
2.18–2.08 (m, 1H)	2.15 (m, 1H)			
1.95 (br d, $J = 17.4$ Hz, 1H)	1.96 (br d, 17.8 Hz, 1H)			
1.72–1.60 (m, 2H)	1.69 (m, 1H), 1.65 (m, 1H)			
1.56–1.47 (m, 1H)	1.53 (m, 1H)			
1.42 (s, 3H)	1.43 (s, 3H)			
1.41–1.30 (m, 2H)	1.41 (m, 1H), 1.35 (m, 1H)			
1.16 (s, 3H)	1.16 (s, 3H)			
1.15–1.07 (m, 1H)	1.13 (m, 1H)			
1.03 (s, 3H)	1.03 (s, 3H)			
^{13}C NMR (101 MHz, Acetone-d ₆)	^{13}C NMR (125 MHz, Acetone-d ₆)			
185.1	185.1			
167.7	167.7			
157.2	157.2			
145.6	145.7			
128.5	128.5			
122.4	122.4			
110.8	110.8			
70.2	70.2			
41.9	41.9			
37.9	37.9			
37.5	37.5			
34.4	34.4			
33.4	33.4			
32.4	32.5			
30.4	30.4			
27.4	27.4			
26.9	26.9			
21.5	21.5			
17.6	17.6			
Optical Rotation	Optical Rotation			
$[\alpha]_{D}^{25}$ –123.2 (<i>c</i> 0.10, MeOH)	$[\alpha]_{\rm D}^{21}$ +8.3 (<i>c</i> 0.10, MeOH)			

Comparison of Synthetic and Natural Aspewentin C (Table S3)

Methods for the Determination of Enantiomeric Excess (Table S4)

entry	compound	assay conditions	retention time of major isomer (min)	retention time of minor isomer (min)	% ee
1	O O O O O O O O O O O O O O O O O O O	HPLC Chiralcel OD-H 10% IPA in hexanes isocratic, 1.0 mL/min 220 nm	6.43	5.93	91
2	ent-7	GC G-TA 60 °C isotherm 5 min then ramp 2 °C/min	24.20	23.92	92
3	о о о о о о о о о о о о о о о о о о о	HPLC Chiralcel OD-H 5% IPA in hexanes isocratic, 1.0 mL/min 210 nm	8.08	9.99	91
4	o ge	HPLC Chiralpak AS 0.2% IPA in hexanes isocratic, 1.0 mL/min 210 nm	9.50	7.89	92
5	OMe 12	SFC Chiralcel OB-H 3% MeOH in CO ₂ isocratic, 3.0 mL/min 254 nm	8.82	6.41	94
6	(-)-Aspewentin B (1)	HPLC Chiralpak AD 0.5% IPA in hexanes isocratic, 1.0 mL/min 210 nm	7.22	6.56	98

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Infrared spectrum (Thin Film, NaCl) of compound 8a.







Infrared spectrum (Thin Film, NaCl) of compound SI-3.







Infrared spectrum (Thin Film, NaCl) of compound 8b.







Infrared spectrum (Thin Film, NaCl) of compound 8c.







Infrared spectrum (Thin Film, NaCl) of compound 8d.







Infrared spectrum (Thin Film, NaCl) of compound SI-7.







Infrared spectrum (Thin Film, NaCl) of compound SI-8.







Infrared spectrum (Thin Film, NaCl) of compound 8e.







Infrared spectrum (Thin Film, NaCl) of compound 8f.







Infrared spectrum (Thin Film, NaCl) of compound 8g.







Infrared spectrum (Thin Film, NaCl) of compound SI-12.







Infrared spectrum (Thin Film, NaCl) of compound 8h.







¹³C NMR (126 MHz, CDCl₃) of compound 8i.

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Infrared spectrum (Thin Film, NaCl) of compound 8j.







Infrared spectrum (Thin Film, NaCl) of compound 9a.







Infrared spectrum (Thin Film, NaCl) of compound 10.







Infrared spectrum (Thin Film, NaCl) of compound 7.







Infrared spectrum (Thin Film, NaCl) of compound 9c.







Infrared spectrum (Thin Film, NaCl) of compound 9d.







Infrared spectrum (Thin Film, NaCl) of compound 9e.



¹³C NMR (126 MHz, CDCl₃) of compound **9e**.





Infrared spectrum (Thin Film, NaCl) of compound 9f.







Infrared spectrum (Thin Film, NaCl) of compound 9g.







Infrared spectrum (Thin Film, NaCl) of compound 9h.







Infrared spectrum (Thin Film, NaCl) of compound 9i.







Infrared spectrum (Thin Film, NaCl) of compound 9j.







Infrared spectrum (Thin Film, NaCl) of compound SI-15.







Infrared spectrum (Thin Film, NaCl) of compound 15.







Infrared spectrum (Thin Film, NaCl) of compound 13.







Infrared spectrum (Thin Film, NaCl) of compound 16.







Infrared spectrum (Thin Film, NaCl) of compound 12.







Infrared spectrum (Thin Film, NaCl) of compound 11.






Infrared spectrum (Thin Film, NaCl) of compound 17.







Infrared spectrum (Thin Film, NaCl) of compound 1.







Infrared spectrum (Thin Film, NaCl) of compound 18.











Infrared spectrum (Thin Film, NaCl) of compound 20.







Infrared spectrum (Thin Film, NaCl) of compound SI-16.













Infrared spectrum (Thin Film, NaCl) of compound SI-18.



Representative Chiral HPLC, SFC, and GC Traces

Data File C:\CHEM32\1\DATA\YL11\YL-XI-223A_OD_10IPA15-220.D

Sample Name: YL-XI-223A_OD

Acq. Operator	:	YL Seq. Line : 7
Acq. Instrument	:	HPLC 1 Location : Vial 31
Injection Date	:	11/22/2014 8:50:26 PM Inj: 1
		Inj Volume : 5.0 µl
Different Inj Vo	plu	ume from Sequence ! Actual Inj Volume : 10.0 µl
Acq. Method	:	C:\CHEM32\1\METHODS\10IPA15_220.M
Last changed	:	11/22/2014 8:42:16 PM by MW
Analysis Method	:	C:\CHEM32\1\METHODS\10IPA20_254_0_5MLMIN.M
Last changed	:	1/26/2015 5:59:15 PM by SJH
		(modified after loading)
Method Info	:	10% IPA 20 min 254 nm 0.5 mL/min





Area	Percent	Report	

Sorted By	:	Sign	nal		
Multiplier:		:	1.0	0000	
Dilution:		:	1.0	0000	
Do not use Multiplier	б.	Dilution	Factor	with	ISTDs

Signal 1: VWD1 A, Wavelength=220 nm, TT

Peak	RetTime	Type	Width	Are	ea	Heig	ght	Area
#	[min]		[min]	mAU	*s	[mAU]	8
1	5.916	BB	0.1171	107.3	35893	14.1	12435	50.4407
2	6.442	BB	0.1282	105.4	48306	12.0	69412	49.5593

Totals : 212.84200 26.81847

Summed Peaks Report

Data File C:\CHEM32\1\DATA\YL11\YL-XI-223B-2_OD_10IPA15-220.D Sample Name: YL-XI-223B-2 OD Acq. Operator : YL Seq. Line : 4 Location : Vial 33 Acq. Instrument : HPLC 1 Injection Date : 11/25/2014 10:50:17 PM Inj: 1 Inj Volume : 5.0 µl Different Inj Volume from Sequence ! Actual Inj Volume : 10.0 µl Acq. Method : C:\CHEM32\1\METHODS\10IPA15 220.M Last changed : 11/22/2014 8:42:16 PM by MW Analysis Method : C:\CHEM32\1\METHODS\10IPA20 254 0 5MLMIN.M Last changed : 1/26/2015 5:59:15 PM by SJH (modified after loading) Method Info : 10% IPA 20 min 254 nm 0.5 mL/min





Area Percent Report _____ Sorted By : Signal Multiplier: : Dilution: : 1.0000 : 1.0000 Do not use Multiplier & Dilution Factor with ISTDs Signal 1: VWD1 A, Wavelength=220 nm, TT Peak RetTime Type Width Area Height Area # [min] [min] mAU *s [mAU] % 1 5.933 MM 0.1368 11.08131 1.35039 4.6650 6.433 MM 0.1432 226.45950 26.36230 95.3350 2 Totals : 237.54081 27.71269 Summed Peaks Report

HPLC 1 1/26/2015 5:59:24 PM SJH

Data File C:\HPCHEM\2\DATA\YIYANG\9223-2.D



GC4 5/5/2015 11:15:35 AM KEK

Data File C:\HPCHEM\2\DATA\YIYANG\X045-1.D



GC4 5/5/2015 11:14:47 AM KEK

Data File C:\CHEM32\2\DATA\YL11\NV1 2014-10-25 21-22-40\YL-XI-177RAC_OD_5IPA20-210.D Sample Name: YL-XI-177rac-OD

Acq. Operator	:	YL	Seq. Line : 14	
Acq. Instrument	:	HPLC 2	Location : Vial 1	
Injection Date	:	10/25/2014 11:42:10 PM	Inj: 1	_
			Inj Volume : 5.0 µl	
Acq. Method	:	C:\CHEM32\2\DATA\YL11\NV1	2014-10-25 21-22-40\5IPA20_210.M	
Last changed	:	10/25/2014 9:19:06 PM by M	IW	1
Analysis Method	:	C:\CHEM32\2\METHODS\2IPA2C	_210.M	
Last changed	:	10/25/2014 10:23:26 PM by	MW	
Method Info	:	2% IPA 20 min 210 nm	1 mL/min	





_____ Area Percent Report _____ : Signal Sorted By Multiplier: : 1.0000 Dilution: : 1.0000 Do not use Multiplier & Dilution Factor with ISTDs Signal 1: VWD1 A, Wavelength=210 nm, TT Peak RetTime Type Width Area Height Area # [min] [min] mAU *s [mAU] 8 18.037 BB0.2061 1410.35120104.1317050.337429.904 BB0.2629 1391.4469080.7422749.6626 2801.79810 184.87397 Totals : _____ Summed Peaks Report

Summed Feaks Report

Data File C:\CHEM32\2\DATA\YL11\NV1 2014-10-25 21-22-40\YL-XI-219_OD_5IPA20-210.D Sample Name: YL-XI-219-OD

Acq. Operator	: YL Seq. Line : 16
Acq. Instrument	: HPLC 2 Location : Vial 2
Injection Date	: 10/26/2014 12:14:06 AM Inj : 1
	Inj Volume : 5.0 µl
Acq. Method	: C:\CHEM32\2\DATA\YL11\NV1 2014-10-25 21-22-40\5IPA20_210.M
Last changed	: 10/25/2014 9:19:06 PM by MW
Analysis Method	: C:\CHEM32\2\METHODS\2IPA20_210.M
Last changed	: 10/26/2014 12:42:41 AM by MW
	(modified after loading)
Method Info	: 2% IPA 20 min 210 nm 1 mL/min





2 9.990 BB 0.2448 68.03126 4.20617 4.2679

Totals :

1594.01527 118.82023

Summed Peaks Report

Data File C:\CHEM32\...TA\YL11\YL11 2014-11-01 14-21-30\YL-X-211FLASH-2_AS_02IPA20-210YL.D Sample Name: YL-X-211flash-AS Acq. Operator : YL Seq. Line : 7 Acq. Instrument : HPLC 2 Location : Vial 23 Injection Date : 11/1/2014 4:08:18 PM Inj : 1 Inj Volume : 5.0 µl

Inj Volume : 5.0 µl Acq. Method : C:\CHEM32\2\DATA\YL11\YL11 2014-11-01 14-21-30\02IPA20_210YL.M Last changed : 11/1/2014 10:34:26 AM by MW Analysis Method : C:\CHEM32\2\METHODS\02IPA20_210YL.M Last changed : 11/1/2014 10:34:26 AM by MW Method Info : 0.2% IPA 20 min 210 nm 1 mL/min YL





Area Percent Report

Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Do not use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=210 nm, TT

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	mAU *s	[mAU]	8
1	7.945	BV	0.4982	2409.75024	69.29615	49.8495
2	9.612	VB	0.6070	2424.29688	56.40124	50.1505

Totals :

4834.04712 125.69739

Summed Peaks Report

HPLC 2 11/1/2014 4:43:23 PM MW

Data File C:\CHEM32\2\DATA\YL11\YL11 2014-11-01 14-21-30\YL-XI-229FLASH_AS_02IPA20-210YL.D Sample Name: YL-XI-229flash-AS

Acq. Operator	:	YL	Seq. Line : 5			
Acq. Instrument	:	HPLC 2	Location : Vial 25			
Injection Date	:	11/1/2014 3	3:25:55 PM Inj: 1	-		
			Inj Volume : 5.0 µl			
Acq. Method	:	C:\CHEM32\2	2\DATA\YL11\YL11 2014-11-01 14-21-30\02IPA20_210YL.	М		
Last changed	:	11/1/2014 1	10:34:26 AM by MW		r	
Analysis Method	:	C:\CHEM32\2	2\METHODS\02IPA20_210YL.M		Į	
Last changed	:	11/1/2014 1	10:34:26 AM by MW			
Method Info	:	0.2% IPA	20 min 210 nm 1 mL/min YL	L	-	



_____ Area Percent Report _____ : Signal Sorted By Multiplier: : 1.0000 Dilution: : 1.0000 Do not use Multiplier & Dilution Factor with ISTDs Signal 1: VWD1 A, Wavelength=210 nm, TT Peak RetTime Type Width Area Height Area # [min] [min] mAU *s [mAU] 8 17.888 VB0.4110171.554865.250784.179929.497 VB0.59453932.7294992.0554495.8201 4104.28435 97.30622 Totals : _____ Summed Peaks Report

HPLC 2 11/1/2014 4:43:51 PM MW

Data File C:\CHEM32\1\DATA\YL\2014-08-16 20-31-55\YL-XI-027RAC_S1C6_12MIN3_3.D Sample Name: YL-XI-027rac



Data File C:\CHEM32\1\DATA\YL\2014-08-16 20-31-55\YL-XI-069-2_S1C6_12MIN3_3.D Sample Name: YL-XI-069-2



Data File C:\HPCHEM\1\DATA\GROUP\YYL12115.D

Sample Name: YYL-XII-115



Instrument 1 5/3/2015 5:22:59 PM XM

Data File C:\HPCHEM\1\DATA\GROUP\YYL12115.D

Sample Name: YYL-XII-115

	Area Percent	Report	
Sorted By : Multiplier : Dilution : Use Multiplier & Dilution	Signal 1.0000 1.0000 Factor with	ISTDs	
Signal 1: DAD1 A, Sig=250	,10 Ref=360,1	100	
Signal 2: DAD1 B, Sig=254	,10 Ref=360,3	100	
Signal 3: DAD1 C, Sig=210	,10 Ref=360,3	100	
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %
1 6.505 BV 0.1434 2 7.192 VB 0.1608	5666.01025 5689.62744	604.05304 540.63428	49.8960 50.1040
Totals :	1.13556e4	1144.68732	
Results obtained with en	hanced integ:	cator!	
Signal 4: DAD1 D, Sig=230	,10 Ref=360,3	100	
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %
1 6.505 BV 0.1437 2 7.192 VB 0.1612	5709.97949 5738.43896	607.32703 543.26447	49.8757 50.1243
Totals :	1.14484e4	1150.59149	
Results obtained with en	hanced integ:	rator!	
Signal 5: DAD1 E, Sig=280	,10 Ref=360,3	100	
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %
1 6.505 BV 0.1464 2 7.192 VB 0.1618	3253.07544 3279.59814	343.87250 309.11072	49.7970 50.2030
Totals :	6532.67358	652.98322	
Results obtained with en	hanced integ: =======	rator!	

*** End of Report ***

(±)-Aspewentin B (1)

Data File C:\HPCHEM\1\DATA\GROUP\YYL1111.D

Sample Name: YYL-XI-111



Instrument 1 5/3/2015 5:21:37 PM XM

Data File C:\HPCHEM\1\DATA\GROUP\YYL1111.D

Sample Name: YYL-XI-111

	Area Percent	Report		
Sorted By : Multiplier : Dilution : Use Multiplier & Dilution	Signal 1.0000 1.0000 Factor with	n ISTDs		
Signal 1: DAD1 A, Sig=250	,10 Ref=360,	100		
Signal 2: DAD1 B, Sig=254	,10 Ref=360,	100		
Signal 3: DAD1 C, Sig=210	,10 Ref=360,	100		
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %	
1 6.559 BB 0.1458 2 7.225 BB 0.1634	44.17721 5276.64453	4.61094 498.90460	0.8303 99.1697	
Totals :	5320.82174	503,51554		
Results obtained with en	hanced integ	grator!		
Signal 4: DAD1 D, Sig=230	,10 Ref=360,	100		
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %	
1 6.558 BB 0.1430 2 7.225 PB 0.1635	44.21828 5309.22021	4.64623 501.35626	0.8260	
Totals :	5353.43849	506.00250		
Results obtained with en	hanced integ	grator!		
Signal 5: DAD1 E, Sig=280	,10 Ref=360,	100		
Peak RetTime Type Width # [min] [min]	Area [mAU*s]	Height [mAU]	Area %	
1 6.560 BB 0.1407 2 7.225 BB 0.1645	24.88921 3047.15698	2.67128 285.48328	0.8102	
Totals :	3072.04619	288.15456		
Results obtained with en	hanced integ	grator!		
	*** End of	Report ***		

(-)-Aspewentin B (1)