The Horeau Principle

The Nature of Statistical Amplification in Enantioselective Synthesis



The Horeau Principle: A Statistical Basis for Asymmetric Amplification

- I.) Origins and explanation of the Horeau Principle.
 - A.) Wolfgang Langenback's observations.
 - B.) Horeau's original experiments and conclusions.
 - C.) Critical assumptions, costs, and guiding concepts of Horeau's work.
- II.) The Horeau Principle via chemical duplication.
 - A.) Duplication as a method of chiral resolution or enantioenrichment.
 - B.) Applications of duplication to the elucidation of substrate ee.
 - C.) Duplications and their application to synthesis.
- III.) Extensions of the Horeau Principle beyond duplication.
 - A.) The Horeau Principle via multiple enantioselective catalysis.
 - B.) Adding complexity to the Horeau Principle.
- IV.) Conclusions.

Reviews:

Duplication and purification - Jacques, J.; Collet, A.; Wilen, S. H. *Enantiomers, Racemates and Resolution*, **1981**, p. 430-434. The Horeau Principle explained - Rautenstrauch, V. *Bull. Soc. Chim. Fr.* **1994**, *131*, 515-524. Double Asymmetric synthesis - Baba, S. E.; Sartor, K.; Poulin, J.; Kagan, H. *Bull. Soc. Chim. Fr.* **1994**, 131, 525-533. In the context of non-linear effects - Girard, C.; Kagan, H. *Angew. Chem. Int. Ed.* **1998**, *37*, 2922-2959. The historical perspective - Heller, G. *Angew. Chem. Int. Ed.* **2000**, *39*, 495-499. Mathematical treatment and assumptions - Chandrasekhar, S. **2005**, *31*, 779-783.

Langenback's Observations: The Origin of the Horeau Principle An Idea Before its Time

· Langenback's original publication in 1936 postulated:

"With every synthesis of an optically active compound from inactive starting materials, a degredation in optical purity takes place... The infinite repetition of these processes over a geological time period would have led to a complete loss of optical activity... if the degredation were not compensated for by an increase in optical purity in a different process."

• To understand and indentify this phenomenon, Langenback investigated a simple dimerization process:



Langenbeck, W. Z. Phys. Chem. A **1936**, *177*, 401-408.

Langenback's Observations: The Origin of the Horeau Principle An Idea Before its Time



Heller, G. Angew. Chem. Int. Ed. **2000**, *39*, 495-499. Langenbeck, W. Z. Phys. Chem. A **1936**, *177*, 401-408.

Horeau's Experiments: Statistical Enantiomeric Enrichment Dimerization as a Resolution Technique

• In 1973, Horeau performed experiments very similar to Langenbeck's duplications, but with the aid of separation techniques.



Vigneron, J. P.; Dhaenes, M.; Horeau, A. *Tetrahedron*, **1973**, *29*, 1055-1059. Rautenstrauch, V. *Bull. Soc. Chim. Fr.* **1994**, *131*, 515-524.

Horeau's Experiments: Statistical Enantiomeric Enrichment Math and Theory

• For any scalemic mixture composed of molecules baring a single chiral center, the respective enantiomers are represented by the letters R and S. An achiral, bifunctional duplication reagent is represented by A.

- If the scalemic mixture is enriched with the R enantiomer, the total composition of R in the mixture can be represented as x, while the total composition of S is 1-x (where R+S=1).
- Criticial Assumptions:
 - The outcome of the first reaction must have *no* influence upon the second reaction. This means *no* chiral recognition between RA or SA and either incoming monomer.
 - There must be *no* difference in rate between enantiomers during the formation of RA or SA.
 - There must be *no* interference by side reactions.
 - Reactions must be irreversible.
- Following the above assumptions, we can conclude:

$$dr = \frac{\text{Homochiral Yield}}{\text{Heterochiral Yield}} = \frac{x^2 + (1-x)^2}{2 \cdot x \cdot (1-x)}$$

$$ee_p = \frac{\text{RR} - \text{SS}}{\text{RR} + \text{SS}} = \frac{x^2 - (1-x)^2}{x^2 + (1-x)^2}$$

$$er_p = \frac{\text{RR}}{\text{SS}} = \frac{x^2}{(1-x)^2} = \left(\frac{x}{(1-x)}\right)^2 = (er_j)^2$$

Vigneron, J. P.; Dhaenes, M.; Horeau, A. *Tetrahedron*, **1973**, *29*, 1055-1059. Rautenstrauch, V. *Bull. Soc. Chim. Fr.* **1994**, *131*, 515-524.



 If we take the value *ee_i* to be the initial enantiomeric excess of the starting material, we can find alternative expressions for *ee_p* and dr:

$$ee_p = \frac{2 \cdot ee_i}{1 + (ee_i)^2}$$
 $dr = \frac{1 + (ee_i)^2}{1 - (ee_i)^2}$

Horeau's Experiments: Statistical Enantiomeric Enrichment Costs and Benefits

• The ee of the dimer increases beyond that of the monomer quickly. This is an example of positive nonlinear behavior.

$$er_{p} = \frac{RR}{SS} = \left(\frac{R}{S}\right)^{2} = (er_{i})^{2} \qquad \qquad er_{p} \ge er_{i} \qquad \qquad ee_{p} = \frac{2 \cdot ee_{i}}{1 + (ee_{i})^{2}}$$

• The de (and hence dr) of the dimer increases more slowly, with high dr only achieved at large ee_i values.



Enantiomeric Enrichment via Duplication Applications and Synthesis

• Synthesis and enantioenrichment of diisopinocamphenylborane from scalemic (α)-pinene.



Brown, H. C.; Singaram, B. *J. Org. Chem.* **1984**, *49*, 945-947. Prabhakar, K.; Brown, H. C. *J. Org. Chem.* **1985**, *50*, 3203-3206.

• Dimerization of (S)-methylbenzylamine was used in order to assay *ee* without need for a chiral auxiliary.



Grotjahn, D.; Joubran, C.; Tetrahedron: Asymmetry 1995, 6, 745-752.

- ¹H NMR showed 99:1 dr in favor of the homochiral diastereomer for (+)-methylbenzylamine.
- (±)-methylbenzylamine gave a 1:1 dr.
- The amine ee_i was determined to be 99.1%

Enantiomeric Enrichment via Duplication Applications and Synthesis

• Low selectivity carboalumination produced substrates with modest ee values. Duplication made them synthetically useful.



D'Arrigo, P.; Servi, S. J. Org. Chem. 1997, 62, 6394-6396.

Enantiomeric Enrichment via Duplication Duplication without Achiral Linkers

• The total synthesis of (–)-Wodeshiol by Corey benefits from Horeau-type *ee* amplification during the key dimerization:



Hoffmann, R.; Schopfer, U. Helv. Chim. Acta. 2002, 85, 4424-4441.

Enantiomeric Enrichment via Duplication Controversy over the Synthesis of Carpenter Bee Hormone

• **1981**: Mori and Tanida report the total synthesis of four stereoisomers of 2,8-Dimethyl-1,7-dioxaspiro[5,5]undecane.



• **1984**: Isaksson and coworkers separate the four naturally occuring stereoisomers via liquid chromatography. Their optical rotation values do not match those of Mori *et al.* exactly.

[α] ²⁴	(+)-1	(–)-1	(+)- <i>2</i>	(–)-2
Mori Values	+ 51.7°	- 51.6°	-	_
Isaksson Values	+ 44.6°	- 44.3°	+ 44.0°	- 44.6°

"We thus conclude that our isolated enantiomers are at least 98% optically pure. Mori and Tanida reported [larger] specific rotations... although their chiral starting material was of only 92% ee... The reason for this is not clear to us."

• 1986: Mori and Tanida report multiple (re)syntheses of (+)-1 and (-)-1 in response to Isaksson's paper.

"[Isaksson] challenged our higher values, because we employed [starting material] of only 92% ee... Apparently they thought that [this] should yield (+)-1 of 92% ee."

Mori, K.; Tanida. *Tetrahedron*, **1981**, *37*, 3221-3225. Isaksson, R.; Liljefors, T. *J. Chem. Soc. Chem. Commun.* **1984**, 137-138. Mori, K.; Watanabe, H. *Tetrahedron*, **1986**, *42*, 295-304.

Enantiomeric Enrichment via Duplication Controversy over the Synthesis of Carpenter Bee Hormone

• 1981, 1986: Total synthesis and reexamination of 2,8-Dimethyl-1,7-dioxaspiro[5,5]undecane by Mori.



"In conclusion... the specific rotations reported for **1** and **2** by Isaksson et al. must be in error. The reason for this is not clear to us."

Mori, K.; Tanida. *Tetrahedron*, **1981**, *37*, 3221-3225. Isaksson, R.; Liljefors, T. *J. Chem. Soc. Chem. Commun.* **1984**, 137-138. Mori, K.; Watanabe, H. *Tetrahedron*, **1986**, *42*, 295-304.

• Do Horeau-type distributions and effects play a role in the generation of two or more stereocenters from a prochiral molecule?

- For the case of a molecule **M** bearing two nonequivalent prochiral centers labeled 1 and 2, we assume total conversion of the *intermediates*:



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• Additional definitions for the diastereomeric ratio, and a term (i) to weigh the relative contributions of Routes I and II:

dr =
$$\frac{a}{b}$$
 such that $a + b = 1$
i = $\frac{[1R, P] + [1S, P]}{[1R, P] + [1S, P] + [P, 2R]' + [P, 2S]'}$

- Do Horeau-type distributions and effects play a role in the generation of two or more stereocenters from a prochiral molecule?
 - For the case of a molecule **M** bearing two nonequivalent prochiral centers labeled 1 and 2, we assume total conversion of the *intermediates*:



• Limiting cases of Kagan's calculations for multiple chiral reactions.



Assumptions:

Route I and Route II operate simultaneously and competitively, but have identical selectivities at corresponding steps.

$$\mathbf{r} = \mathbf{r}'$$
 $\mathbf{r}_{\alpha} = \mathbf{r}_{\alpha}'$ $\mathbf{r}_{\beta} = \mathbf{r}_{\beta}'$

$$ee_{A} = \frac{i \left[(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\alpha}')(1+r') - (1-r_{\beta}')(1-r') \right]}{i \left[(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\alpha}')(1+r') + (1-r_{\beta}')(1-r') \right]} \qquad ee_{A} = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r)}$$

$$ee_{B} = \frac{i \left[(1-r_{\alpha})(1+r) - (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') - (1-r_{\alpha}')(1+r') \right]}{i \left[(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') + (1-r_{\alpha}')(1+r') \right]} \qquad ee_{B} = (2i-1)\frac{(1-r_{\alpha})(1+r) - (1+r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)}$$

$$dr = \frac{i \left[(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1+r') + (1-r_{\alpha}')(1+r') \right]}{i \left[(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') + (1-r_{\alpha}')(1+r') \right]} \qquad dr = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)}$$

• Limiting cases of Kagan's calculations for multiple chiral reactions.



Assumptions:

Route I and Route II are identical, or else one of the Routes dominates the reaction to the point of always occuring first:

i = 1
$$\mathbf{r} = \mathbf{r}'$$
 $\mathbf{r}_{\alpha} = \mathbf{r}_{\alpha}'$ $\mathbf{r}_{\beta} = \mathbf{r}_{\beta}'$

$$ee_{A} = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r)} \qquad ee_{A} = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r)}$$

$$ee_{B} = (2i - 1)\frac{(1-r_{\alpha})(1+r) - (1+r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)} \qquad ee_{B} = \frac{(1-r_{\alpha})(1+r) - (1+r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)} \qquad dr = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)}$$

• Limiting cases of Kagan's calculations for multiple chiral reactions.



Assumptions:

All stereoselectivities are identical, regardless of order of transformation or other existing stereocenters:

i = 1
$$r = r' = r_{\alpha} = r_{\alpha}' = r_{\beta} = r_{\beta}'$$



• Summary:

• In the case of a double enantioselective reaction, the *ee* of the major (A) and minor (B) diastereomers and the dr are:

$$ee_{A} = \frac{i \left[(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\alpha}')(1+r') - (1-r_{\beta}')(1-r') \right]}{i \left[(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\alpha}')(1+r') + (1-r_{\beta}')(1-r') \right]}$$

$$ee_{B} = \frac{i \left[(1-r_{\alpha})(1+r) - (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') - (1-r_{\alpha}')(1+r') \right]}{i \left[(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') + (1-r_{\alpha}')(1+r') \right]}$$

$$dr = \frac{i \left[(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') + (1-r_{\alpha}')(1-r') \right]}{i \left[(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r) \right] + (1-i) \left[(1+r_{\beta}')(1-r') + (1-r_{\alpha}')(1+r') \right]}$$

• When the selectivities for the reactions are independent of the initial reaction site (Same selectivities along Route I and II):

$$ee_{A} = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1+r_{\alpha})(1+r) + (1-r_{\beta})(1-r)} \qquad ee_{B} = (2i-1)\frac{(1-r_{\alpha})(1+r) - (1+r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)} \qquad dr = \frac{(1+r_{\alpha})(1+r) - (1-r_{\beta})(1-r)}{(1-r_{\alpha})(1+r) + (1+r_{\beta})(1-r)}$$

• When the Routes and their selectivities are identical - OR - When one Route is highly favored ($\mathbf{r} \neq \mathbf{r}_{\alpha} \neq \mathbf{r}_{\beta}$):

$$dr = \frac{ee_B - r}{r - ee_A}$$

• When the selectivities are equal, and independent of all else ($\mathbf{r} = \mathbf{r}_{\alpha} = \mathbf{r}_{\beta}$):

$$ee_A = \frac{2r}{1 + r^2}$$
 $ee_B = 0$ $dr = \frac{1 + r^2}{1 - r^2}$

Catalyst Control

• Amplifications originating from double enantioselective reactions upon a bis-prochrial molecule:



• For a situation in which the chiral catalsyt or reagent employed favors the formation of the *R* stereocenter.

- Intial reaction creates an excess of (1R, P), (P, 2R)' relative to (1S, P), (P, 2S)'.
- Futher reaction of (**1R**, **P**) favors (**1R**, **2R**). Any (**1R**, **2S**) produced has *no effect* on *ee_A*. (The same holds for the analogous path in Route II)
- Any reaction of (**1S**, **P**) generated in the first step favors (**1S**, **2R**), minimizing (**1S**, **2S**). (Again, also for Route II)

Rautenstrauch, V. *Bull. Soc. Chim. Fr.* **1994**, *131*, 515-524. Baba, S.; Sartor, K.; Kagan, H. *Bull. Soc. Chim. Fr.* **1994**, 525-533.

• Obtaining enantioenriched 1,2-diols via asymmetric dihydroxylation of the corresponding terminal olefins afforded lower than anticipated *ee* values in some cases.

			Ligand (DHQD) ₂ –	Oxidant	% ee	p
			-PHAL	K ₃ Fe(CN) ₆	85.7	,
		Ōн	-PHAL	l ₂	83.2	
но	AD	но	-PYR	K ₃ Fe(CN) ₆	95.6	i
-	Conditions		-PYR	l ₂	89.0)
		4	-AQN	K ₃ Fe(CN) ₆	84.0)
Alkylation Conditions			-AQN	l ₂	52.8	
V		ŌН	Ligand (DHQD) ₂ –	Oxidant	% ee _p	dr
0~4/		о су он	-PHAL	K ₃ Fe(CN) ₆	98.9	6.7
			-PHAL	l ₂	97.0	4.1
	Conditions		-PYR	K ₃ Fe(CN) ₆	99.4	9.5
(1)		$\dot{\mathbf{Y}}$	-PYR	l ₂	70.4	1.4
		ОН	-AQN	K ₃ Fe(CN) ₆	98.1	5.1
		ОН	-AQN	l ₂	97.6	4.6

Hoye, T.; Mayer, M. J. Org. Chem. 1998, 63, 8554-8557.

• Horeau-type amplification of 1,2-diols via double asymmetric dihydroxylation gave much better *ee* values for many conditions... But at a cost.



n	Ligand (DHQD) ₂ –	Oxidant	% ee _p	dr	% <i>EE_i</i> (calculated)	% Yield lost to <i>meso</i>
0	-PHAL	K ₃ Fe(CN) ₆	99.2	7.9	88.1	11.2
0	-PYR	K ₃ Fe(CN) ₆	84.8	1.9	55.6	34.6
0	-AQN	K ₃ Fe(CN) ₆	99.9	24.5	96.0	3.9
3	-PHAL	I ₂	97.0	4.1	78.0	19.6
3	-PYR	K ₃ Fe(CN) ₆	99.4	9.5	90.0	9.5
3	-PYR	I ₂	70.4	1.4	41.2	41.2
5	-PHAL	K ₃ Fe(CN) ₆	97.6	4.6	80.0	18.0
5	-PYR	I ₂	89.7	2.3	62.3	30.6

Hoye, T.; Mayer, M. J. Org. Chem. 1998, 63, 8554-8557.

• Stepwise transformations show similar mathematical behavior in the absence of substrate control.



 Knowing the "local" ee values at each of the stereocenters individually allows for Horeau predictions, which can be checked against experimental findings:

Calculated
Values
$$\begin{cases} ee_p = \frac{2 \cdot ee_i}{1 + ee_i^2} = 95.6\% & 96.7\% = ee_p \\ dr = \frac{1 + ee_i^2}{1 - ee_i^2} = 3.42 & 3.32 = dr \end{cases}$$
 Experimentally
Values

• Stepwise transformations show similar mathematical behavior in the absence of substrate control.



• Even in cases of sequential enantioselective reactions, amplification of product *ee* may result.

• Is it possible to estimate the enantioselectivity of a chiral catalyst from its racemic mixture?



• Equal opportunity to interact with either sense of the catalyst at either step leads to a racemate. In the absence of substrate influence, equal quantities of the two diastereomers are expected. . .



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· Is it possible to estimate the enantioselectivity of a chiral catalyst from its racemic mixture?



- What would occur if, after the initial reaction, the catalyst did *not* release the substrate? What if it could continue to act upon the remaining prochiral site?
- The heterochiral diastereomer would then be afforded *only* as a result of "catalyst errors", and (dr ≠ 1) would be expected.



- · Is it possible to estimate the enantioselectivity of a chiral catalyst from its racemic mixture?
- For a single catalyst of a racemic mixture interacting with both prochiral sites of a symmetric substrate:



• If the catalyst operates on both prochiral sites with equal selectivity, then $ee_1 = ee_2$ holds true.



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• Is it possible to estimate the enantioselectivity of a chiral catalyst from its racemic mixture?

- Guiding conditions and critical assumptions:
 - 1.) For any given substrate with two prochiral sites, a single catalyst molecule must perform *both* reactions (No scrambling).
 - 2.) After a single reaction, the newly formed stereocenter must have *no influence* on further selectivity (Zero substrate control).
 - 3.) The catalyst selectivity must be *identical* at both prochiral sites, regardless of the order of reaction ($ee_1 = ee_2$).



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• Both racemic and enantioenriched catalysts were employed, and *ee*₁ values were calculated from *de* data.



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Multiple Enantioselective Transformations nth Order Amplifications



Crispino, G.; Sharpless, K. B. *Tetrahedron Lett.* **1992**, *33*, 4273-4273. Crispino, G.; Ho, P. T.; Sharpless, K.B. *Science* **1993**, *259*, 64-66. Rautenstrauch, V. *Bull. Soc. Chim. Fr.* **1994**, *131*, 515-524.

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Summary and Conclusion

- Dimerization of a chrial compound, whether via an achiral linker or direct coupling, will result in a products boasting an *ee* value higher than the starting material employed. This occurs by virtue of the forming *meso* diastereomer, which allows for removal of racemic material.
- Multiple enantioselective transformations will result in an increase in product *ee* relative to the value expected for a single reaction. This occurs by virtue of the heterochiral diastereomer acting as a 'buffer' against formation of the undesired enantiomer.
- In both cases, the cost paid for an increase in *ee* is always material lost in the form of undesired diastereomers. The lower *ee_i* is, the higher this cost becomes.
- Horeau-type amplifications in *ee* are exponential in relation to the number of duplications or enantioselective transformations.
- All of the preceding cases assume zero substrate control! The equations discussed are approximations and may not be applicable to all cases. However, they serve as an excellent starting point for many systems, and have numerous extensions with interesting applications.

Dimerizations or Double Enantioselective Transformations with all r = ee_i



Double Enantioselective Transformations with

 $\mathbf{r} \neq \mathbf{r}_{\alpha} \neq \mathbf{r}_{\beta}$

