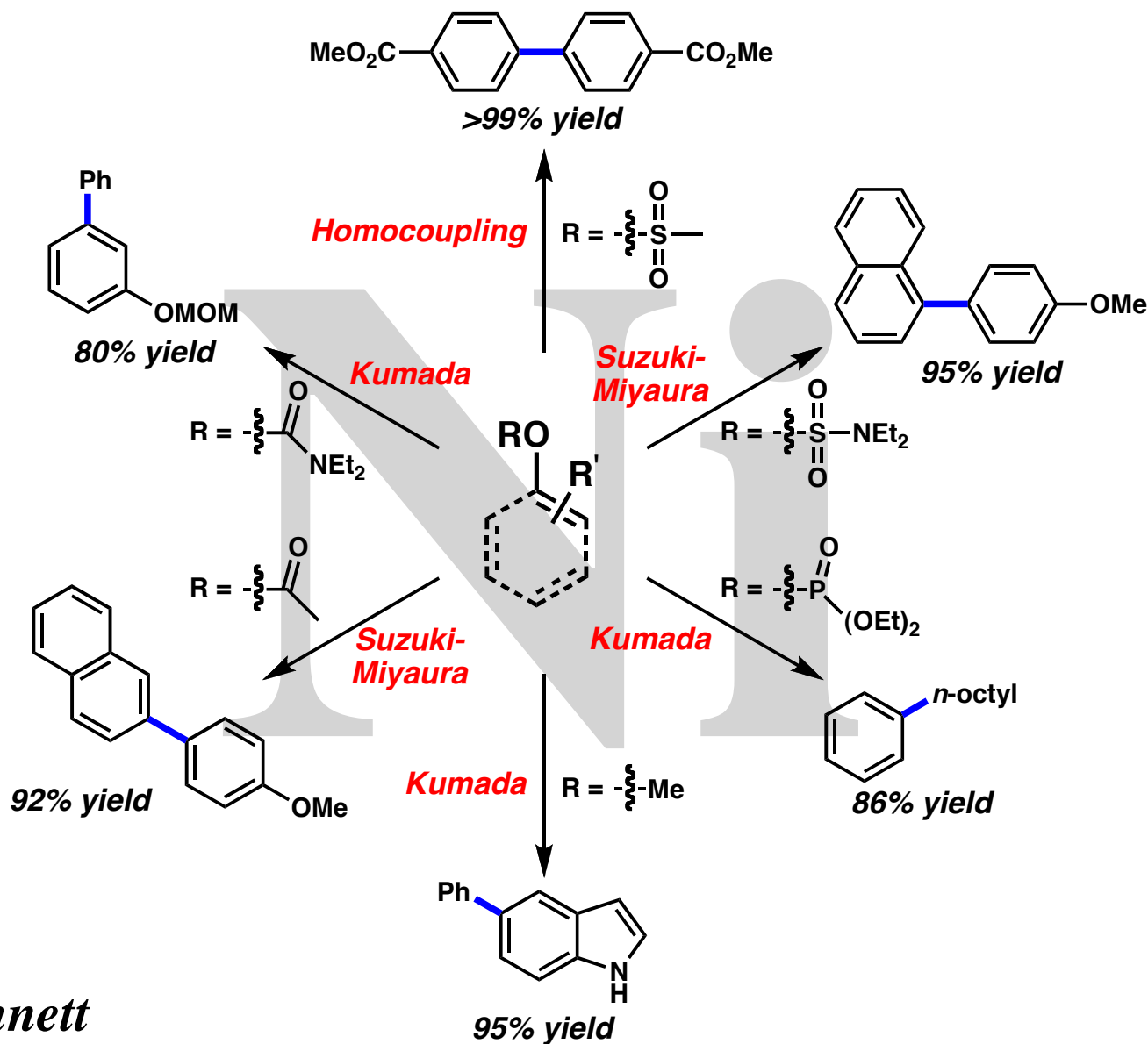
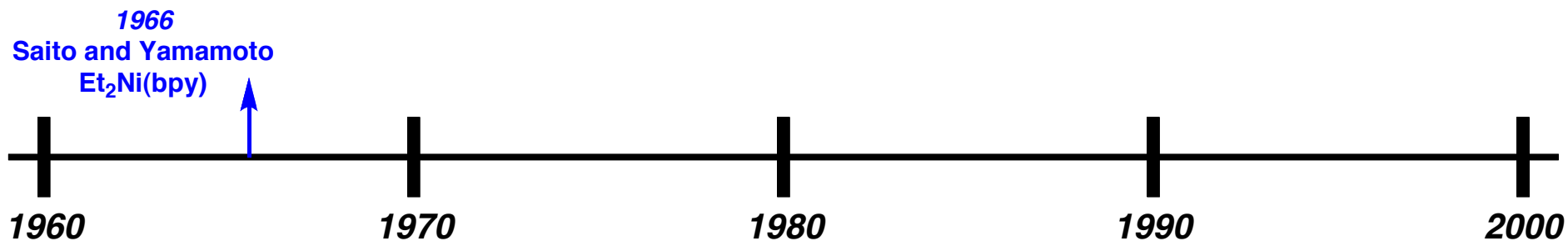


From C–O to C–C through Nickel Coupling

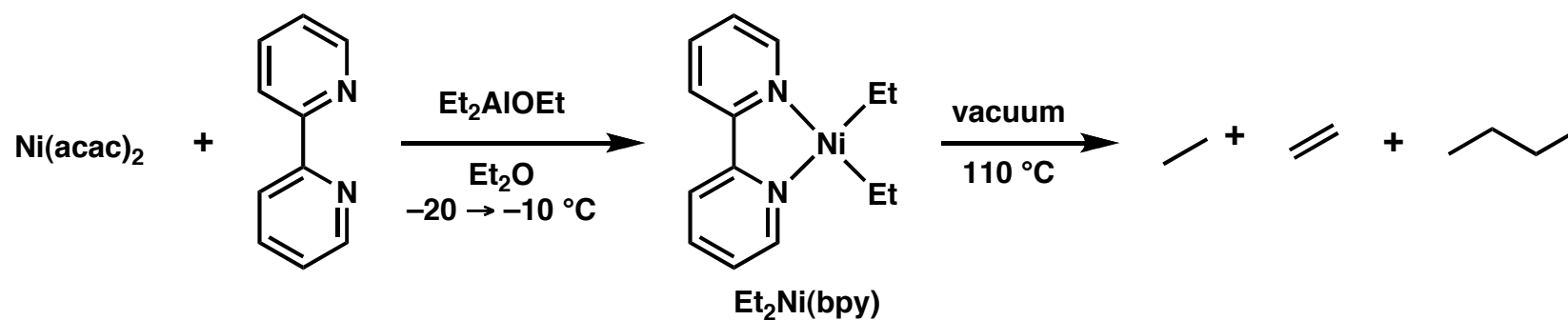


Ni Chemistry Timeline

Initial Studies



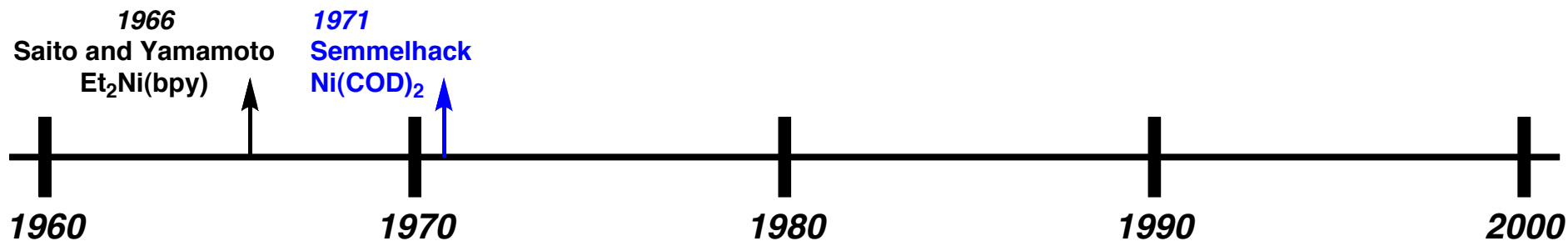
Saito, T. and Yamamoto, A.



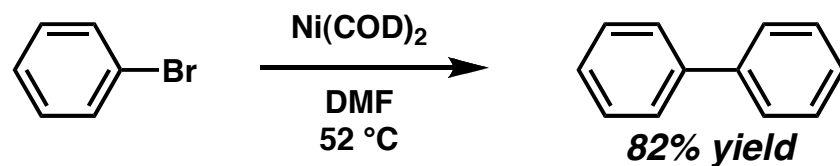
J. Am. Chem. Soc. **1966**, 88, 5198.

Ni Chemistry Timeline

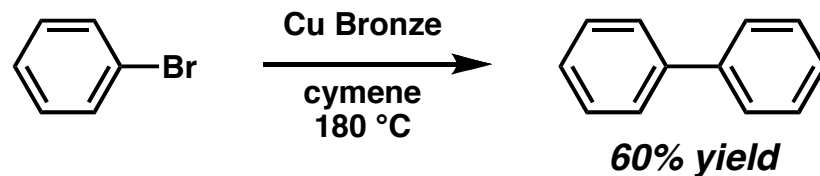
Initial Studies



Semmelhack, M. F.



Ullmann Coupling Comparison

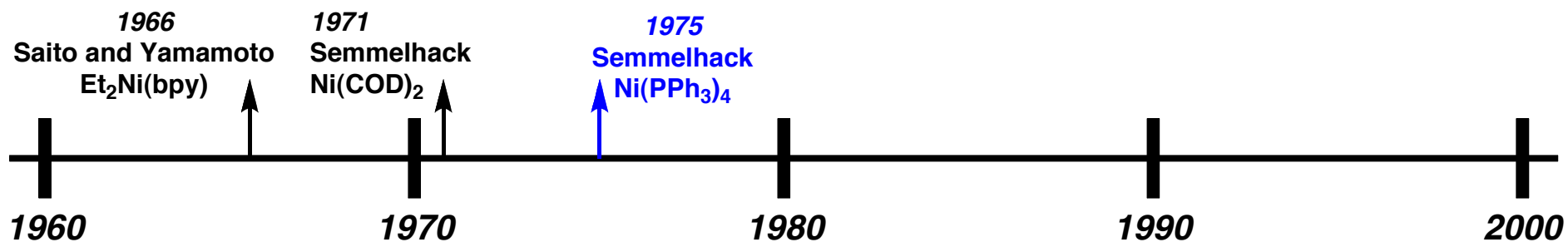


Ni: Semmelhack, M. F. *J. Am. Chem. Soc.* **1971**, 93, 5908.

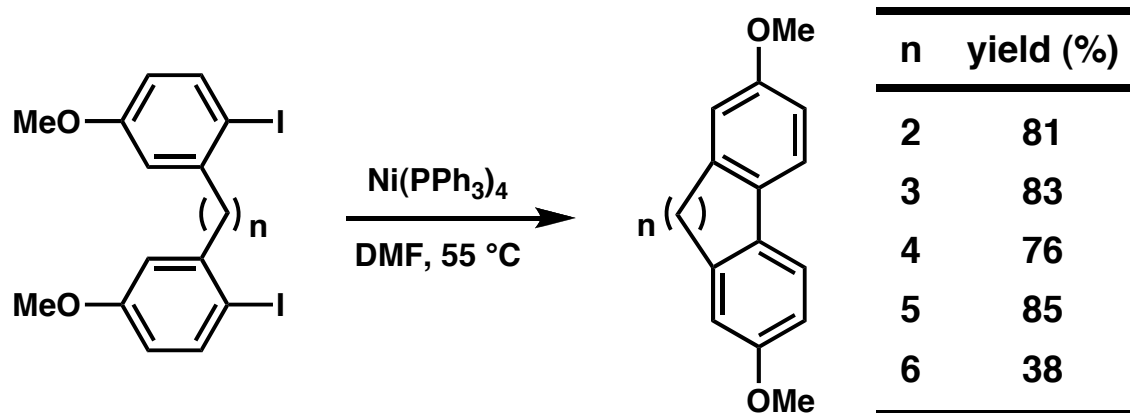
Ullmann Coupling: Fanta, P. E. *Chem. Res.* **1964**, 64, 613.

Ni Chemistry Timeline

Initial Studies



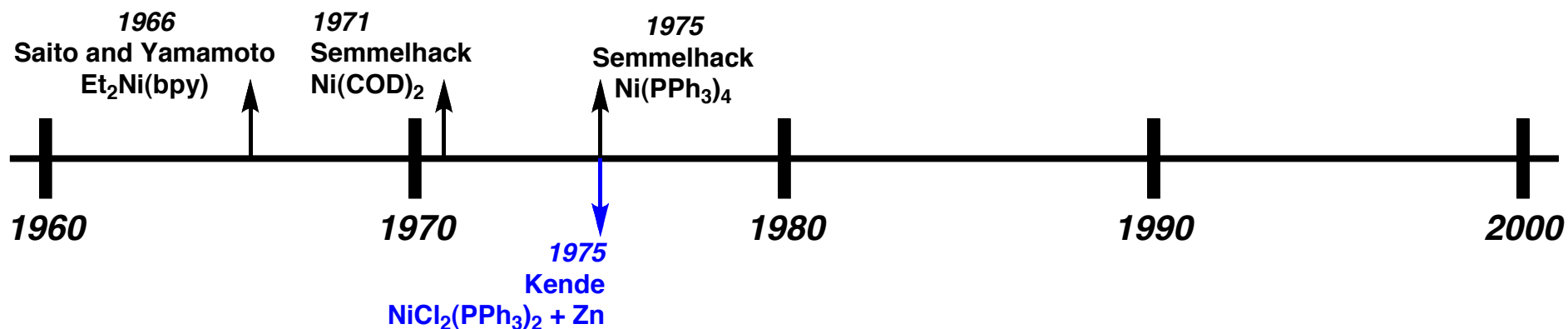
Semmelhack, M. F.



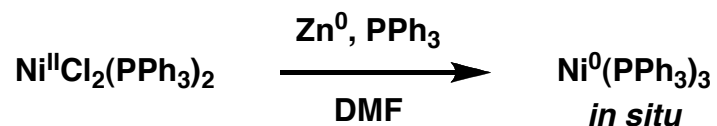
J. Am. Chem. Soc. **1975**, 97, 3873.
J. Am. Chem. Soc. **1981**, 103, 646.

Ni Chemistry Timeline

Initial Studies



Kende, A. S.

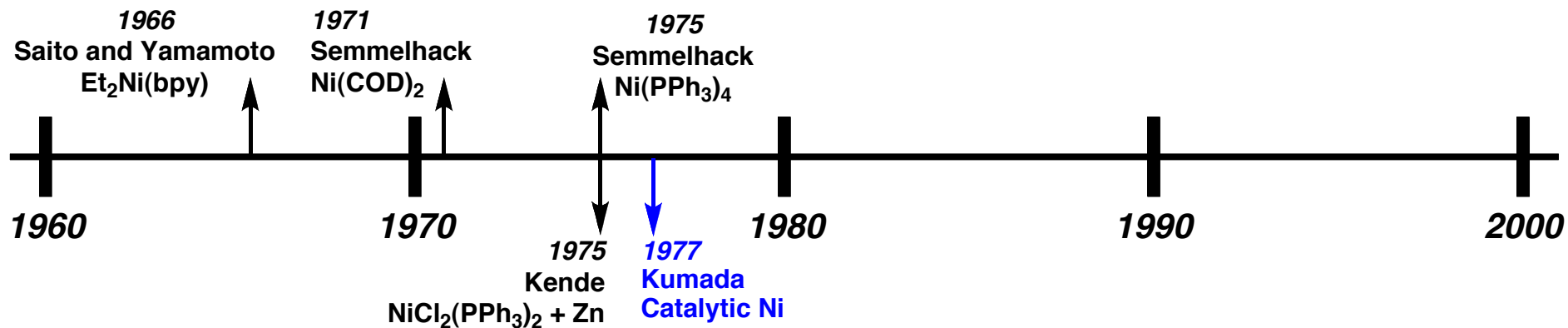


— Tolman Research: $\text{Ni}^0(\text{PPh}_3)_3$ active catalyst
 — Kende formed *in situ*

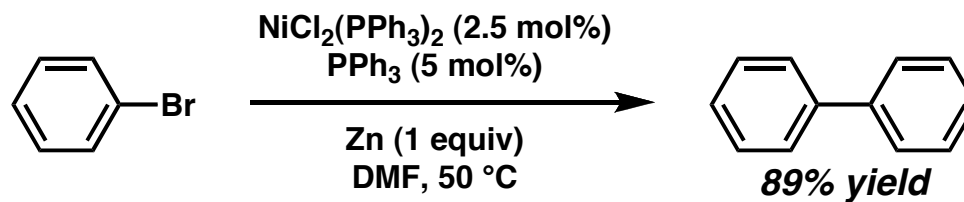
Tolman, C. A. *J. Am. Chem. Soc.* **1974**, 96, 53.
 Kende, A. S. *Tetrahedron Lett.* **1975**, 3375.

Ni Chemistry Timeline

Initial Studies



Kumada, M.



Why Phenol/Enol Derived Electrophiles?

Availability

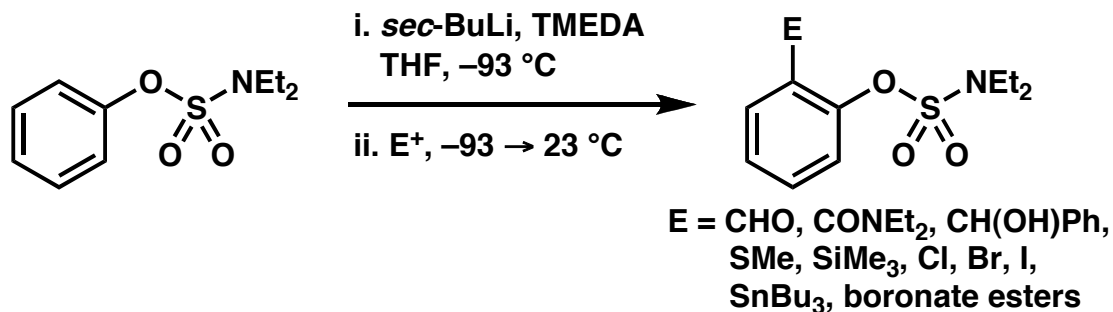
- 50,000 commercial phenol and aryl polyol
- Enol electrophile from trapping enolates

Reactivity

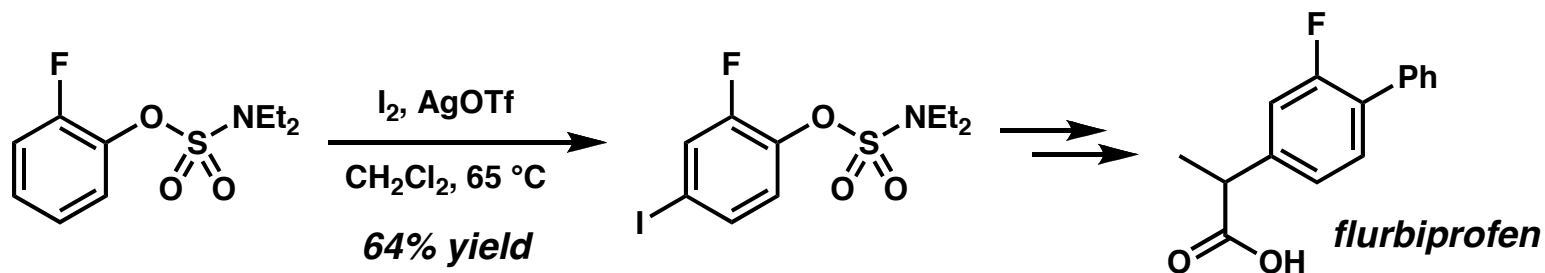
- Directed Ortho Metalation (DoM)

Possible with:

Sulfamates
Ethers
Esters
Carbamates



- *p*-Direction



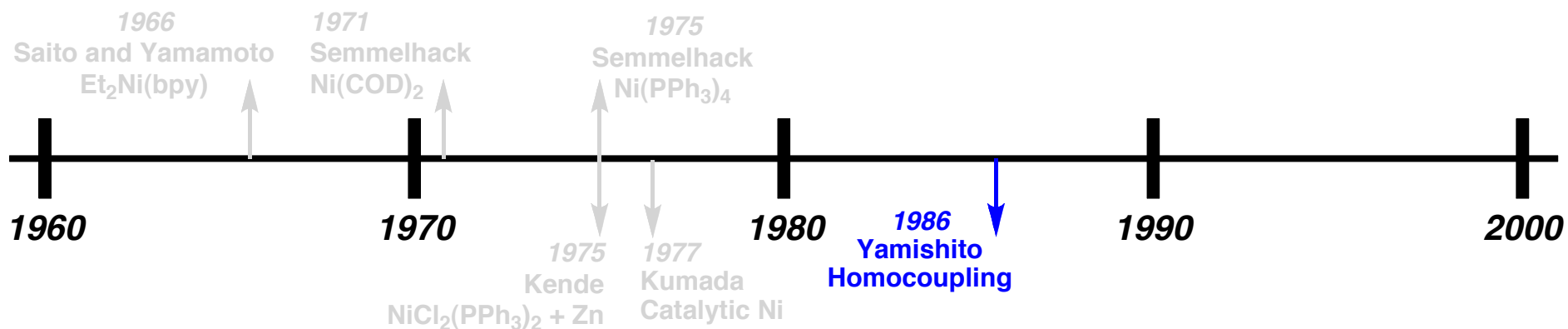
Garg, N. K. and Percec, V. *Chem. Rev.* **2011**, ASAP, DOI: 10.1021/cr100259t.

DoM Review: Snieckus, V. *Chem. Rev.* **1990**, 90, 879.

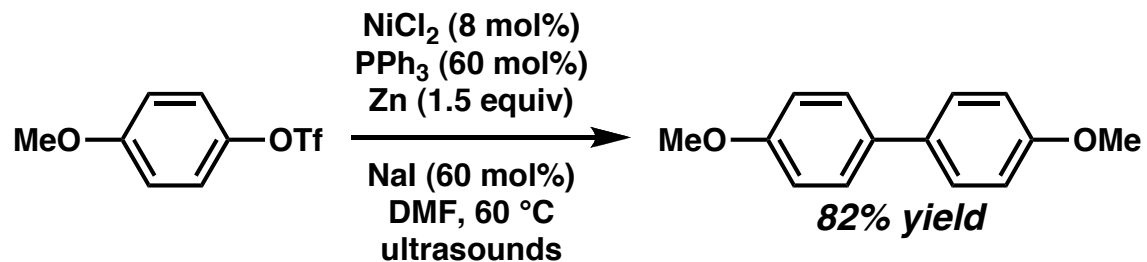
***p*-Direction Example:** Garg, N. K. *J. Am. Chem. Soc.* **2009**, 131, 17748.

Ni Chemistry Timeline

Sulfonate Highlights

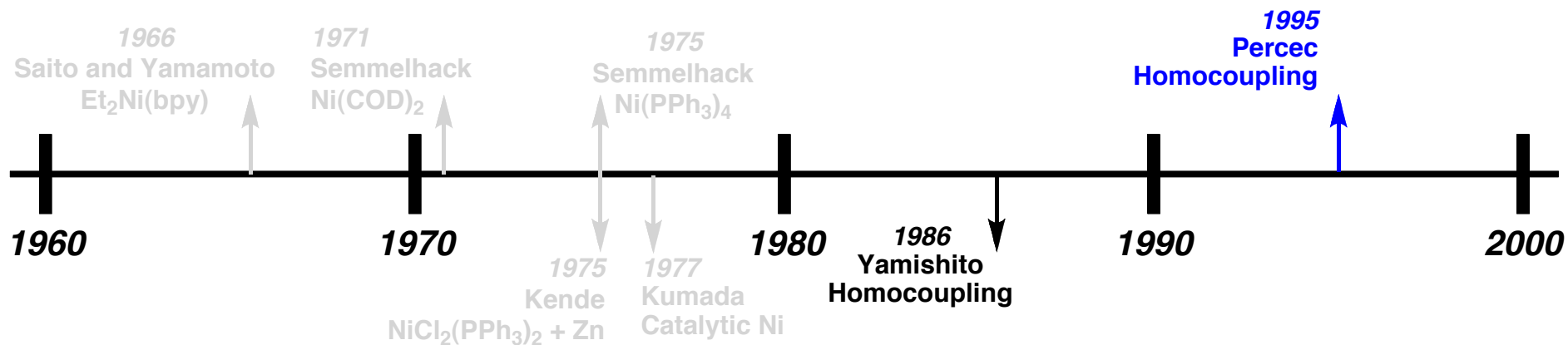


Yamishito, J.

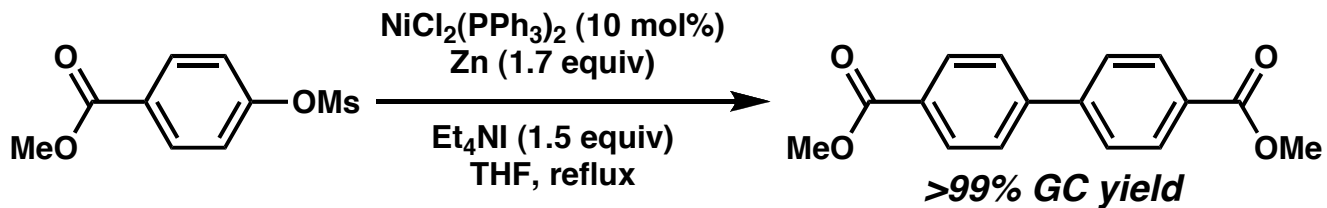


Ni Chemistry Timeline

Sulfonate Highlights

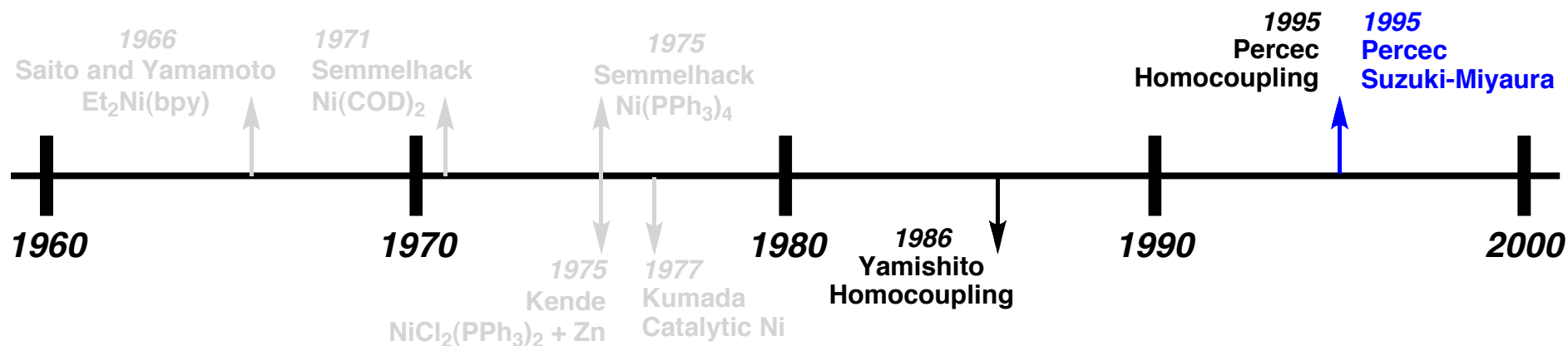


Percec, V.

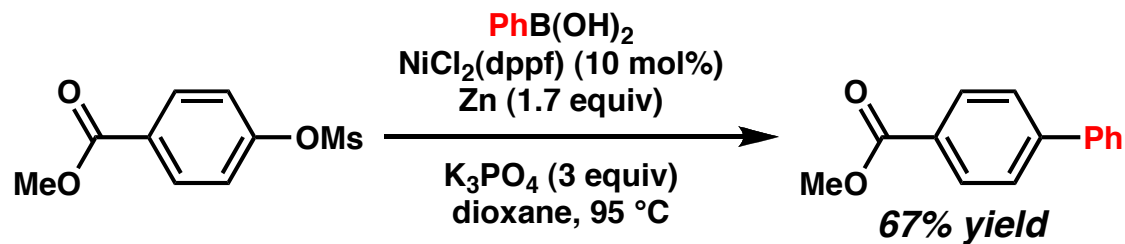


Ni Chemistry Timeline

Sulfonate Highlights

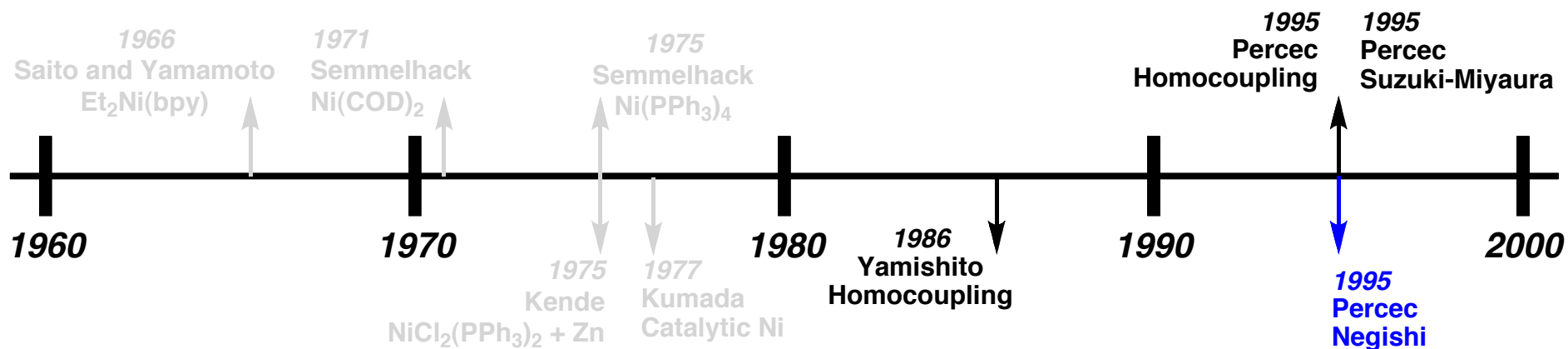


Percec, V.

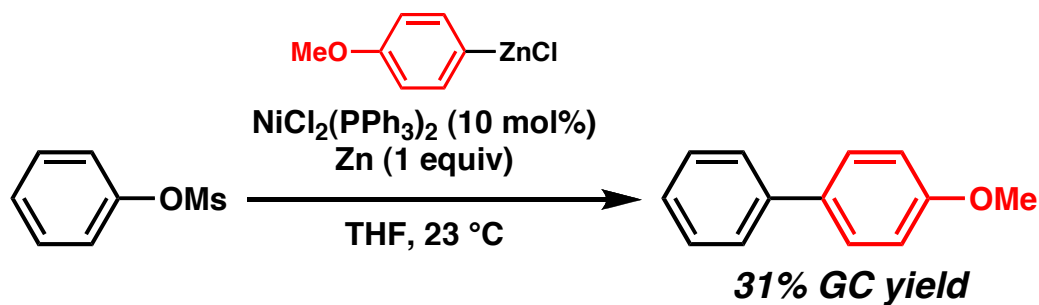


Ni Chemistry Timeline

Sulfonate Highlights

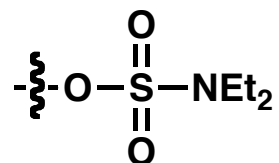


Percec, V.

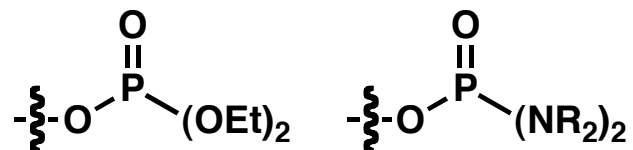


Overview

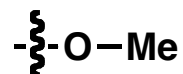
Sulfamates



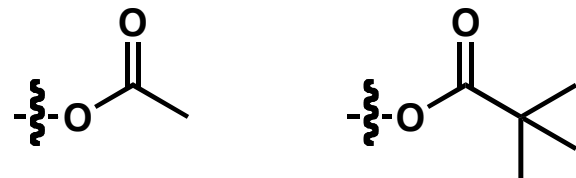
Phosphonates and Phosphoramides



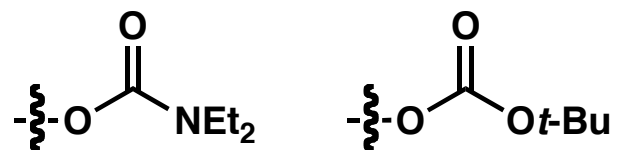
Ethers



Esters

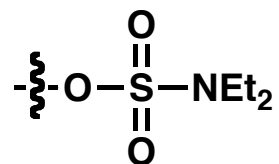


Carbamates and Carbonates

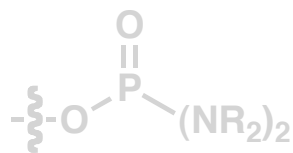


Overview

Sulfamates



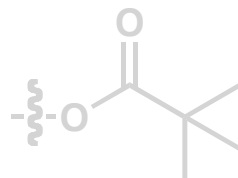
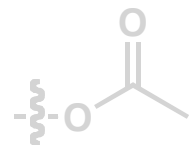
Phosphonates and Phosphoramides



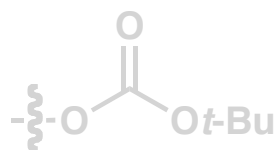
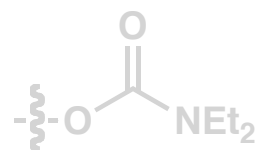
Ethers



Esters



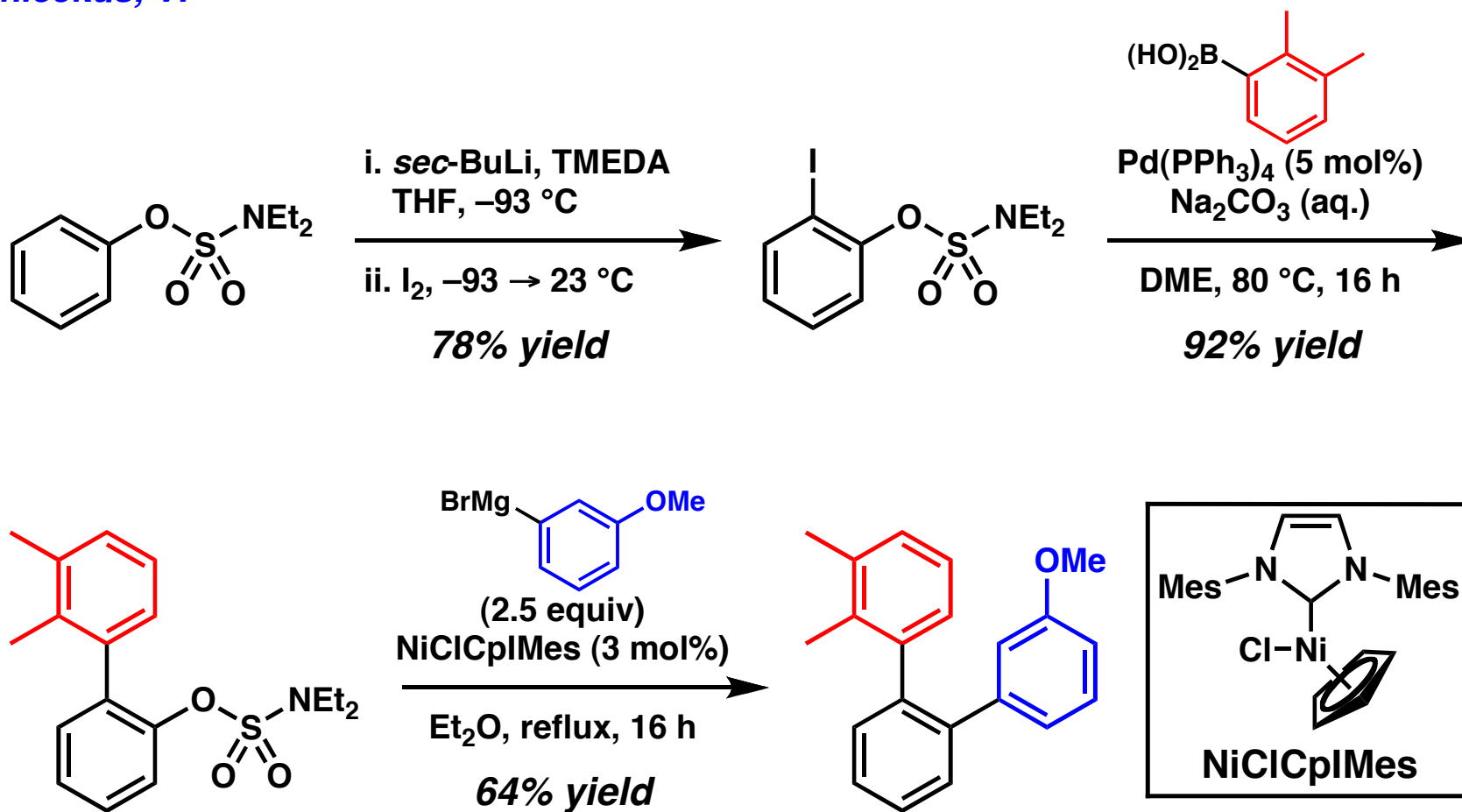
Carbamates and Carbonates



Sulfamate Ni Cross Coupling

Directed Ortho Metalation (DoM)

Snieckus, V.

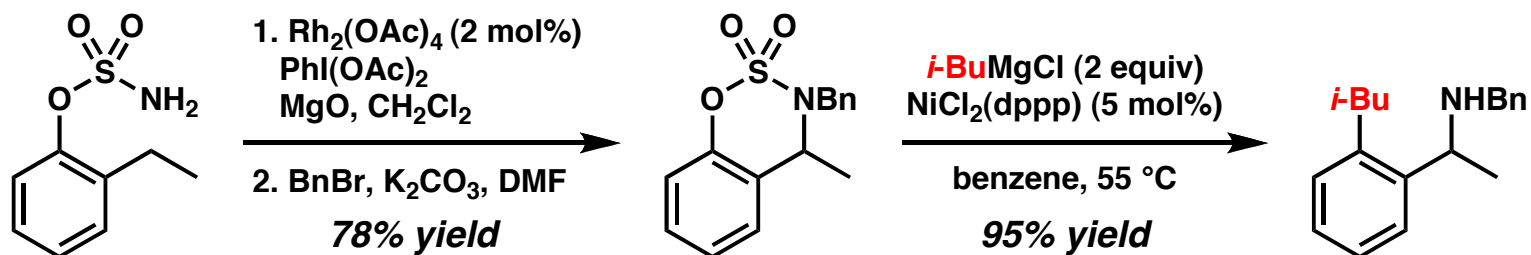


Sulfamate Ni Cross Coupling

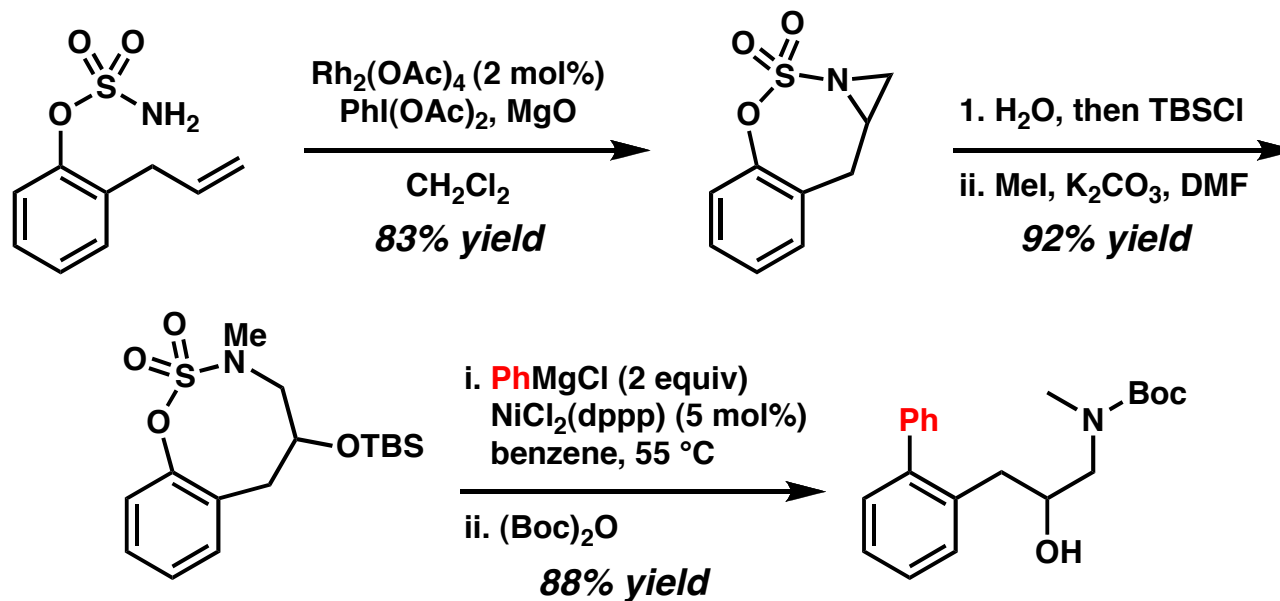
Kumada Coupling

Du Bois, J.

C-H Insertion



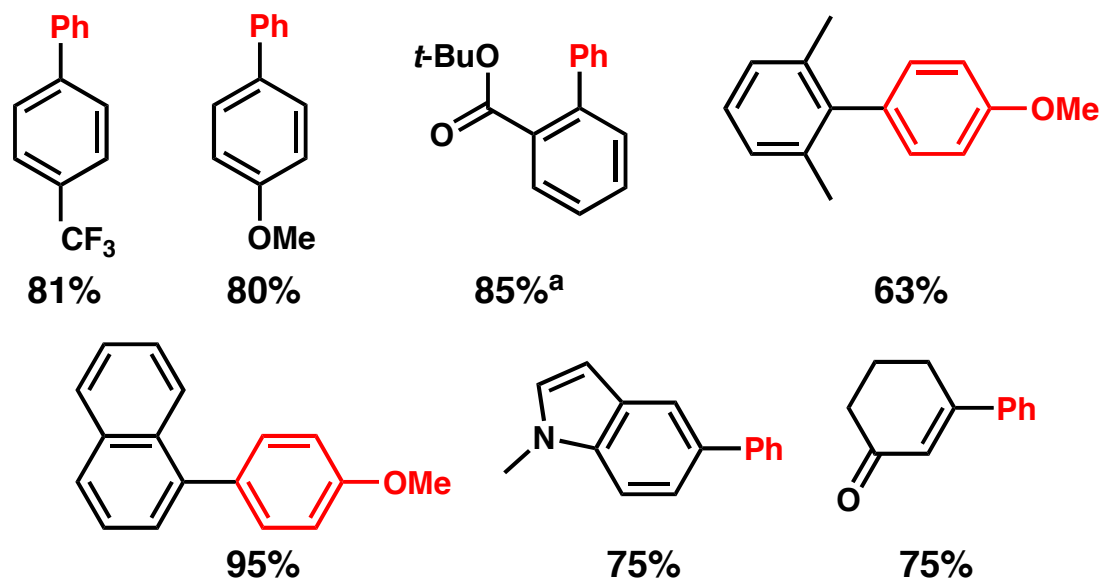
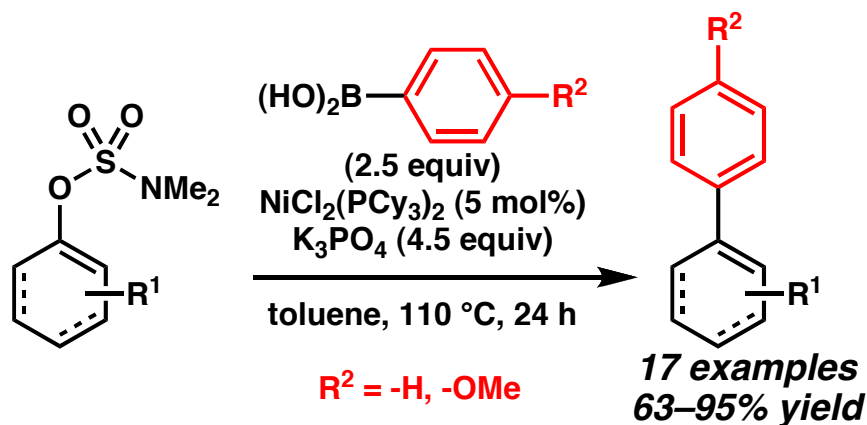
Aziridine Formation



Sulfamate Ni Cross Coupling

Suzuki-Miyaura Coupling

Garg, N. K.

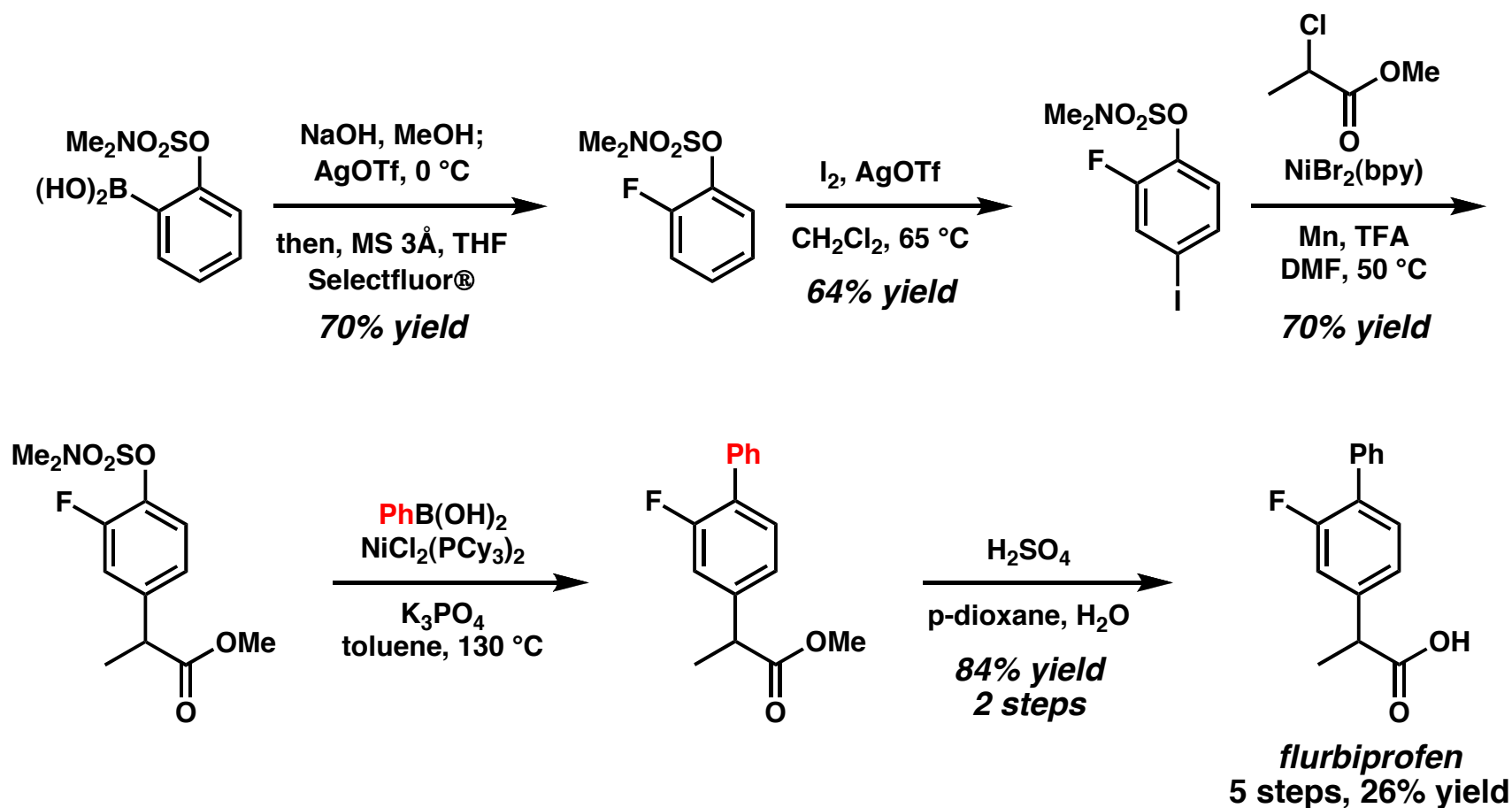


^a 10 mol% $\text{NiCl}_2(\text{PCy}_3)_2$, 4 equiv $\text{ArB}(\text{OH})_2$, 7.2 equiv K_3PO_4 , toluene, 130 °C

Sulfamate Ni Cross Coupling

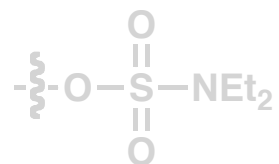
Total Synthesis of Flurbiprofen

Garg, N. K.

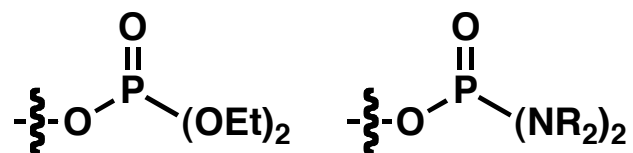


Overview

Sulfamates



Phosphonates and Phosphoramides



Ethers



Esters



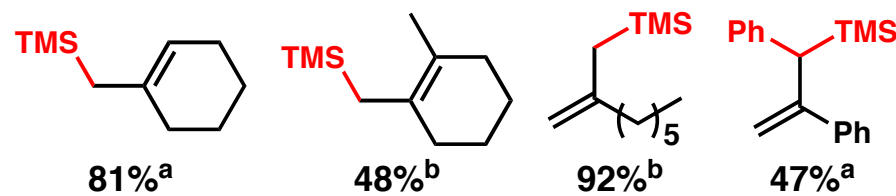
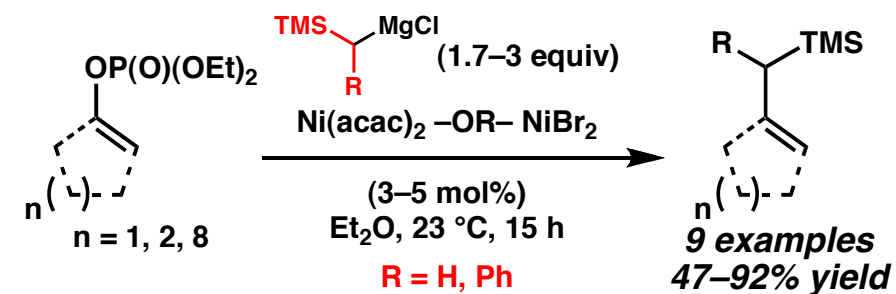
Carbamates and Carbonates



Phosphonate and Phosphoramidate Cross Coupling

Kumada Coupling: Vinyl Phosphonates

Kumada, M.



^a Ni(acac)_2 . ^b NiBr_2

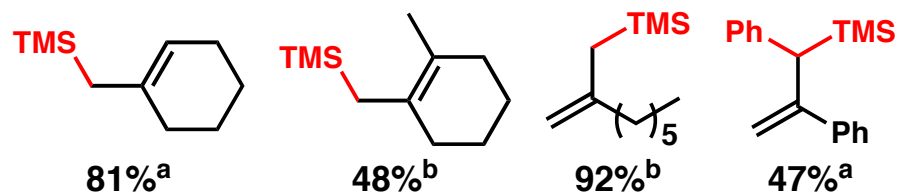
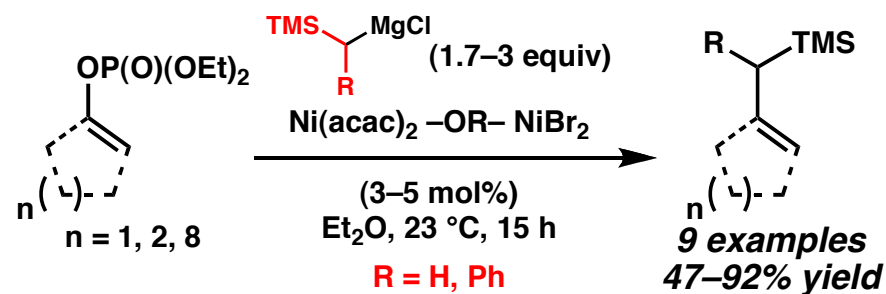
Kumada, M. *Synthesis* **1981**, 1001.
 Claesson, M. *Tetrahedron Lett.* **1983**, 24, 5137.

Nicolaou, K. C. *Chem. Commun.* **1998**, 1757.
 Backvall, J.-E. *J. Org. Chem.* **1999**, 64, 1745.

Phosphonate and Phosphoramidate Cross Coupling

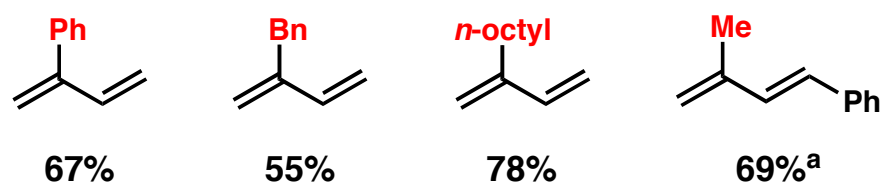
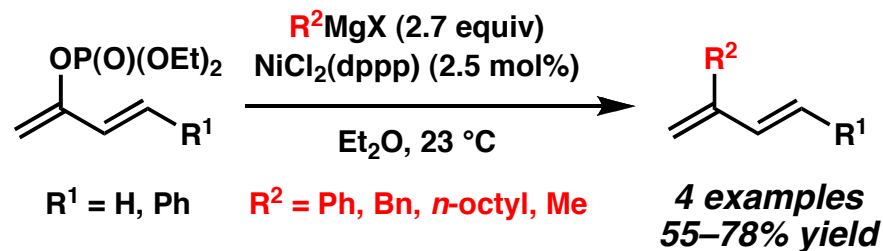
Kumada Coupling: Vinyl Phosphonates

Kumada, M.



^a Ni(acac)_2 . ^b NiBr_2

Claesson, A.



^a $\text{NiCl}_2(\text{dppe})$.

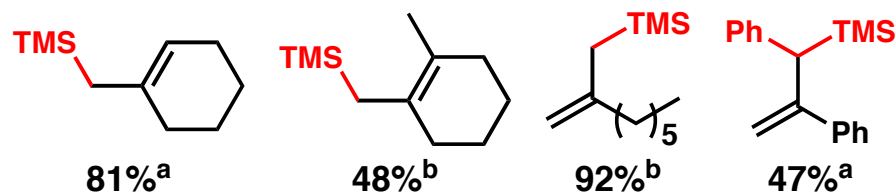
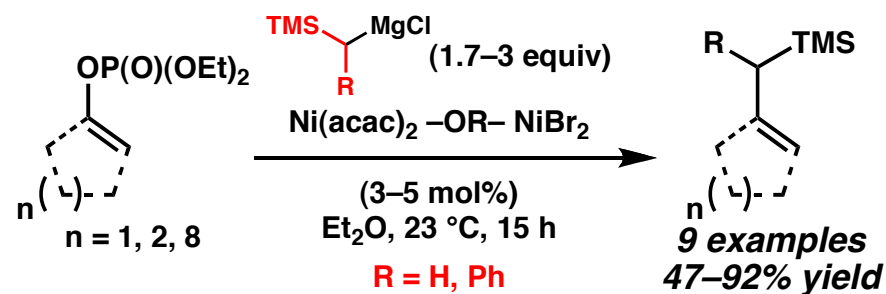
Kumada, M. *Synthesis* **1981**, 1001.
 Claesson, M. *Tetrahedron Lett.* **1983**, 24, 5137.

Nicolaou, K. C. *Chem. Commun.* **1998**, 1757.
 Backvall, J.-E. *J. Org. Chem.* **1999**, 64, 1745.

Phosphonate and Phosphoramidate Cross Coupling

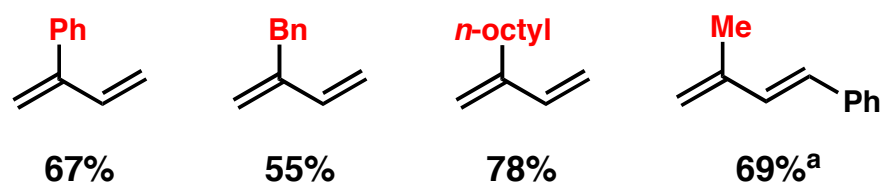
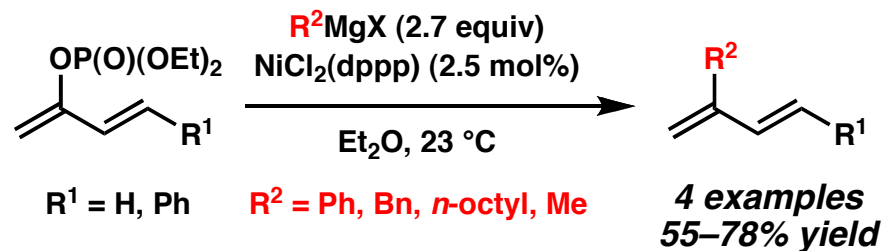
Kumada Coupling: Vinyl Phosphonates

Kumada, M.



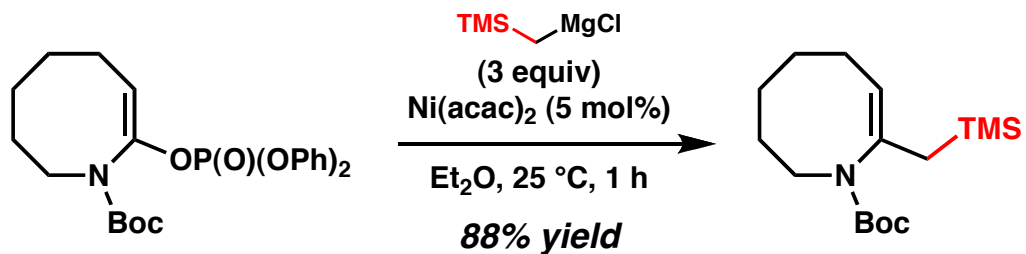
^a Ni(acac)_2 . ^b NiBr_2

Claesson, A.



^a $\text{NiCl}_2(\text{dppe})$.

Nicolaou, K. C.



Kumada, M. *Synthesis* **1981**, 1001.

Claesson, M. *Tetrahedron Lett.* **1983**, 24, 5137.

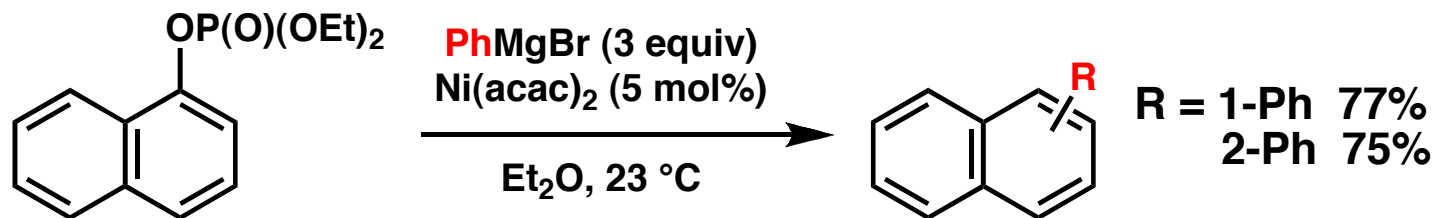
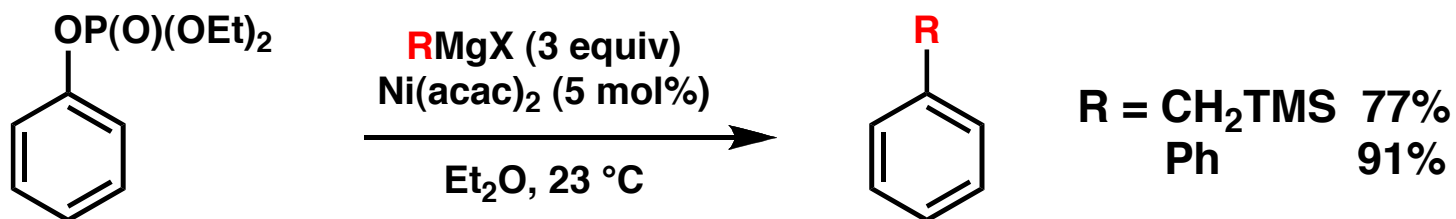
Nicolaou, K. C. *Chem. Commun.* **1998**, 1757.

Backvall, J.-E. *J. Org. Chem.* **1999**, 64, 1745.

Phosphonate and Phosphoramidate Cross Coupling

Kumada Coupling: Aryl Phosphonates

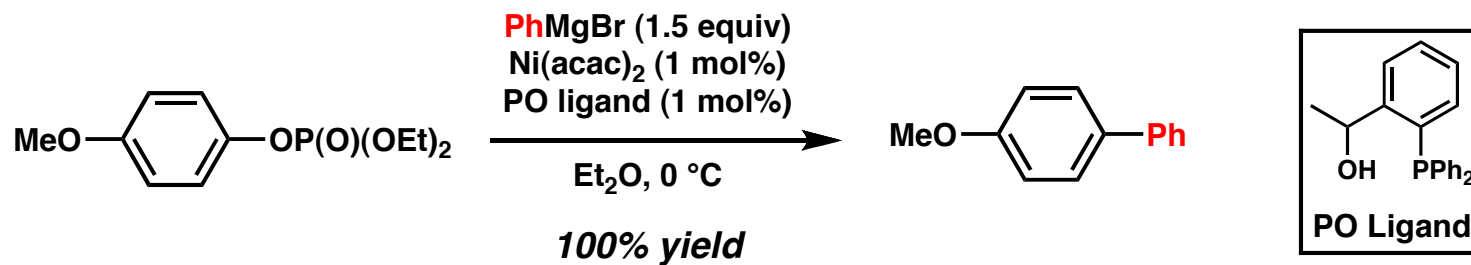
Kumada, M.



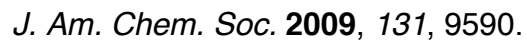
Phosphonate and Phosphoramidate Cross Coupling

Hydroxyphosphine Promoted Kumada Coupling

Nakamura, E.

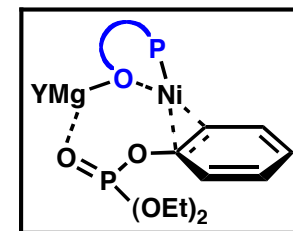


Nakamura, E.

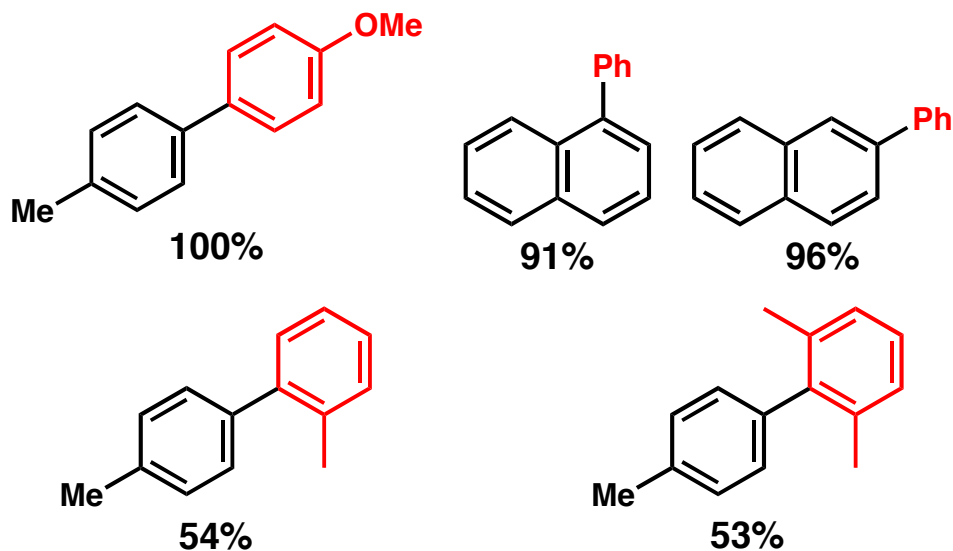
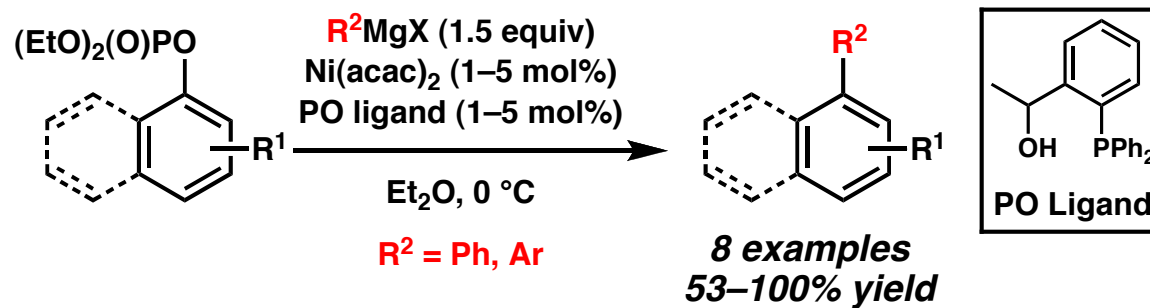


Phosphonate and Phosphoramidate Cross Coupling

Hydroxyphosphine Promoted Kumada Coupling



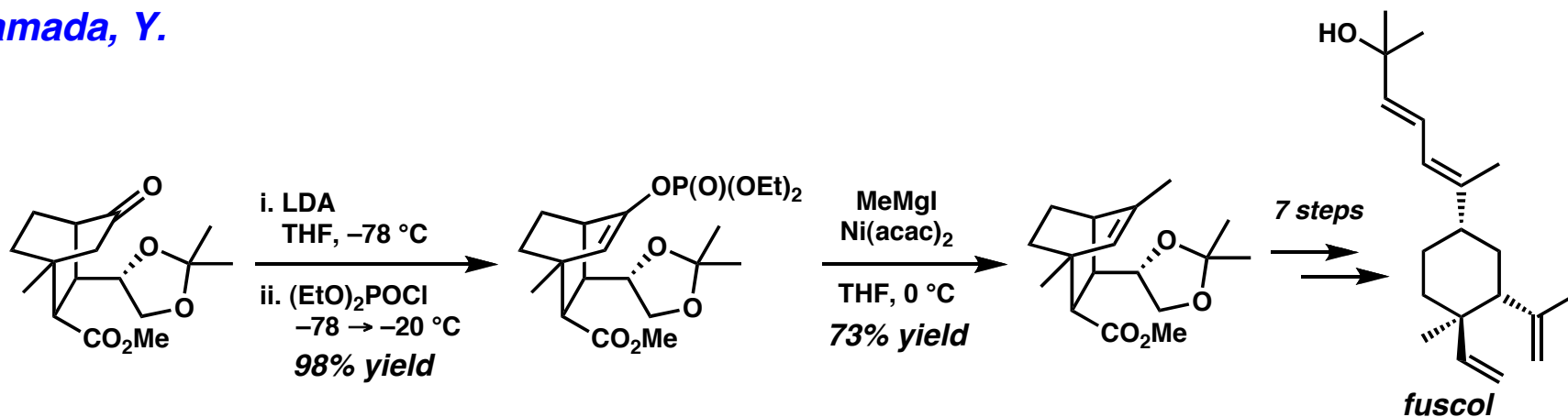
Nakamura, E.



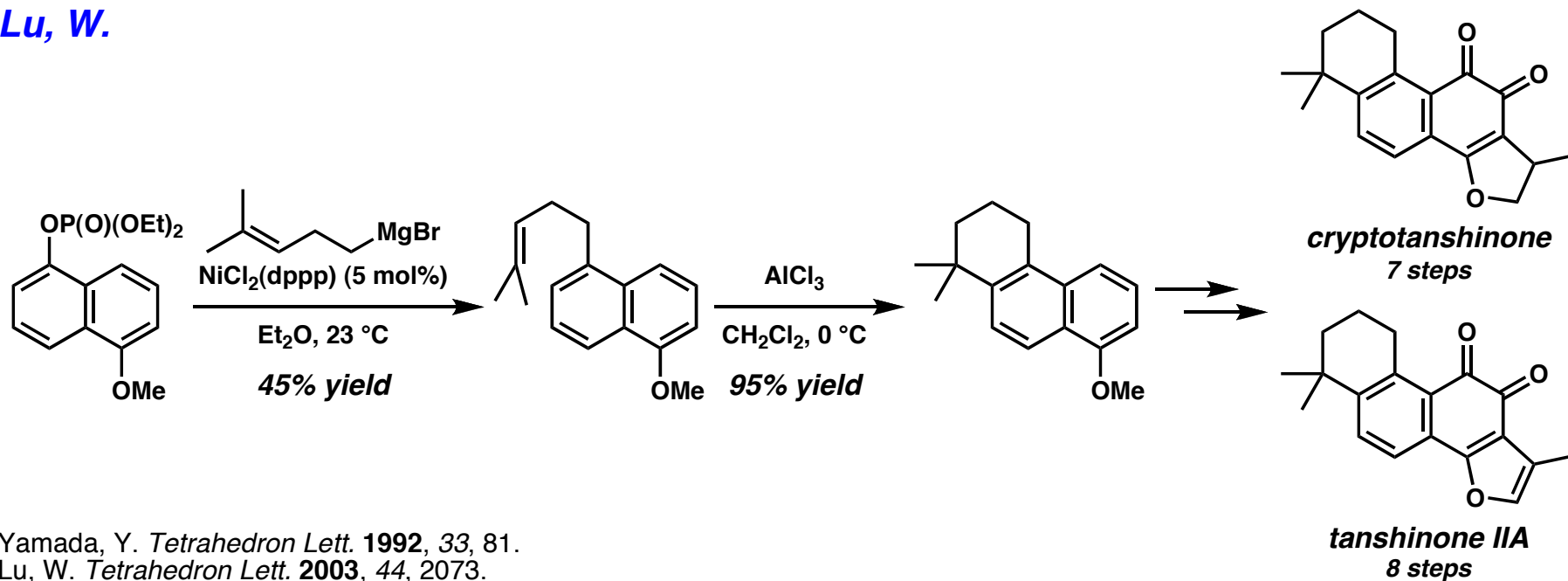
Phosphonate and Phosphoramidate Cross Coupling

Kumada Coupling: Total Synthesis

Yamada, Y.



Lu, W.



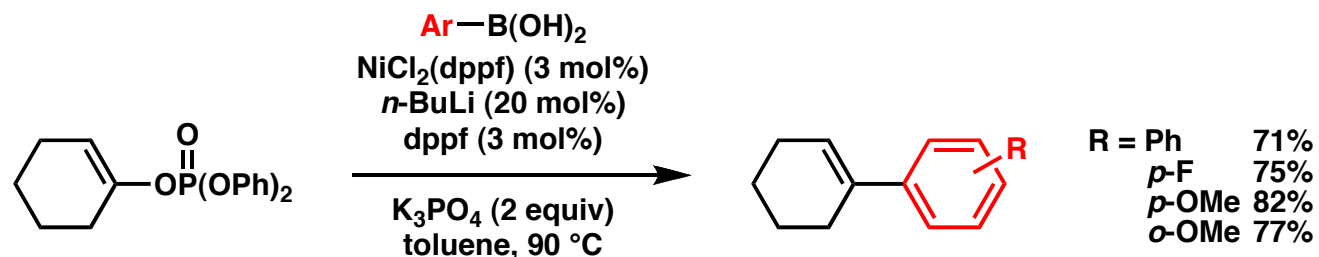
Yamada, Y. *Tetrahedron Lett.* **1992**, 33, 81.

Lu, W. *Tetrahedron Lett.* **2003**, 44, 2073.

Phosphonate and Phosphoramidate Cross Coupling

Suzuki-Miyaura Coupling: Vinyl Phosphonates

Yang, Z.



Yang, Z. *Tetrahedron Lett.* **1999**, 40, 3321.

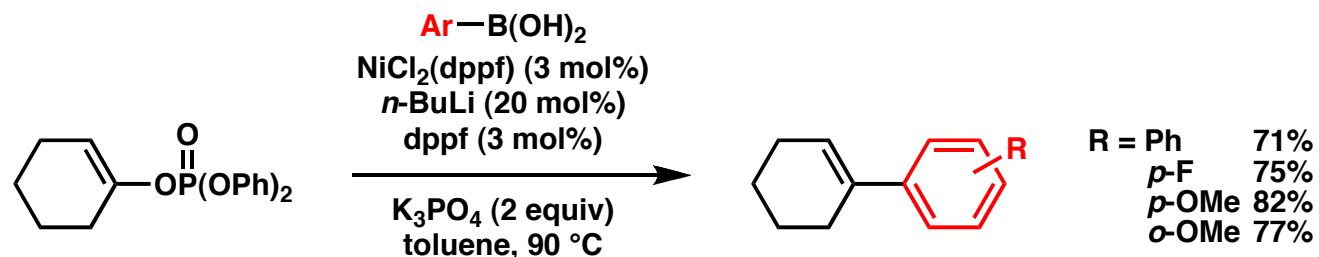
Yamada, Y. *Chem. Commun.* **2006**, 4137.

Yamada, Y. *J. Org. Chem.* **2007**, 72, 6465.

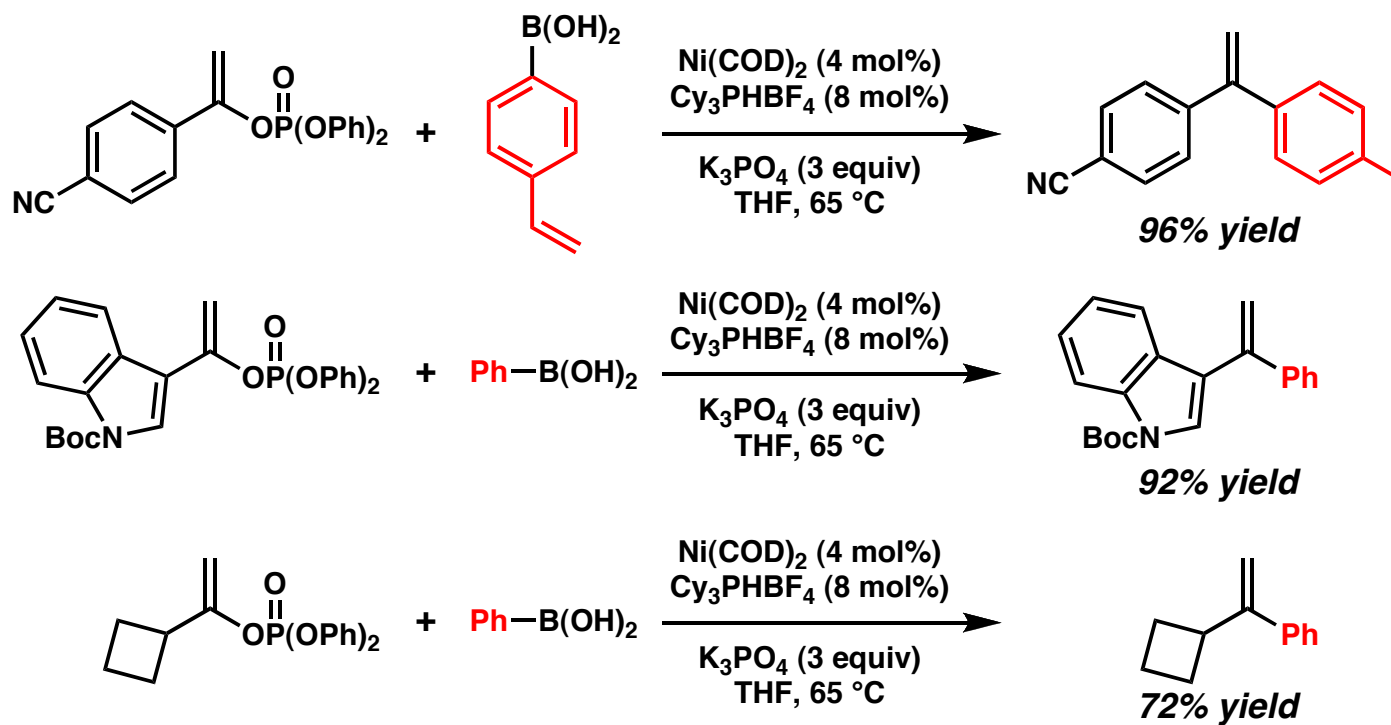
Phosphonate and Phosphoramidate Cross Coupling

Suzuki-Miyaura Coupling: Vinyl Phosphonates

Yang, Z.



Yamada, Y.



Yang, Z. *Tetrahedron Lett.* **1999**, 40, 3321.

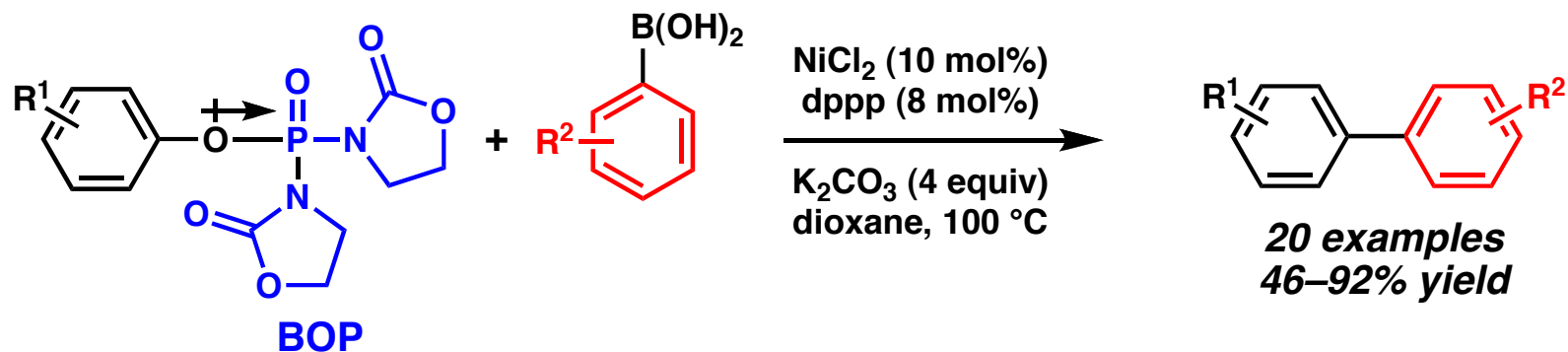
Yamada, Y. *Chem. Commun.* **2006**, 4137.

Yamada, Y. *J. Org. Chem.* **2007**, 72, 6465.

Phosphonate and Phosphoramidate Cross Coupling

Suzuki-Miyaura Coupling: Aryl Phosphoramidates

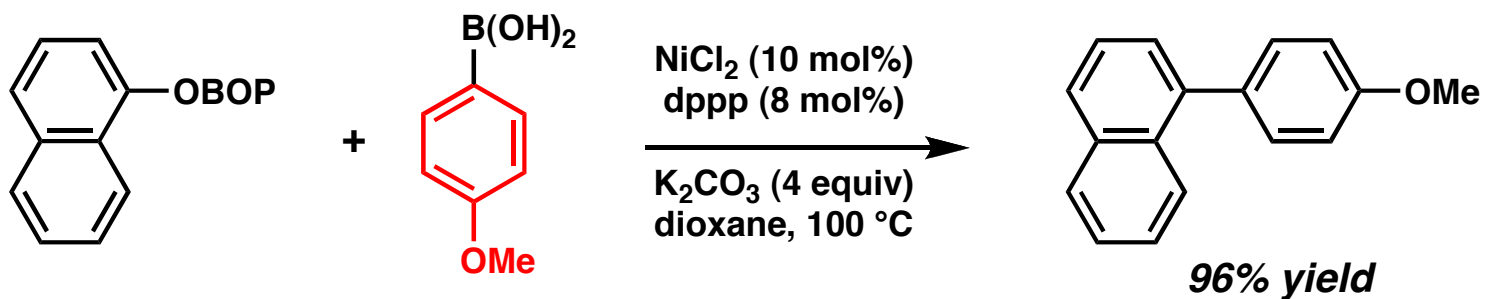
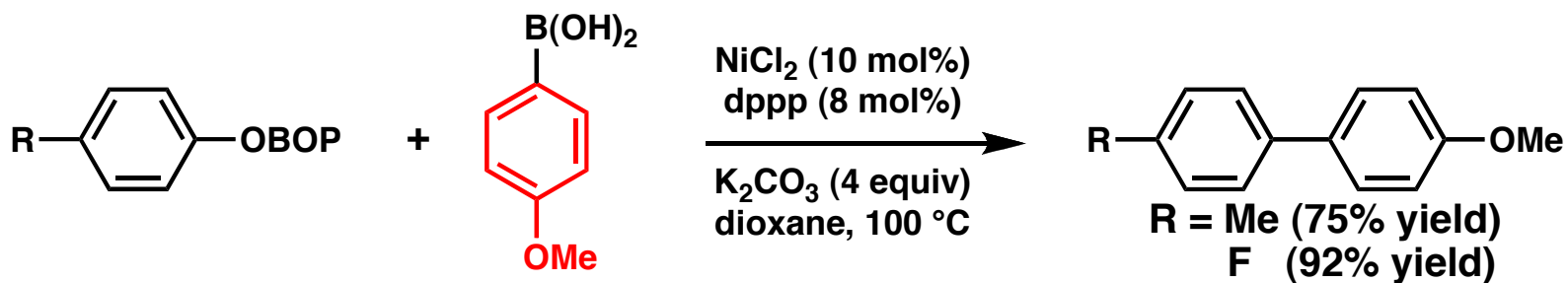
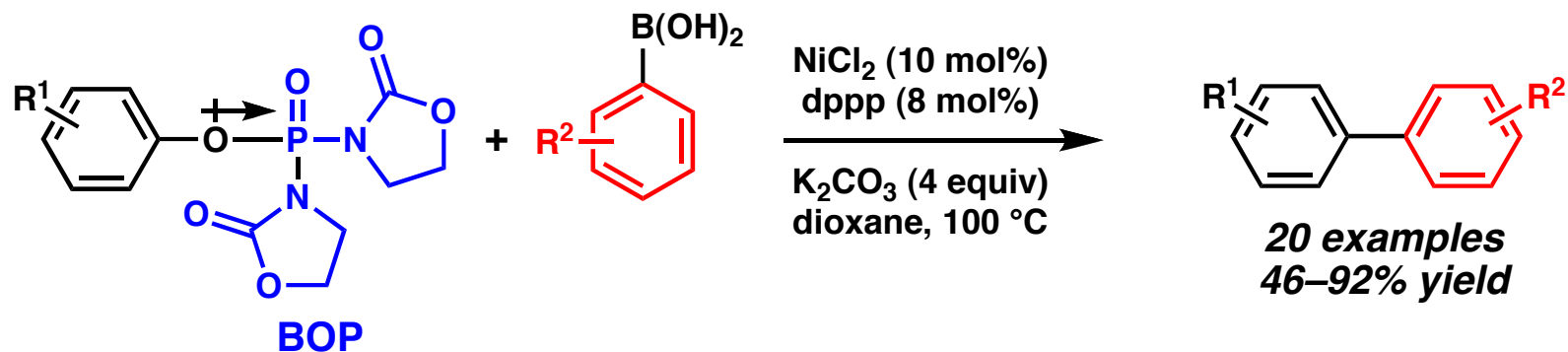
Han, F.-S.



Phosphonate and Phosphoramidate Cross Coupling

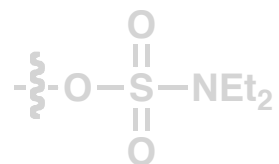
Suzuki-Miyaura Coupling: Aryl Phosphoramides

Han, F.-S.



Overview

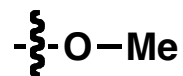
Sulfamates



Phosphonates and Phosphoramides



Ethers



Esters



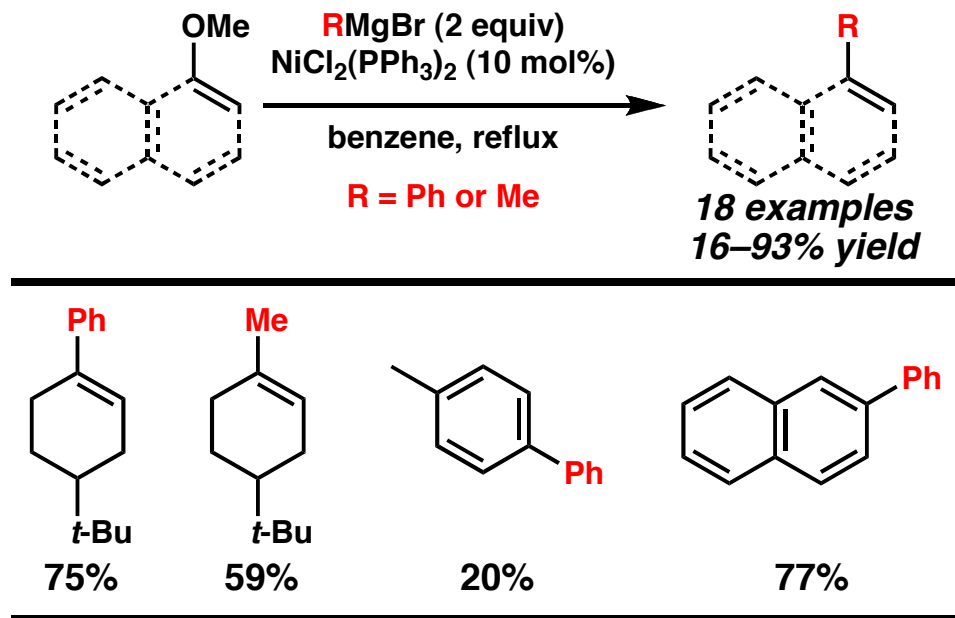
Carbamates and Carbonates



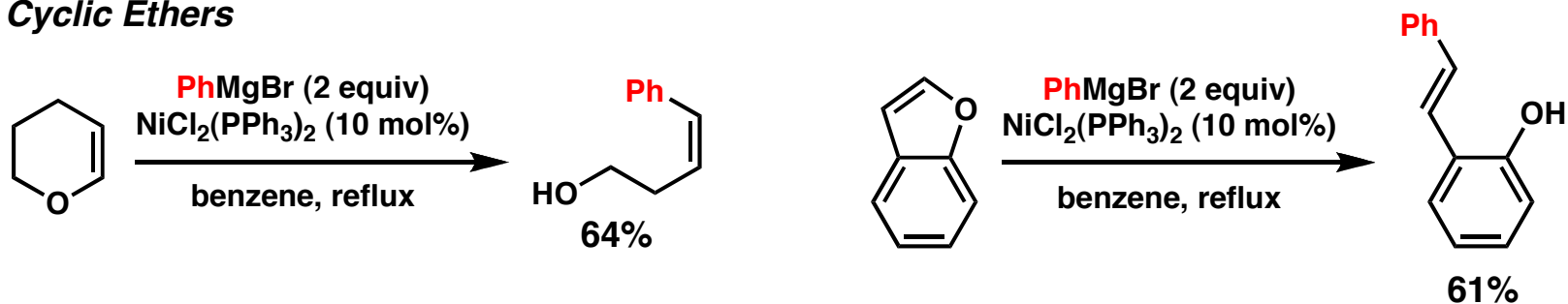
Ether Cross Coupling

Kumada Coupling

Wenkert, E.



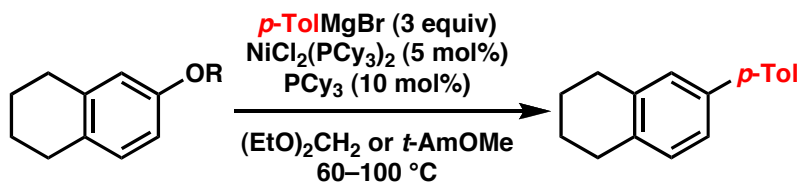
Cyclic Ethers



Ether Cross Coupling

Kumada Coupling: Aryl and Benzyl Ether Screen

Dankwardt, J. W.



R	GC yield	R	GC yield
-Me	94%	-MOM	92%
-(CH ₂) ₂ OMe	77%	-TMS	70%
-(CH ₂) ₂ NMe ₂	99%		

Aryl Ethers: Dankwardt, J. W. *Angew. Chem., Int. Ed.* **2004**, 43, 2428.

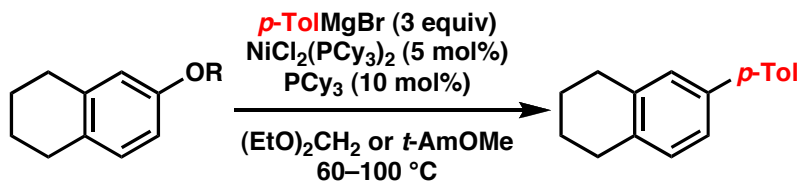
Aryl Ethers: Shi, Z.-J. *Chem. Commun.* **2008**, 1437.

Benzyl Ethers: Shi, Z.-J. *J. Am. Chem. Soc.* **2008**, 130, 3268.

Ether Cross Coupling

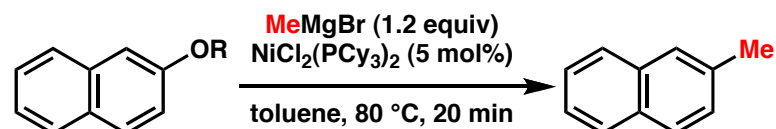
Kumada Coupling: Aryl and Benzyl Ether Screen

Dankwardt, J. W.



R	GC yield	R	GC yield
-Me	94%	-MOM	92%
-(CH ₂) ₂ OMe	77%	-TMS	70%
-(CH ₂) ₂ NMe ₂	99%		

Shi, Z.-J.



R	yield	R	yield
-Me	92%	- <i>t</i> -Bu	88%
-Et	92%	-MOM	69%
- <i>n</i> -Bu	96%	-TMS	95%
- <i>i</i> -Bu	91%	-Ph	95%

Aryl Ethers: Dankwardt, J. W. *Angew. Chem., Int. Ed.* **2004**, 43, 2428.

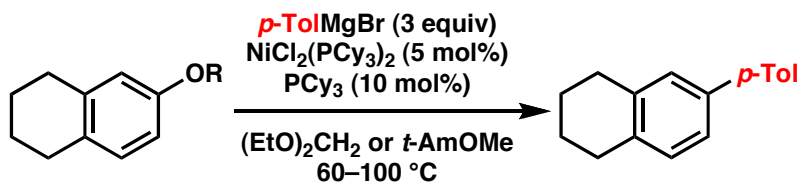
Aryl Ethers: Shi, Z.-J. *Chem. Commun.* **2008**, 1437.

Benzyl Ethers: Shi, Z.-J. *J. Am. Chem. Soc.* **2008**, 130, 3268.

Ether Cross Coupling

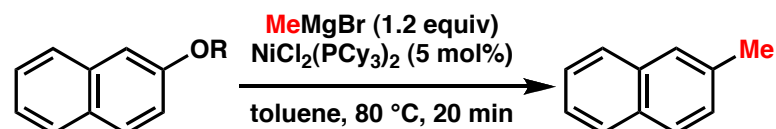
Kumada Coupling: Aryl and Benzyl Ether Screen

Dankwardt, J. W.



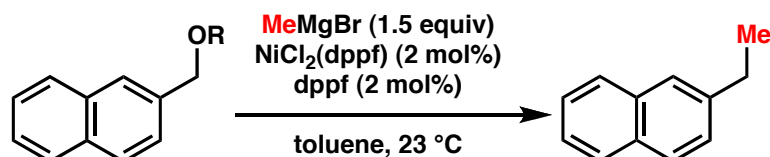
R	GC yield	R	GC yield
-Me	94%	-MOM	92%
-(CH ₂) ₂ OMe	77%	-TMS	70%
-(CH ₂) ₂ NMe ₂	99%		

Shi, Z.-J.



R	yield	R	yield
-Me	92%	- <i>t</i> -Bu	88%
-Et	92%	-MOM	69%
- <i>n</i> -Bu	96%	-TMS	95%
- <i>i</i> -Bu	91%	-Ph	95%

Shi, Z.-J.



R	yield	R	yield
-Me	80%	-Ph	75%
-Et	73%	-TMS	76%
- <i>t</i> -Bu	79%	-Ac	18% ^a

^a GC yield

Aryl Ethers: Dankwardt, J. W. *Angew. Chem., Int. Ed.* **2004**, 43, 2428.

Aryl Ethers: Shi, Z.-J. *Chem. Commun.* **2008**, 1437.

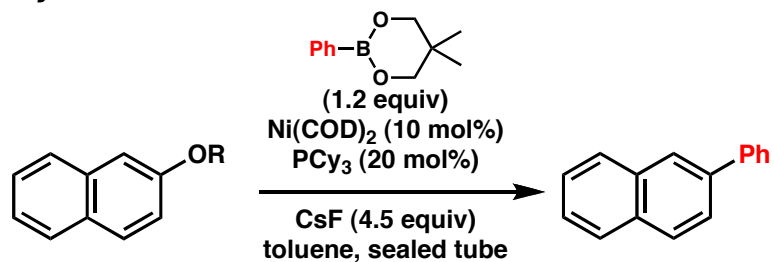
Benzyl Ethers: Shi, Z.-J. *J. Am. Chem. Soc.* **2008**, 130, 3268.

Ether Cross Coupling

Suzuki-Miyaura

Chatani, N.

Aryl Ether Screen



R	yield	R	yield
-Me	93%	- <i>i</i> -Pr	16%
-Et	74%	-Ac	10%

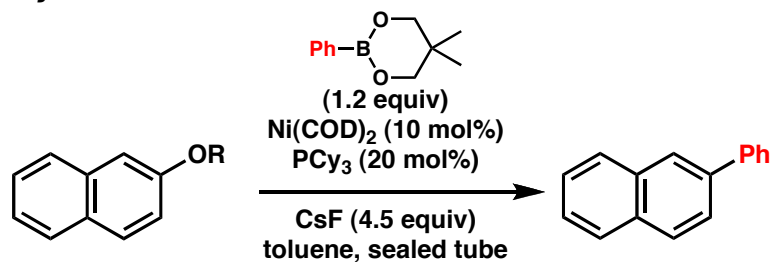
Aromatic Ethers: *Angew. Chem., Int. Ed.* **2008**, 47, 4866.
Vinyl Ethers: *Org. Lett.* **2009**, 11, 4890.

Ether Cross Coupling

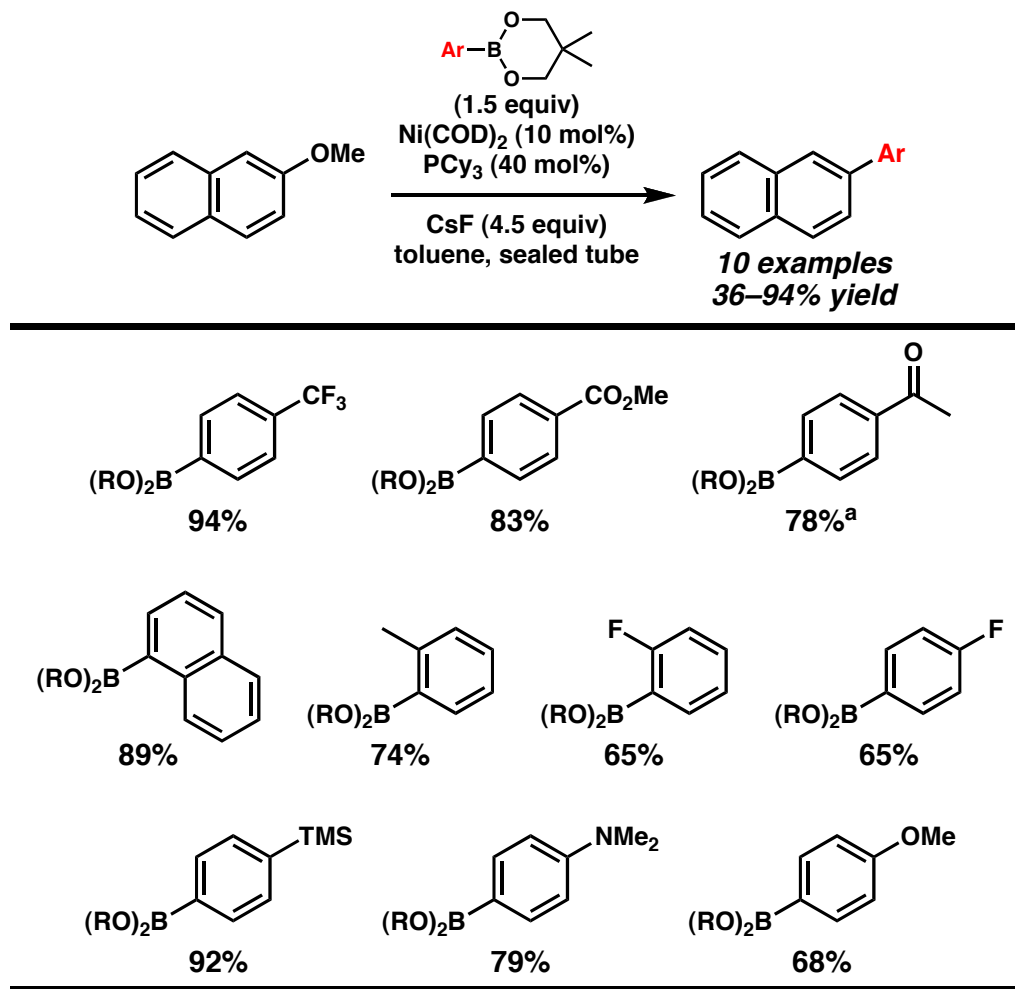
Suzuki-Miyaura

Chatani, N.

Aryl Ether Screen



R	yield	R	yield
-Me	93%	-i-Pr	16%
-Et	74%	-Ac	10%



^a 20 mol% Ni(COD)₂, 80 mol% PCy₃.

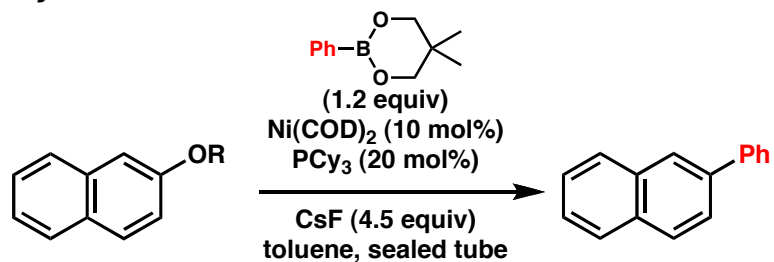
Aromatic Ethers: *Angew. Chem., Int. Ed.* **2008**, 47, 4866.
 Vinyl Ethers: *Org. Lett.* **2009**, 11, 4890.

Ether Cross Coupling

Suzuki-Miyaura

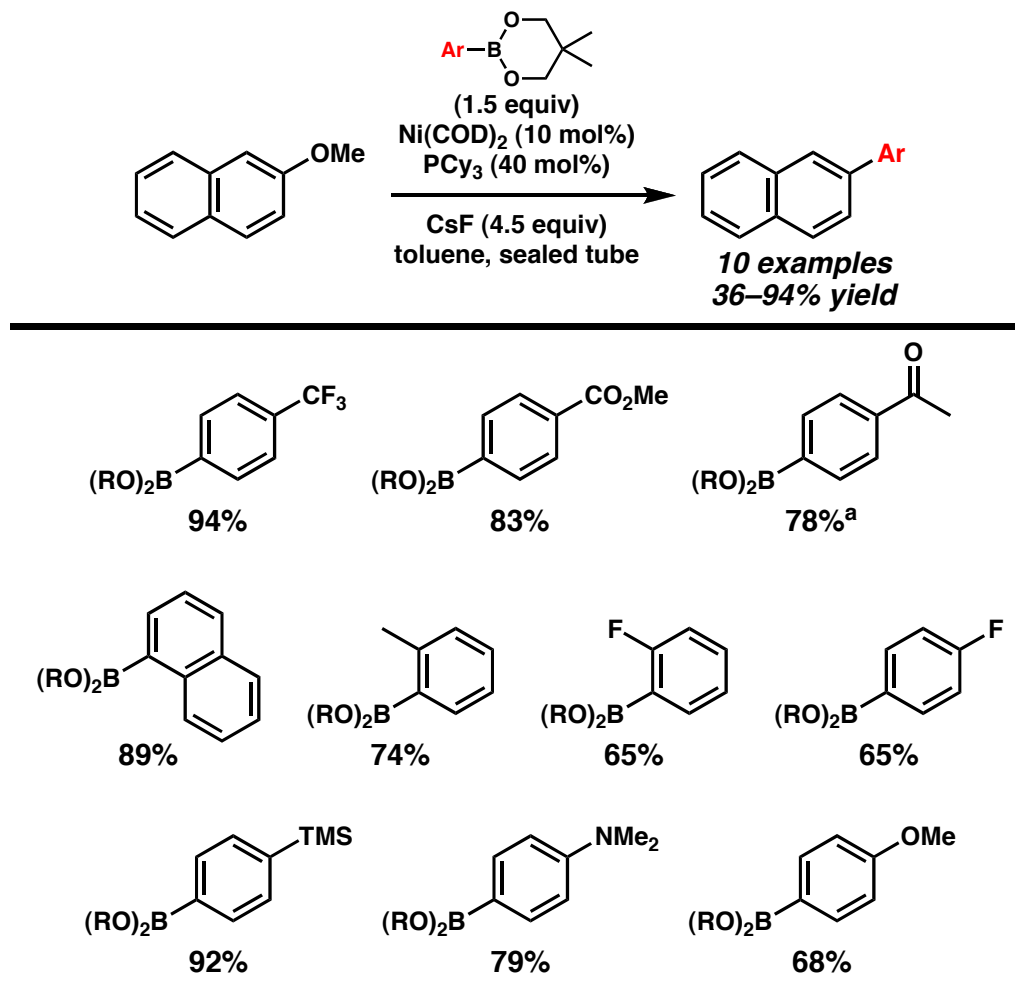
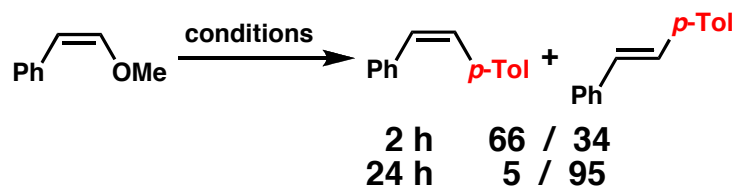
Chatani, N.

Aryl Ether Screen



R	yield	R	yield
-Me	93%	-i-Pr	16%
-Et	74%	-Ac	10%

Vinyl Ethers



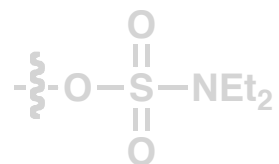
^a 20 mol% Ni(COD)_2 , 80 mol% PCy_3 .

Aromatic Ethers: *Angew. Chem., Int. Ed.* **2008**, 47, 4866.

Vinyl Ethers: *Org. Lett.* **2009**, 11, 4890.

Overview

Sulfamates



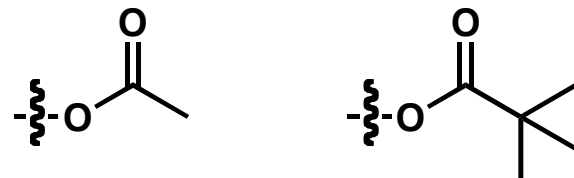
Phosphonates and Phosphoramides



Ethers



Esters



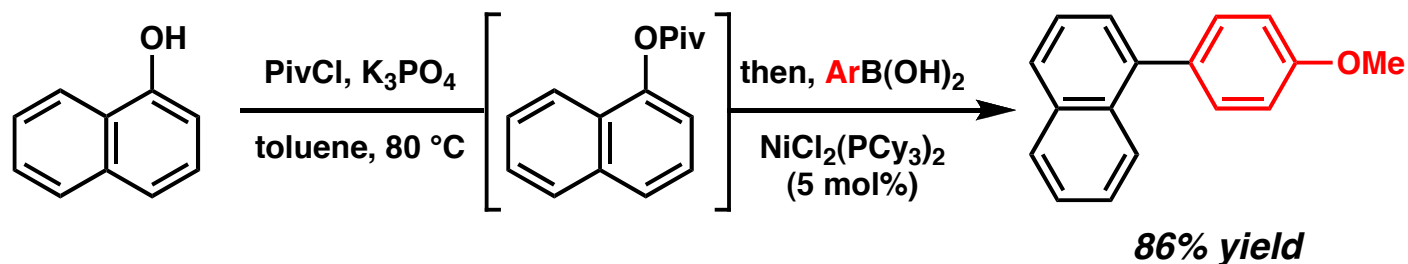
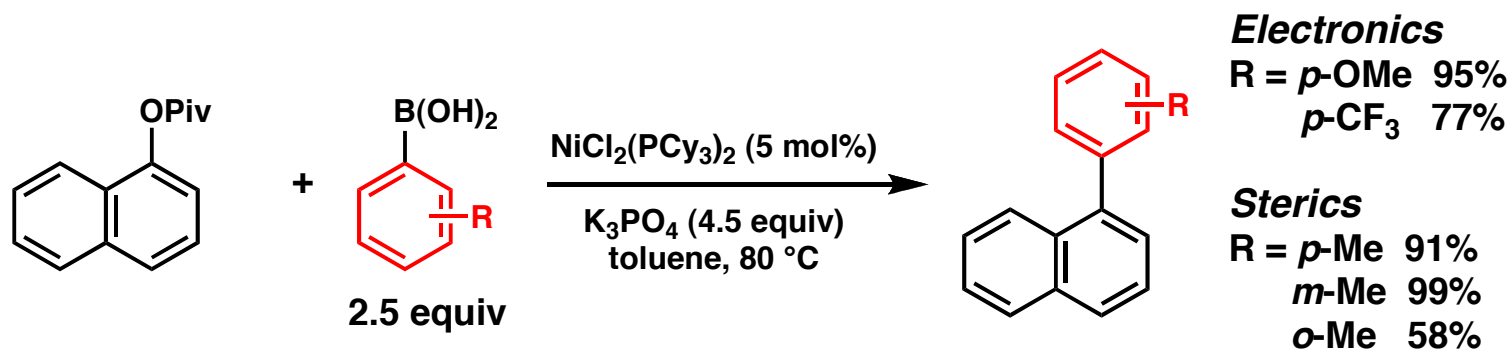
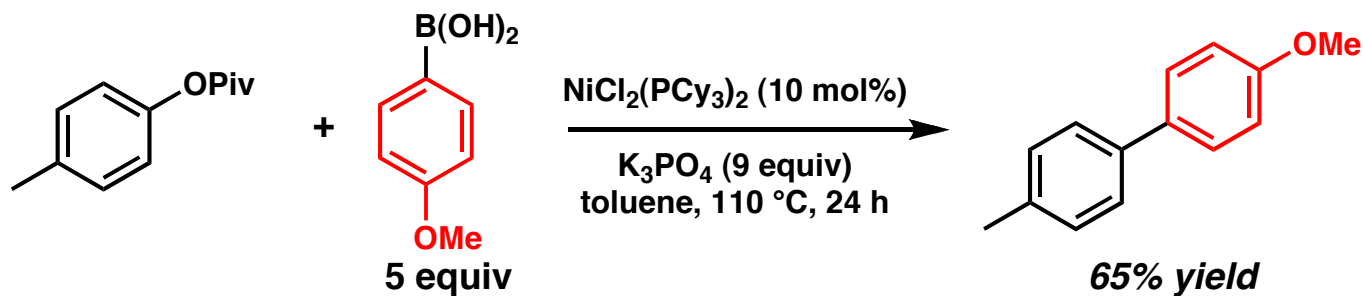
Carbamates and Carbonates



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

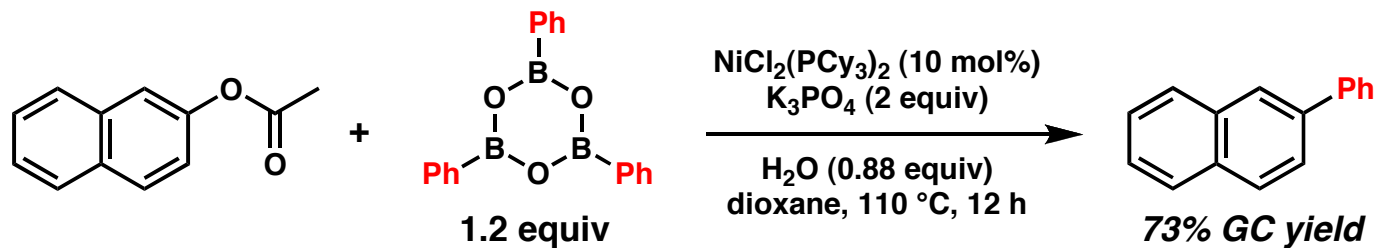
Garg, N. K.



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

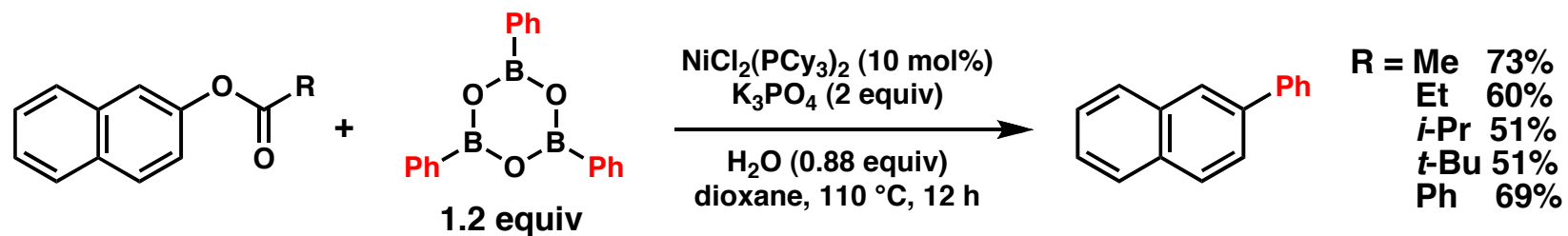
Shi, Z.-J.



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

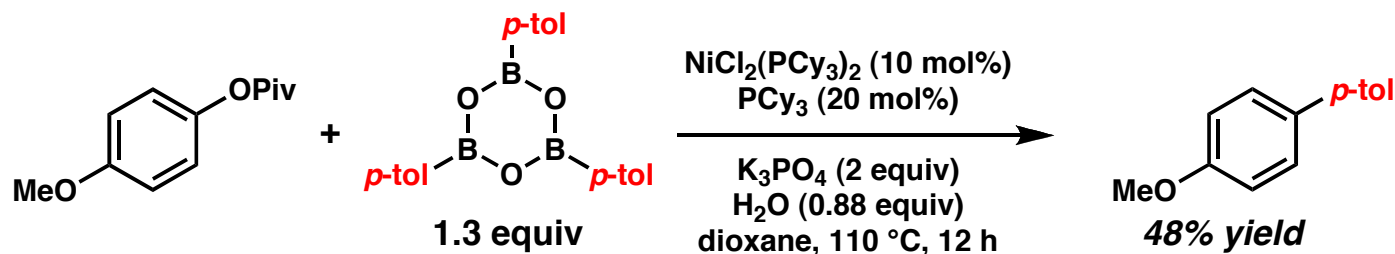
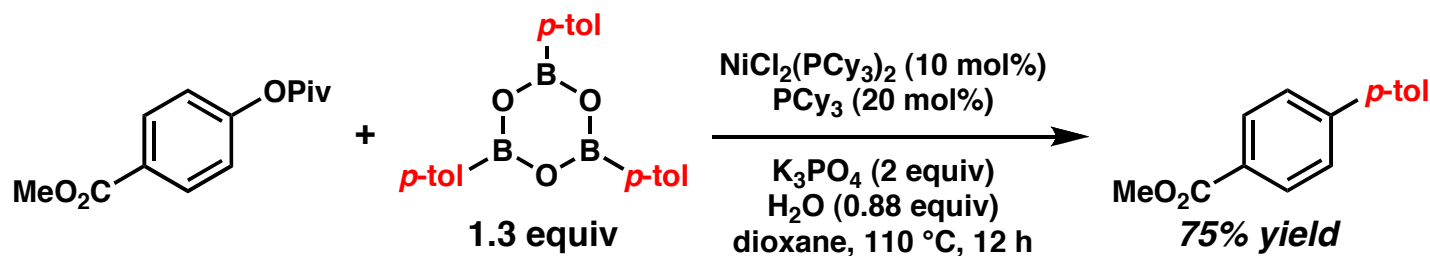
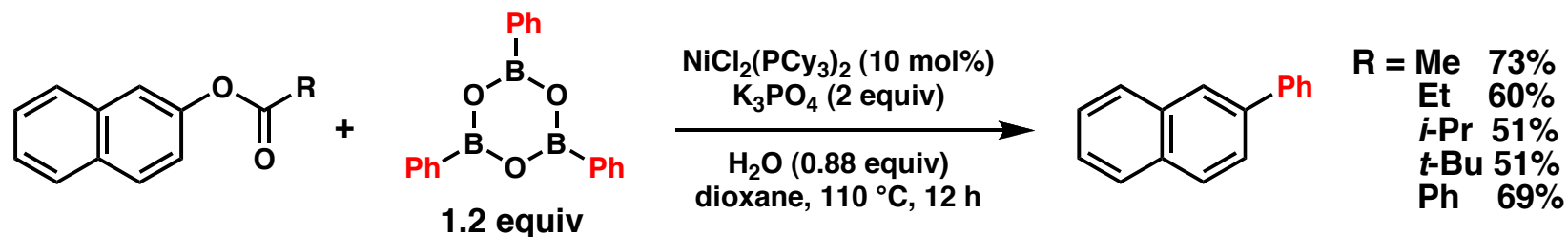
Shi, Z.-J.



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

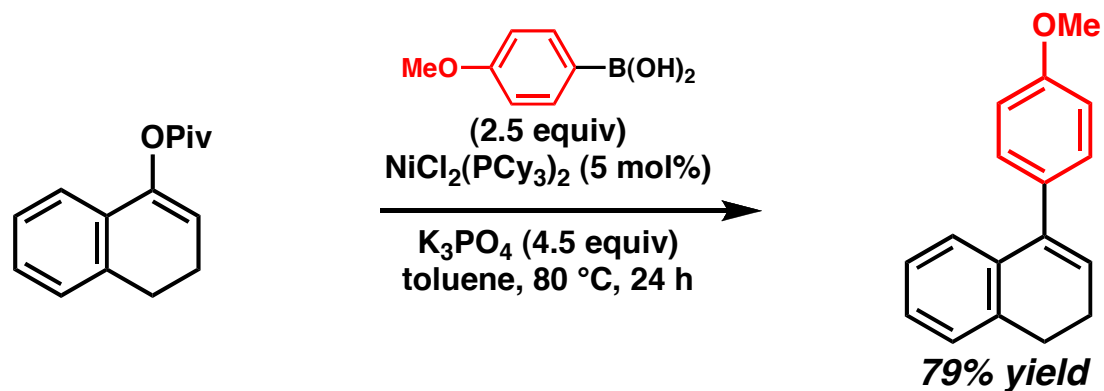
Shi, Z.-J.



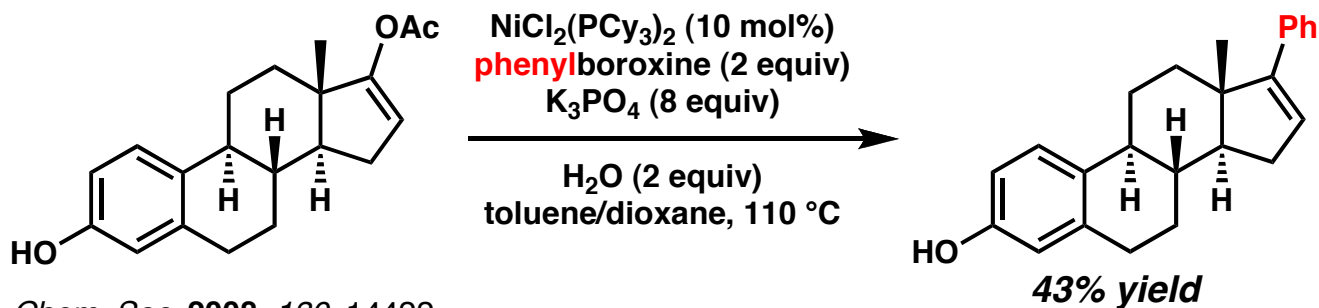
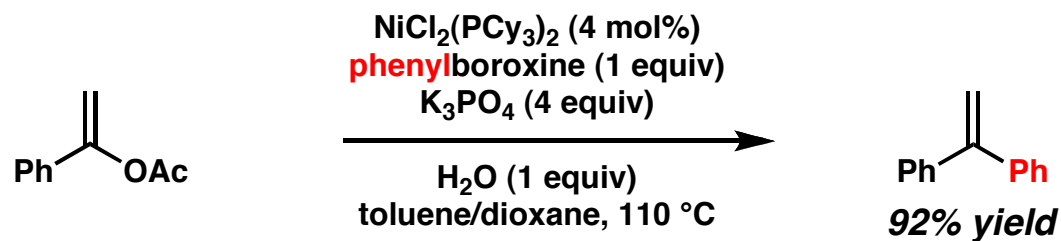
Ester Cross Coupling

Suzuki-Miyaura Coupling: Vinyl Esters

Garg, N. K.



Shi, Z.-J.



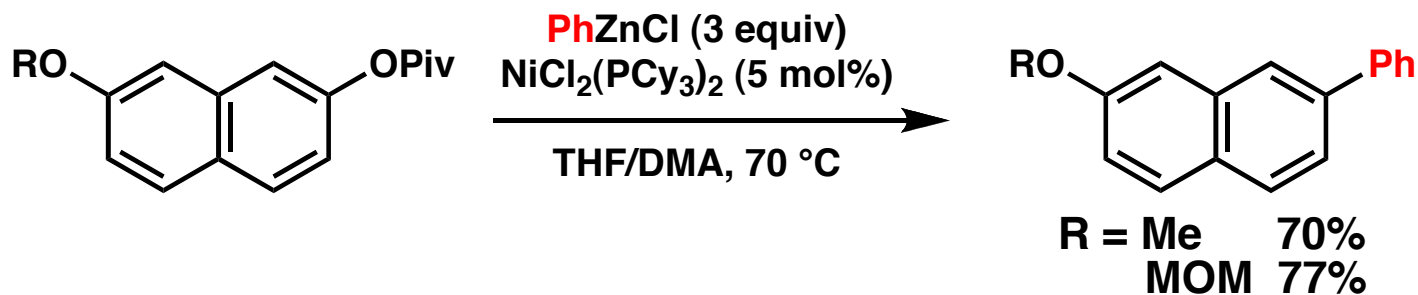
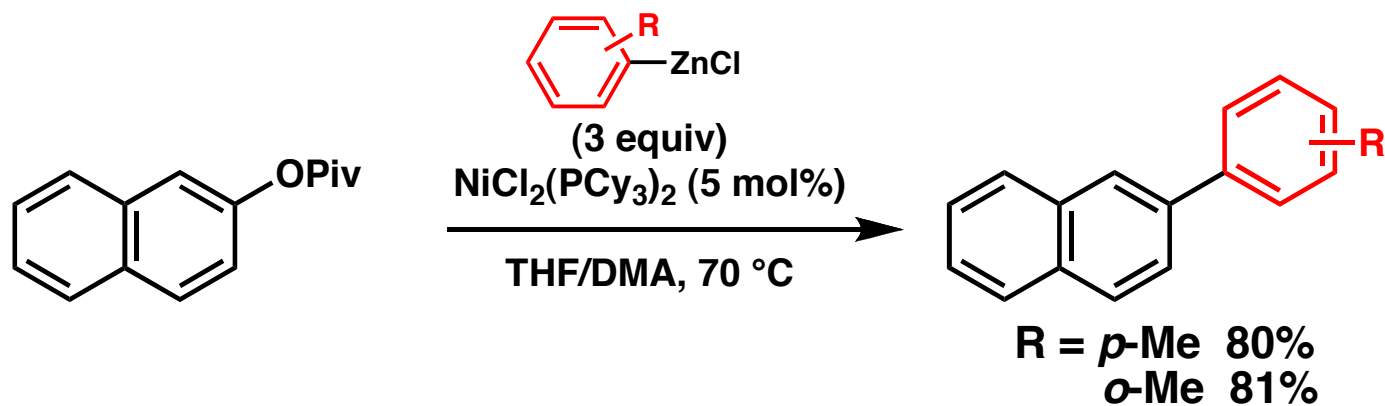
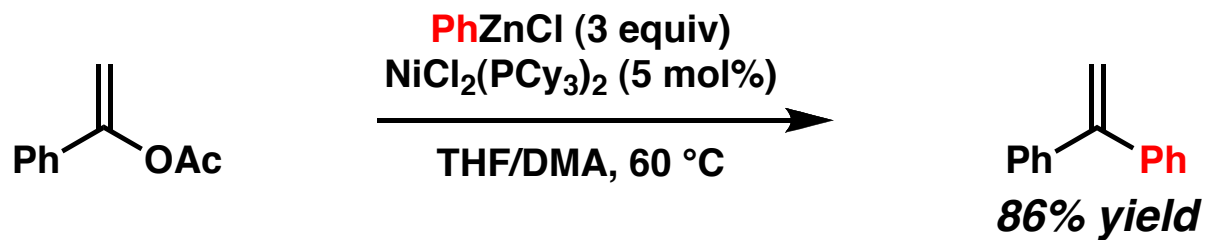
Garg, N. *J. Am. Chem. Soc.* **2008**, 130, 14422.

Shi, Z.-J. *Chem. Eur. J.* **2010**, 16, 5844.

Ester Cross Coupling

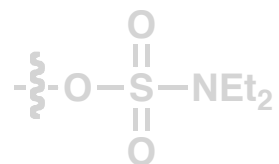
Negishi Coupling

Shi, Z.-J.

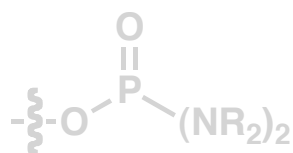


Overview

Sulfamates



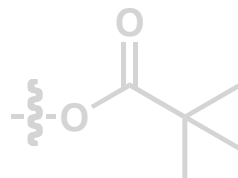
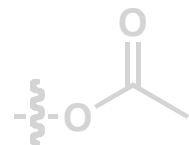
Phosphonates and Phosphoramides



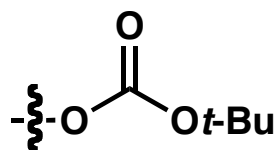
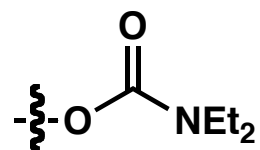
Ethers



Esters



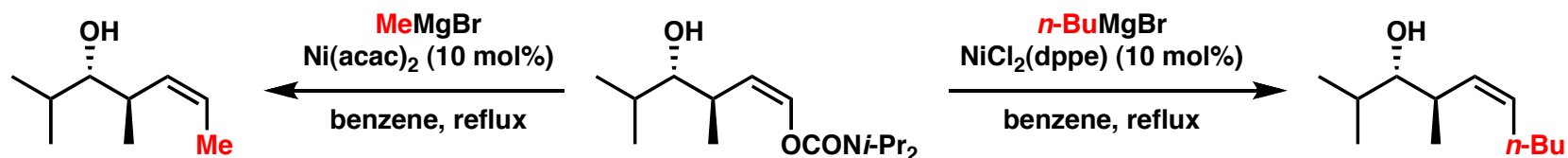
Carbamates and Carbonates



Carbamate and Carbonate Cross Coupling

Kumada Coupling: Vinyl Carbamate

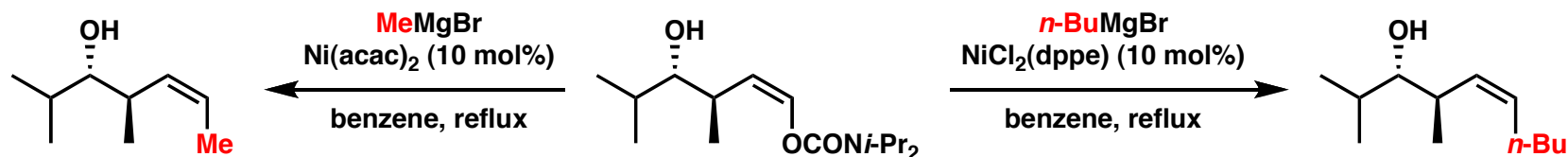
Kocienski, P.



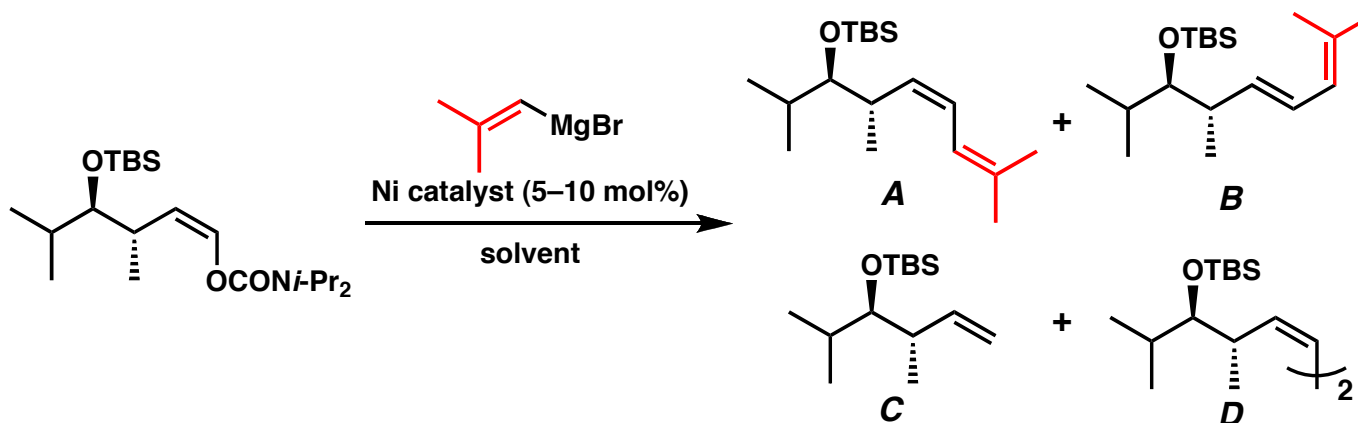
Carbamate and Carbonate Cross Coupling

Kumada Coupling: Vinyl Carbamate

Kocienski, P.



Betzer, J.-F.



Ni catalyst	solvent	yield	A	B	C	D
Ni(acac) ₂	Et ₂ O	57%	100	0	0	0
Ni(acac) ₂	toluene	63%	89	5	0	6
NiCl ₂ (dppp)	toluene	56%	89	7	4	0

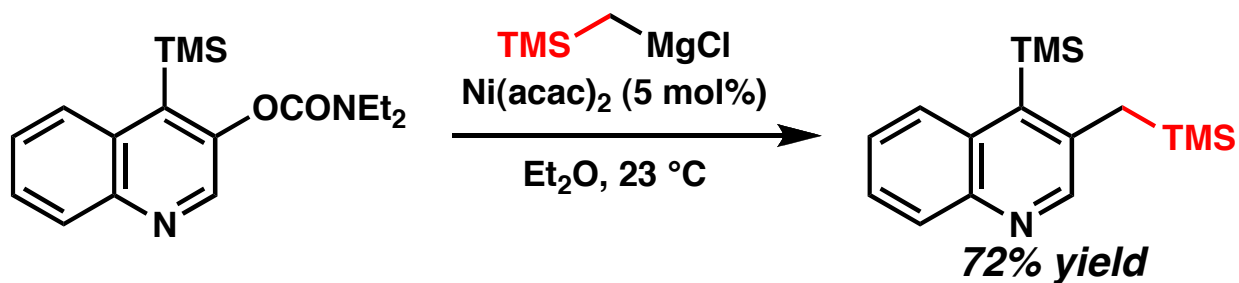
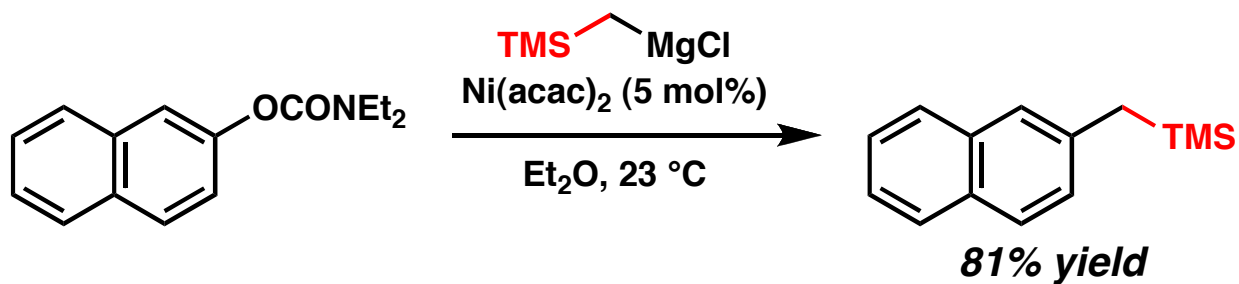
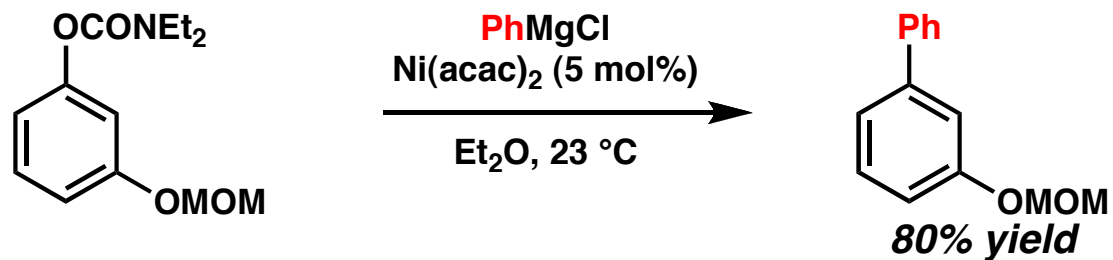
Kocienski, P. *Sylett* **1989**, 52.

Betzer, J.-F. *Tetrahedron Lett.* **2003**, 44, 7553.

Carbamate and Carbonate Cross Coupling

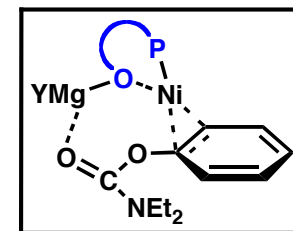
Kumada Coupling: Aryl Carbamate

Snieckus, V.

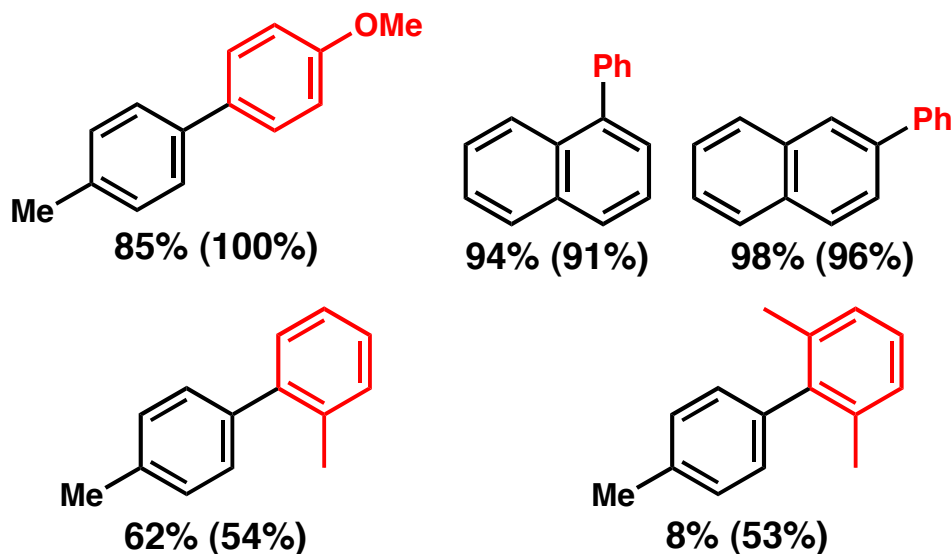
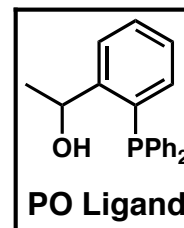
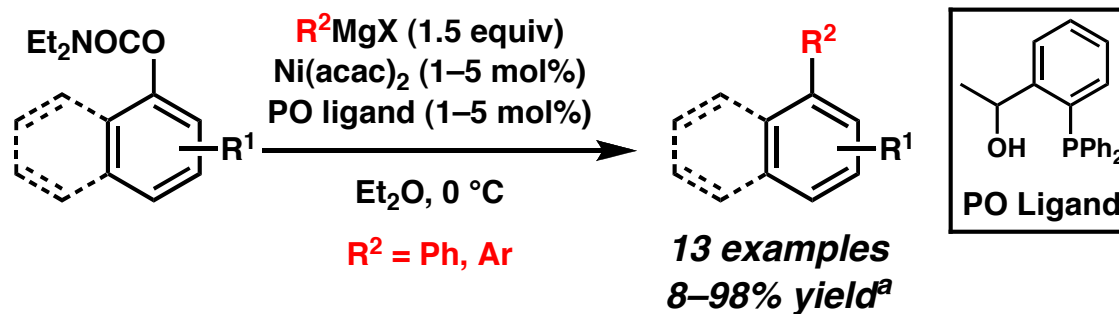


Carbamate and Carbonate Cross Coupling

Hydroxyphosphine Promoted Kumada Coupling



Nakamura, E.

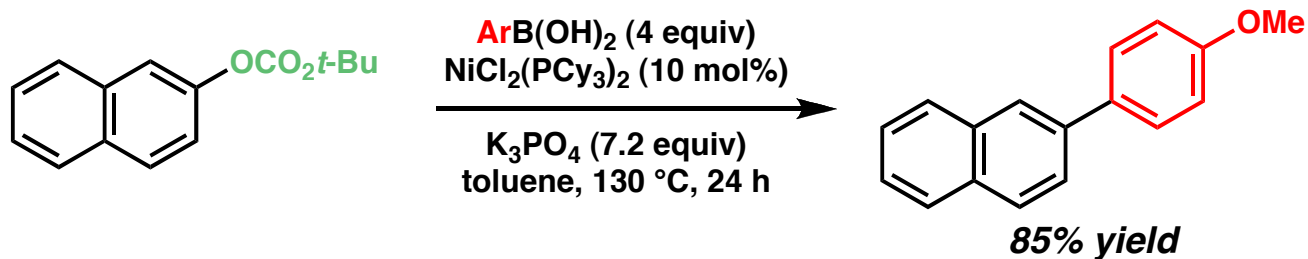
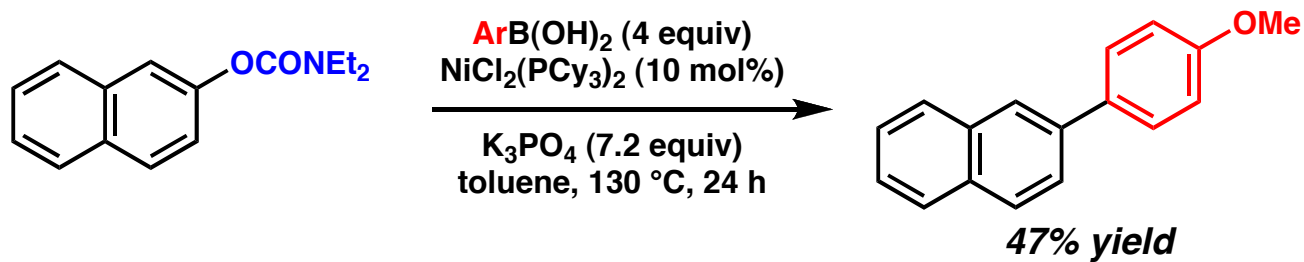
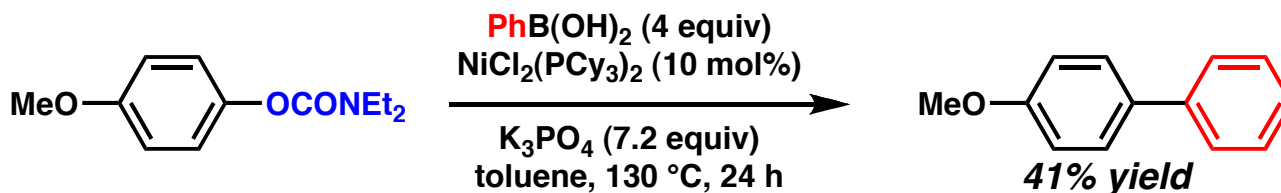


^a Phosphonate comparison in parentheses.

Carbamate and Carbonate Cross Coupling

Suzuki-Miyaura Coupling: Aryl Carbamates and Carbonates

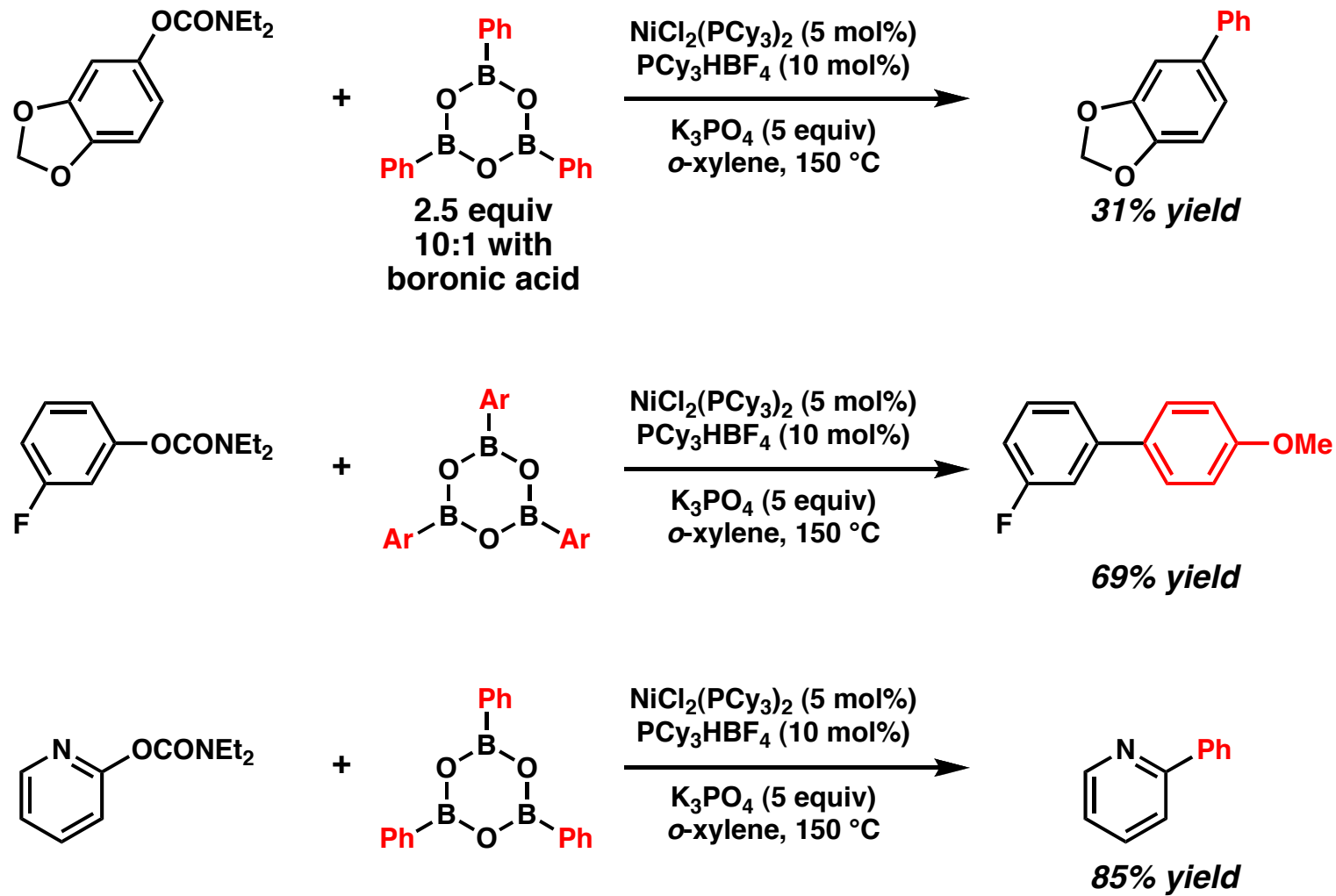
Garg, N. K.



Carbamate and Carbonate Cross Coupling

Suzuki-Miyaura Coupling: Aryl Carbamates

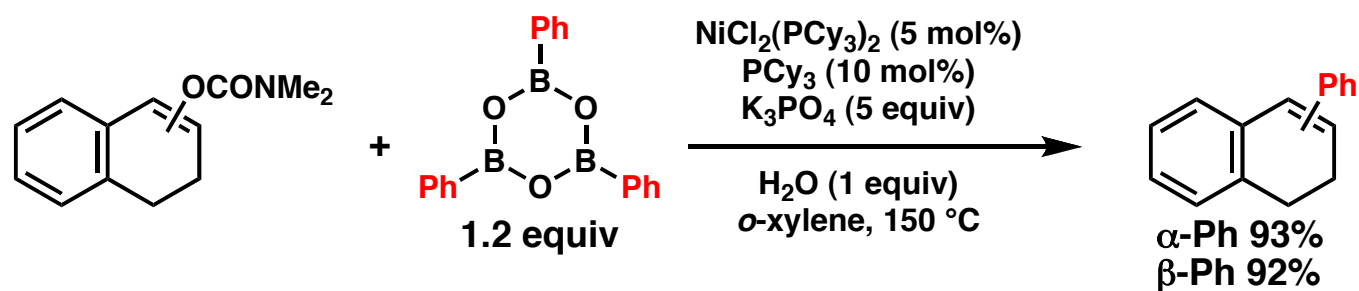
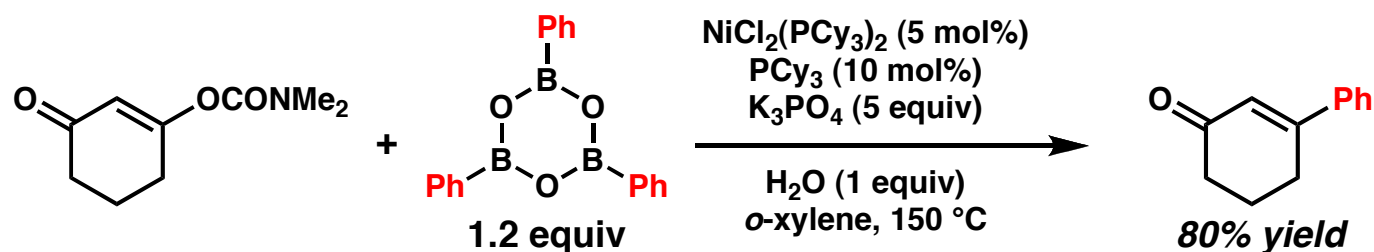
Snieckus, V.



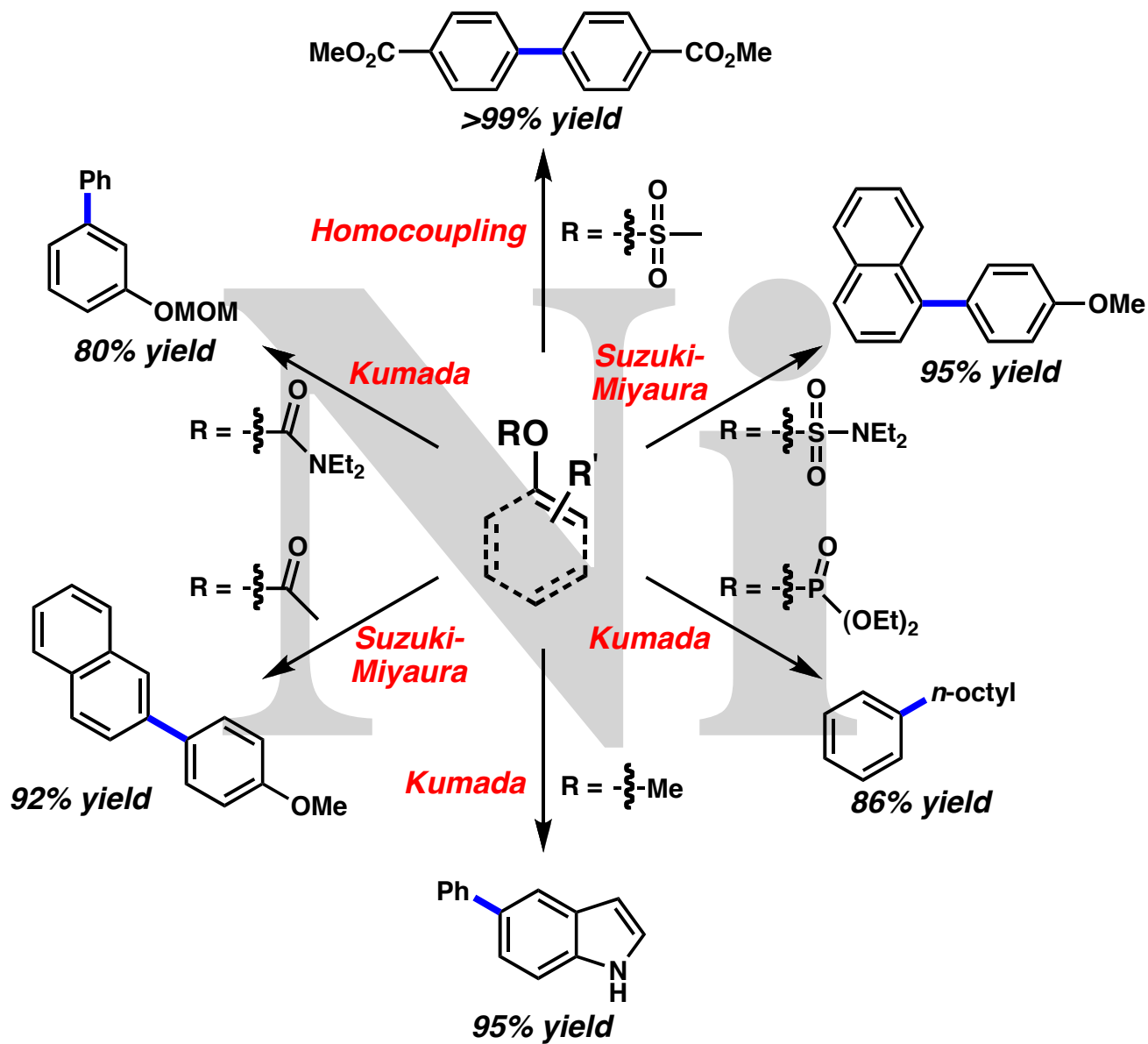
Carbamate and Carbonate Cross Coupling

Suzuki-Miyaura Coupling: Vinyl Carbamates

Shi, Z.-J.



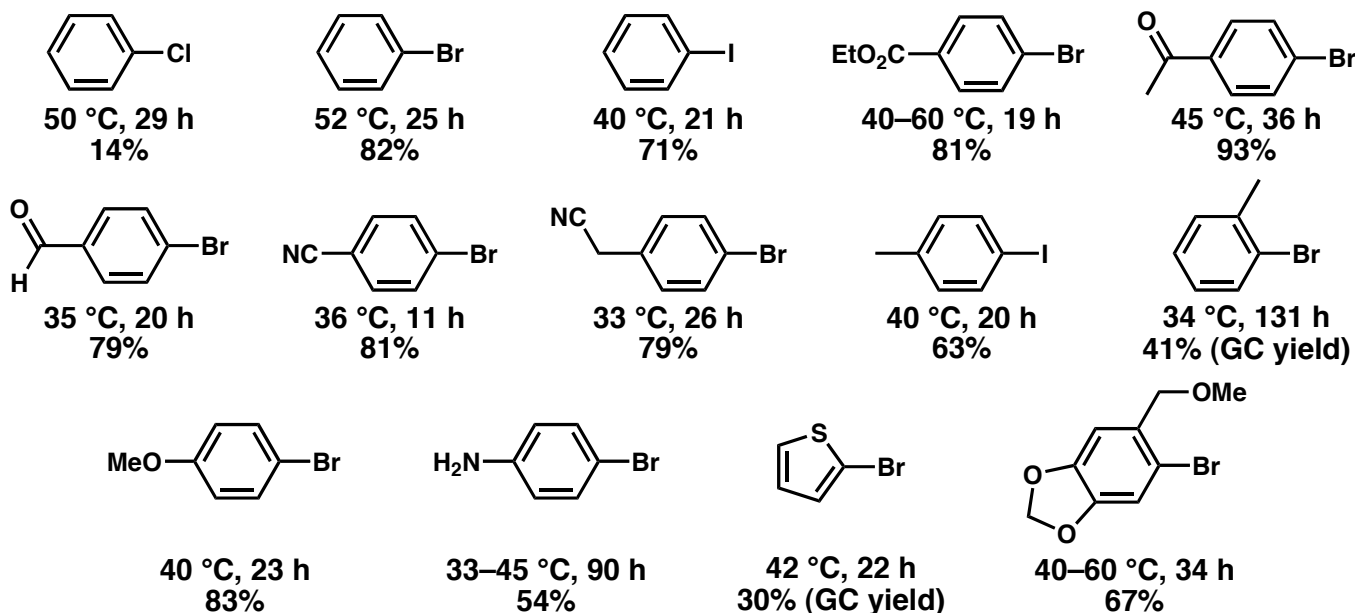
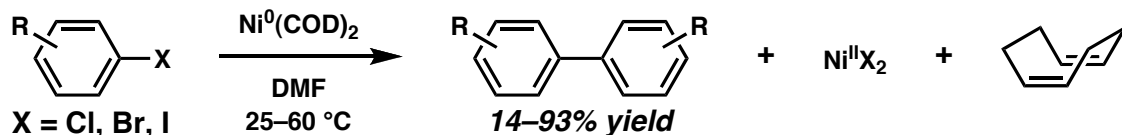
Summary



History of Nickel Coupling

First Ni Homocoupling

Semmelhack, M. F.



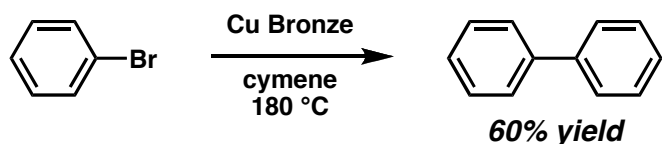
Reactivity

- I > Br > Cl
- EWG > EDG
- o-substitution not well tolerated
- -NO₂, -OH, -CO₂H not tolerated
- Reduction products with protic substrates

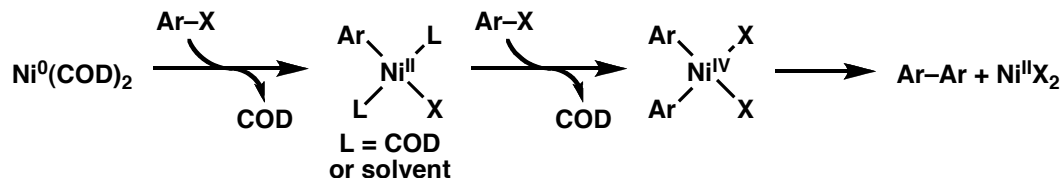
Limitations

- Stoichiometric Ni
- Substrate issues
- Ni decomposition @ high temp
- DMF only solvent

Ullmann Coupling Comparison



Proposed Mechanism

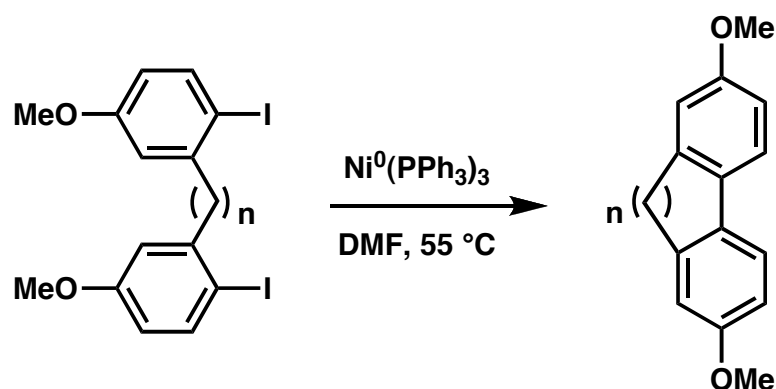


Ni: Semmelhack, M. F. *J. Am. Chem. Soc.* **1971**, 93, 5908.
Ullmann Coupling: Fanta, P. E. *Chem. Res.* **1964**, 64, 613.

History of Nickel Coupling

Transition to Other Ni Sources

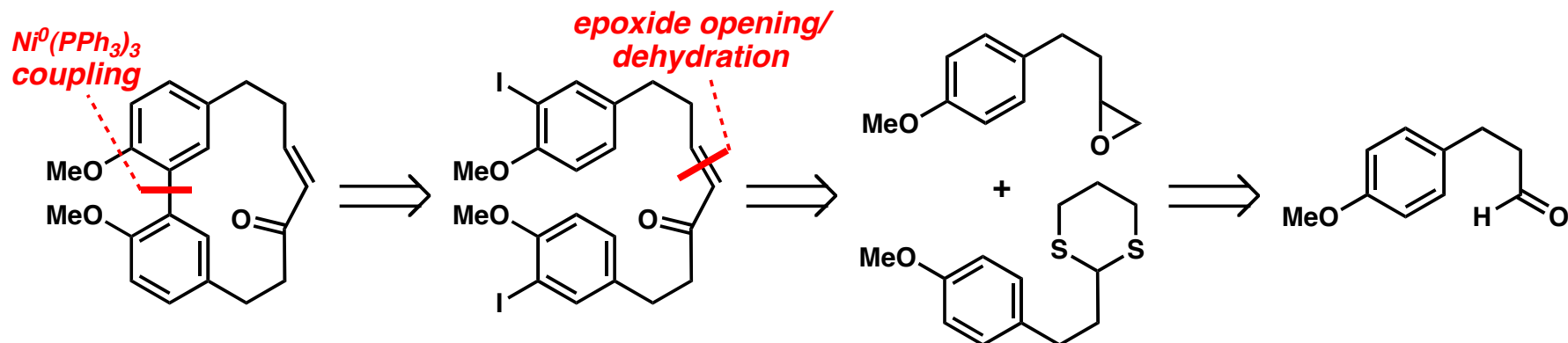
Semmelhack, M. F.



n	yield (%)
2	81
3	83
4	76
5	85
6	38

- Stiochiometric Ni
- $\text{Ni}^0(\text{PPh}_3)_3$ necessary
- $\text{Ni}^0(\text{COD})_2$ alone or with PPh_3 or PEt_3 resulted in no coupling

Synthesis of Dimethyl Ether Alnusone



Kende, A. S.



- Tolman Research: $\text{Ni}^0(\text{PPh}_3)_3$ active catalyst
- Kende formed *in situ*

Semmelhack, M. F. *J. Am. Chem. Soc.* **1975**, 97, 3873.
 Semmelhack, M. F. *J. Am. Chem. Soc.* **1981**, 103, 6460.
 Tolman, C. A. *J. Am. Chem. Soc.* **1972**, 94, 2669.

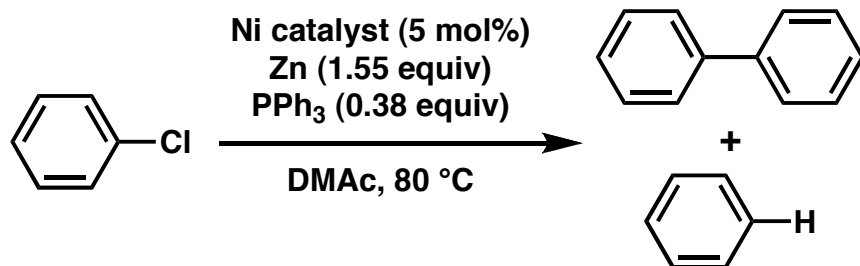
Tolman, C. A. *J. Am. Chem. Soc.* **1974**, 96, 53.
 Kende, A. S. *Tetrahedron Lett.* **1975**, 3375.

History of Nickel Coupling

Ni^{II} Catalyst and Reducing Metal Analysis

Colon, I.

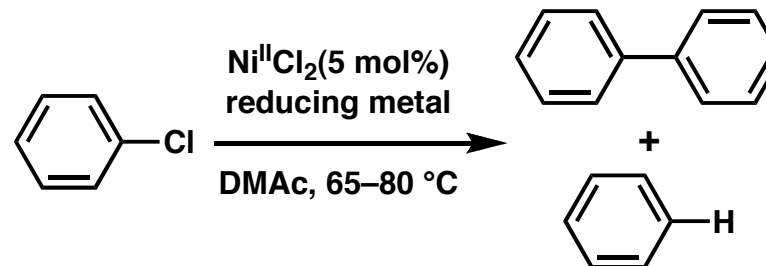
Ni Catalyst Screen



Ni catalyst	time (h)	GC yield (%)	
		biphenyl	benzene
NiCl_2	2	99	1
NiBr_2	2	99	1
$\text{NiI}_2 \cdot 6\text{H}_2\text{O}$	2	58	42
$\text{Ni}(\text{OOCMe})_2 \cdot 4\text{H}_2\text{O}$	17	66	34
$\text{Ni}(\text{acac})_2 \cdot 2\text{H}_2\text{O}$	17	82	18
$\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$	24	0	0
NiO	24	0	0
NiF_2	24	0	0

Strem: NiCl_2 = \$34 / 50 g
 NiBr_2 = \$78 / 50 g

Reductant Screen

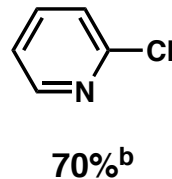
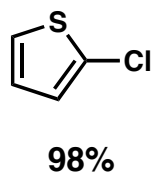
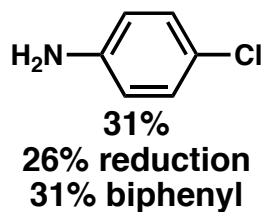
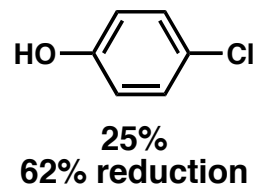
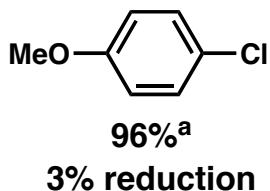
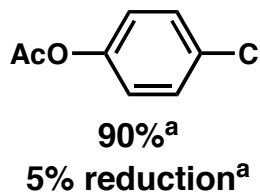
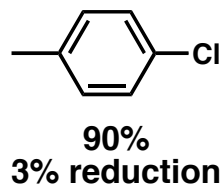
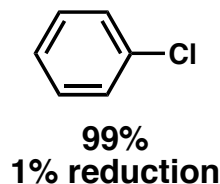
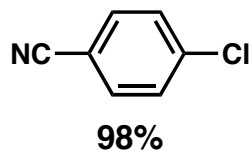
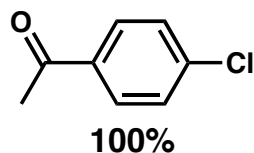
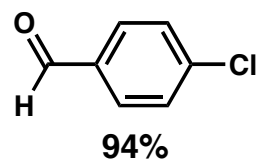
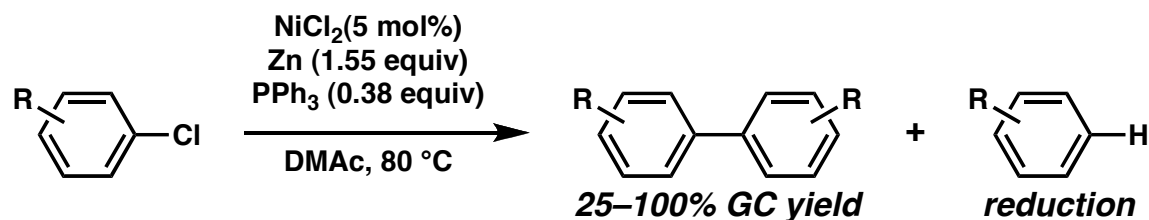


metal	E^0 (V)	time (h)	GC yield (%)	
			biphenyl	benzene
Fe	−0.44	24	0	0
Zn	−0.76	2	99	1
Mn	−1.18	24	>99	0
Al	−1.66	5	22	76
Mg	−2.37	2	96	4
Na	−2.71	24	6	6
Ca	−2.87	6	22	77

- Zn essential for reactivity
- $\text{Ni}(\text{PPh}_3)_4 + \text{ArCl} \rightarrow$ stalls
 - Add Zn $\rightarrow$ completion

History of Nickel Coupling

Homocoupling of Aryl Chlorides



Reactivity

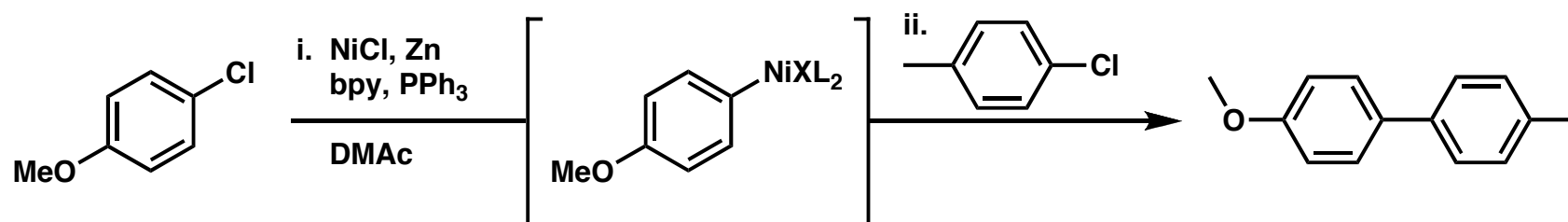
- Aryl Cl best substrates
- Aryl I and Br produce more reduction and are slower
- EWG > EDG
- -NO₂, -CO₂H not tolerated

^a 5 mol% bpy added. ^b Isolated yield.

History of Nickel Coupling

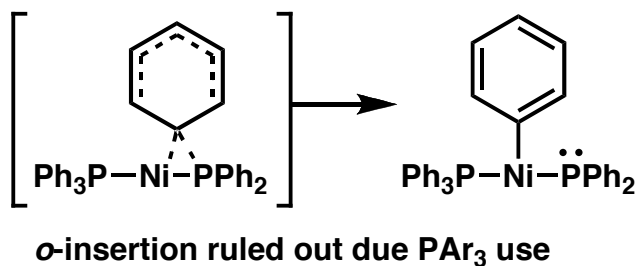
Impact of Other Additives

First Cross Coupling



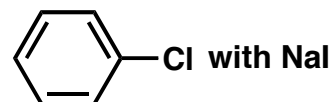
Phosphines

- PAr_3 best
- P(alkyl)_3 slower and lower yields
- dppe ineffective
- Ph from PPh_3 transferred for e^- rich substituents



Halide Ions

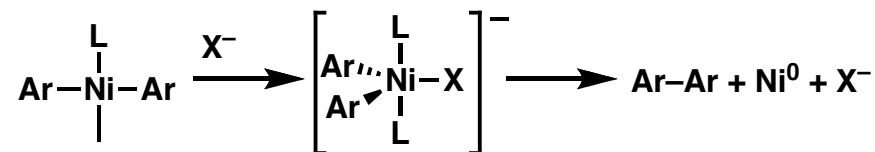
- $\text{I}^- > \text{Br}^- \gg \text{Cl}^- > \text{F}^-$
- Accelerates reaction substantially:



completely homocoupled in 2 min!

- Two Suggested Explanations for Results

- Pentacoordinate anionic complex



- Accelerated Ni Reduction

- Bridges Zn and Nickel
- Increase dielectric properties of solvent

Colon, I. *J. Org. Chem.* **1986**, 51, 2627.

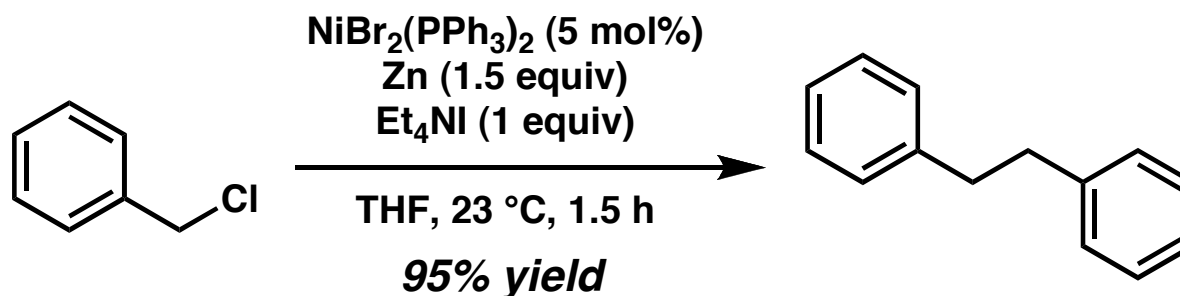
Phosphine Pentacoordinate Intermediate: Favero, G. *J. Organomet. Chem.* **1978**, 149, 395.

History of Nickel Coupling

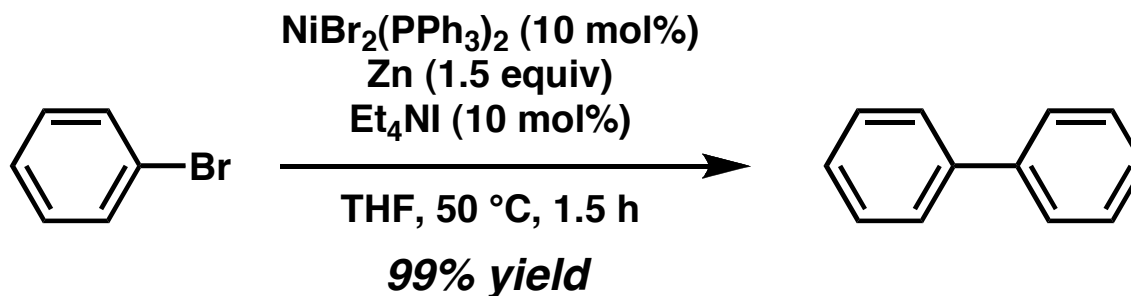
Et₄Ni: A More Soluble Halogen Source

Iyoda, M.

Benzyl Halide Couplings



Aryl Halide Couplings



Conditions Screened

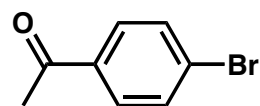
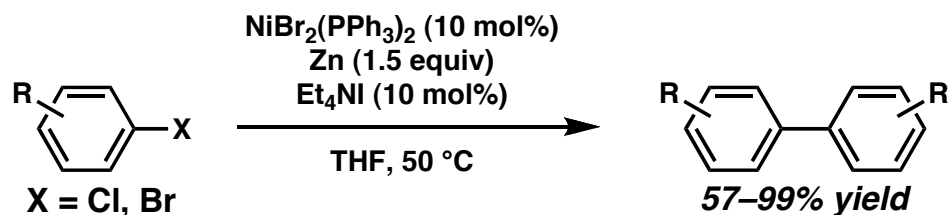
- Ni Sources: NiCl2(PPh3)2, **NiBr2(PPh3)2**, NiI2(PPh3)2
- Solvents: **THF**, DMF, CH3CN, Acetone
- Loadings of Et4NI: 10 mol% *p*- & *m*-substituted; 1 equiv for *o*-substituted
- Excess PPh3 not necessary

Benzyl Coupling: Iyoda, M. *Chem. Lett.* **1985**, 51, 2627.

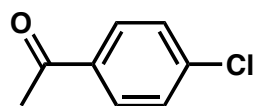
Aryl Coupling: Iyoda, M. *Bull. Chem. Soc. Jpn.* **1990**, 63, 80.

History of Nickel Coupling

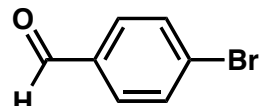
Aryl Halide Homocoupling



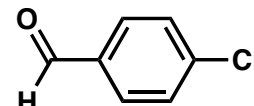
20 h, 71%



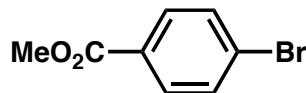
20 h, 73%



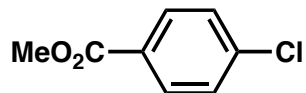
20 h, 75%



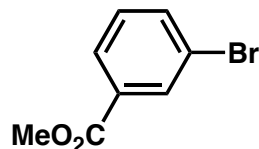
20 h, 70%



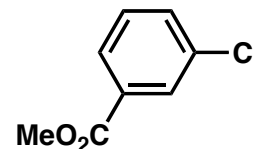
4 h, 90%



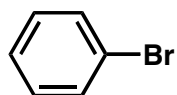
20 h, 85%



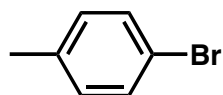
20 h, 85%



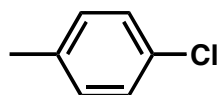
20 h, 81%



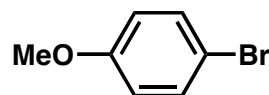
1.5 h, 99%



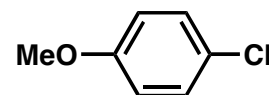
4 h, 89%



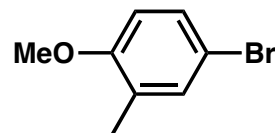
5 h, 81%



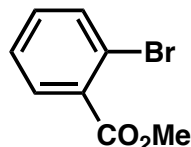
4 h, 72%



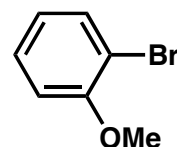
20 h, 67%



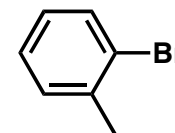
10 h, 57%



24 h, 90%^a



46 h, 81%^b



5 h, 83%^a

Notes

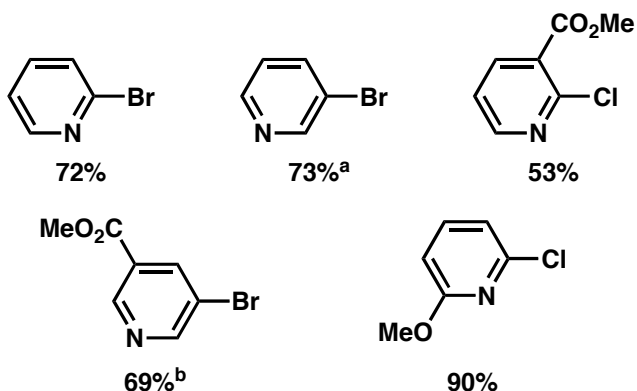
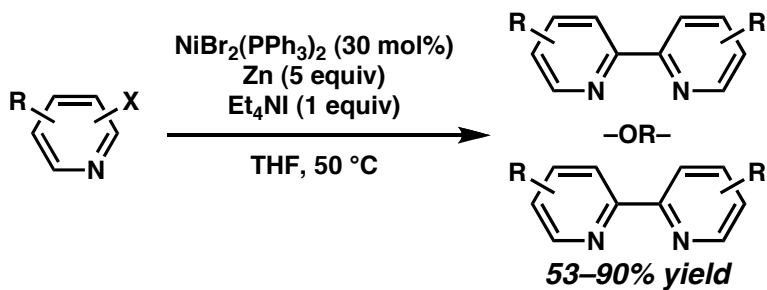
- $\text{I} \gg \text{Br} > \text{Cl}$
- Aryl I do not require Et_4NI
- o-substituted require high Ni loadings

^a 20 mol% $\text{NiBr}_2(\text{PPh}_3)_2$, 1 equiv Et_4NI . ^b 50 mol% $\text{NiBr}_2(\text{PPh}_3)_2$.

History of Nickel Coupling

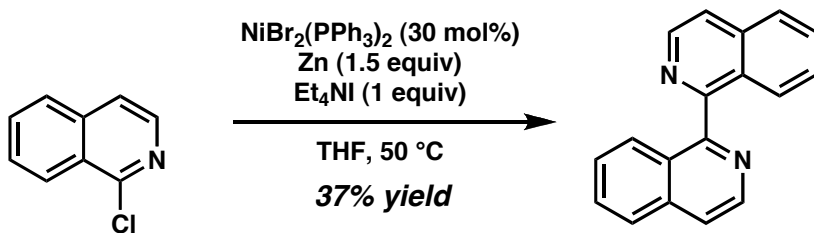
Heteroaromatic Homocoupling

Pyridines

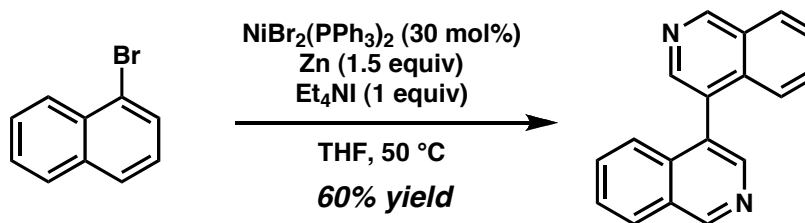
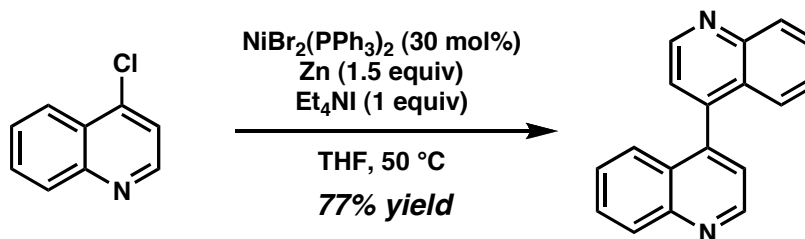
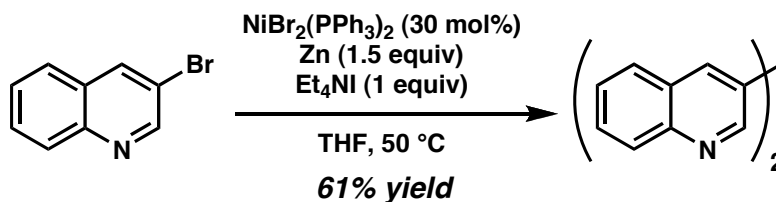
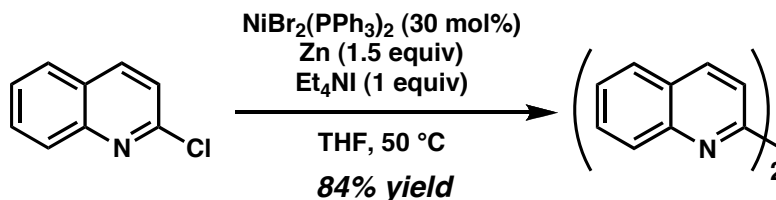


^a 3 equiv Et_4NI . ^b 30 mol% Et_4NI .

Isoquinolines



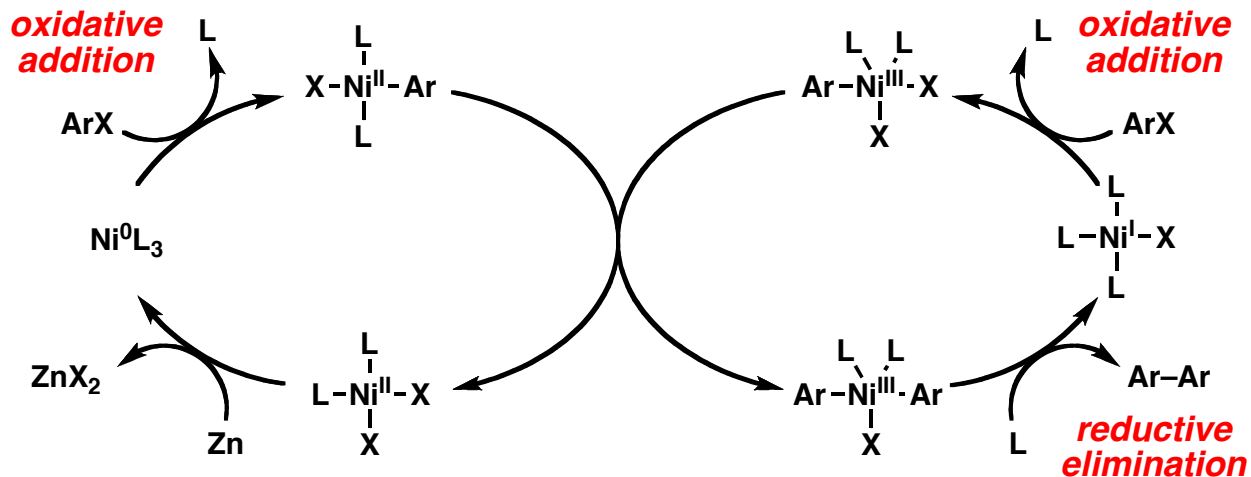
Quinolines



History of Nickel Coupling

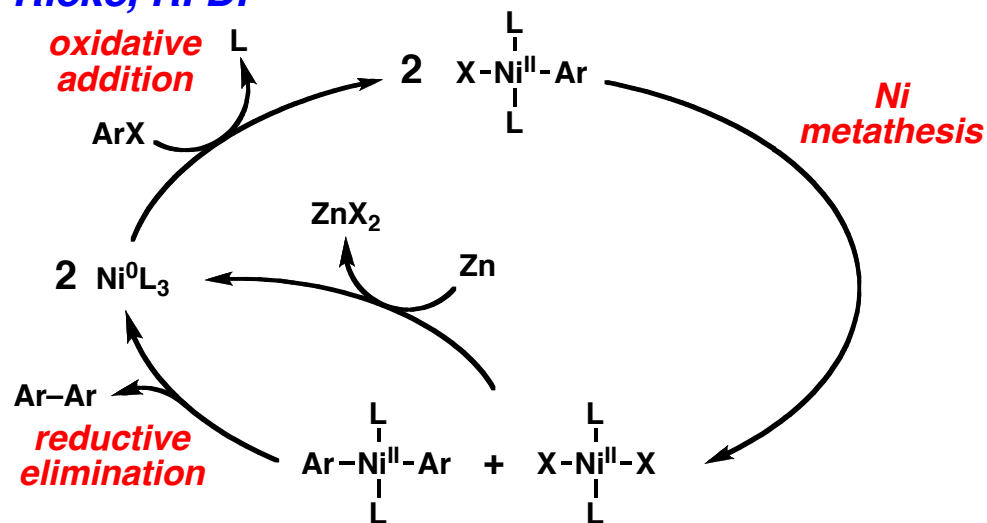
Proposed Mechanisms

Tsou, T. T. and Kochi, J. K.



J. Am. Chem. Soc. **1979**, 101, 6319. *J. Am. Chem. Soc.* **1979**, 101, 7547.

Rieke, R. D.

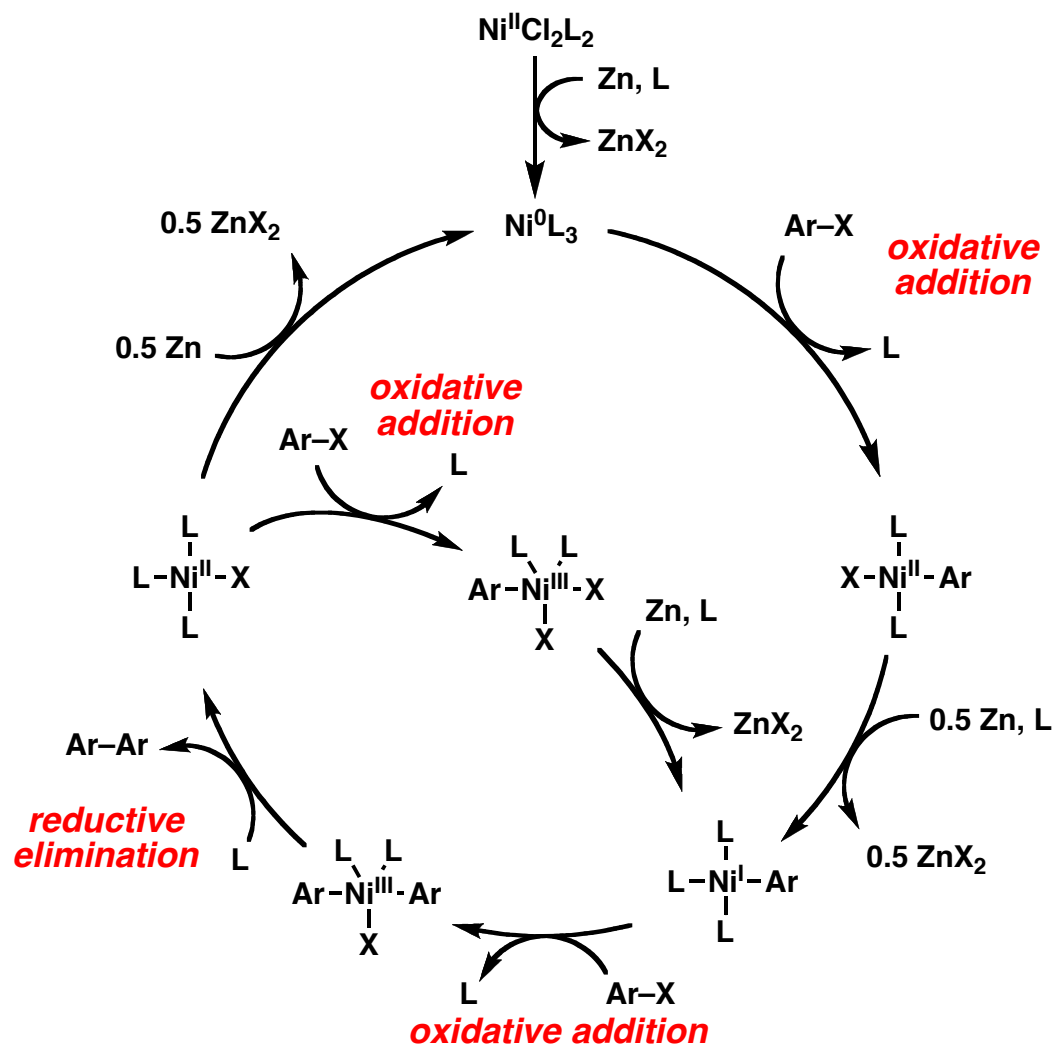


J. Am. Chem. Soc. **1977**, 99, 4089. *J. Org. Chem.* **1983**, 48, 840.

History of Nickel Coupling

Proposed Mechanisms

Bontempelli, G.

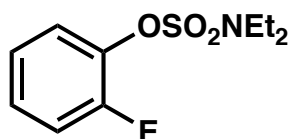


Sulfamate Ni Cross Coupling

Kumada Coupling

Snieckus, V.

Directed o-Metalation (DoM)



E = CHO, CONEt₂, CH(OH)Ph, SMe, SiMe₃, Cl, Br, I, SnBu₃, boronate esters

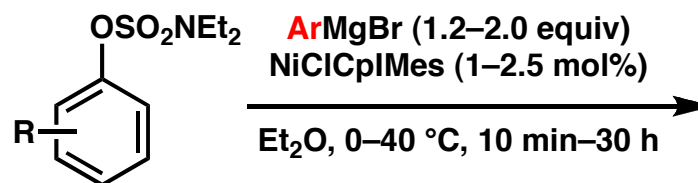
- Sulfamate is orthogonal functionality to Suzuki-Miyaura coupling with halides

Ni Catalyst Screen

- Ni(acac)₂, NiCl₂, NiCl₂/P(*t*-Bu)₃, NiCl₂/dppf, NiClCpPPh₃, NiClCpIMes
- Cp Complexes provided best yields
- NHC bulky ligands likely accelerate reductive elimination
- NiClCpIMes air stable

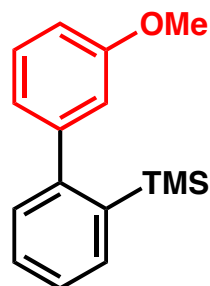
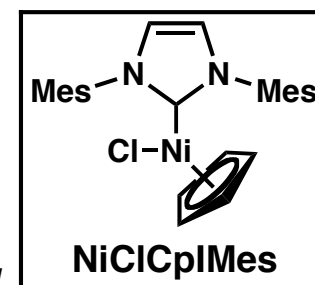
Solvent Screen

- PhMe, Et₂O, THF

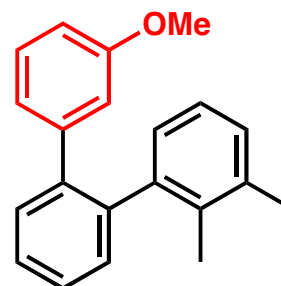


Ar = Ph, C₆H₄(3-OMe), 2-Mes, *p*-Tol

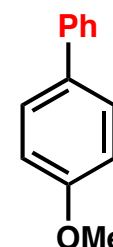
13 examples
47–86% yield



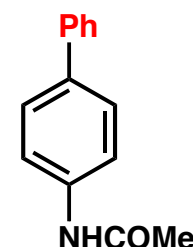
40 °C, 12 h
76%



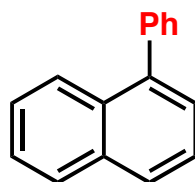
40 °C, 30 h
64%



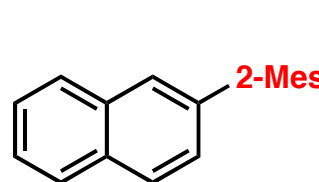
40 °C, 19 h
47%



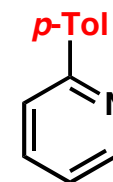
23 °C, 1 h
50%



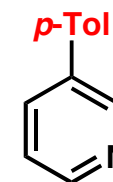
0 °C, 10 min
84%



40 °C, 18 h
69%



23 °C, 18 h
71%



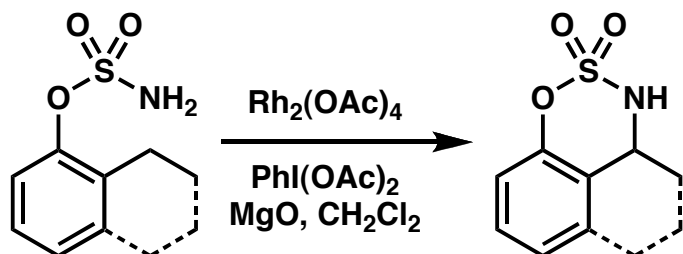
23 °C, 8 h
82%

Sulfamate Ni Cross Coupling

Kumada Coupling

Du Bois, J.

C-H Insertion to Form Sulfamate



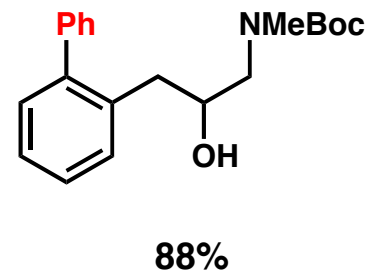
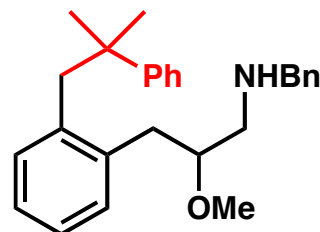
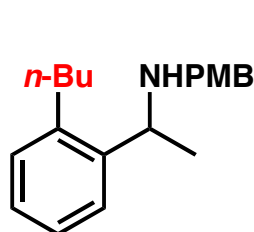
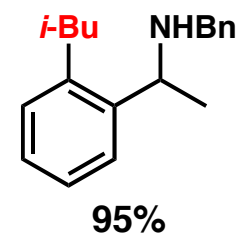
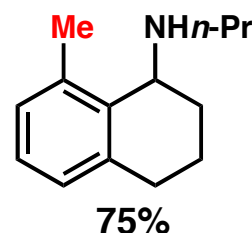
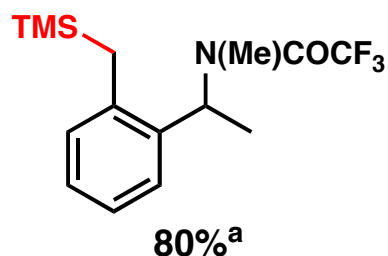
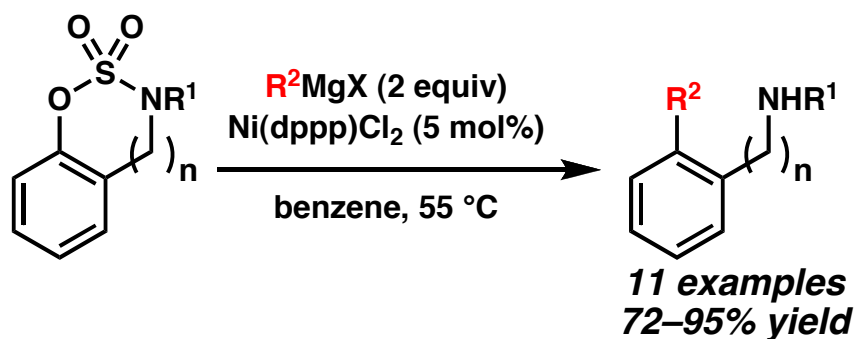
- NH → NR to allow Kumada coupling
- NH under Kumada conditions results acyl moiety cleavage

Ni Catalyst Screen

- Ni(dppe)Cl₂, Ni(PPh₃)Cl₂, Ni(acac)₂, Ni(dppf)Cl₂, Ni(dppp)Cl₂, Ni(COD)+dppp
- Ni(dppp)Cl₂ optimal
- Other Ni^{II} phosphine left starting material

Solvent Screen

- benzene, CF₃C₆H₅, Et₂O, THF, DME, dioxane
- THF, DME, dioxane inhibit reaction



^a 10 mol % Ni(dppp)Cl₂.

Sulfamate Ni Cross Coupling

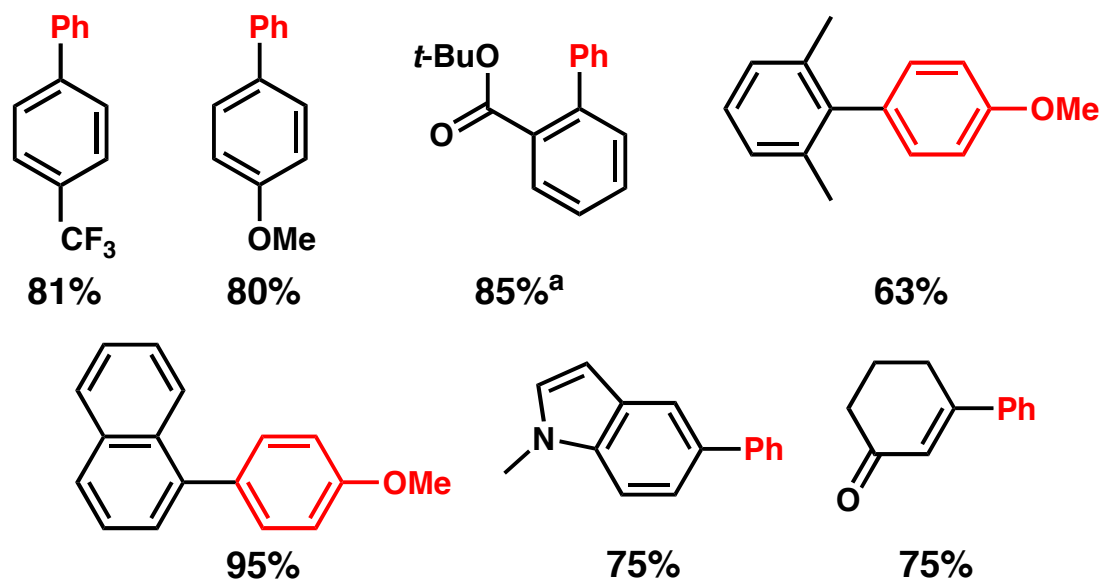
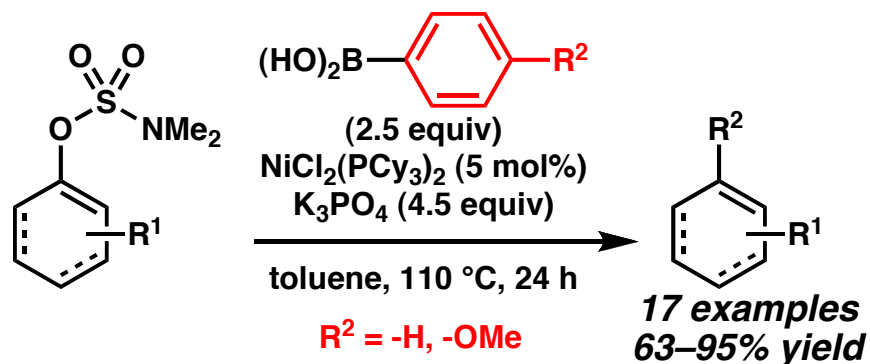
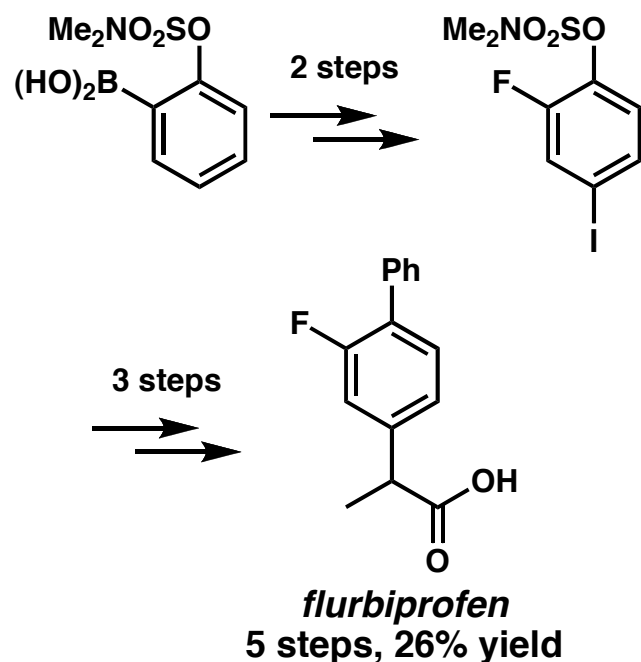
Suzuki-Miyaura Coupling

Garg, N. K.

Functionalization of Aryl Sulfamates

- Directed *o*-Metalation (DoM)
- *p*-directed electrophilic aromatic substitution

Total Synthesis of Flurbiprofen

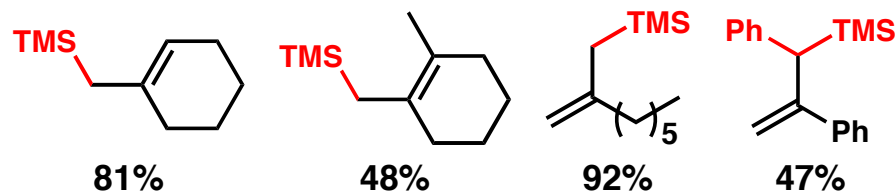
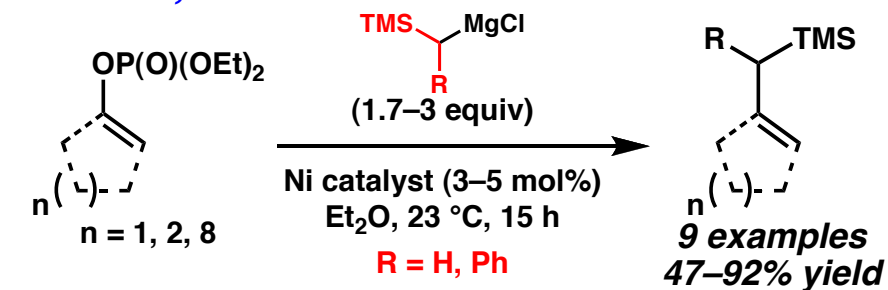


^a 10 mol% NiCl2(PCy3)2, 4 equiv ArB(OH)2, 7.2 equiv K3PO4, toluene, 130 °C

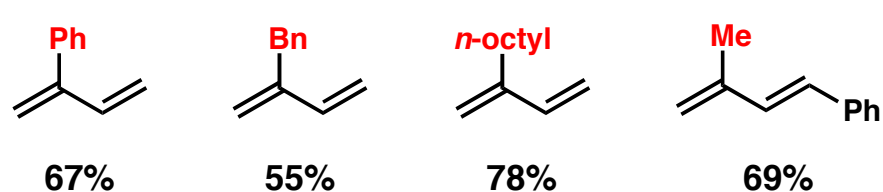
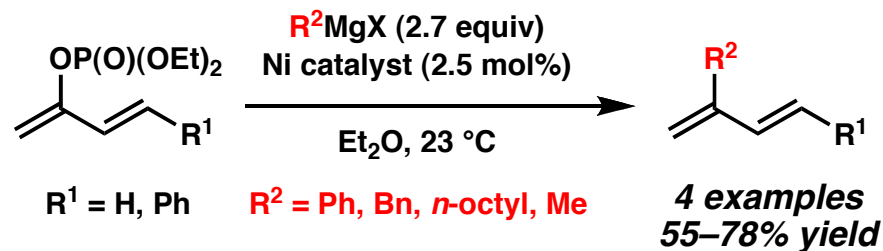
Phosphate Cross Coupling

Kumada Coupling: Vinyl Phosphates

Kumada, M.



Claesson, A.



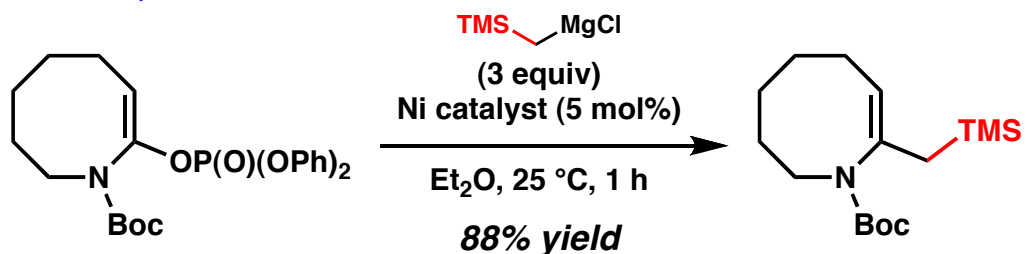
Notes

- Ni Catalysts: Ni(acac)_2 , NiBr_2
- Ni^{II} + Phosphines also work (no examples)

Notes

- Ni Catalysts: $\text{NiCl}_2(\text{dppe})$, $\text{NiCl}_2(\text{dppp})$
- 2° Grignards inoperable: CyMgBr and 1-methylbutylMgBr

Nicolaou, K. C.



Kumada, M. *Synthesis* **1981**, 1001.

Claesson, M. *Tetrahedron Lett.* **1983**, 24, 5137.

Nicolaou, K. C. *Chem. Commun.* **1998**, 1757.

Phosphate Cross Coupling

Kumada Coupling: Vinyl Phosphates

Bäckvall, J.-E.

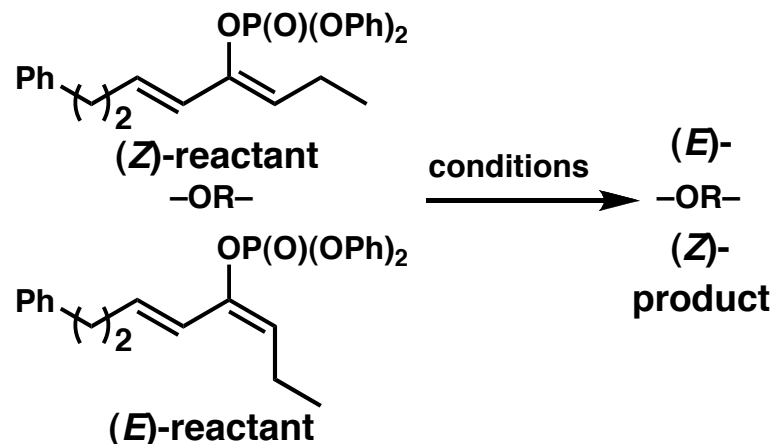
Notes

- Ni Catalyst: $\text{NiCl}_2(\text{dppe})$ and $\text{NiCl}_2(\text{dppp})$
- 2° Grignard reagents possible for cyclic vinyl phosphates with longer coupling times
- No product from 2-octylMgBr

Acyclic Diene

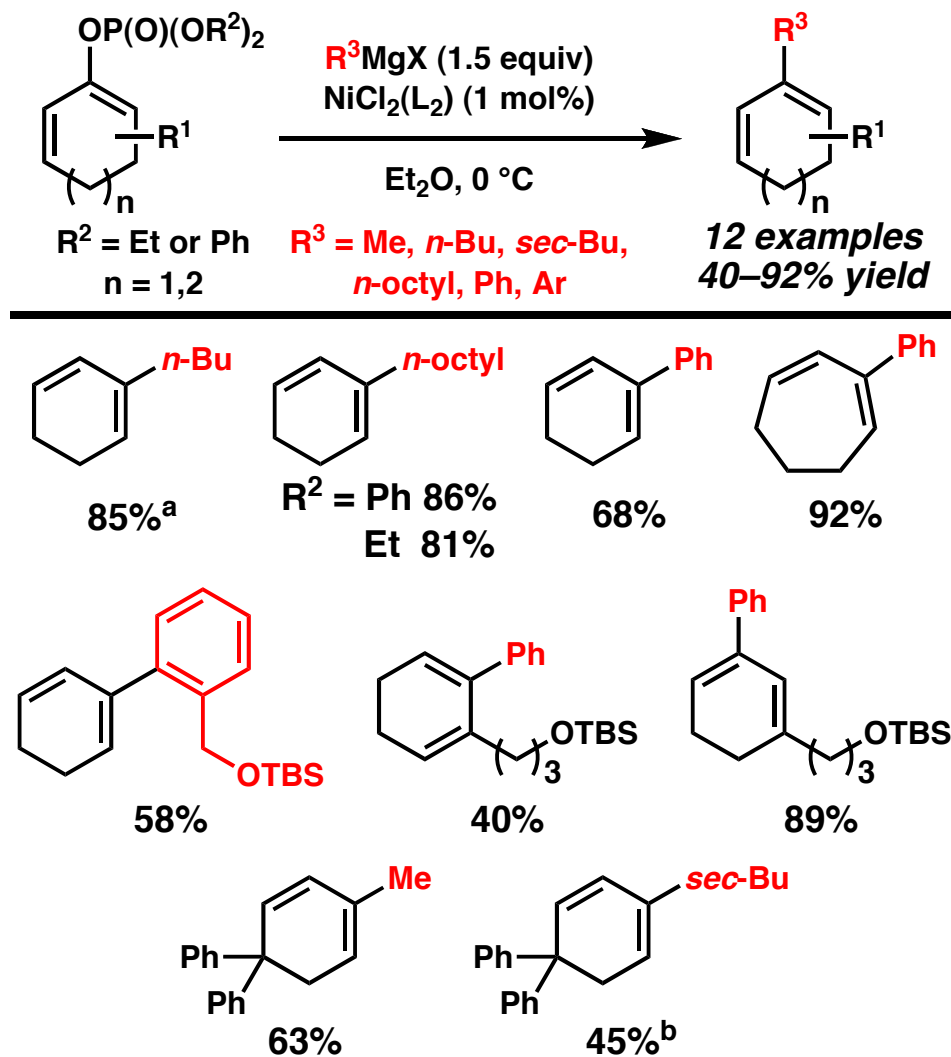
- Yield similar to Claesson report
- 2° Grignard reagents inoperable

E/Z Selectivity Study



- Both reactant isomers favor (E)-product
- (E)-reactant $\rightarrow$ (Z)-product (5:95 = E:Z)
- (Z)-reactant $\rightarrow$ (E)-product (57:43 = E:Z)

J. Am. Chem. Soc. **1979**, 101, 2246.



^a GC yield. ^b THF, 40 °C.

Phosphate Cross Coupling

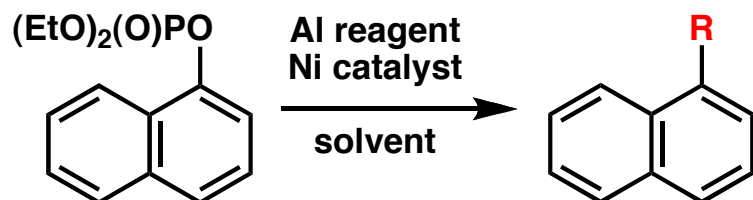
Kumada Coupling: Aryl Phosphates

Kumada, M.

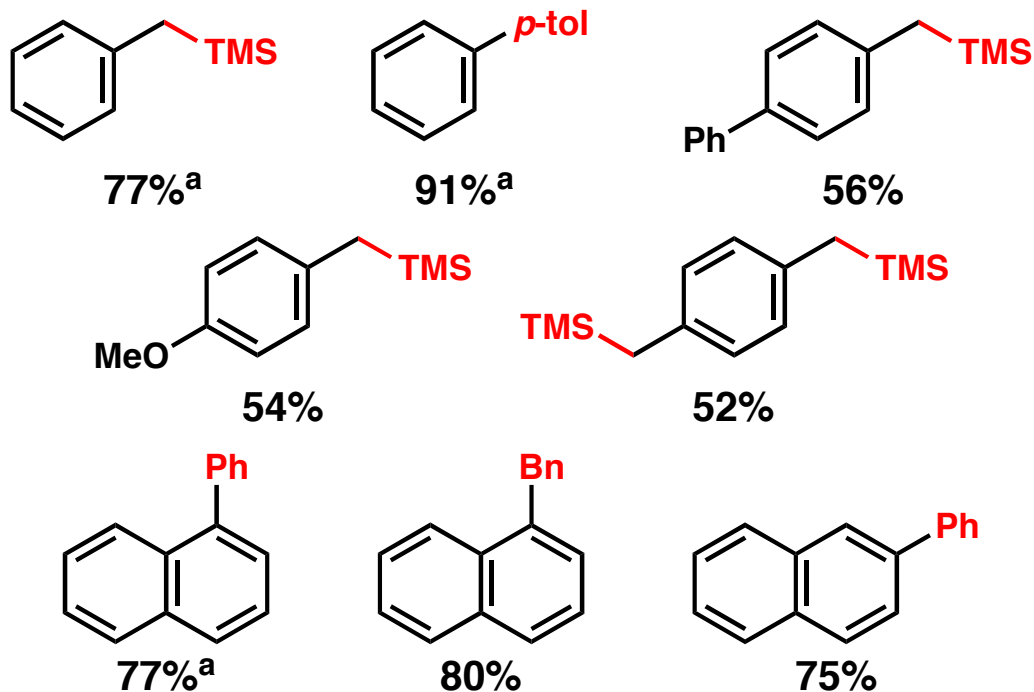
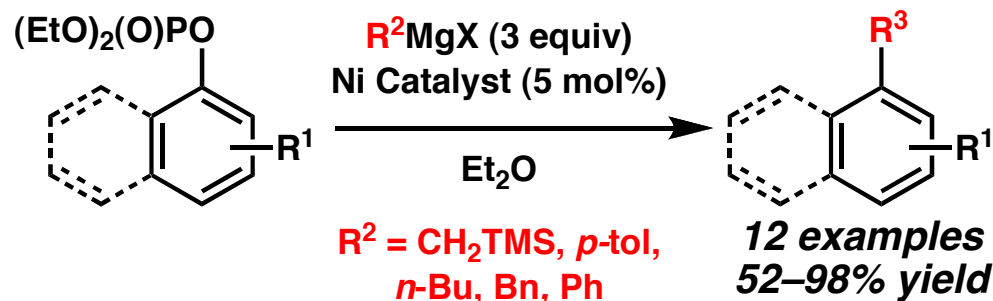
Catalyst Screen

- Ni Catalyst: $\text{Ni}(\text{acac})_2$, $\text{NiCl}_2(\text{PPh}_3)_2$, and $\text{NiCl}_2(\text{dppp})$
- Inoperable Catalysts: $\text{Pd}(\text{PPh}_3)_4$, PdCl_2 , CoCl_3 , $\text{Fe}(\text{acac})_3$, CuBr

Aluminum Reagents



Al reagent	solvent	yield
Et_3Al	Et_2O	95%
$n\text{-Bu}-\text{CH}=\text{CH}-\text{Al}(\text{i-Bu})_2$	THF	99%



^a GC yield.

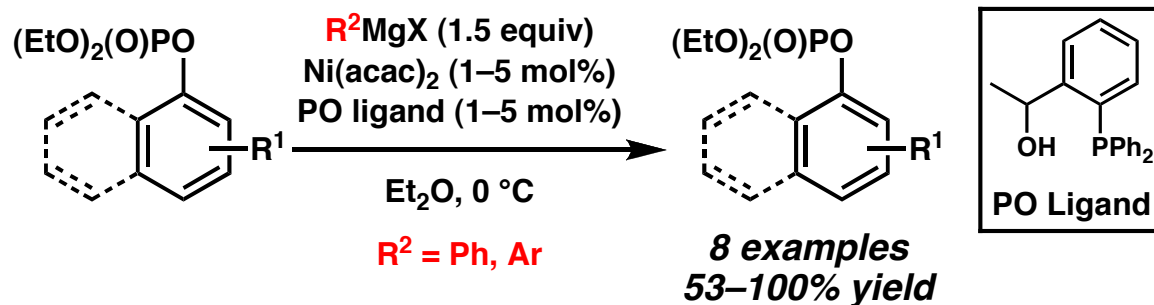
Phosphate Cross Coupling

Kumada Coupling: Aryl Phosphates

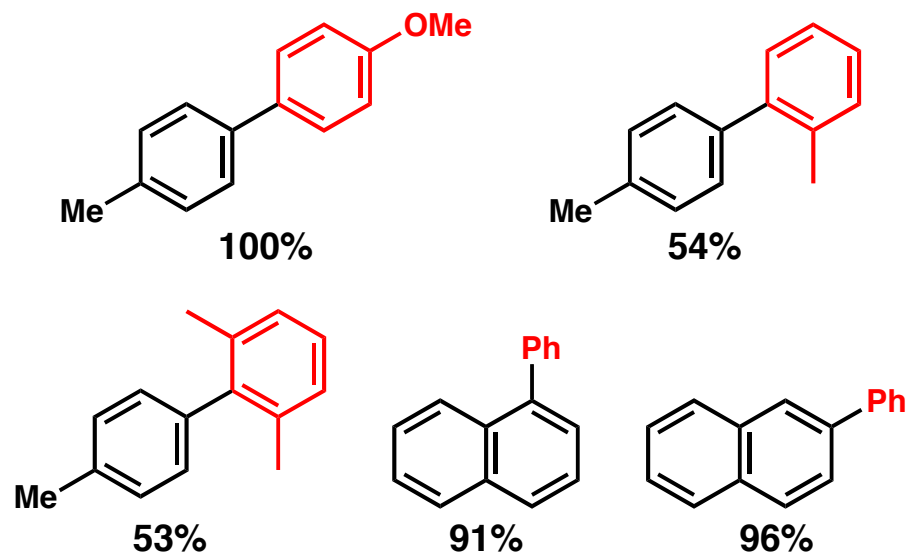
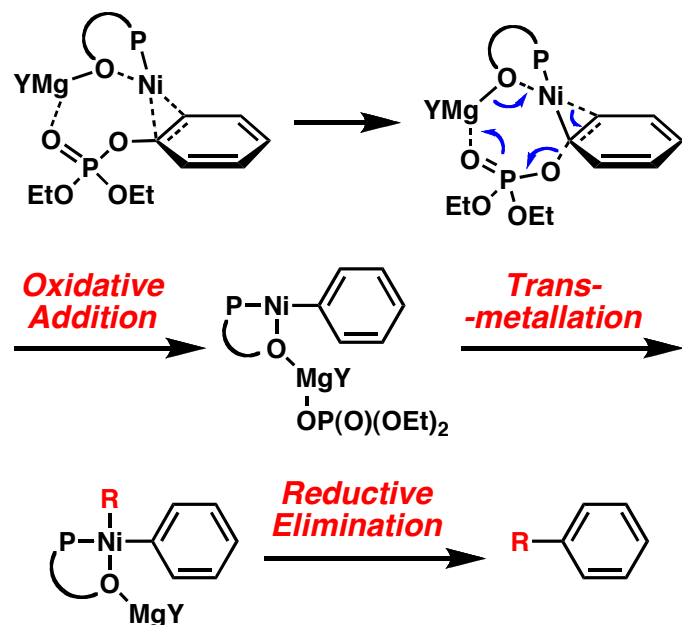
Nakamura, E.

Notes

- Ligand provides cooperative effect
- 12 ligands screened
- Effective for aryl F, Cl, Br, I, OTf, OCONEt₂



Push-Pull Mechanism



^a GC yield. ^b THF, 40 °C.

Aryl and Vinyl Ether Cross Coupling

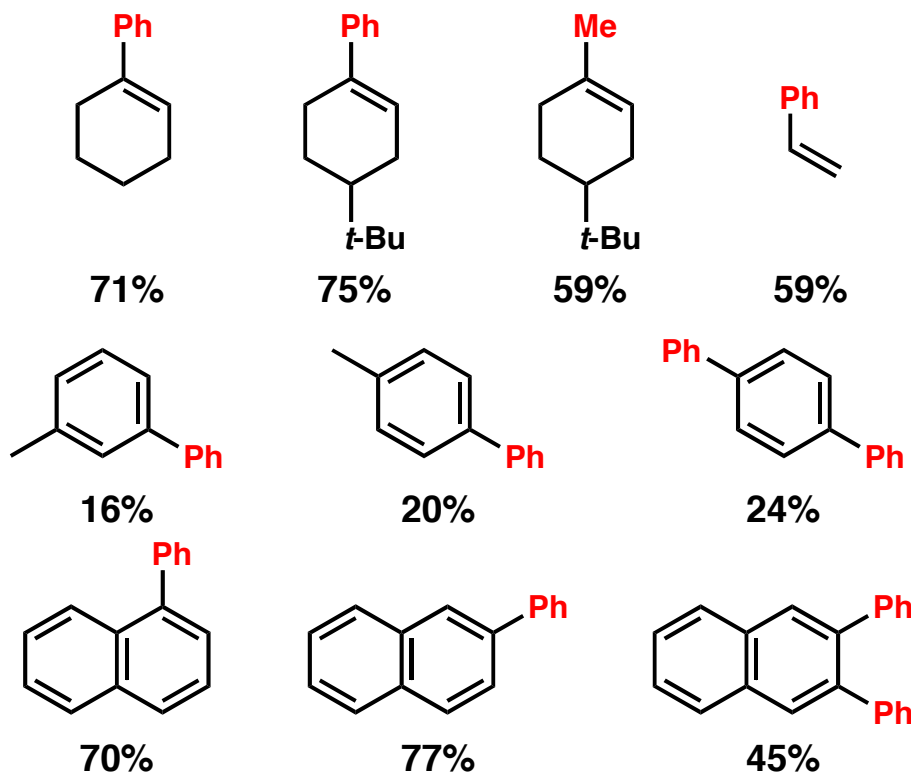
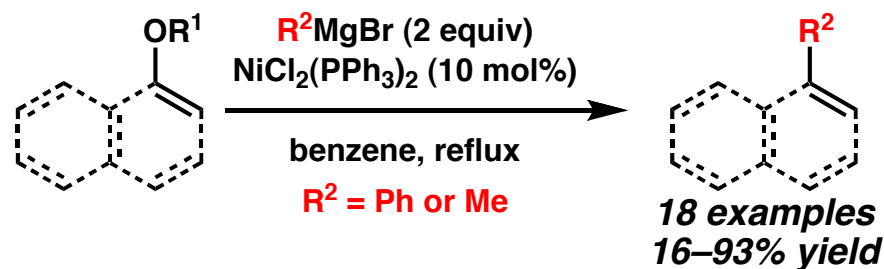
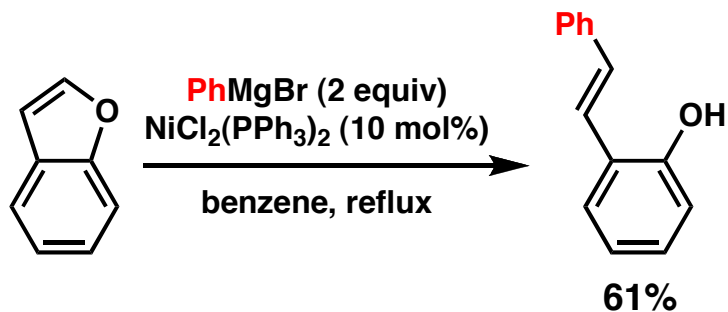
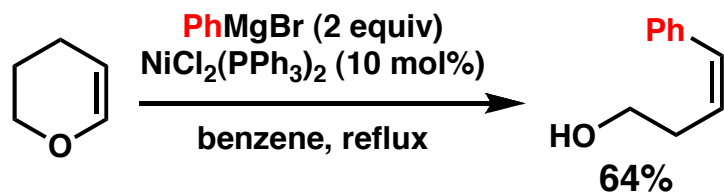
Kumada Coupling

Wenkert, E.

Grignard Effects

- PhMgBr effective with vinyl and aryl ethers
- MeMgBr is unreactive with aryl ethers

Cyclic Ethers



Aryl and Vinyl Ether Cross Coupling

Kumada Coupling: sp^2-sp^2 C-C Formation

Dankwardt, J. W.

Ni Catalyst Screen

- Numerous Ni^{II} catalysts investigated
- Ni(acac)₂/2 PCy₃ and Ni(acac)₂/2 PPhCy₂ are best

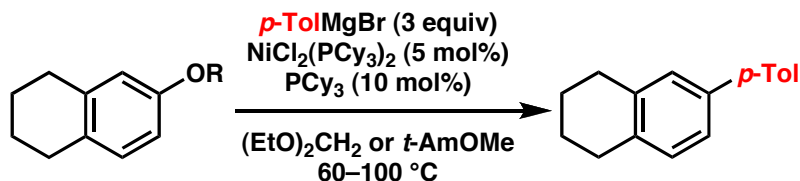
Ligand Screen

- PMe₃, PEt₃, *Pi*-Bu₃, *Pi*-Pr₃, PCy₃, PPhCy₂, PPh₂Cy, PPh₃
- PCy₃ and PPhCy₂ are best
- Intermediate cone angle and good σ -donating

Solvent Screen

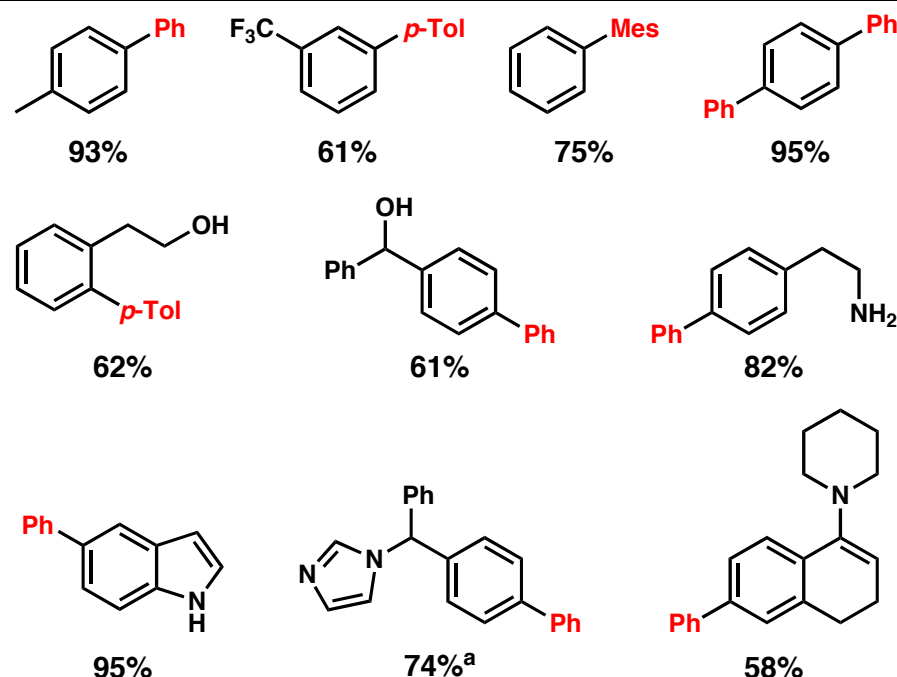
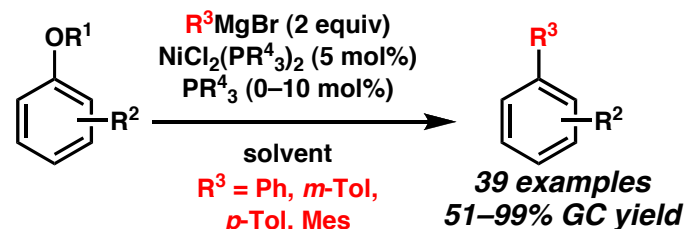
- DME, THF, toluene, *i*-Pr₂O, (EtO)₂CH₂, *t*-AmOMe, *n*-Bu₂O
- Non-polar ethers or toluene preferred
- MgBr(OMe) (catalyst poison) crashes out in non-polar solvents

Aryl Ether Screen



R	GC yield	R	GC yield
-Me	94%	-MOM	92%
-(CH ₂) ₂ OMe	77%	-TMS	70%
-(CH ₂) ₂ NMe ₂	99%		

Angew. Chem., Int. Ed. 2004, 43, 2428.



^a Isolated yield.

Reaction Scope Notes

- Amenable to functionalized benzene, naphthalene, imidazole, pyridines, and indoles
- Tolerant of free alcohols and amines
- Limited to aryl Grignard reagents

Aryl and Vinyl Ether Cross Coupling

Kumada Coupling: sp^3-sp^2 C-C Formation

Shi, Z.-J.

Catalyst Screen

- Fe, Co, Ni, Pd catalysts investigated
- $\text{NiCl}_2(\text{PCy}_3)_2$ and $\text{Ni}(\text{acac})_2/2 \text{ PCy}_3$ are best

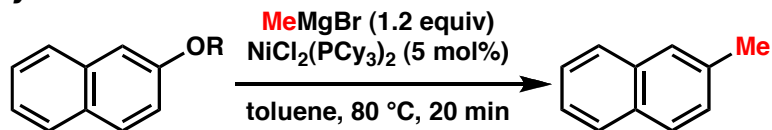
Ligand Screen

- PCy_3 , PPh_3 , dppe, dppf
- PCy_3 best

Solvent Screen

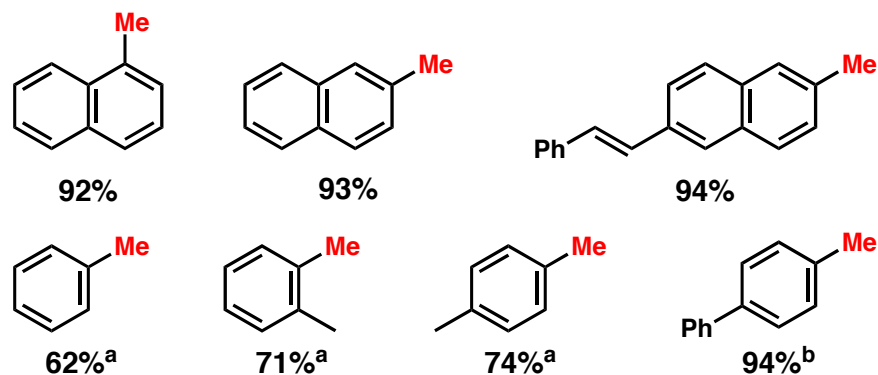
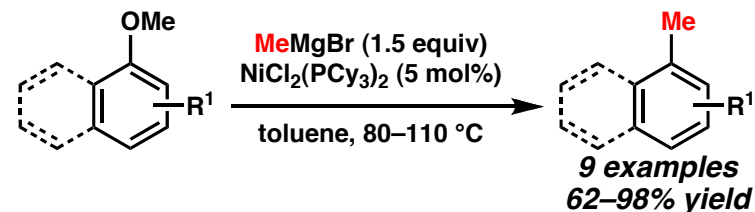
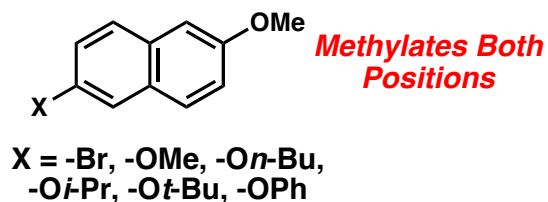
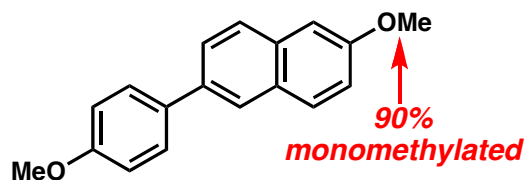
- Dioxane, THF, toluene, Et_2O
- Toluene and Et_2O are best solvents

Aryl Ether Screen



R	yield	R	yield
-Me	92%	- <i>t</i> -Bu	88%
-Et	92%	-MOM	69%
- <i>n</i> -Bu	96%	-TMS	95%
- <i>i</i> -Bu	91%	-Ph	95%

Differentiation Studies



^a GC yield. ^b *t*-butyl phenyl ether

Reaction Scope Notes

- More favorable for naphthalene and biphenyl
- Only MeMgBr explored

Aryl and Vinyl Ether Cross Coupling

Kumada Coupling: sp^3-sp^3 C-C Formation

Shi, Z.-J.

Catalyst Screen

- Fe, Ni, Pd catalysts investigated
- $\text{NiCl}_2(\text{dppf})/\text{dppf}$ and $\text{Ni}(\text{acac})_2/\text{dppf}$ are best

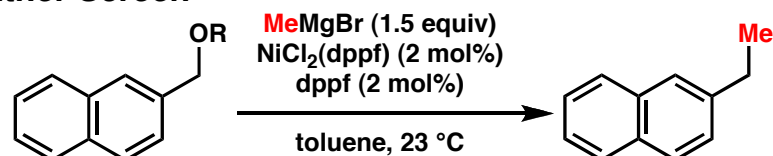
Ligand Screen

- PCy_3 , PPh_3 , dppe , dppf , dppb , dppp , dppe
- PCy_3 best

Solvent Screen

- THF, toluene, Et_2O
- Toluene and Et_2O are best solvents

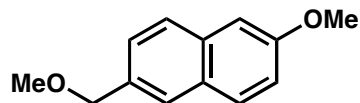
Ether Screen



R	yield	R	yield
-Me	80%	-Ph	75%
-Et	73%	-TMS	76%
- <i>t</i> -Bu	79%	-Ac	18% ^a

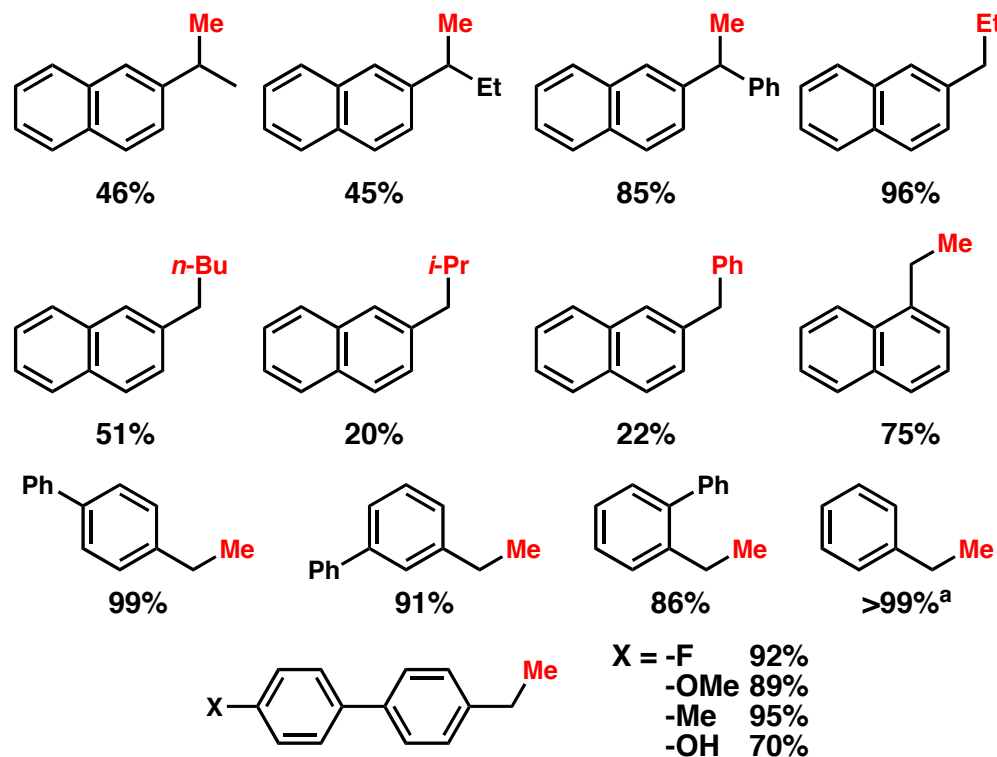
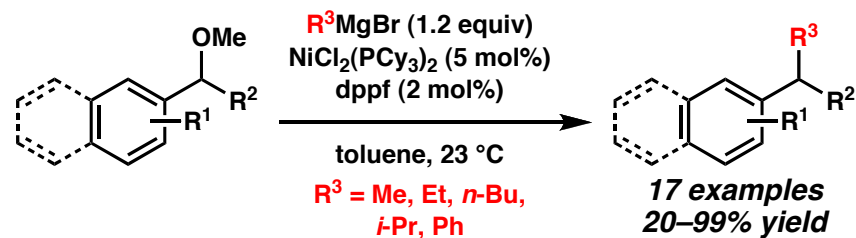
^a GC yield

Differentiation Studies



- Methylate Sequentially: sp^3-sp^2 then sp^2-sp^2
- Methylate Simultaneously: Use $\text{dppf} + \text{PCy}_3$

J. Am. Chem. Soc. **2008**, *130*, 3268.



^a GC yield.

Aryl and Vinyl Ether Cross Coupling

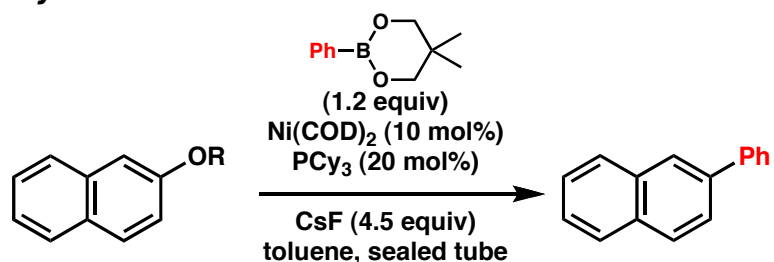
Suzuki-Miyaura

Chatani, N.

Organoboron Screen

— PhB(OH)₂, PhBF₃K, NaBPh₄

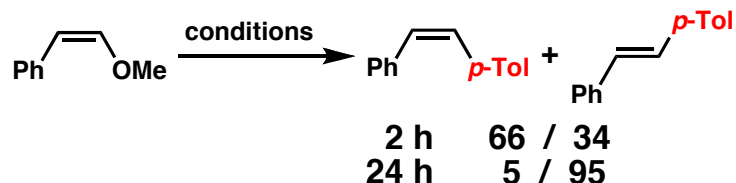
Aryl Ether Screen



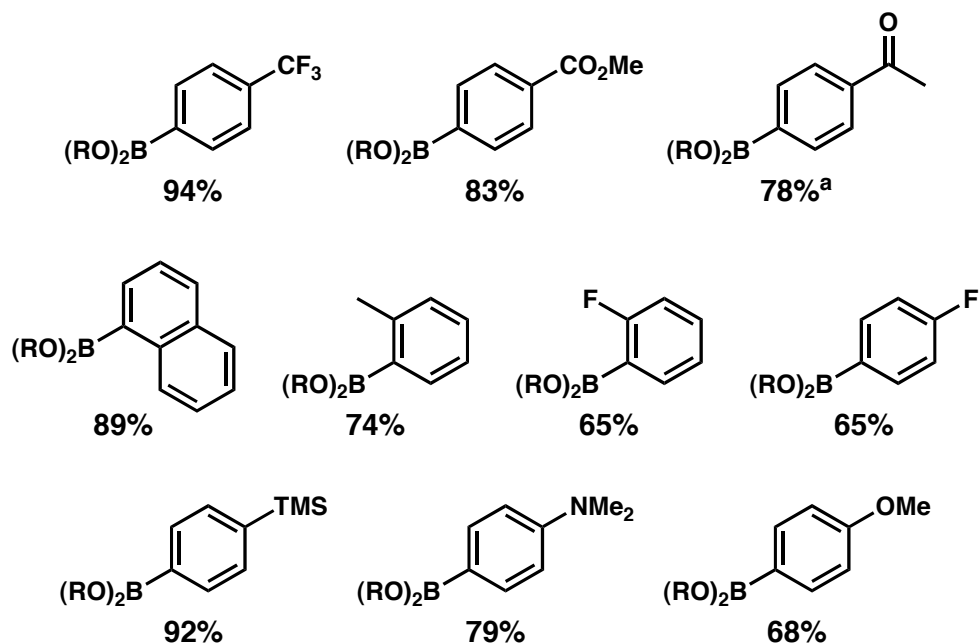
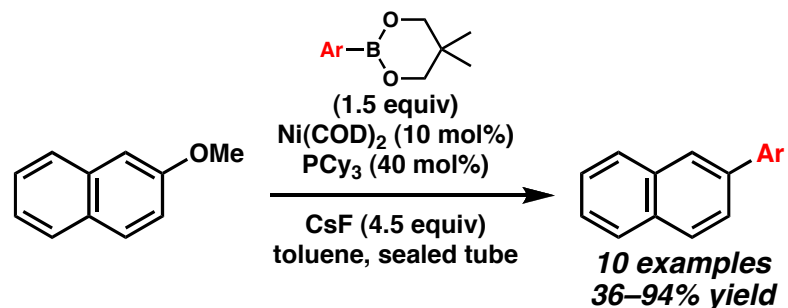
R	yield	R	yield
-Me	93%	- <i>i</i> -Pr	16%
-Et	74%	-Ac	10%

Electrophile Screen

- Naphthalene and phenanthrene good electrophilic substrates (57–93%)
- Only EWG functionalized benzene couple (14–55%)
- Olefins couple with poor E/Z selectivity



Aromatic Ethers: *Angew. Chem., Int. Ed.* **2008**, 47, 4866.
 Vinyl Ethers: *Org. Lett.* **2009**, 11, 4890.



^a 20 mol% Ni(COD)₂, 80 mol% PCy₃.

Price Comparison

By Functional Group

	R = CF ₃ \$310 / mol p-tol \$70 / mol Me \$20 / mol
	\$180 / mol
	\$130 / mol
I-Me	\$20 / mol
	R = t-Bu \$10 / mol Me \$7 / mol
	\$40 / mol
	\$220 / mol

By Price

Increasing Price

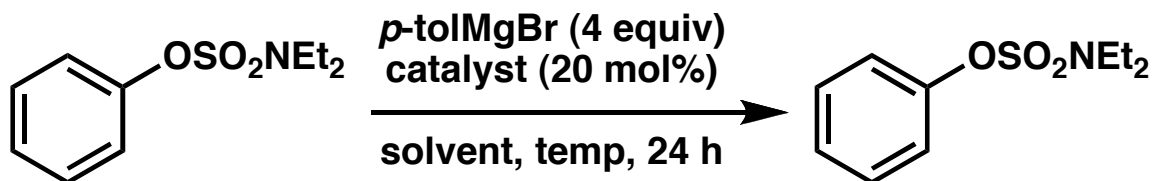
	\$7 / mol
	\$10 / mol
I-Me	\$20 / mol
	\$20 / mol
	\$40 / mol
	\$70 / mol
	\$130 / mol
	\$180 / mol
	\$220 / mol
	\$310 / mol

Approximate Cost from Sigma-Aldrich
see also Garg, N. *J. Am. Chem. Soc.* **2008**, 130, 14422.

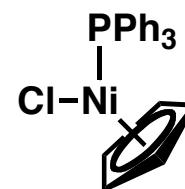
Sulfamate Ni Cross Coupling

Kumada Ni Catalyst Screen

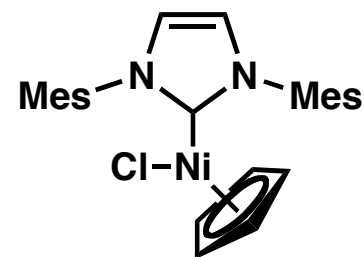
Snieckus, V.



entry	catalyst	solvent	temp (°C)	GC yield (%)
1	Ni(acac) ₂	PhMe	120	29
2	NiCl ₂	PhMe	120	72
3	NiCl ₂ / P(<i>t</i> -Bu) ₃	PhMe	120	63
4	NiCl ₂ / dppf	PhMe	120	74
5	NiClCpPPh ₃	PhMe	120	91
6	NiClCpIMes	PhMe	40	85
7	NiClCpIMes	Et ₂ O	40	>99



NiClCpIMes

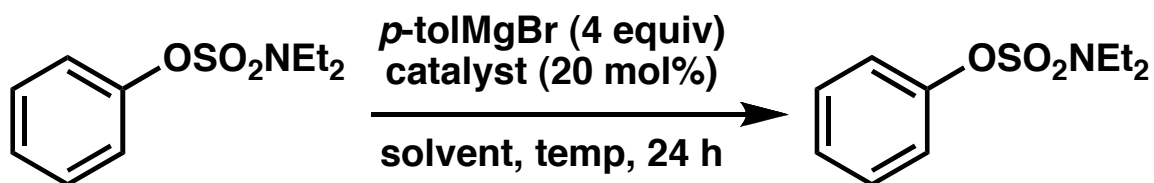


NiClCpIMes

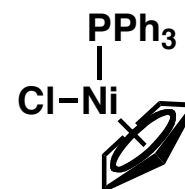
Sulfamate Ni Cross Coupling

Kumada Ni Catalyst Screen

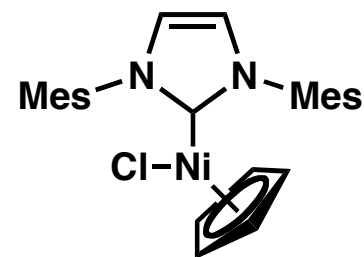
Snieckus, V.



entry	catalyst	solvent	temp (°C)	GC yield (%)
1	Ni(acac) ₂	PhMe	120	29
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3	NiCl ₂ / P(<i>t</i> -Bu) ₃	PhMe	120	63
4	NiCl ₂ / dppf	PhMe	120	74
5	NiClCpPPh ₃	PhMe	120	91
6	NiClCpIMes	PhMe	40	85
7	NiClCpIMes	Et ₂ O	40	>99



NiClCpIMes

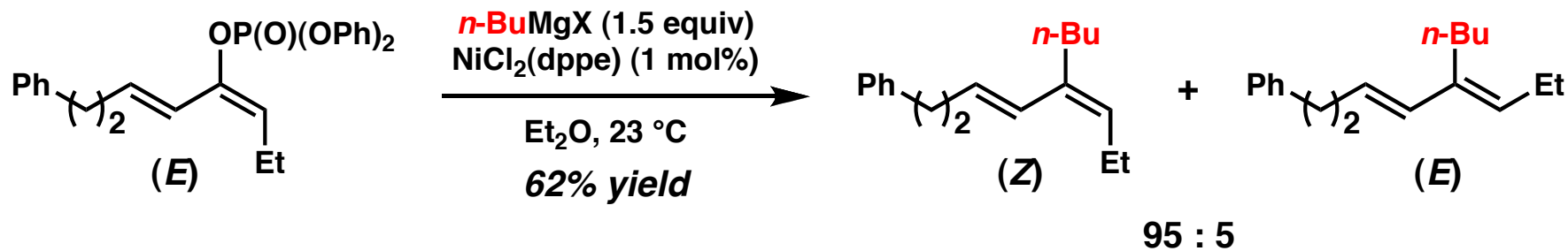
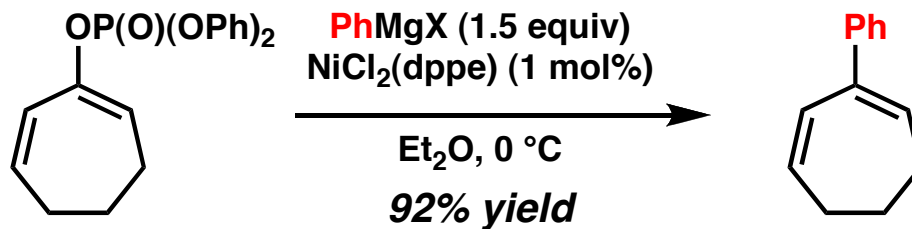
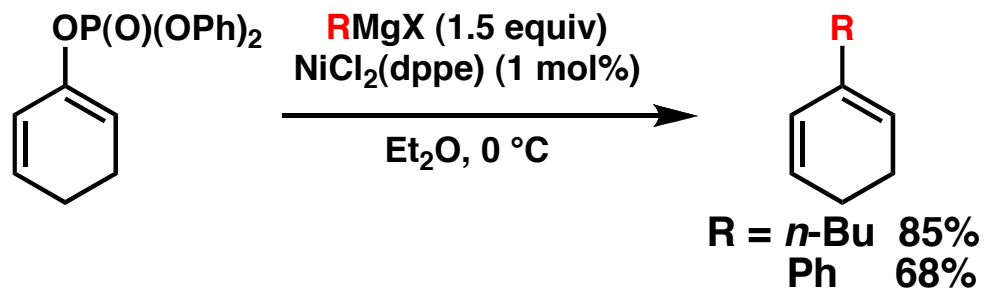


NiClCpIMes

Phosphonate and Phosphoramidate Cross Coupling

Kumada Coupling: Vinyl Phosphonates

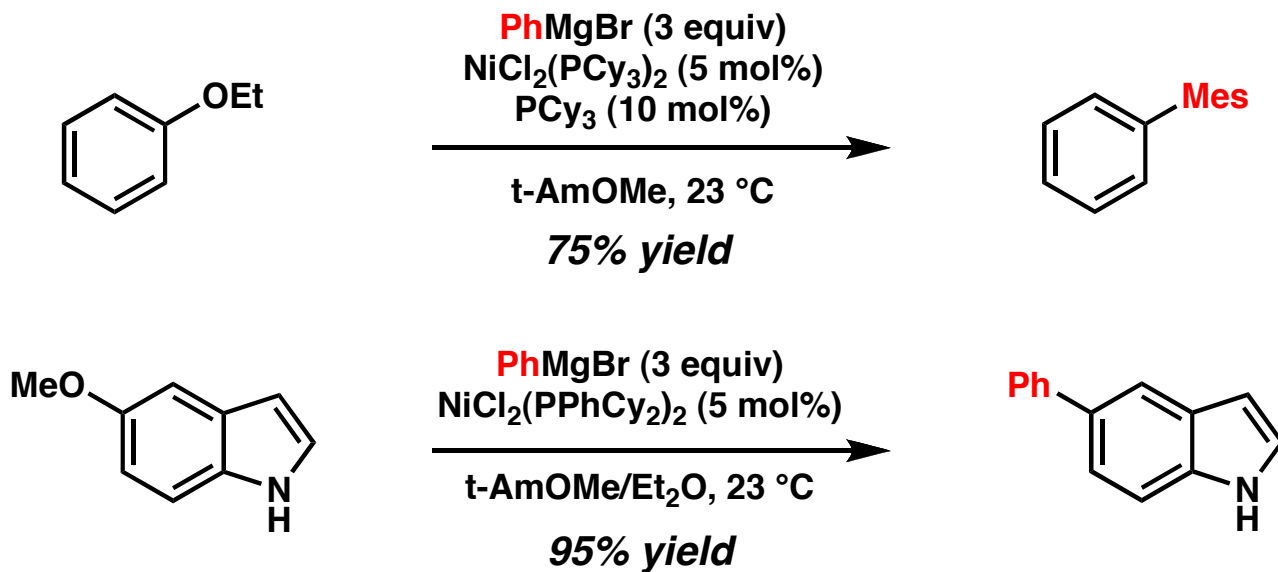
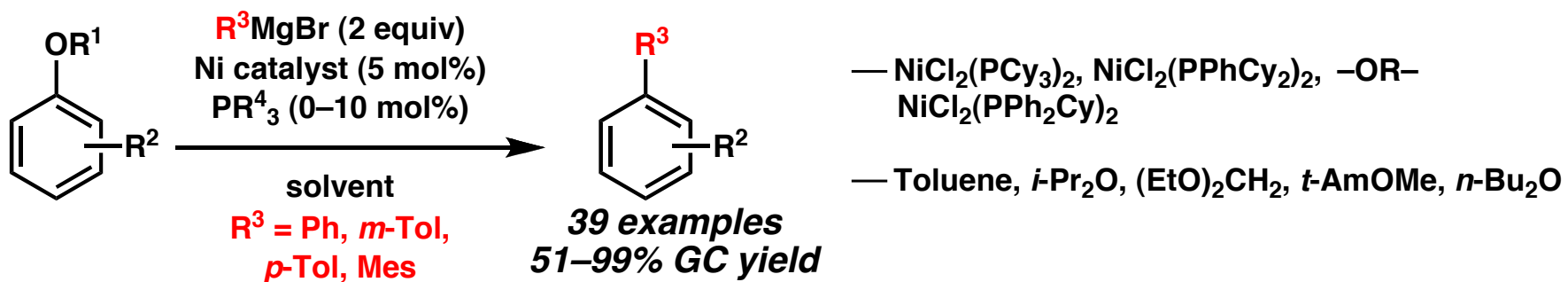
Bäckvall, J.-E.



Ether Cross Coupling

Kumada Coupling: sp^2-sp^2 C–C Formation

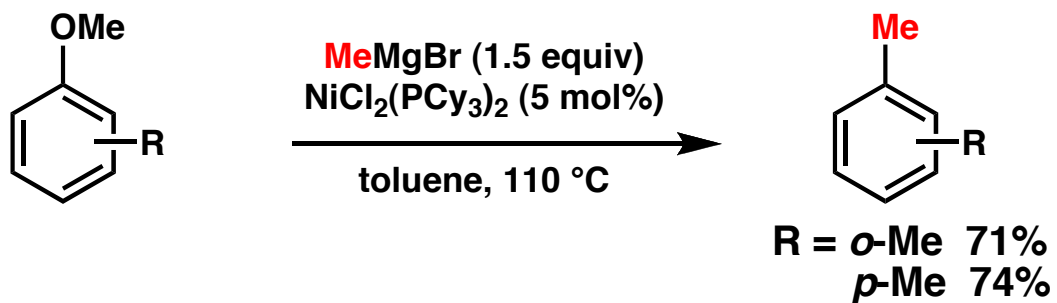
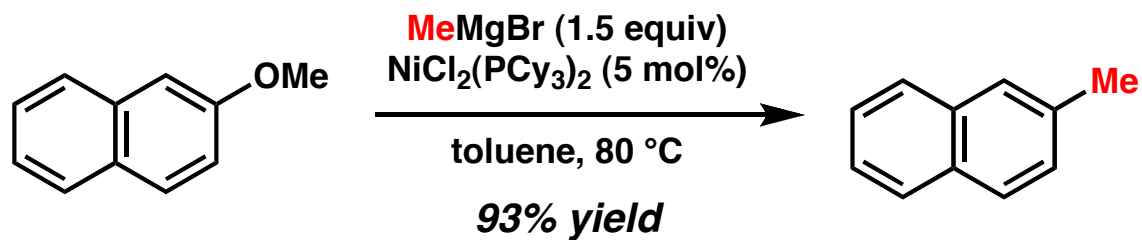
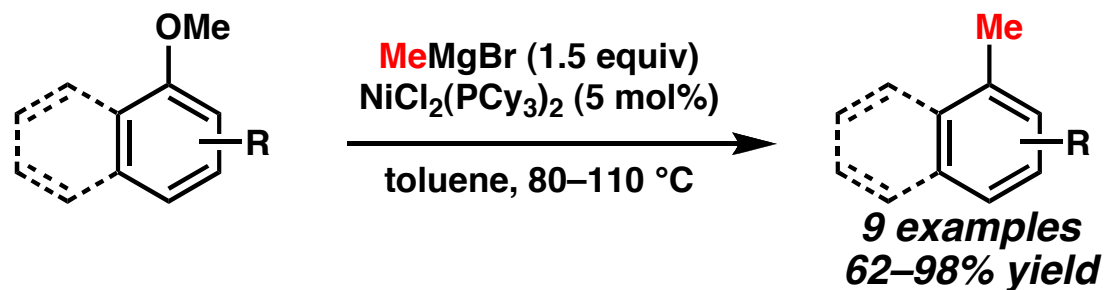
Dankwardt, J. W.



Ether Cross Coupling

Kumada Coupling: sp^3 - sp^2 C-C Formation

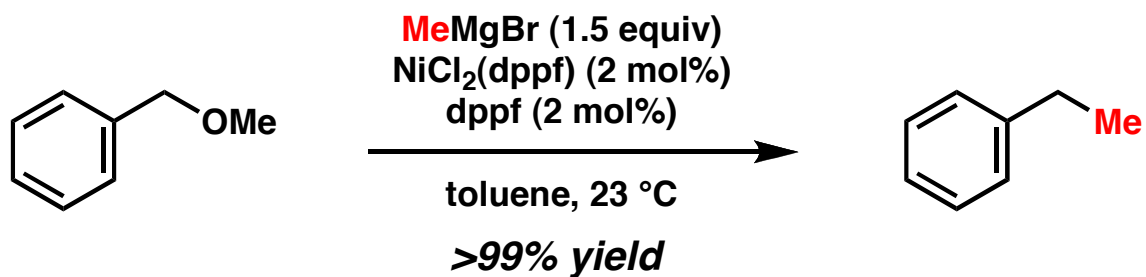
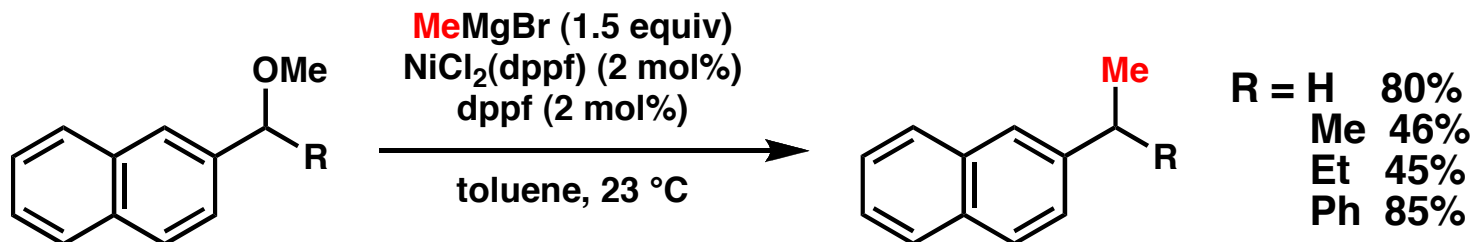
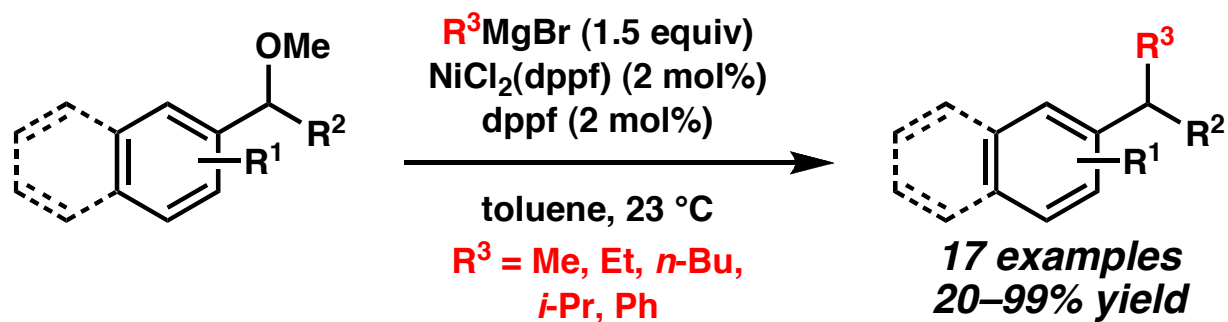
Shi, Z.-J.



Ether Cross Coupling

Kumada Coupling: sp^3-sp^3 C-C Formation

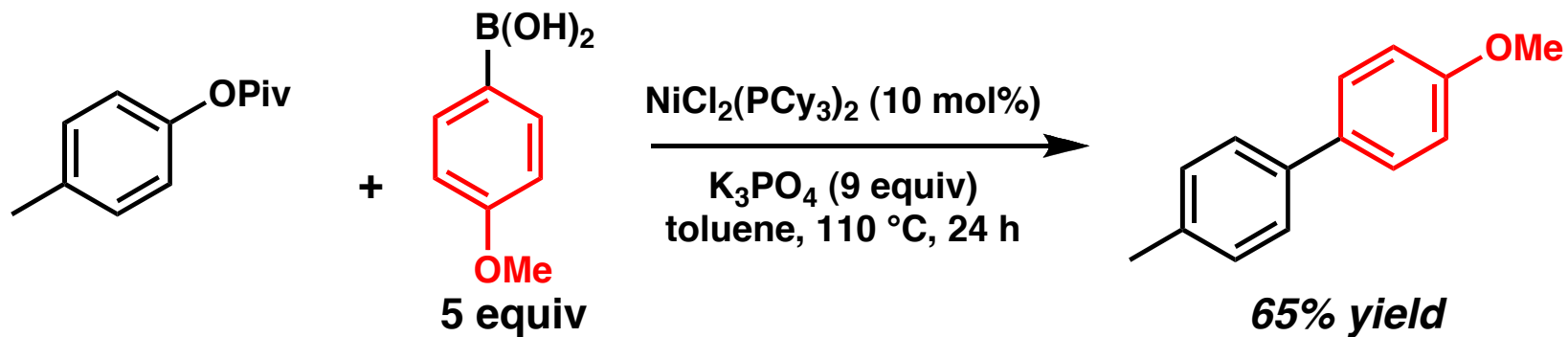
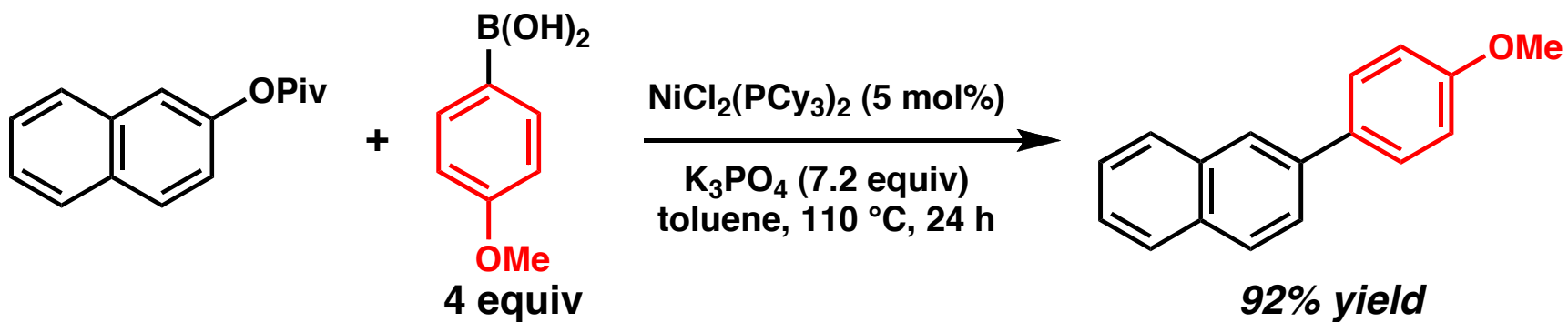
Shi, Z.-J.



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

Garg, N. K.



Ester Cross Coupling

Suzuki-Miyaura Coupling: Aryl Esters

Garg, N. K.

