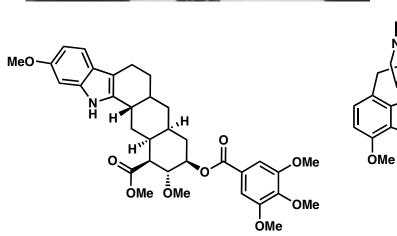


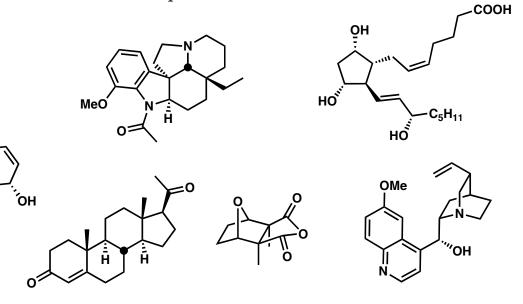
Gilbert Stork: Explorations into Position Selectivity and Stereocontrol, a Synthetic Renaissance

Jenn Stockdill

29 January 2007

A Stoltz Group Literature Presentation





#### ~Gilbert Stork~

- Born: 31 December 1921 in Brussels, Belgium
- Secondary education in France (Paris and Nice)
- Moved to the United States in 1939
- B.S. at University of Florida in 1942
- Ph.D. at University of Wisconsin in 1945
- Harvard: 1946-1948 (Instructor) and 1948-1953 (Assistant Professor)
- Columbia University: 1953-1955 (Associate Professor), 1955-1967 (Professor), 1967-1993 (Eugene Higgins Professor), 1993-present (Eugene Higgins Professor Emeritus)
- 1957 Award in Pure Chemistry, American Chemical Society
- 1959 Guggenheim Foundation Fellow
- 1961 D.Sc. (honorary), Lawrence University
- 1961 Baekeland Medal, North Jersey Section, ACS
- 1962 Harrison Howe Award
- 1966 Edward Curtis Franklin Memorial Award, Stanford University
- 1967 Award for Creative Work in Synthetic Organic Chemistry, ACS
- 1971 Gold Medal, Synthetic Organic Chemical Manuufacturers Association
- 1973 Nebraska Award
- 1978 Roussel Prize, Paris
- 1979 D.Sc. (honorary), Université Pierre et Marie Curie
- 1980 Nichols Medal, New York Section, ACS
- 1982 Edgar Fahs Smith Award, Philadelphia Section, ACS
- 1982 Willard Gibbs Medal, Chicago Section, ACS
- 1982 Award in Chemical Sciences, National Academy of Sciences



#### Winifred and Gilbert Stork

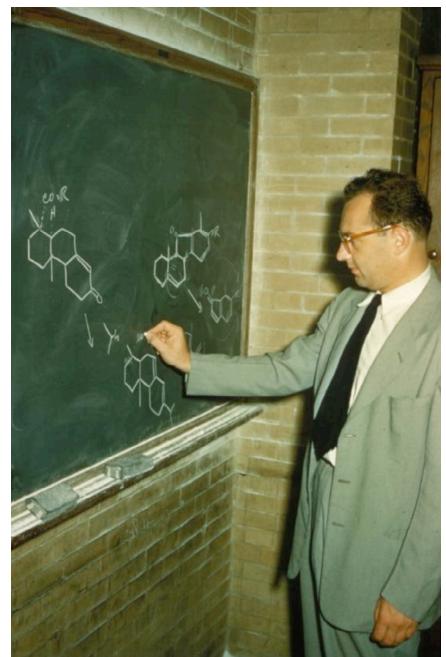
- 1982 D.Sc. (honorary), University of Rochester
- 1983 National Medal of Science
- 1983 Pauling Award
- 1985 Tetrahedron Prize
- 1986 Remsen Award, Maryland Section, ACS
- 1986 Cliff S. Hamilton Award, Nebraska
- 1987 Monie A. Ferst Award and Medal, Georgia Tech
- 1988 D.Sc. (honorary), Emory University
- 1991 Roger Adams Award
- 1992 George Kenner Award, Liverpool
- 1992 Robert Robinson Lectureship Award, U. K.
- 1992 D.Sc. (honorary), Columbia University
- 1993 Robert A. Welch Award
- 1996 Wolf Prize, Israel
- 1997 D.Sc. (honorary), University of Wisconsin

"Not many people have had a greater impact on modern organic chemistry than Gilbert Stork. His original and profound intellectual style strongly influence the way organic chemists now think about synthesis. In particular, the invention of a new reagent to solve a specific problem in total synthesis, and the inclusion of stereospecificity in a synthetic design are Stork trademarks."

From: http://www.chemistry.msu.edu/Portraits/PortraitsHH\_Detail.asp?HH\_LName=Stork

#### Outline

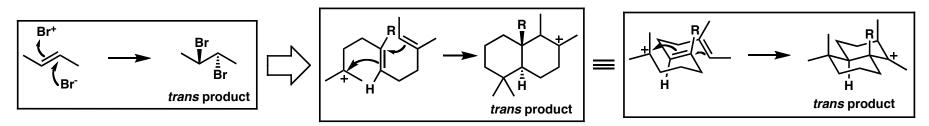
- Stereochemistry of polyene cyclizations
  - Application to cyclopropyl ketones
  - Bicyclo[2.2.1]heptanes
- Enamine alkylation & acylation of ketones and aldehydes
  - Aspidospermine
  - Byssochlamic acid
  - Yohimbine alkaloids
- ♦ Regiospecific formation and trapping of enolates ◊ via Addition of 2 e<sup>-</sup> and 1H<sup>+</sup> to an enone
  - Lupeol
  - *dl*-D-Homotestosterone and *dl*-progesterone
  - Adrenosterone
  - $\Diamond$  Conjugate addition of an organometallic reagent to an enone
    - Lycopodine
    - Prostaglandin  $F_{2\alpha}$
- Regiospecific deprotonation & trapping with saturated ketones
   β-Vetivone
- Cyano-epoxide and allylic-epoxide carbocyclizations
   Histrionicotoxin
- Cyclization of alkynyl ketones
  - Gibberellic acid
- Protected cyanohydrins as acyl carbanion equivalents
   Taxol
  - Taxol
  - Prostaglandin  $F_{2\alpha}$
- Vinyl radical cyclization
  - rac-Patchouli alcohol
  - (+)-Digitoxigenin
- Radical cyclization to form a temporary ring
   Mixed acetal linkage
  - 12a-deoxytetracycline
  - Calcitriol
  - **♦** Temporary silicon connection



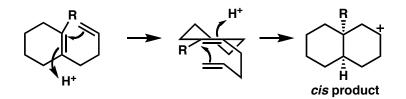
Key Review: Medicinal Research Reviews 1999, 19, 370-387.

#### Polyene Cyclizations

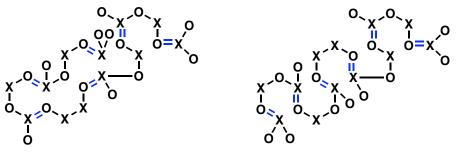
• The idea: a cationic olefic cyclization should proceed analogously to the dibromination of an olefin.



• Note: When the precursor to the cationic cyclization contains a 6-membered ring, the product will have a *cis*-fused ring junction.



- Harvard Colloquium March 14, 1950: Stork suggests a *concerted* polyene cyclization as the mechanism for steroid formation in natural triterpene steroids based on their *trans-anti-trans* ring junctions.
- JACS 1953, 75, 2023: Woodward proposes this idea as the mechanism for squalene cyclization to for cholesterol (below)
- JACS 1955, 77, 5068: Stork's Propsal of synthetic routes to Lupeol, Lanosterol, and Cholesterol
- Stork, ACS Meeting, San Diego, CA March 2005: "The sincerest form of flattery?"



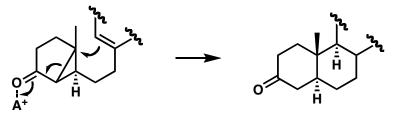
"We now wish to propose the alternative mechanism shown in Fig. 1B, as a more likely one to be involved in this transformation."

~R.B. Woodward and Konrad Bloch

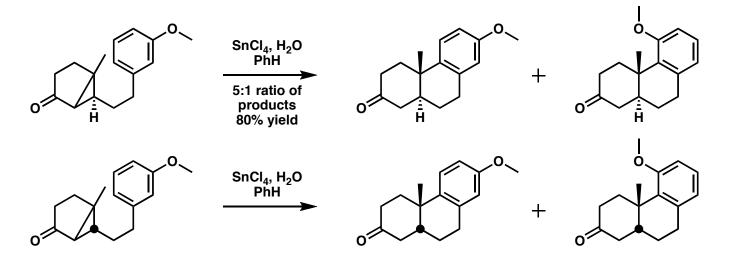
# Acyl Cyclopropanes in Cationic Cyclizations

• C(3)-keto groups in steroid systems could be synthesized directly and the ring junction formed stereoselectively. (*JACS* 1969, *91*, 2371)

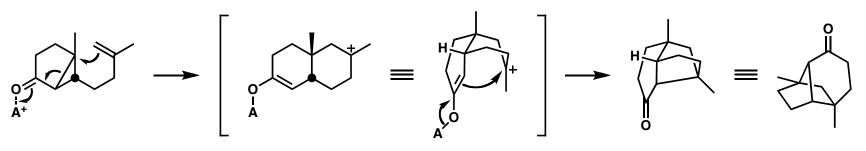
For acyl cyclopropane synthesis, see: Stork JACS 1961, 83, 4678.



• Cyclopropyl cleavage was determined to be concerted with bond formation by stereochemical studies (JACS 1969, 91, 2373)



• Synthesis of bicyclo[2.2.1]heptanes (JACS 1969, 91, 2407):

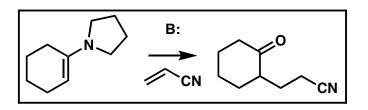


### Enamine Alkylation and Acylation

**Key Features:** 

- Allows alkylation in presence of unsubstituted  $\alpha$ -methylenes, giving acess to mono-alkylated products
- Allows alkylation with electrophilic olefins, including Michael acceptors
- Alkylation of  $\alpha$ -monosubstituted ketones gives further alkylation at the  $\alpha'$  position
- Alkyl halide electrophiles must be particularly active

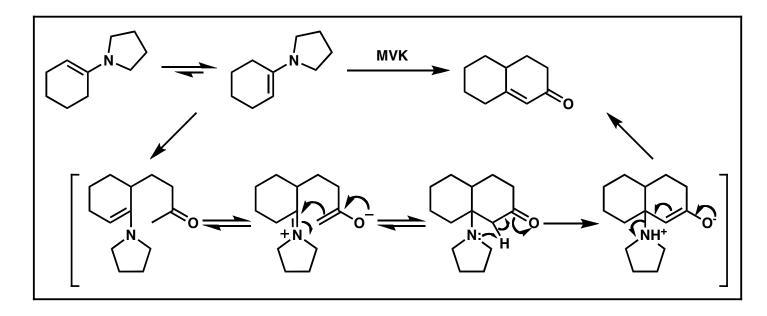
(JACS 1954, 76, 2029, JACS 1956, 78, 5128, and JACS 1963, 85, 207)



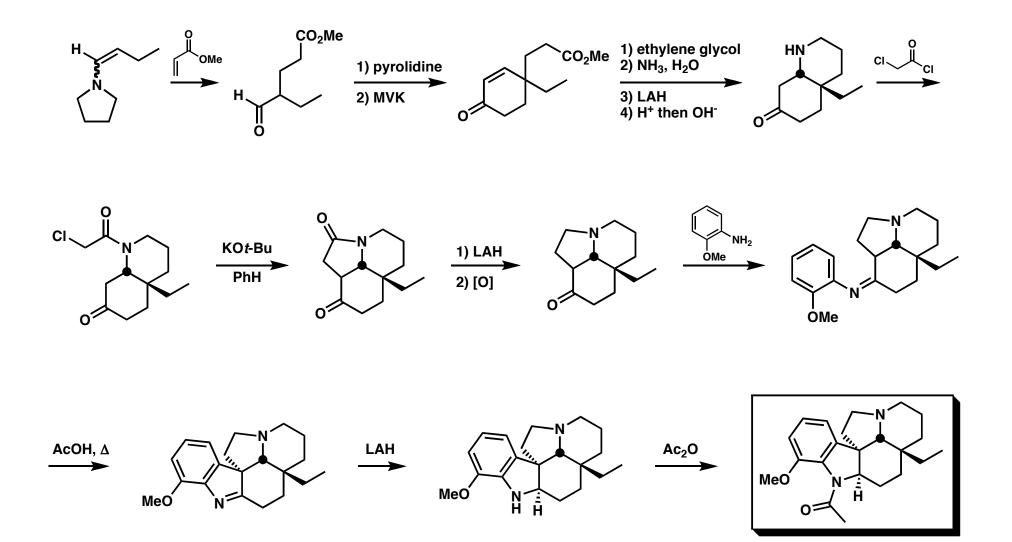
Later, the *Metalloenamine Alkylation* was developed...features:

- Works with unreactive alkyl halides
- Allows mono-alkylation
- Can use Grignards or LDA and an electrophile
- works with *N*,*N*-dimethylhydrazones as well (further developed by Corey & Enders)

(JACS 1963, 85, 2178 and JACS 1971, 93, 5938)

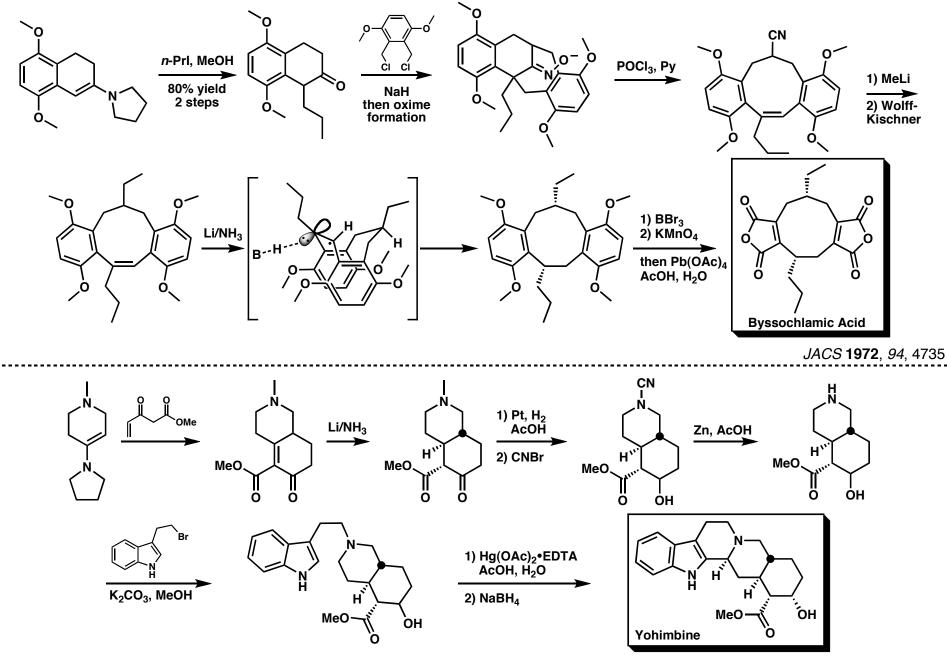


### Total Synthesis of Aspidospermine



JACS 1963, 85, 2872

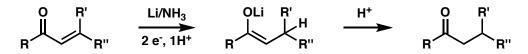
#### Byssochlamic Acid and Yohimbine



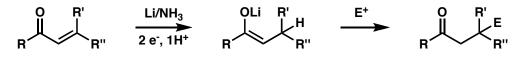
JACS 1972, 94, 5109

#### Regiospecific Formation/Trapping of Enolates Adding 2e<sup>-</sup> and 1H<sup>+</sup> to an Enolate

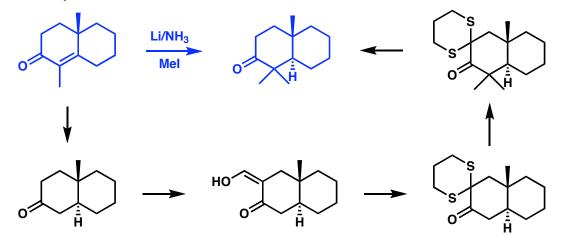
- Regiospecific alkylation was a big problem, and it often required blocking the undesired site.
- Metal/ammonia reduction of an  $\alpha$ , $\beta$ -unsaturated keton proceeds through an intermediate enolate, which is formed regiospecifically!



• If it is a Li enolate, alkylation is faster than protonation.



• This is a good way to save steps:



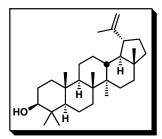
*JACS* **1965**, *87*, 275

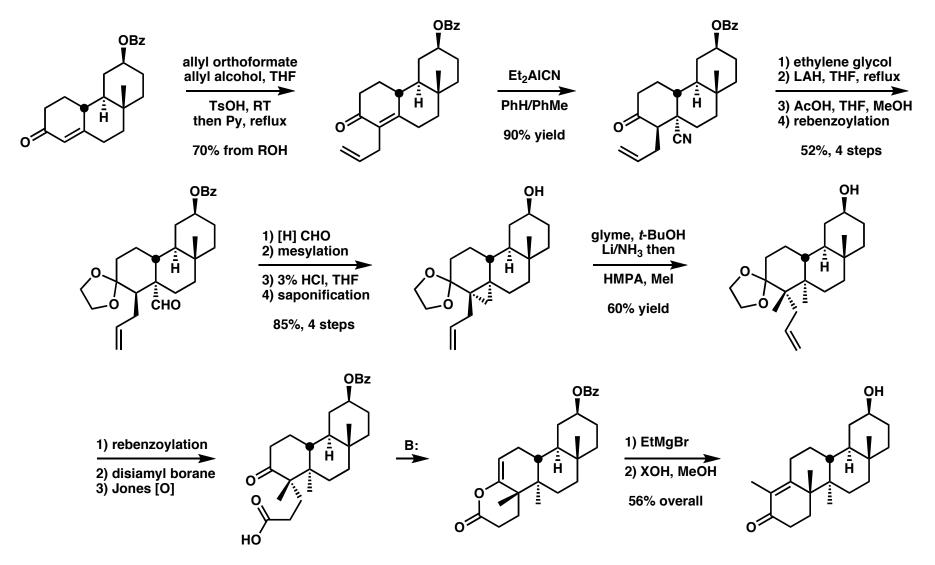
• Also a good technique for reductive alkylation of enediones, useful in the synthesis of corticosteroids. *JACS* **1980**, *102*, 1218 and *JACS* **1980**, *102*, 1219

• Examples in total synthesis (that I don't have time to discuss): *dl*-D-homotestosterone and *dl*-progesterone (*JACS* 1967, *89*, 5464) and Adrenosterone (*JACS* 1982, *104*, 3767).

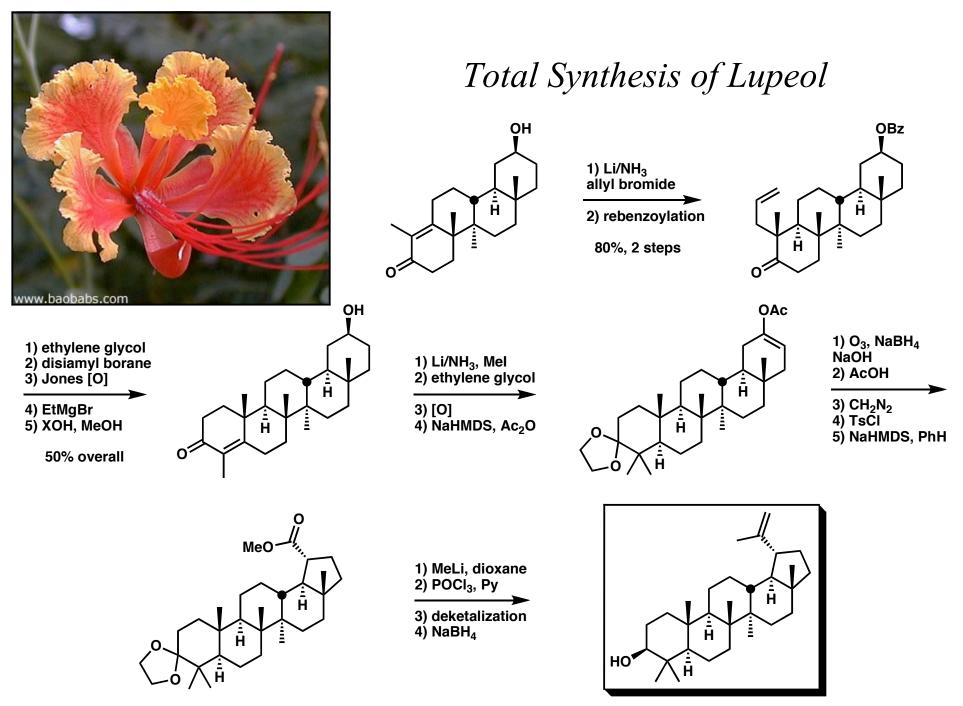
JACS 1961, 83, 2965

#### Total Synthesis of Lupeol





JACS 1971, 93, 4945

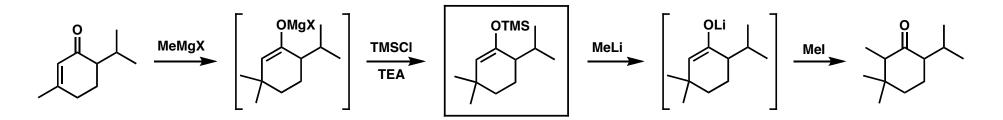


JACS 1971, 93, 4945

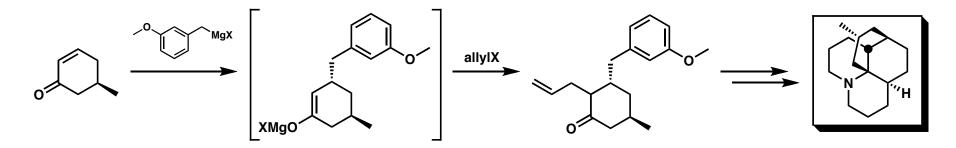
### Regiospecific Formation/Trapping of Enolates

Conjugate Addition of an Organometallic Reagent into an Enolate

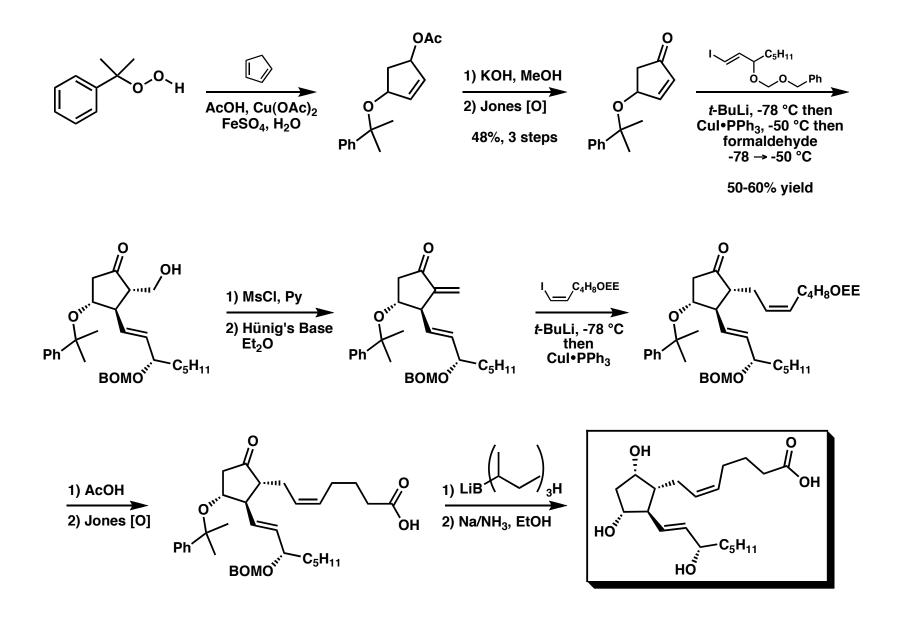
- Another alternative developed to avoid blocking groups, this time allowing effective  $\beta$ -alkylation by conjugate addition. *JACS* **1968**, *90*, 4462 and *JACS* **1968**, *90*, 4464
- Trapping the enolate as a silvl enol ether was key to the general applicability of this method, allowing addition of the trapped enolates to  $\alpha$ , $\beta$ -unsaturated systems.



• From the total synthesis of Lycopodine (Pure Appl. Chem. 1968, 17, 383):



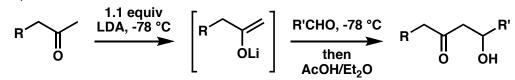
### Total Synthesis of Prostaglandin $F2\alpha$



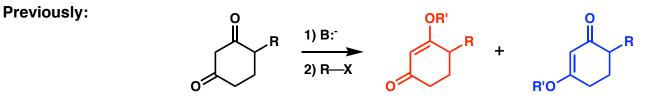
JACS 1975, 97, 6260

# Regiospecific Deprotonation-Trapping with Saturated Carbonyls

- Key finding thus far: Lithium enolates are special!
- Eventually discovered that even kinetic enclates capable of undergoing quilibration to the thermodynamic enclate retain their regiochemical fidelity.
- For methyl ketone enolates, this development made *inter*molecular aldol reactions possible by trapping with an aldehyde. *JACS* **1974**, *39*, 3459.

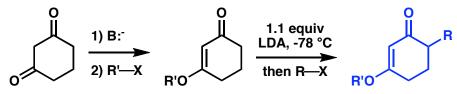


• A very useful and well-known application of the regiospecific deprotonation: alkylation of β-alkoxy enones (Stork-Danheiser!) *JOC* **1973**, *38*, 1775

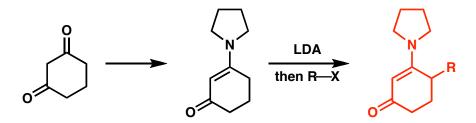


Mixture of products!



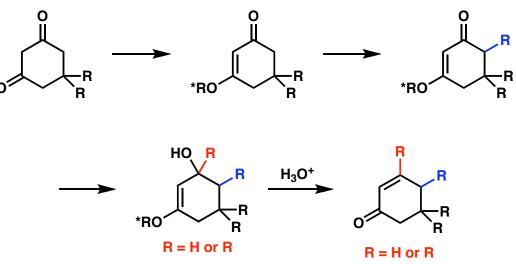


AND the other product is still available through enamine alkylation chemistry!

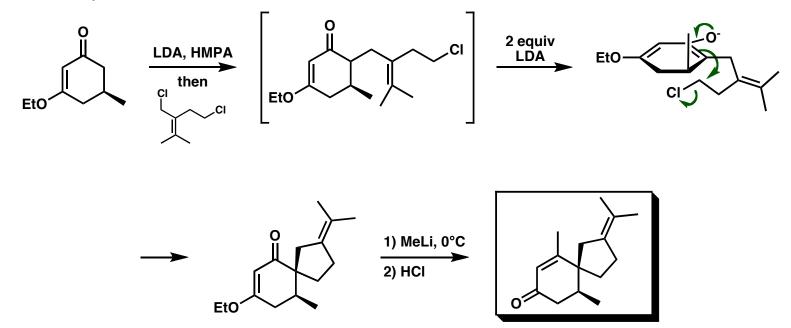


*Poly-Alkylations and Synthesis of β-Vetivone* 

 In Stork-Danheiser chemistry followed by dehydration, the conjugated enone is always formed, no matter how stable the β,γ-isomer is.



• Total synthesis of β-vetivone. JACS 1973, 95, 3414



### *Cyano-Epoxide Cyclizations*

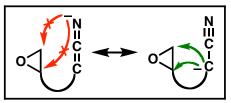
- Key: The  $\alpha$  anion of a nitrile is unusually effective in opening even tetrasubstituted epoxides in a position- and stereocontrolled and predictable manner.
- Nitrile features (JACS 1974, 96, 5268):

- 1) High nucleophilicity as a result of the low acidity of the  $\alpha$ -methylene.
- 2) Displacement by the anion delocalized on the heteroatom is geometrically impossible for normal ring sizes.
- Enolates either do not cyclize or give the O-alkylated product.
- If the cyclization is an *endo*  $S_N$ 2-type epoxide opening, the 6-membered ring forms faster than the corresponding 5-membered ring system would because no distortion is needed to acheive the necessary angle of approach. JOC 1974, 96, 5270
- The 4-membered ring is similarly well-poised, but for an exo cyclization. It proceeds stereospecifically to the *trans* product.

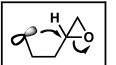


- Given an equal degree of substitution on either side of the epoxide, the smaller ring (exo product) will always form.
- A cyclopropane ring will always form in preference to a cyclobutane ring, regardless of the substitution on the epoxide.









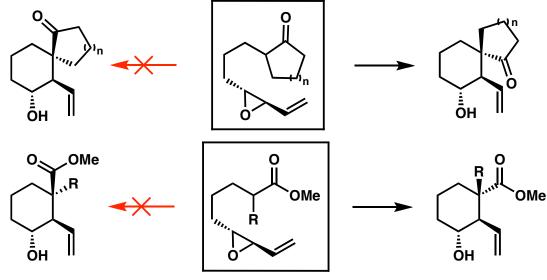




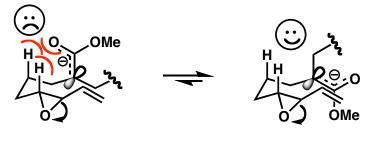


# Allylic Epoxide Cyclizations

- Ketone enolates can be used in conjunction with *trans*-substituted allylic epoxides to form cyclized products.
- This is particularly useful if you want to make a 6-membered ring out of an equally substituted epoxide.
- The cyclizations proceed analogously when a disubstituted ethynyl group is used in place of the olefin. (*JACS* **1990**, *112*, 1661)



• Stereochemical Interpretation:

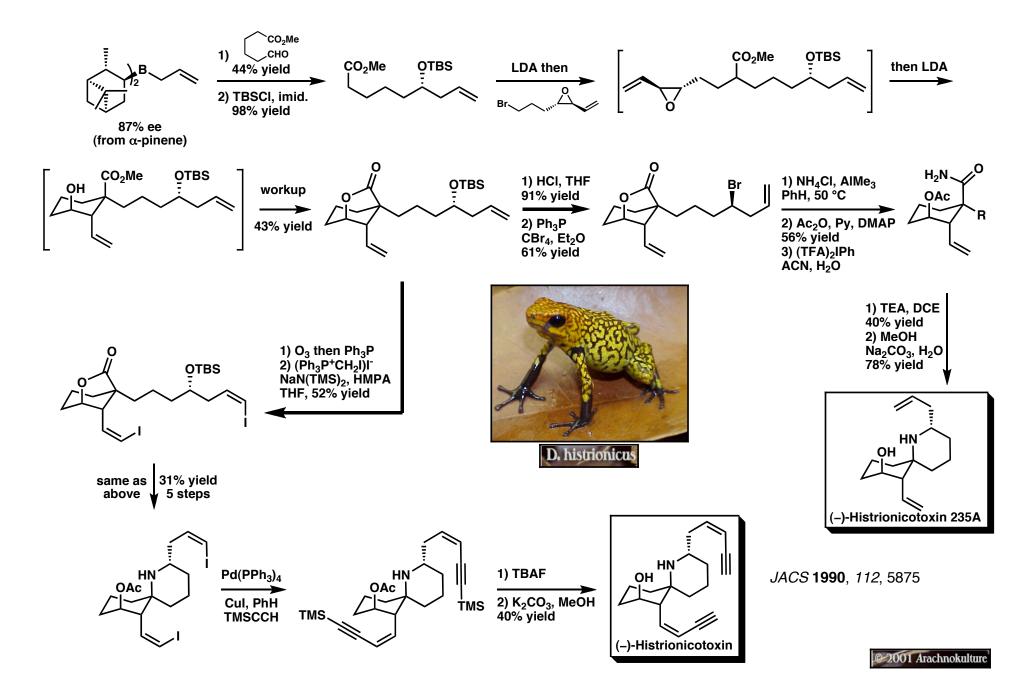


axial enolate

equatorial enolate

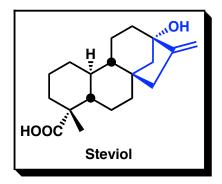
In both cases, the carbanion orbital is perpendicular to the plane of the carboxylate, so all orbitals are correctly alligned.

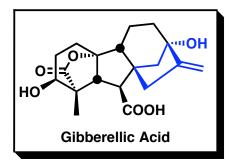
#### Total Syntheses of (–)-Histrionicotoxin and (–)-Histrionicotoxin 235A



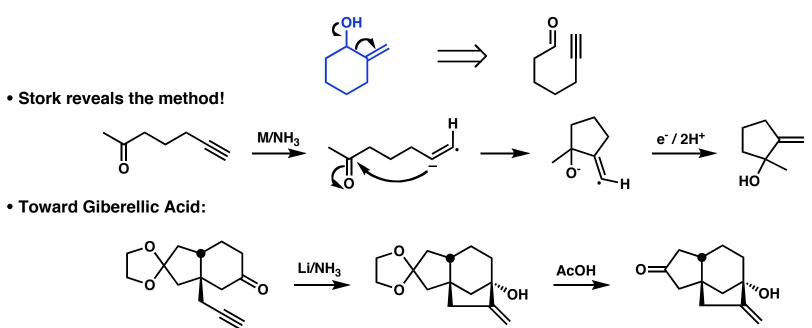
# Cyclization of Alkynyl Ketones

• Inspiration for this work came from natural products such as Steviol and Gibberellic Acid.





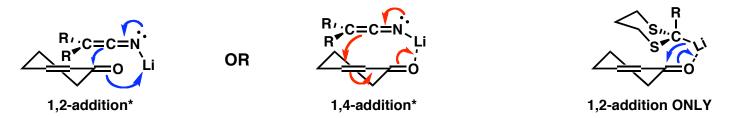
• A retrosynthetic analysis reveals the synthon:



JACS 1965, 87, 1148, JACS 1979, 101, 7107, and JACS 1979, 101, 7109

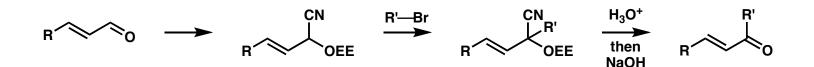
### Protected Cyanohydrins as Acyl Carbanion Equivalents

• Protected cyanohydrins undergo conjugate addition to enone systems. A common acyl carbanion equivalent, the dithiane, generally undergoes 1,2-addition (*JACS* 1971, *93*, 5286 and *JACS* 1974, *96*, 5272):

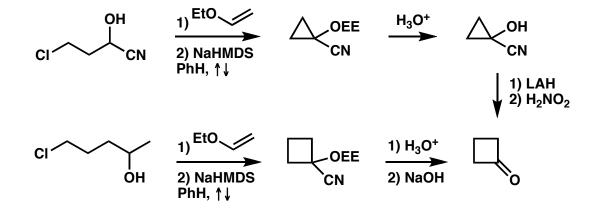


\*NOTE: Steric hinderance in the R groups or near the carbonyl of the enone lead to more 1,4-addition.

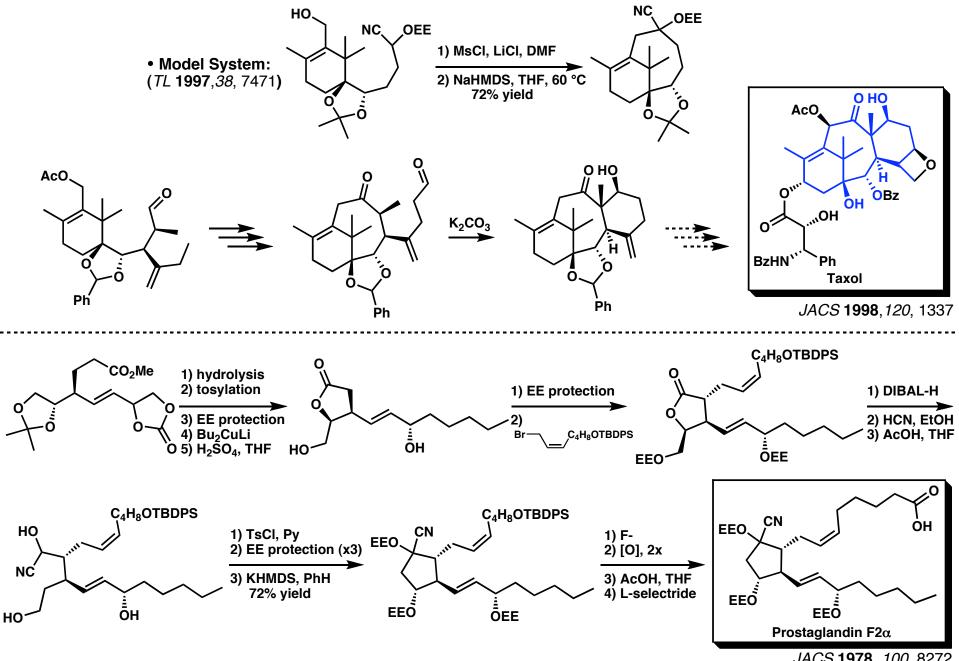
•  $\alpha,\beta$ -Unsaturated aldehydes can be made into  $\alpha,\beta$ -unsaturated ketones:



• Protected cyanohydrins are useful for small ring synthesis (TL 1975, 389):



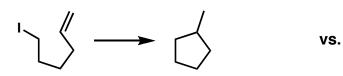
### Application to the ABC Rings of Taxol and to Prostaglandin $F2\alpha$

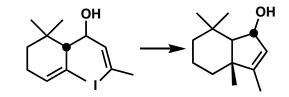


JACS 1978, 100, 8272

#### The Vinyl Radical Cyclization

• Goal: To build rings with a predictably placed olefin, which could be used as a functional group handle, unlike the alkyl radical cyclization, which results in a saturated ring. (*JACS* 1982, *104*, 2321)

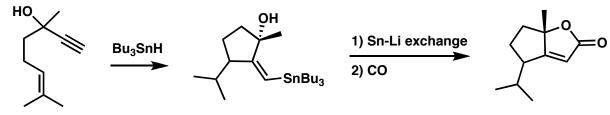




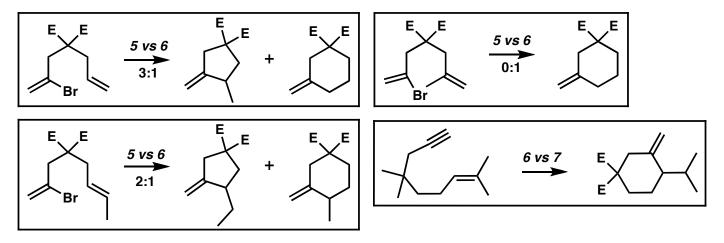
Other Advantages:

- Smaller steric hinderance than even an alkyl radical.
- More reactive than an alkyl radical.
- Earlier transition state may decrease the effect of sterics even further.
- Can be generated from vinyl halides (generated by alkylation with a substituent containing a vinyl halide, Corey reductive halogenation of ethynyl carbinols, or addition of HBr to a terminal acetylene) or by addition of Bu<sub>3</sub>SnH across an alkyne. (*JACS* 1987, *109*, 2829)

The vinyl stannane product can be reduced to an olefin or further manipulated.

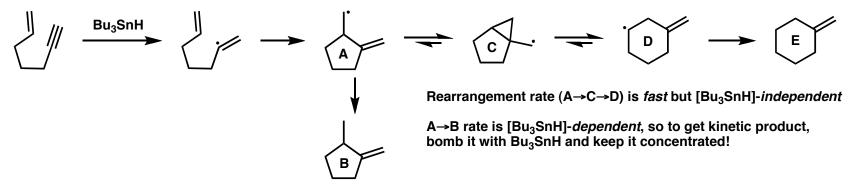


• Kinetics vs. Thermodynamics... similar to ratios observed for alkyl radical cyclizations.



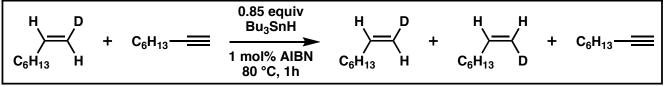
### The Vinyl Radical Cyclization

• The product ratio is a bit more tunable in the case of vinyl stannane-derived radicals. (*JACS* **1987**, *109*, 2829 and *TL* **1986**, *27*, 4529)

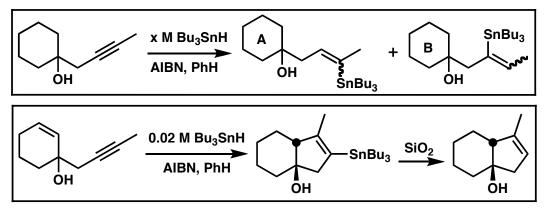


• The selectivity for cyclization of the vinyl radical onto the olefin vs. the cyclization of an alkyl radical onto the alkyne is somewhat surprising and has important implications. (Stannyl radical addition to the olefin is reversible.)

"Enyne cyclizations are successful because cyclization is faster, and reversal slower, for vinyl radicals than for their alkyl counterparts, so that the whole cyclization pathway is via the adduct of the stannyl radical to the acetylene rather than to the olefin."



• The regioselectivity of stannyl radical formation and subsequent cyclization implies that its addition to the *triple* bond is *also* reversible!

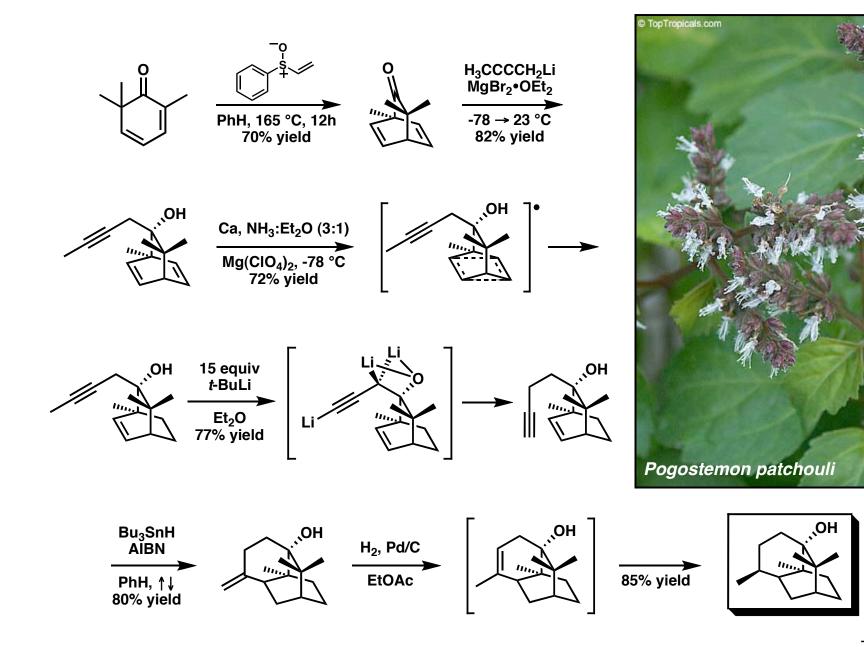


For x = 0.02 M Bu<sub>3</sub>SnH (normal conditions), *no* product is observed.

For x = 0.77 M Bu<sub>3</sub>SnH, a 1.2:1 ratio of A:B is observed in addition to some SM.

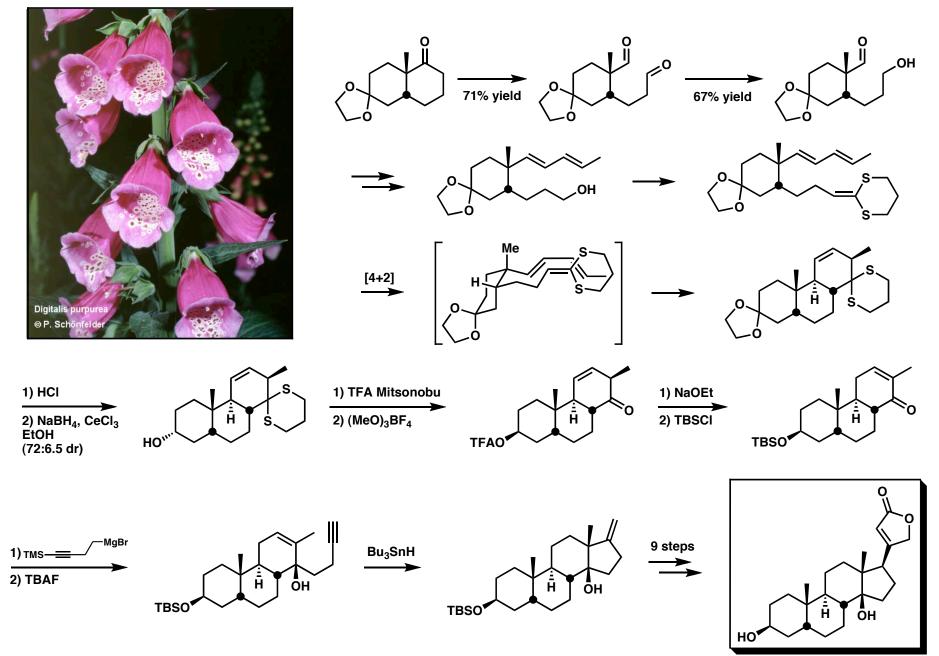
Reactivity is now observed at *low* tin concentrations, thus confirming that the *reversibility* of the stannyl radical radical addition to the triple bond leads to the preferential formation of the more rapidly formed ring (here 5 vs. 4).

# Total Synthesis of Patchouli Alcohol



*TL* **1995**,*36*, 7607

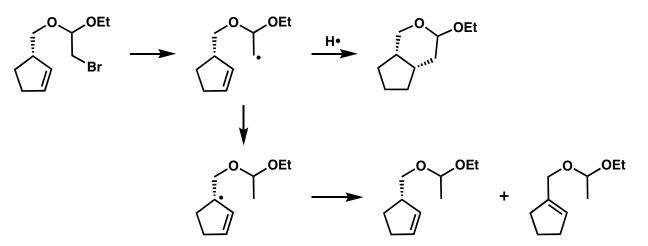
#### Total Synthesis of (+)-Digitoxigenin



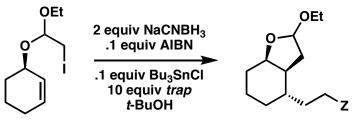
JACS 1996, 118, 10660

### Position and Stereochemical Control by Radical Cyclization to Form a Temporary Ring ~ Mixed Acetal Linkages

• Goal: Take advantage of the position selectivity and stereochemical fidelity of the previously developed radical cyclizations, but allow the ring to be disassembled after the relevant stereocenters are set. *JACS*, **1983**, *105*, 3741 and *JACS*, **1985**, *107*, 500

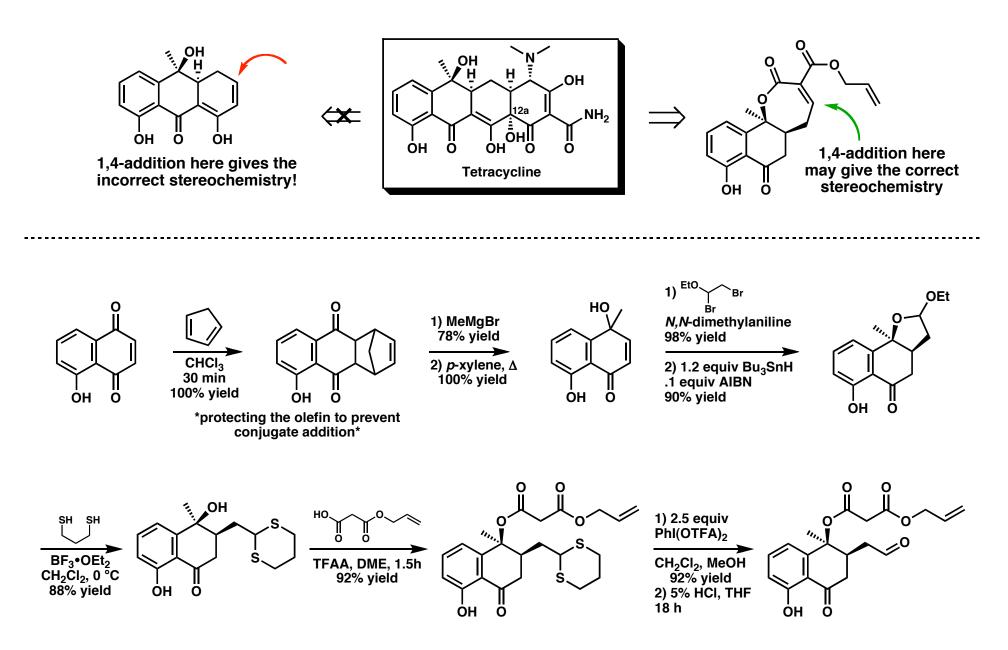


• Trapping with an electrophile enables multiple new stereocenters to be set during the cyclization. *JACS*, **1986**, *108*, 303



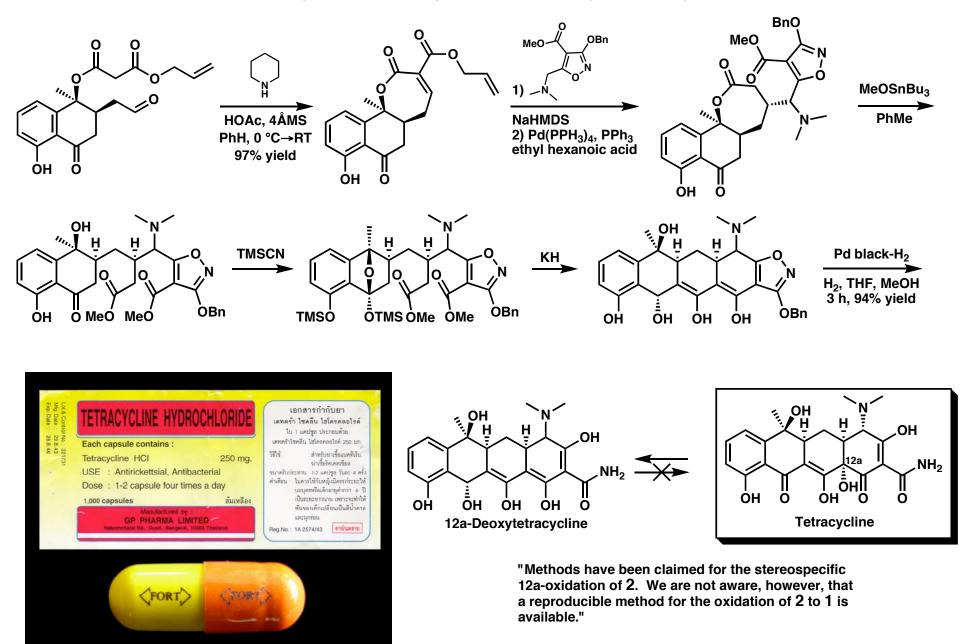
- $Z = CN, CO_2Me, COEt, PO(OEt)_2, SO_2Me, SO_2Ph$
- Mini-Review: Radical Cyclization in the Control of Regio- and Stereochemistry. Bull. Chem. Soc. Jpn. 1988, 61, 149-154.
- Synthetic Applications: Prostaglandin F2α JACS, 1986, 108, 6384. Reserpine JACS, 2005, 127, 16255.

#### Total Synthesis of 12a-Deoxytetracycline

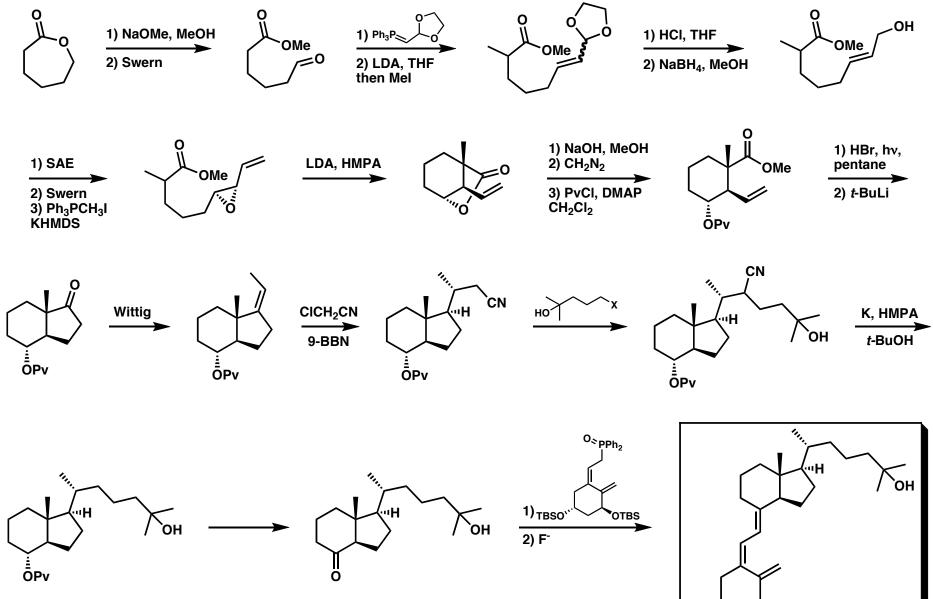


JACS, 1996, 118, 5304

#### Total Synthesis of 12a-Deoxytetracycline



### Total Synthesis of Calcitriol



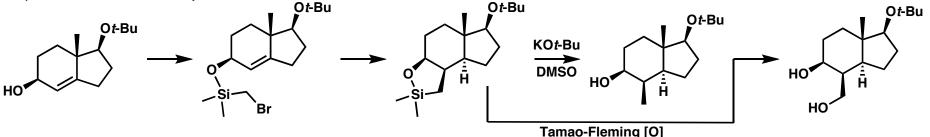
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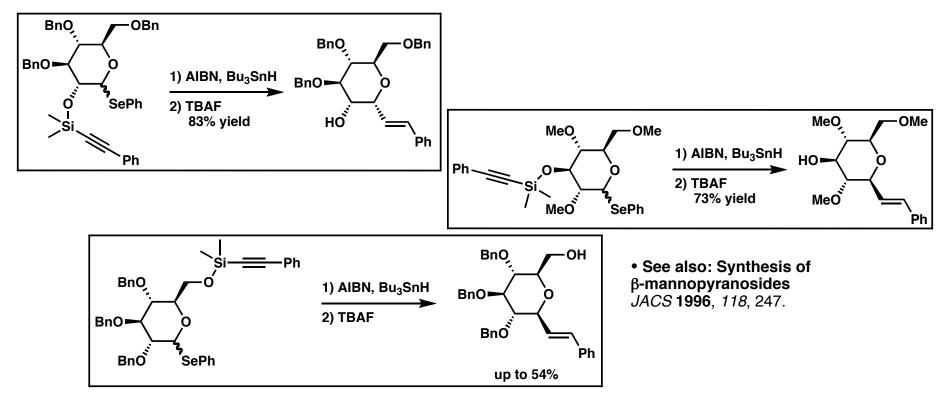
Pure & Appl. Chem., **1992**, *64*, 1809 Bull. Korean Chem. Soc., **1997**, *18*, 137 Aust. J. Chem., **1992**, *45*, 275

### Position and Stereochemical Control by Radical Cyclization to Form a Temporary Ring ~ Temporary Silicon Connection

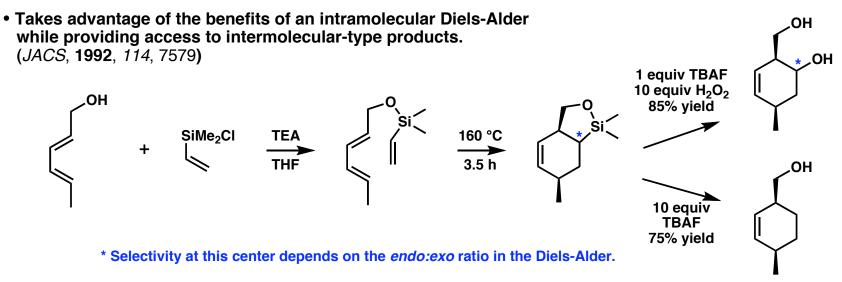
• Same Goal: (Take advantage of the position selectivity and stereochemical fidelity of the previously developed radical cyclizations, but allow the ring to be disassembled after the relevant stereocenters are set.) (*JACS*, **1986**, *108*, 6826)



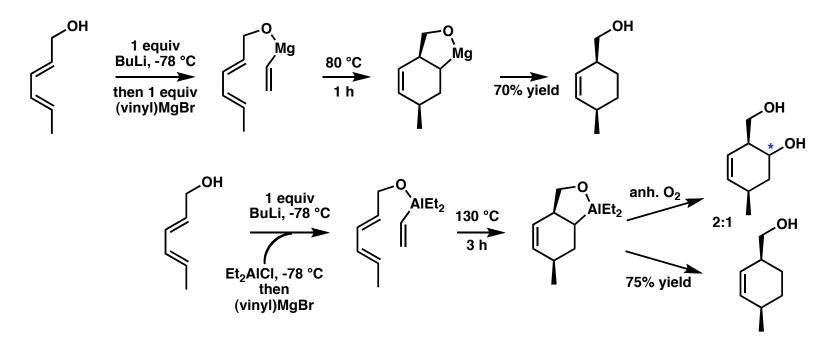
• Allows stereocontrol to be acheived at the anomeric center of a sugar. (JACS, 1991, 113, 7054)



# Temporary Silicon Connection in the Diels-Alder Reaction



• This transformation can also be effected with a temporary Mg/Al connection. (JACS, 1995, 117, 6595)





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