

Enantioselective Construction of Acyclic Quaternary Carbon Stereocenters: Palladium-Catalyzed Decarboxylative Allylic Alkylation of Fully Substituted Amide Enolates

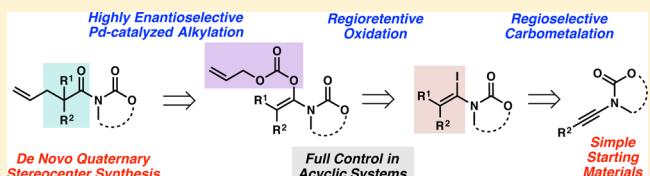
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S Supporting Information

ABSTRACT: We report a divergent and modular protocol for the preparation of acyclic molecular frameworks containing newly created quaternary carbon stereocenters. Central to this approach is a sequence composed of a (1) regioselective and -retentive preparation of allyloxycarbonyl-trapped fully substituted stereodefined amide enolates and of a (2) enantioselective palladium-catalyzed decarboxylative allylic alkylation reaction using a novel bisphosphine ligand.



1. INTRODUCTION

The quaternary carbon stereocenter is a structure that represents a minimalistic molecular framework with enhanced (bio)chemical stability and embedded propensity to encode directionality in the three-dimensional space. Over the past decade, several efficient approaches addressing *de novo* construction of cyclic quaternary carbon stereocenters have been developed;¹ however, only a select few provide general access to such molecular structures in an acyclic setting.² The prevailing majority of these methods heavily rely on the use of stereodefined trisubstituted alkenes as substrates for enantioselective allylic substitution,³ conjugate addition,⁴ or nucleophilic allylation reactions.⁵

An alternative strategy, with potentially a broader synthetic application, is enantioselective electrophilic functionalization of fully substituted acyclic enolates. In this context, regioselective formation of functionalized and fully substituted enolates is the center of renewed interest,⁶ with several attractive strategies having been reported for the preparation of β,β -disubstituted enolates of esters⁷ and amides⁸ (albeit, largely using enantiomerically enriched α,α -disubstituted carbonyl derivatives) (Scheme 1A).

The Marek group has previously reported a robust stereoselective approach to the formation of polysubstituted stereodefined acyclic enolates using a highly regioselective carbometalation of α -heterosubstituted alkynes followed by a regioretentive *in situ* oxidation reaction.⁹ This approach proved to be successful in preparation of new compounds bearing quaternary carbon stereocenters using aldol, Mannich, and allylic alkylation reactions, achieving good to excellent overall yields and exceptional diastereoselectivities (Scheme 1B). Additionally, a highly diastereoselective aldol reaction employ-

ing aliphatic aldehydes was possible using stereodefined fully substituted silyl ketene amines and titanium-mediated Mukaiyama aldol-type reaction conditions.^{9d} Recent studies demonstrated efficient preparation of stereochemically defined acyclic fully substituted ketone enolates from simple vinyl carbamates in a single-pot operation (Scheme 1C).¹⁰

As most of the above-mentioned strategies relied on the use of chiral auxiliaries, the remaining challenge was development of catalytic enantioselective transformations^{11,12} to establish acyclic quaternary carbon centers via electrophilic C-functionalization of fully substituted amide enolates. The Stoltz group has extensively contributed to the design of highly efficient catalysts for highly enantioselective allylic alkylation-type reactions,¹³ especially those involving fully substituted cyclic allyl enol carbonates (Scheme 1D)¹⁴ en route to the synthesis of complex natural products.¹⁵ We envisaged developing a unified carbometalation–enantioselective catalysis approach to access *de novo* quaternary carbon stereocenters. Our approach would involve implementing achiral acyclic amide enolates in combination with enantioselective palladium-catalyzed decarboxylative allylic alkylation technology (Scheme 1E).

Of special interest are the α -quaternary amide and imide derivatives that this outlined strategy would produce. Activated amides are among the most widely utilized carboxylic acid derivatives and have seen historical synthetic use¹⁶ as well as more recent applications in catalytic transformations.¹⁷ Additionally, we surmised that having the ability to tune the electronic and steric nature of the amido functionality would be critical for the success of the asymmetric catalysis. Thus, based

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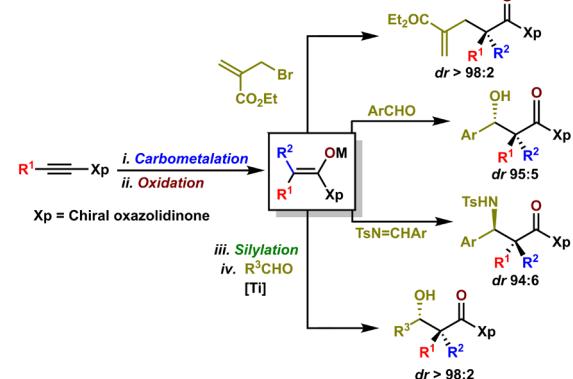


Scheme 1. Preparation of Polysubstituted Enolates and Construction of Quaternary Stereocenters

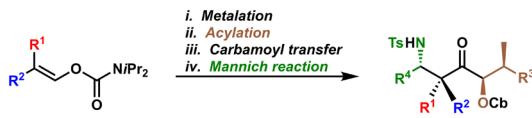
A. β,β' -disubstituted enolate of ester and amide



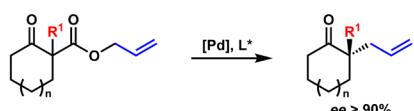
B. Combined carbometalation – oxidation then aldol, Mannich, or alkylation processes



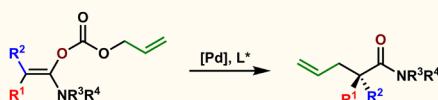
C. Combined metalation – acylation / carbamoyl transfer / Mannich reaction



D. Allylic alkylation reaction of fully substituted cyclic allyl enol carbonates



E. Palladium-catalyzed enantioselective decarboxylative allylic alkylation of stereodefined acyclic enolates (this research)

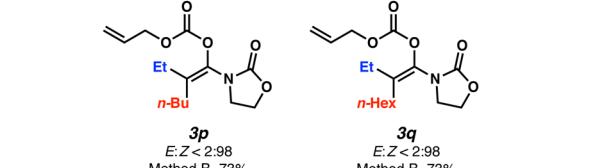
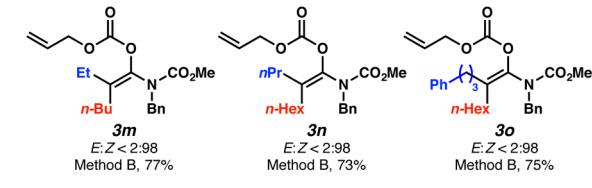
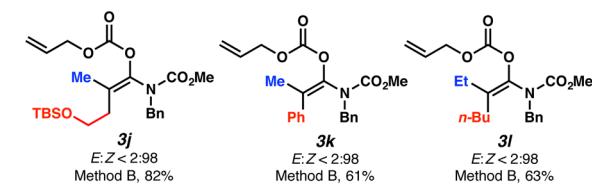
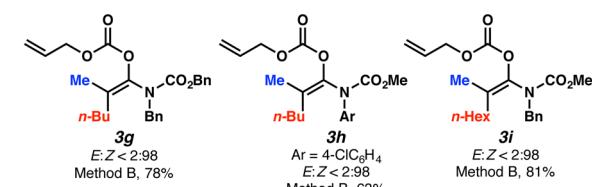
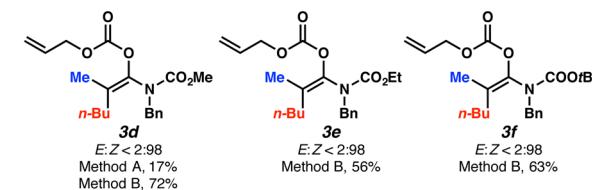
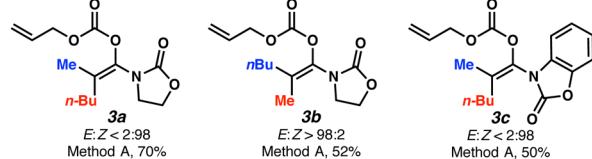
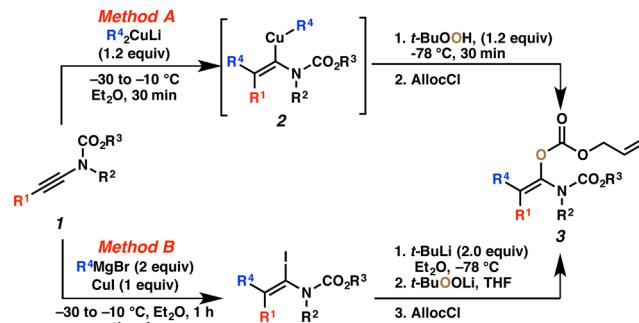


on our past experience with catalytic enantioselective allylic alkylations of lactam-derived enolates, we intentionally focused our studies on amido-type enolate chemistry.^{14b,d} This supposition indeed proved to be true, as will be outlined below.

2. RESULTS AND DISCUSSION

2.1. Stereodefined Acyclic Enol Carbonates. Our first target was an efficient approach to stereodefined acyclic allyloxycarbonyl-protected polysubstituted amide enolates. Given the multiple potential challenges and pitfalls of our strategy, we purposely limited the scope of the study to substrates with the highest probability of success. For the formation of stereodefined acyclic allyloxycarbonyl-protected polysubstituted amide enolates, we envisioned that heteroatoms in proximity to the sites of reactivity could interfere with the regio- and stereoselectivity of the multistep organometallic process, and thus generally limited our exploration to alkyl and aryl substituted systems. To this end, we have developed two complementary methods. The initial method (Method A, Scheme 2) consists of a regioselective carbometalation reaction of ynecarbamates 1 using one equivalent of the Gilman reagent

Scheme 2. Preparation of Stereodefined Acyclic Allyloxycarbonyl Polysubstituted Amide Enolates 3



“Conditions for each method are as follow. Method A: ynecarbamate (1.0 equiv), CuBr·DMS (1.0–1.2 equiv), R^4Li in hexanes or Et_2O (2.0–2.4 equiv), in Et_2O (ca. 0.1–0.5 M) at -30 to -10 °C for 0.5 h, then ca. 5.5 M $t\text{-BuOOH}$ in decane (1.0–1.2 equiv) at -80 °C for 0.5 h, then AllocCl (2.5–4.0 equiv) at -80 to 23 °C for 1 h, then 1 M HCl. Method B: step 1, ynecarbamate (1.0 equiv), Cul (1.0 equiv), R^4MgBr in Et_2O (2.0 equiv), in Et_2O (ca. 0.1–0.5 M) at -30 to -10 °C for 1 h, then I_2 (2.0 equiv) in THF at -20 to 0 °C for 15 min, then sat. $Na_2S_2O_3$; step 2, vinyl iodide (1 equiv), $t\text{-BuLi}$ (2.0–2.1 equiv), in Et_2O

Scheme 2. continued

(0.1–0.3 M) at -80°C for 15 min, then *in situ* prepared *t*BuOO^{Li} [from ca. 5.5 M *t*BuOOH in decane (1.8–2.1 equiv) and MeLi in Et₂O (2.0–2.3 equiv), in THF at -80°C for 0.5 h] at -80 to -40°C for 1 h, then AllocCl (4.0–5.0 equiv) at -50 to 23°C for 1 h, then 1 M HCl.

(R⁴₂CuLi) to result in the formation of nonsymmetric vinyl alkyl cuprate species 2.¹⁸ Upon treatment with a single equivalent of *tert*-butyl hydroperoxide (*t*-BuOOH), a selective protonation of the residual R⁴ moiety of 2 occurs to give a reactive heterocuprate intermediate.⁹ It subsequently undergoes an intramolecular S_N2 reaction (1,2-metalate rearrangement) to provide the desired stereodefined copper(I) enolate. The final electrophilic trapping by allyl chloroformate (AllocCl) gives allyloxycarbonyl-protected polysubstituted amide enolate 3. Method A is effective for substrates bearing cyclic carbamates such as 2-oxazolidinone and 2-benzoxazolinone (3a–3c) but is much less efficient for acyclic carbamates. In the latter case of acyclic yne-carbamate-derived products (e.g., 3d), the regioselectivity of the carbometalation reaction is poor and therefore, the overall efficiency of preparation of our desired allyloxycarbonyl-protected polysubstituted amide enolates 3 was less than optimal. As an alternative, a copper-promoted carbomagnesiation reaction using Normant-type reagent (RMgBr/CuI in a 2:1 ratio)¹⁸ proved to be highly regioselective in diethyl ether as solvent. Trapping of the resulting vinyl magnesium species with molecular iodine leads to the formation of fully substituted vinyl iodide 4 in excellent yields as a single constitutional isomer (Method B, Scheme 2). The vinyl iodide is then directly converted to the corresponding vinyl lithium species by iodine–lithium exchange and regioretention oxidized with an externally prepared oxenoid¹⁹ (*t*-BuOOLi). The resulting stereodefined fully substituted enolate is then reacted with AllocCl to deliver well-diversified products 3d–3q in excellent yields as single isomers as shown in Scheme 2.

2.2. Asymmetric Pd-Catalyzed Decarboxylative Allylic Alkylation. With a library of acyclic substrates 3 in hand, we evaluated the enantioselective palladium-catalyzed decarboxylative allylic alkylation (Table 1). Initially, the enantiomeric excess was determined directly by using the decarboxylative allylic alkylation reaction products 5 (on chiral SFC and HPLC); however, we opted to include an additional olefin metathesis step that results in products 6, which were more easily separable and gave a greater UV signal strength on chiral SFC. We first examined a subset of solvents using the two PHOX ligands (L₃ and L₂), known to achieve excellent enantioselectivities for the palladium-catalyzed asymmetric alkylation of cyclic ketones. Surprisingly, for acyclic substrates 3a–3d, neither of these ligands achieved satisfactory level of enantiodiscrimination (for the full list of solvents and detailed results, see Table S1 in Supporting Information). The first promising results were obtained with C₂-symmetric bisphosphines ligands L₃–L₇ in THF (Table 1, entries 1–5, and Table S2 in the Supporting Information).²⁰ Increasing the bulk around the phosphine (L₈), which has been shown to have had a positive impact on allylation of trisubstituted amide enolates with stilbene-derived diamine-linked P,P-ligands,²¹ inhibited the reaction with our fully substituted enolates 3a and 3d (Table 1, entries 6 and 20). Interestingly, when the two alkyl groups (R¹ and R⁴) on the stereodefined acyclic allyloxycarbonyl polysubstituted amide enolate were permu-

Table 1. Evaluation of Ligands for Palladium-Catalyzed DAA of Acyclic Allyloxycarbonyl Polysubstituted Amide Enolates 3^a

The reaction scheme shows the conversion of substrate 3 to product 5 or 6. Substrate 3 is an allyloxycarbonyl-protected polysubstituted amide enolate. It reacts with Pd(dba)₂ (4–10 mol %) and a ligand (L, 7–12.5 mol %) in a solvent at 23 °C for 24–72 h. An optional step involves methyl acrylate (10 equiv) and Grubbs's II (5 mol %) in CH₂Cl₂ at 40 °C for 24 h to yield product 6. The ligands L₁–L₈ are shown below, each with its chemical structure and specific substituents:

- L₁: R = H, Ar = Ph; Ar₂P = Ph₂P(OEt)₂
- L₂: R = CF₃, Ar = 4-CF₃C₆H₄
- L₃: Ph₂P(OEt)₂ linked to a cyclohexyl ring via an NH group.
- L₄: Ph₂P(OEt)₂ linked to a naphthalene ring via an NH group.
- L₅: Ph₂P(OEt)₂ linked to a stilbene-like structure via an NH group.
- L₆: R = H, Ar = Ph
- L₇: R = CF₃, Ar = 4-CF₃C₆H₄
- L₈: R = H, Ar = 2-CH₃C₆H₄

entry	substrate	ligand	solvent	ee
1	3a	L ₃	THF	64
2	3a	L ₄	THF	76
3	3a	L ₅	THF	−70
4	3a	L ₆	THF	70
5	3a	L ₇	THF	70
6	3a	L ₈	THF	NR
7	3b	L ₅	THF	46
8	3b	L ₆	THF	−30
9	3b	L ₆	EtOAc	−36
10	3c	L ₄	THF	32
11	3c	L ₆	THF	50
12	3d	L ₃	THF	72
13	3d	L ₄	THF	74
14	3d	L ₅	THF	−60
15	3d	L ₅	EtOAc	−68
16	3d	L ₆	THF	86
17	3d	L ₆	EtOAc	84
18	3d	L ₇	THF	86
19	3d	L ₇	EtOAc	94
20	3d	L ₈	THF	NR

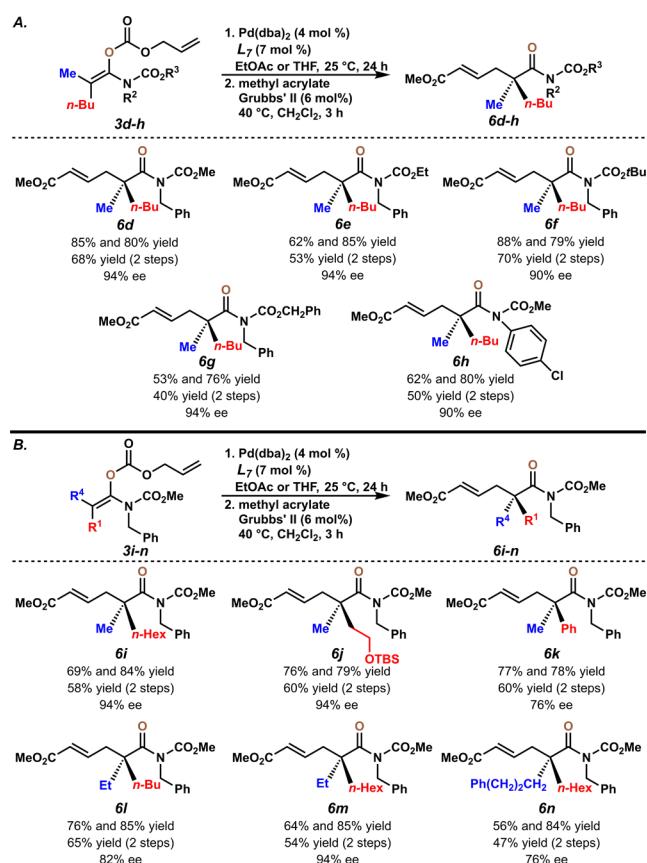
^aConditions: polysubstituted amide (1.0 equiv), Pd(dba)₂ (4–10 mol %), ligand (7–12.5 mol %) in solvent (0.17 M) at 25 °C for 24–72 h in glovebox. Filtration through silica plug, optionally followed by methyl acrylate (10 equiv), Grubbs second generation catalyst (5 mol %) in CH₂Cl₂ (0.03 M) at 40 °C for 24 h in glovebox. For details of individual experiments, see Supporting Information. NR = no reaction.

tated (cf. 3a and 3b), opposite enantiomers were obtained in lower enantiomeric excess (Table 1, cf. entries 3 and 7 vs entries 4 and 8). These observations suggest that the enolate geometry is likely conserved through the course of the reaction and that the highest enantioselectivities are obtained when the smallest substituent is *syn* to the allyloxycarbonyl group. Slightly better enantiodiscrimination was observed when EtOAc was used as solvent instead of THF (Table 1, entries 8 and 9). Benzoxazolidinone 3c²¹ performed unsatisfactorily when compared to the parent oxazolidinone 3a (Table 1, cf. entries 2 and 10 and entries 4 and 11). Far better enantioselectivities

were finally achieved with acyclic carbamate **3d** using C_2 -symmetric bisphosphine ligands **L₃–L₇** (Table 1, entries 12–18) to reach an excellent enantiomeric excess of 94% with the novel electron-deficient ligand **L₇** and EtOAc as solvent (Table 1, entry 19). Using these optimized conditions, we decreased the catalyst loading to 4 mol% of Pd(db₂) and ligand **L₇** loading to 7 mol% in EtOAc with no loss in enantiodiscrimination.

With an optimal asymmetric reaction protocol in hand, we initiated an investigation into the scope of the enantioselective alkylation. The nature of the *N*-substituents R^2 and R^3 of the acyclic carbamate substrate was probed by measuring the enantioselectivity of the reaction as shown in Scheme 3A.

Scheme 3. Access to All-Carbon Quaternary Stereocenters in Unbiased Acyclic Systems with Alloc-Trapped Enolates^a



^aConditions: polysubstituted amide (1.0 equiv), Pd(db₂)₄ (4 mol%), ligand (7 mol%) in solvent (0.17 M) at 20 °C for 24 h in glovebox. Purification (preparative TLC) followed by methyl acrylate (10 equiv), Grubbs second generation catalyst (6 mol%) in CH₂Cl₂ (0.03 M) at 40 °C for 3 h in glovebox. Yields are given for the individual steps along with the overall two-step yield.

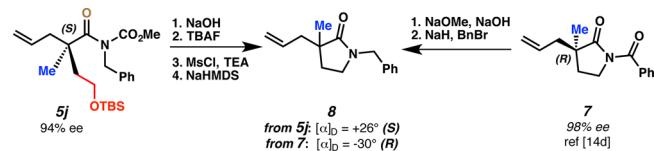
Gratifyingly, the scope of the reaction process appears to be fairly broad with respect to the R^3 substituent on the carbamate group (COOR³, where R³ = Me, Et, *t*-Bu, and Bn, **3d–3g**, respectively) and produced reaction products **6d–6g** of uniformly high enantiomeric excesses. Additionally, changing the R^2 substituent from *N*-benzyl to *N*-4-chlorophenyl produced α -quaternary amide **6h** in 90% ee.

We then turned our attention to altering the substituents at the α -carbon (Scheme 3B, i.e., R¹ and R⁴). Changing the R¹ substituent from a butyl group (**6d**, 94% ee) to a hexyl or

CH₂CH₂OTBS group did not alter the enantioselectivity of the acyclic quaternary carbon center (94% ee for **6i** and **6j**, Scheme 3B). However, when the R¹ substituent is an aromatic group, the enantiodiscrimination is moderately lower (76% ee for **6k**). To a certain extent, replacing the R⁴ substituent with an ethyl group was well tolerated (82% and 94% ee for **6l** and **6m**, respectively), while introduction of a bulkier (CH₂)₃Ph group resulted in diminished enantiomeric excess (76% ee for **6n**). In general, there is a reasonable amount of functional group tolerance as well as structural flexibility at every variable position in the new combined method. In certain cases the chemical yields of the Pd-catalyzed allylic alkylation are somewhat modest. This is likely due to a combination of steric congestion and the opportunity for palladium-enolates to proceed through multiple catalytic pathways (inner sphere versus outer sphere alkylation chemistry),²² some of which do not productively lead to the desired product (e.g., enolate protonation,²³ β -hydride elimination,²⁴ etc.).

To determine the absolute configuration of the newly formed quaternary carbon stereocenters in acyclic compounds **5** and **6** (Scheme 4), we compared the optical rotation of lactam **8**, an

Scheme 4. Determination of Absolute Stereochemistry



easily accessible cyclic product resulting from simple manipulations of **5j** with a previously characterized derivative obtained by enantioselective decarboxylative allylic alkylation of cyclic allyl enol carbonate **7** (see Supporting Information for further details).^{14d} As a result of this chemical structural correlation, we determined that the absolute stereochemistry of acyclic derivative **5j** is *S*. All other acyclic amide products resulting from our new asymmetric alkylation reaction are assigned in Scheme 3 by analogy.

3. CONCLUSION

In conclusion, we have developed a robust method for the preparation of versatile molecular backbones containing a newly created quaternary carbon stereocenter in unbiased acyclic systems. This was accomplished by the unique combination of easily accessible, fully substituted stereodefined amide enolates with the enantioselective catalytic decarboxylative allylic alkylation reaction employing a novel electronically perturbed C_2 symmetrical bisphosphine ligand. Finally, while the full synthetic utility of the enantioenriched α -quaternary amides prepared has yet to be realized, one can glean the implications of this chemistry from both past literature applications of these subunits as well as the simple sequence employed to determine absolute stereochemistry, resulting in the preparation of lactam **8** (Scheme 4). Efforts to exploit this new technology on the context of multistep synthetic chemistry will be reported in due course.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.7b04086.

Experimental procedures and characterization data
([PDF](#))

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) For recent reviews, see: (a) Quasdorf, K. W.; Overman, L. E. *Nature* **2014**, *516*, 181. (b) Marek, I.; Minko, Y.; Pasco, M.; Mejuch, T.; Gilboa, N.; Chechik, H.; Das, J. P. *J. Am. Chem. Soc.* **2014**, *136*, 2682. (c) Das, J. P.; Marek, I. *Chem. Commun.* **2011**, *47*, 4593. (d) Peterson, E. A.; Overman, L. E. *Proc. Natl. Acad. Sci. U. S. A.* **2004**, *101*, 11943. (e) Hong, A. Y.; Stoltz, B. M. *Eur. J. Org. Chem.* **2013**, *2013*, 2745. (f) Hawner, C.; Alexakis, A. *Chem. Commun.* **2010**, *46*, 7295. (g) Bella, M.; Gasperi, T. *Synthesis* **2009**, *2009*, 1583. (h) Cozzi, P. G.; Hilgraf, R.; Zimmermann, N. *Eur. J. Org. Chem.* **2007**, *2007*, 5969. (i) Trost, B. M.; Jiang, C. *Synthesis* **2006**, 369. (j) Christoffers, J.; Baro, A. *Adv. Synth. Catal.* **2005**, *347*, 1473. (k) Denissova, I.; Barriault, L. *Tetrahedron* **2003**, *59*, 10105. (l) Christoffers, J.; Mann, A. *Angew. Chem., Int. Ed.* **2001**, *40*, 4591. (m) Corey, E. J.; Guzman-Perez, A. *Angew. Chem., Int. Ed.* **1998**, *37*, 388. (n) Eppe, G.; Didier, D.; Marek, I. *Chem. Rev.* **2015**, *115*, 9175.
- (2) (a) Singh, S.; Bruffaerts, J.; Vasseur, A.; Marek, I. *Nat. Commun.* **2017**, *8*, 14200. (b) Pupo, G.; Properzi, R.; List, B. *Angew. Chem., Int. Ed.* **2016**, *55*, 6099. (c) Roy, S. R.; Didier, D.; Kleiner, A.; Marek, I. *Chem. Sci.* **2016**, *7*, 5989. (d) Zhang, F.-G.; Eppe, G.; Marek, I. *Angew. Chem., Int. Ed.* **2016**, *55*, 714. (e) Trost, B. M.; Doncke, E. J.; Thaisrivongs, D. A.; Osipov, M.; Masters, J. T. *J. Am. Chem. Soc.* **2015**, *137*, 2776. (f) Lin, H.-C.; Wang, P.-S.; Tao, Z.-L.; Chen, Y.-G.; Han, Z.-Y.; Gong, L.-Z. *J. Am. Chem. Soc.* **2016**, *138*, 14354. (g) Delaye, P.-O.; Didier, D.; Marek, I. *Angew. Chem., Int. Ed.* **2013**, *52*, 5333.
- (3) (a) Shi, Y.; Jung, B.; Torker, S.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2015**, *137*, 8948. (b) Jung, B.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2012**, *134*, 1490. (c) Feng, J.; Garza, V. J.; Krische, M. J. *J. Am. Chem. Soc.* **2014**, *136*, 8911. (d) Fañanás-Mastral, M.; Pérez, M.; Bos, P. H.; Rudolph, A.; Harutyunyan, S. R.; Feringa, B. L. *Angew. Chem., Int. Ed.* **2012**, *51*, 1922. (e) Liu, W.-B.; Reeves, C. M.; Stoltz, B. M. *J. Am. Chem. Soc.* **2013**, *135*, 17298. (f) Trost, B. M.; Xu, J.; Schmidt, T. J. *J. Am. Chem. Soc.* **2009**, *131*, 18343. (g) Zhang, P.; Le, H.; Kyne, R. E.; Morken, J. P. *J. Am. Chem. Soc.* **2011**, *133*, 9716. (h) Potter, B.; Edelstein, E. K.; Morken, J. P. *Org. Lett.* **2016**, *18*, 3286. (i) Ardolino, M. J.; Morken, J. P. *J. Am. Chem. Soc.* **2014**, *136*, 7092. (j) McGrath, K. P.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2014**, *53*, 1910. (b) Dabrowski, J. A.; Villaume, M. T.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2013**, *52*, 8156. (c) Hawner, C.; Müller, D.; Gremaud, L.; Felouat, A.; Woodward, S.; Alexakis, A. *Angew. Chem., Int. Ed.* **2010**, *49*, 7769. (d) Hawner, C.; Li, K.; Cirriez, A.; Alexakis, A. *Angew. Chem., Int. Ed.* **2008**, *47*, 8211. (e) Kikushima, K.; Holder, J. C.; Gatti, M.; Stoltz, B. M. *J. Am. Chem. Soc.* **2011**, *133*, 6902. (f) Gao, Z.; Fletcher, S. P. *Chem. Sci.* **2017**, *8*, 641. (g) Sidera, M.; Roth, P. M. C.; Maksymowicz, R. M.; Fletcher, S. P. *Angew. Chem., Int. Ed.* **2013**, *52*, 7995. (h) Vabre, R.; Island, B.; Diehl, C. J.; Schreiner, P. R.; Marek, I. *Angew. Chem., Int. Ed.* **2015**, *54*, 9996. (b) Turnbull, B. W. H.; Evans, P. A. *J. Am. Chem. Soc.* **2015**, *137*, 6156. (c) Das, J. P.; Chechik, H.; Marek, I. *Nat. Chem.* **2009**, *1*, 128. (d) Alam, R.; Vollgraff, T.; Eriksson, L.; Szabó, K. *J. Am. Chem. Soc.* **2015**, *137*, 11262. (e) Alam, R.; Diner, C.; Jonker, S.; Eriksson, L.; Szabó, K. *J. Angew. Chem., Int. Ed.* **2016**, *55*, 14417. (f) Minko, Y.; Marek, I. *Chem. Commun.* **2014**, *50*, 12597. (g) (a) Qin, Y.-C.; Stivala, C. E.; Zakarian, A. *Angew. Chem., Int. Ed.* **2007**, *46*, 7466. (b) Stivala, C. E.; Zakarian, A. *J. Am. Chem. Soc.* **2008**, *130*, 3774. (c) Gu, Z.; Herrmann, A. T.; Stivala, C. E.; Zakarian, A. *Synlett* **2010**, *2010*, 1717. (d) Kummer, D. A.; Chain, W. J.; Morales, M. R.; Myers, A. G.; Quiroga, O. *J. Am. Chem. Soc.* **2008**, *130*, 13231. (b) Allais, C.; Tsai, A. S.; Nuhant, P.; Roush, W. R. *Angew. Chem., Int. Ed.* **2013**, *52*, 12888. (c) Bixa, T.; Hunter, R.; Andrijevic, A.; Petersen, W.; Su, H.; Dhoro, F. *J. Org. Chem.* **2015**, *80*, 762. (d) Myers, A. G.; Yang, B. H.; Chen, H.; McKinstry, L.; Kopecky, D. J.; Gleason, J. L. *J. Am. Chem. Soc.* **1997**, *119*, 6496. (e) Myers, A. G.; Gleason, J. L.; Yoon, T.; Kung, D. W. *J. Am. Chem. Soc.* **1997**, *119*, 656. (f) Myers, A. G.; Yang, B. H.; Chen, H.; Gleason, J. L. *J. Am. Chem. Soc.* **1994**, *116*, 9361. (g) Mellem, K. T.; Myers, A. G. *Org. Lett.* **2013**, *15*, 5594. (h) Morales, M. R.; Mellem, K. T.; Myers, A. G. *Angew. Chem., Int. Ed.* **2012**, *51*, 4568. (i) Medley, J. W.; Movassaghi, M. *Angew. Chem., Int. Ed.* **2012**, *51*, 4572. (j) Manthorpe, J. M.; Gleason, J. L. *Angew. Chem., Int. Ed.* **2002**, *41*, 2338. (k) Manthorpe, J. M.; Gleason, J. L. *J. Am. Chem. Soc.* **2001**, *123*, 2091. (l) Burke, E. D.; Gleason, J. L. *Org. Lett.* **2004**, *6*, 405. (m) Arpin, A.; Manthorpe, J. M.; Gleason, J. L. *Org. Lett.* **2006**, *8*, 1359. (n) Tiong, E. A.; Gleason, J. L. *Org. Lett.* **2009**, *11*, 1725. (o) For reports of generation of acyclic amide enolate using regioselective oxidation, see: (a) Minko, Y.; Pasco, M.; Lercher, L.; Botoshansky, M.; Marek, I. *Nature* **2012**, *490*, 522. (b) Minko, Y.; Pasco, M.; Lercher, L.; Marek, I. *Nat. Protoc.* **2013**, *8*, 749. (c) Nairoukh, Z.; Kumar, G. K. S. N.; Minko, Y.; Marek, I. *Chem. Sci.* **2017**, *8*, 627. (d) Nairoukh, Z.; Marek, I. *Angew. Chem., Int. Ed.* **2015**, *54*, 14393. (e) Haimov, E.; Nairoukh, Z.; Shterenberg, A.; Berkovitz, T.; Jamison, T. F.; Marek, I. *Angew. Chem., Int. Ed.* **2016**, *55*, 5517. (f) For enamine-based systems, see: (a) Mukherjee, S.; List, B. *J. Am. Chem. Soc.* **2007**, *129*, 11336. (b) Krautwald, S.; Sarlah, D.; Schafroth, M. A.; Carreira, E. M. *Science* **2013**, *340*, 1065. (c) Mo, X.; Hall, D. G. *J. Am. Chem. Soc.* **2016**, *138*, 10762. (d) Ballesteros, A.; Morán-Poladura, P.; González, J. M. *Chem. Commun.* **2016**, *52*, 2905. (e) For other examples, see: (a) Lee, A.-L.; Malcolmson, S. J.; Puglisi, A.; Schrock, R. R.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2006**, *128*, 5153. (b) Mei, T.-S.; Patel, H. H.; Sigman, M. S. *Nature* **2014**, *508*, 340. (c) Patel, H.; Sigman, M. S. *J. Am. Chem. Soc.* **2016**, *138*, 14226. (d) Zhou, G.; Cobb, K. M.; Tan, T.; Watson, M. P. *J. Am. Chem. Soc.* **2016**, *138*, 12057. (e) Denmark, S. E.; Wilson, T. W.; Burk, M. T. *Chem. - Eur. J.* **2014**, *20*, 9268. (f) Kuwano, R.; Miyazaki, H.; Ito, Y. *J. Organomet. Chem.* **2000**, *603*, 18. (g) Bahlinger, A.; Fritz, S. P.; Wennemers, H. *Angew. Chem., Int. Ed.* **2014**, *53*, 8779. (h) Llaveria, J.; Leonori, D.; Aggarwal, V. K. *J. Am. Chem. Soc.* **2015**, *137*, 10958.

- (i) Shintani, R.; Duan, W.-L.; Hayashi, T. *J. Am. Chem. Soc.* **2006**, *128*, 5628. (k) Nawrat, C. C.; Jamison, C. R.; Slutskyy, Y.; MacMillan, D. W. C.; Overman, L. E. *J. Am. Chem. Soc.* **2015**, *137*, 11270. (l) Plaza, M.; Valdés, C. *J. Am. Chem. Soc.* **2016**, *138*, 12061.
- (13) For recent reviews, see: (a) Weaver, J. D.; Recio, A., III; Grenning, A. J.; Tunje, J. A. *Chem. Rev.* **2011**, *111*, 1846. (b) Mohr, J. T.; Stoltz, B. M. *Chem. - Asian J.* **2007**, *2*, 1476. (c) Trost, B. M. *Tetrahedron* **2015**, *71*, 5708.
- (14) (a) Craig, R. A., II; Loskot, S. A.; Mohr, J. T.; Behenna, D. C.; Harned, A. M.; Stoltz, B. M. *Org. Lett.* **2015**, *17*, 5160. (b) Numajiri, Y.; Jiménez-Osés, G.; Wang, B.; Houk, K. N.; Stoltz, B. M. *Org. Lett.* **2015**, *17*, 1082. (c) Reeves, C. M.; Eidamshaus, C.; Kim, J.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2013**, *52*, 6718. (d) Behenna, D. C.; Liu, Y.; Yurino, T.; Kim, J.; White, D. E.; Virgil, S. C.; Stoltz, B. M. *Nat. Chem.* **2012**, *4*, 130.
- (15) (a) Liu, Y.; Han, S.-J.; Liu, W.-B.; Stoltz, B. M. *Acc. Chem. Res.* **2015**, *48*, 740. (b) Liu, Y.; Virgil, S. C.; Grubbs, R. H.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2015**, *54*, 11800. (c) Numajiri, Y.; Pritchett, B. P.; Chiyoda, K.; Stoltz, B. M. *J. Am. Chem. Soc.* **2015**, *137*, 1040. (d) Ling, T.; Rivas, F. *Tetrahedron* **2016**, *72*, 6729.
- (16) (a) *The Amide Linkage: Structural Significance: Chemistry, Biochemistry and Materials Science*; Greenberg, A., Breneman, C. M., Liebman, J. F., Eds.; Wiley-Interscience: New York, 2003. (b) Nahm, S.; Weinreb, S. M. *Tetrahedron Lett.* **1981**, *22*, 3815. (c) Snieckus, V. *Chem. Rev.* **1990**, *90*, 879.
- (17) (a) Topczewski, J. J.; Cabrera, P. J.; Saper, N. I.; Sanford, M. S. *Nature* **2016**, *531*, 220. (b) Wu, Q.-F.; Shen, P.-X.; He, J.; Wang, X.-B.; Zhang, F.; Shao, Q.; Zhu, R.-Y.; Mapelli, C.; Qiao, J. X.; Poss, M. A.; Yu, J.-Q. *Science* **2017**, *355*, 499. (c) Hie, L.; Fine Nathel, N. F.; Shah, T. K.; Baker, E. L.; Hong, X.; Yang, Y.-F.; Liu, P.; Houk, K. N.; Garg, N. K. *Nature* **2015**, *524*, 79. (d) Hie, L.; Baker, E. L.; Anthony, S. M.; Desrosiers, J.-N.; Senanayake, C.; Garg, N. K. *Angew. Chem., Int. Ed.* **2016**, *55*, 15129. (e) Shang, R.; Ilies, L.; Nakamura, E. *J. Am. Chem. Soc.* **2015**, *137*, 7660. (f) Roane, J.; Daugulis, O. *Org. Lett.* **2013**, *15*, 5842. (g) Nishino, M.; Hirano, K.; Satoh, T.; Miura, M. *Angew. Chem., Int. Ed.* **2013**, *52*, 4457. (h) Yokota, A.; Aihara, Y.; Chatani, N. *J. Org. Chem.* **2014**, *79*, 11922. (i) Fruchey, E. R.; Monks, B. M.; Cook, S. P. *J. Am. Chem. Soc.* **2014**, *136*, 13130.
- (18) (a) Chechik-Lankin, H.; Livshin, S.; Marek, I. *Synlett* **2005**, 2098. (b) Minko, Y.; Pasco, M.; Chechik, H.; Marek, I. *Beilstein J. Org. Chem.* **2013**, *9*, 526.
- (19) (a) Panek, E. J.; Kaiser, L. R.; Whitesides, G. M. *J. Am. Chem. Soc.* **1977**, *99*, 3708. For reviews on oxenoid/nitrenoid chemistry, see: (b) Minko, Y.; Marek, I. *Org. Biomol. Chem.* **2014**, *12*, 1535. (c) Boche, G.; Lohrenz, J. C. W. *Chem. Rev.* **2001**, *101*, 697. (d) Starkov, P.; Jamison, T. F.; Marek, I. *Chem. - Eur. J.* **2015**, *21*, 5278.
- (20) (a) Trost, B. M.; Van Vranken, D. L. *Chem. Rev.* **1996**, *96*, 395. (b) Trost, B. M.; Van Vranken, D. L. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 228. (c) Trost, B. M.; Van Vranken, D. L.; Bingel, C. *J. Am. Chem. Soc.* **1992**, *114*, 9327.
- (21) Trost, B. M.; Michaelis, D. J.; Charpentier, J.; Xu, J. *Angew. Chem., Int. Ed.* **2012**, *51*, 204.
- (22) (a) Keith, J. A.; Behenna, D. C.; Sherden, N.; Mohr, J. T.; Ma, S.; Marinescu, S. C.; Nielsen, R. J.; Oxgaard, J.; Stoltz, B. M.; Goddard, W. A., III *J. Am. Chem. Soc.* **2012**, *134*, 19050. (b) Sherden, N. H.; Behenna, D. C.; Virgil, S. C.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2009**, *48*, 6840.
- (23) (a) Marinescu, S. C.; Nishimata, T.; Mohr, J. T.; Stoltz, B. M. *Org. Lett.* **2008**, *10*, 1039. (b) Mohr, J. T.; Nishimata, T.; Behenna, D. C.; Stoltz, B. M. *J. Am. Chem. Soc.* **2006**, *128*, 11348. (c) Kingston, C.; Guiry, P. J. *J. Org. Chem.* **2017**, *82*, 3806.
- (24) (a) Shimizu, I.; Tsuji, J. *J. Am. Chem. Soc.* **1982**, *104*, 5844. (b) Chen, Y.; Romaire, J. P.; Newhouse, T. R. *J. Am. Chem. Soc.* **2015**, *137*, 5875.