



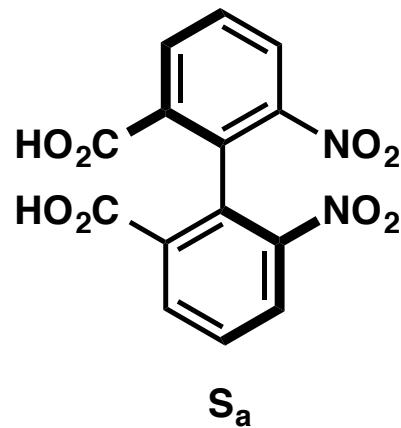
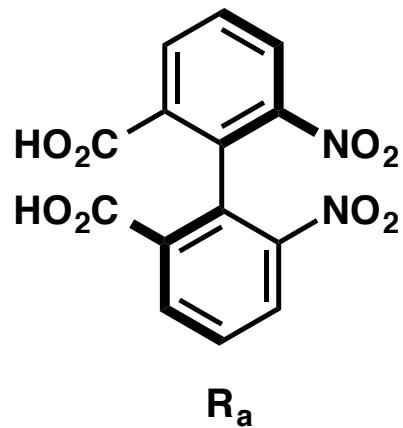
Atropisomerism

Metal-mediated
transformations to set
biaryl chirality

Nathaniel Kadunce
Literature Meeting
12 December 2014

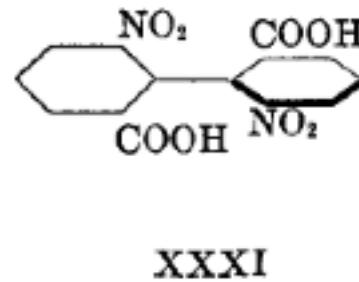
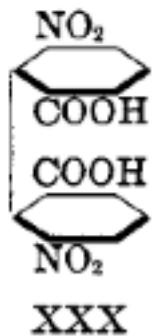
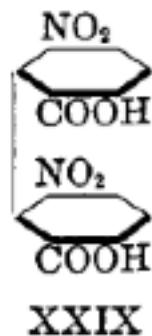
Atropisomerism

- *Atropos-* the “inflexible” or “without turn”
- Arise from hindered rotation about a single bond allowing for isolation of separate conformers



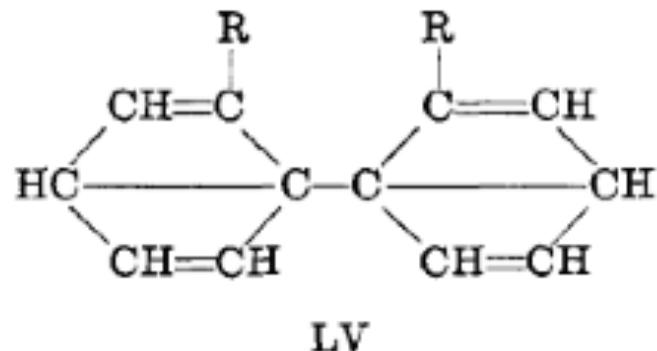
6,6'-dinitro-2,2'-diphenic acid,
the first experimentally described atropisomeric compound

First Evidence



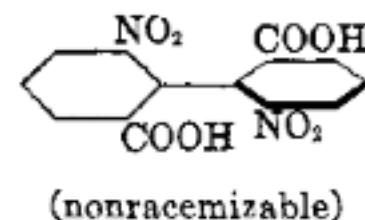
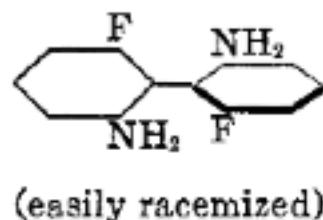
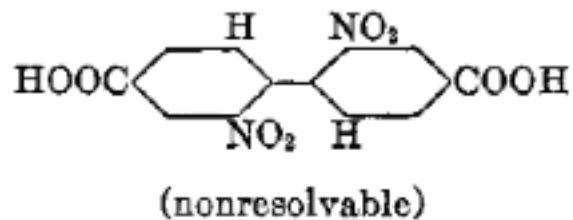
- Kaufler Hypothesis: *cis* and *trans* isomers explain optical properties
- Supported by a series of misassignments in derivative studies
- “It has been suggested by Carothers and by Mascarrelli that an objection to the Kaufler formula which has never been emphasized is the necessity of bending a bond to an angle of 90°. With a Kekule nucleus such a formula cannot be constructed, and each time that chemists adopt assumptions that cannot be reconciled with the Kekule nucleus they encounter difficulties.” –Adams, 1933

First Evidence



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Restricted Rotation



1926- The theory of restricted rotation (Turner, Le Fevre, Bell, and Kenyon)

- “Essentially, the theory states that substituents in 2, 2’, 6, 6’ positions in a diphenyl molecule can, by their interference, restrict the free rotation of the two nuclei around the common axis, thus preventing the rings from becoming coplanar and thereby producing in the molecule an asymmetric configuration.” –Adams, 1933
- Proven by resolution with alkaloids and subsequent racemization upon heating, studied extensively by dynamic NMR

Restricted Rotation

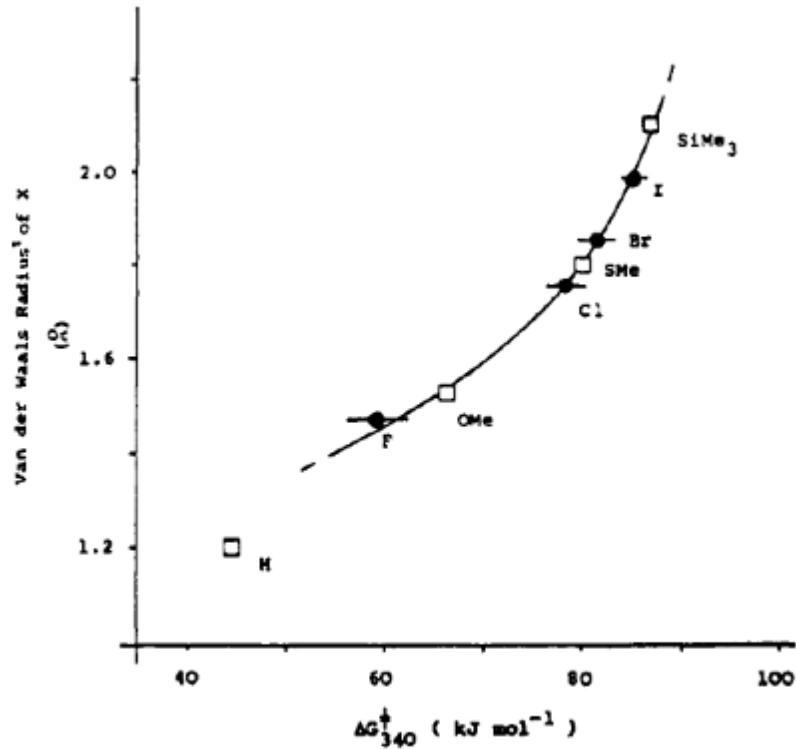
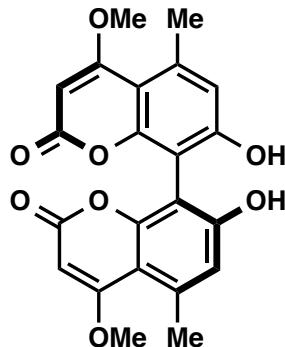


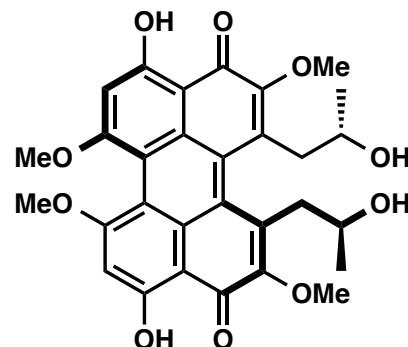
Figure 3. Plot of $\Delta G_{340}^{\ddagger}$ against the van der Waals radius¹ of X in some 6-(2-X-phenyl)-1,1,5-trimethylindans (**1**, Y = Me).

- Extensive dynamic NMR studies by Sternhell show direct correlation between van der Waals radii of substituents and energetic barrier to rotation.
- Increasing number of ortho-substituents and van der Waals radius of each decreases the rate of racemization
- Arbitrary definition of atropisomer is a half life of 1000 seconds at room temperature

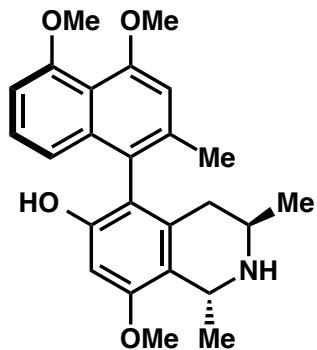
In Nature



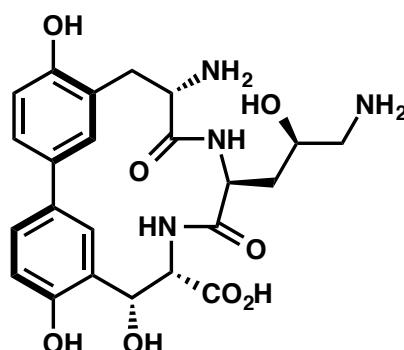
(+)-orlandin
anti-plant growth



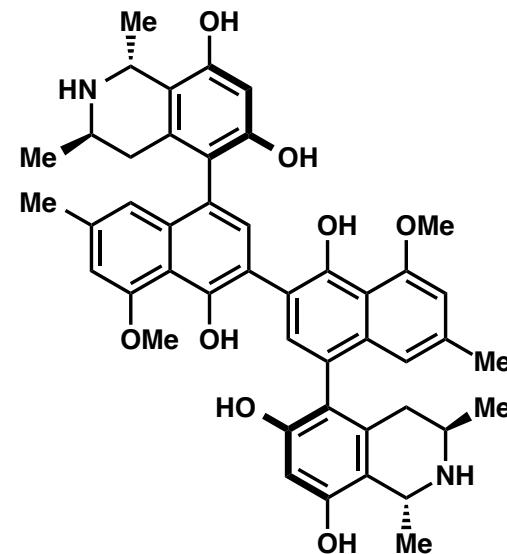
(-)-phleichrome
photodynamic ROS generation



ancistrocladine
anti-malarial

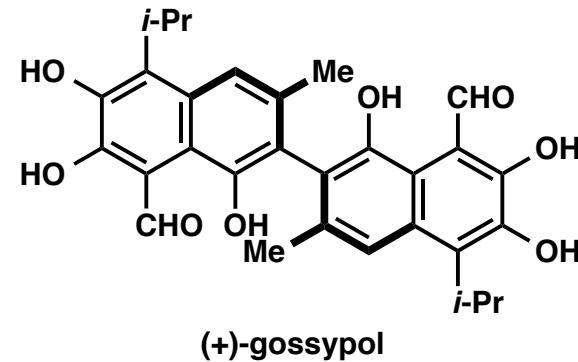
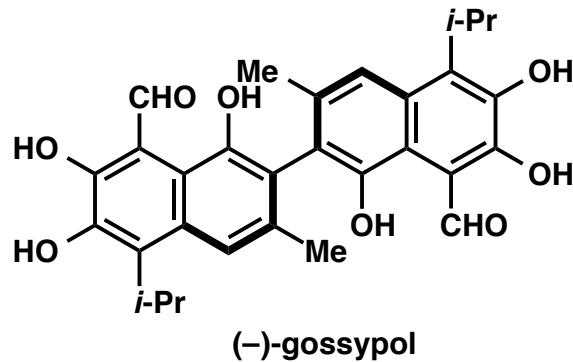


biphenomycin A
antibiotic

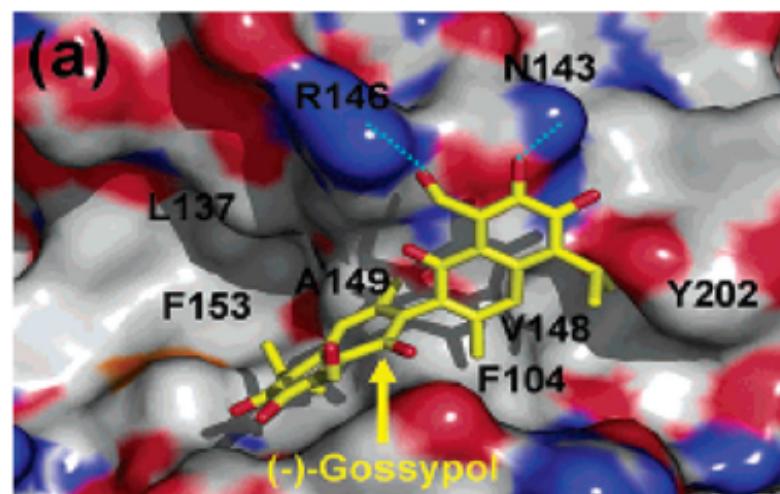


michellamine A
anti-HIV1 and HIV2

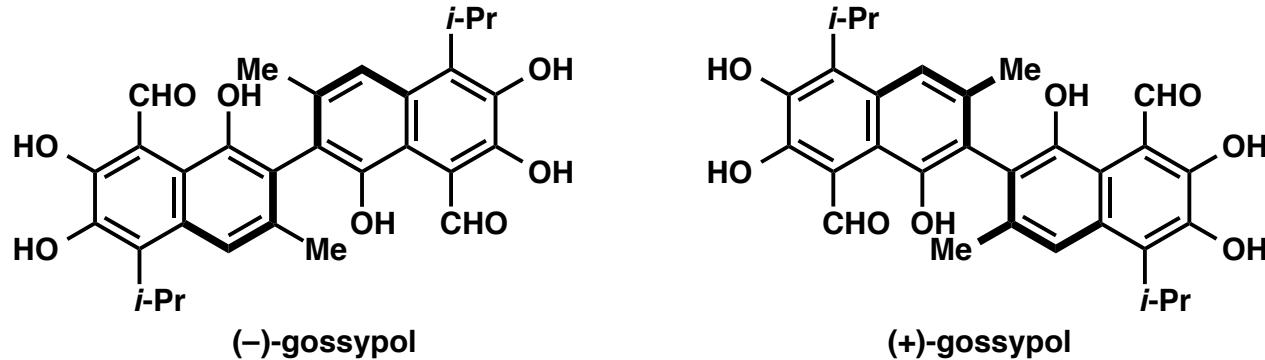
Eudysmic Ratios



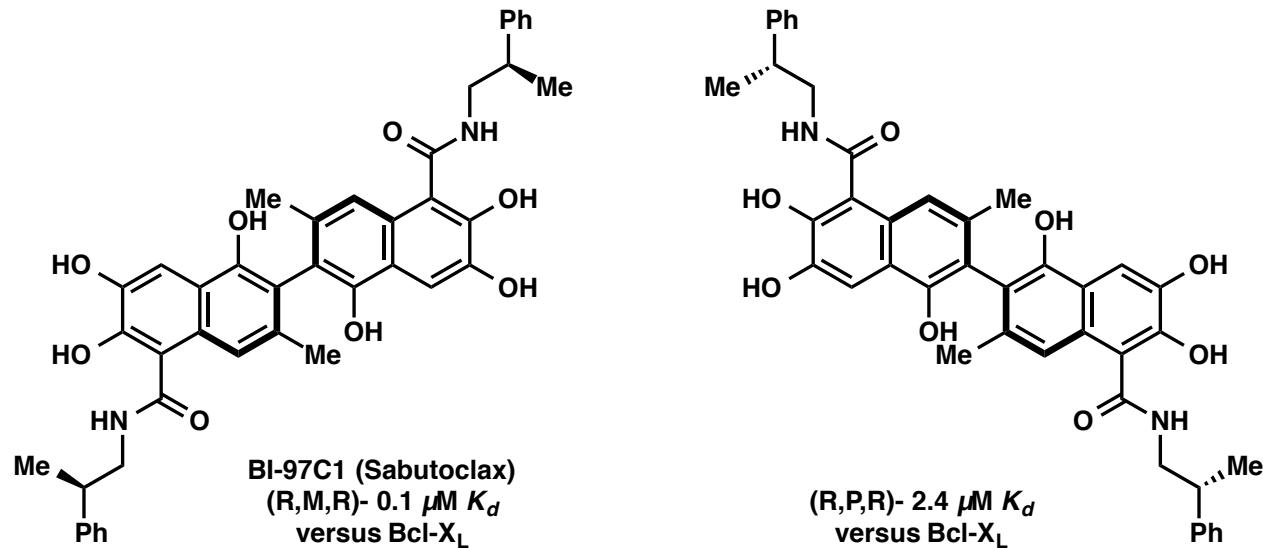
“A potent inhibitor of the antiapoptotic B-cell lymphoma/leukemia-2 (Bcl-2) family of proteins such as Bcl-XL and that the (M)-isomer is some tenfold more cytotoxic than the (P)-isomer.”



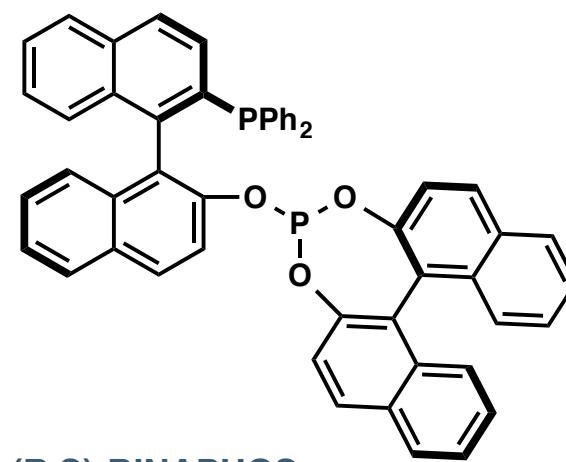
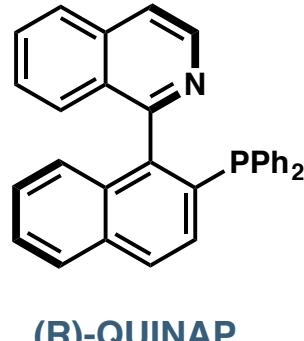
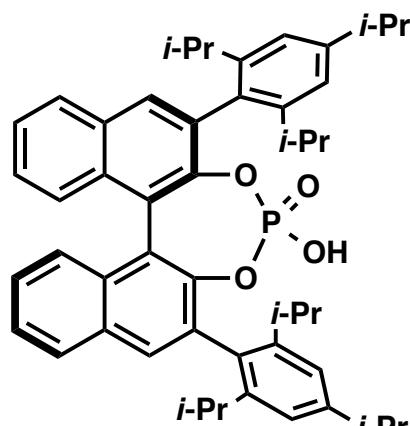
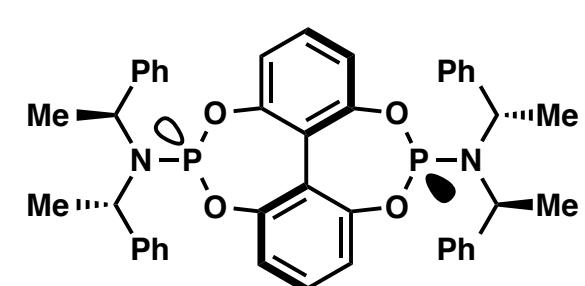
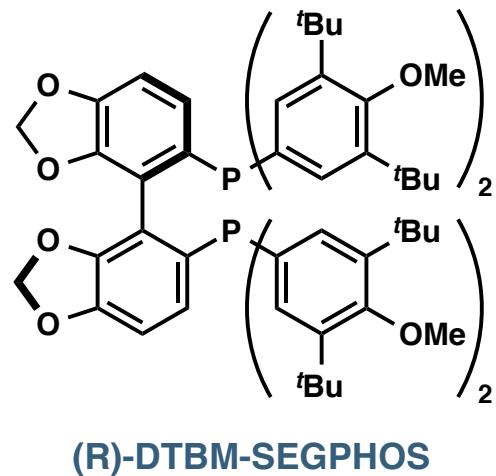
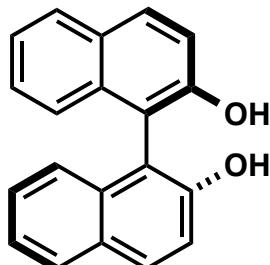
Eudysmic Ratios



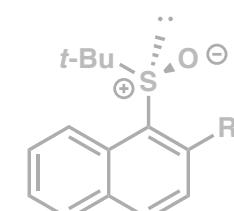
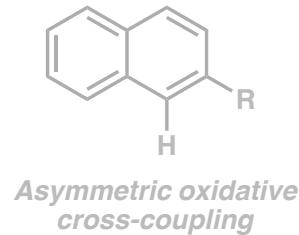
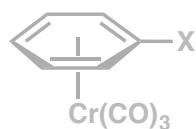
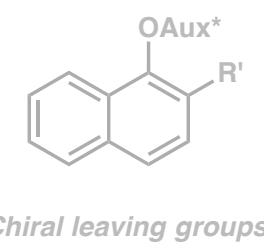
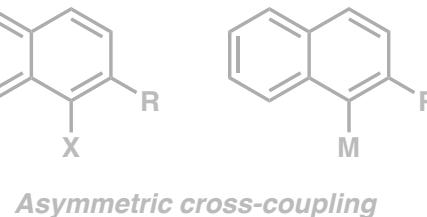
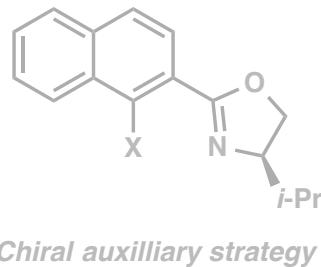
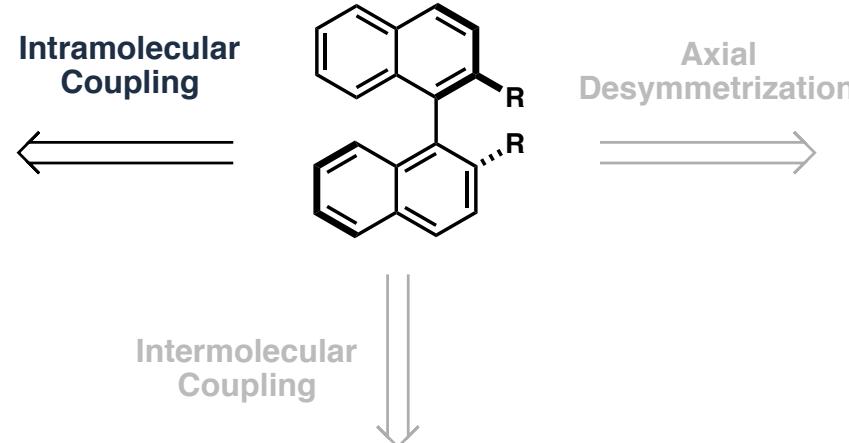
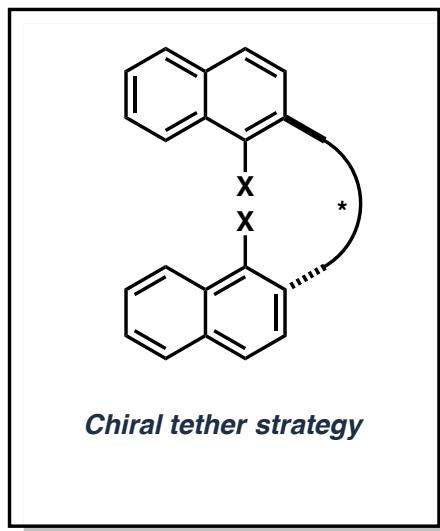
-Derivatization led to increase in activity and in eudysmic ratio from 10 to 24.



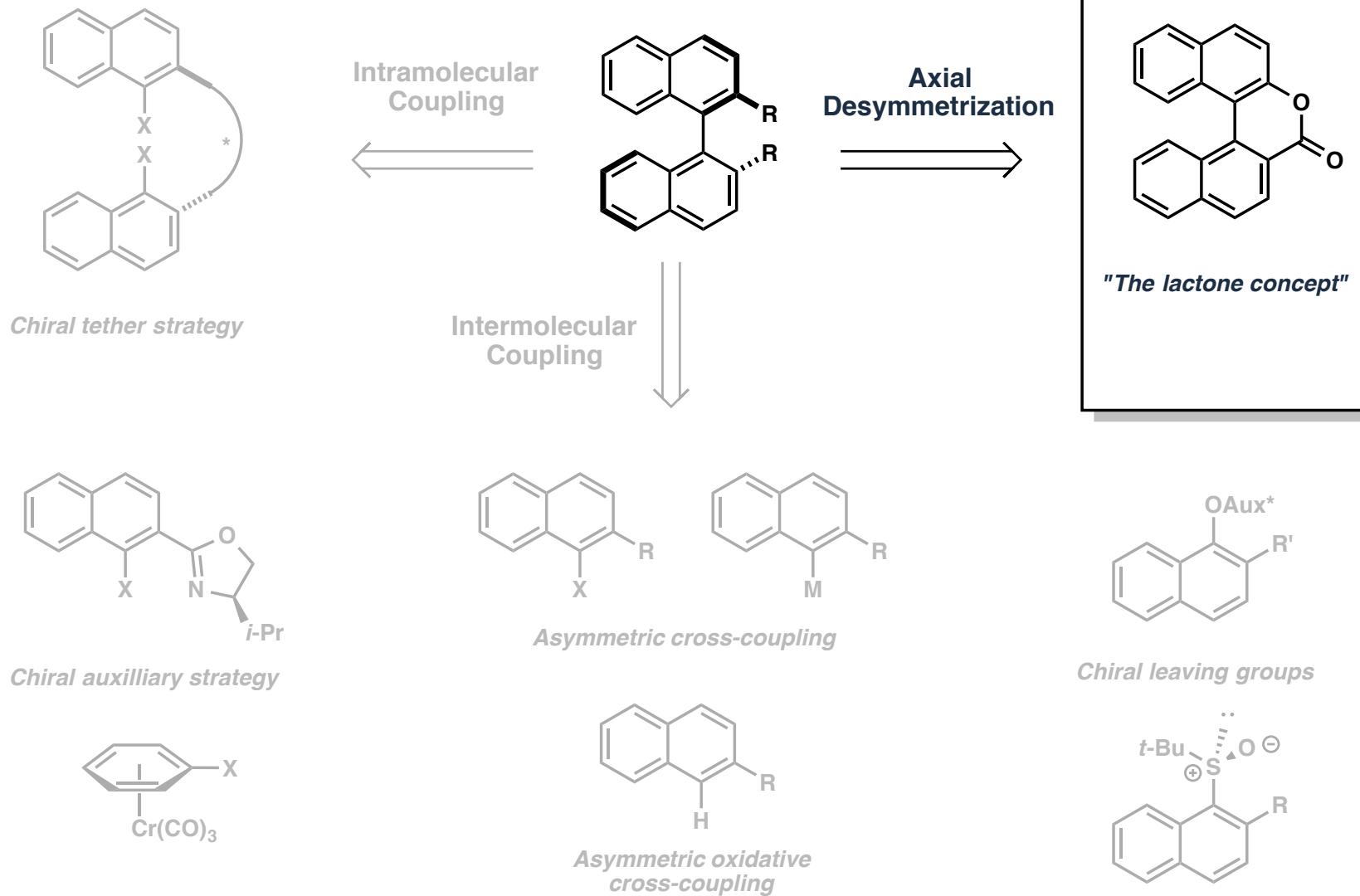
Atropchiral Ligands and Catalysts



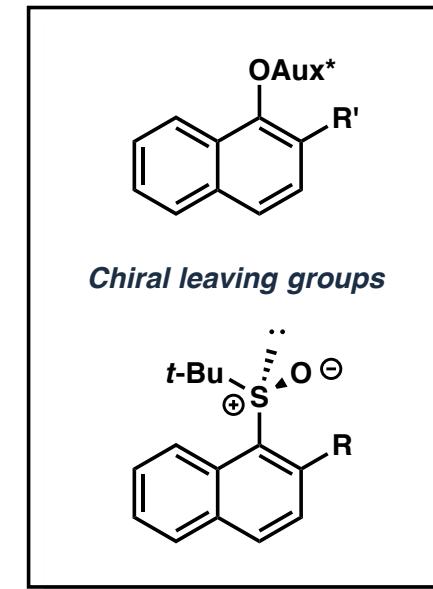
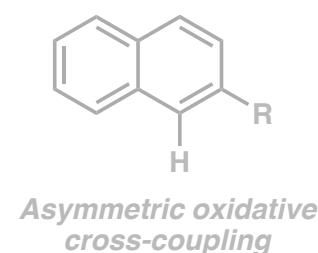
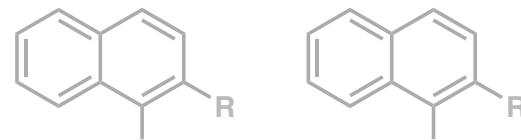
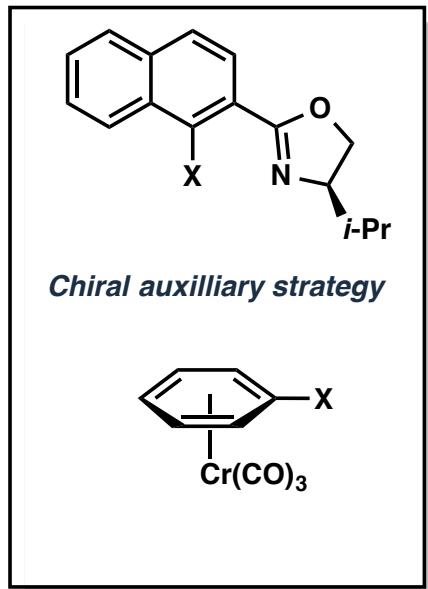
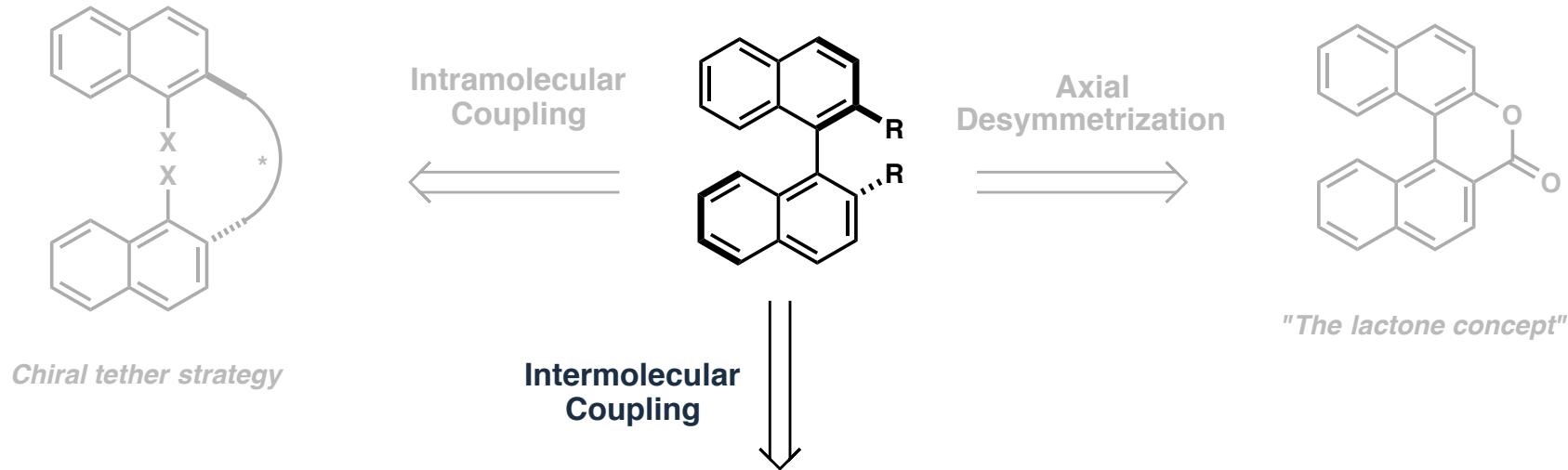
Methods and Approaches



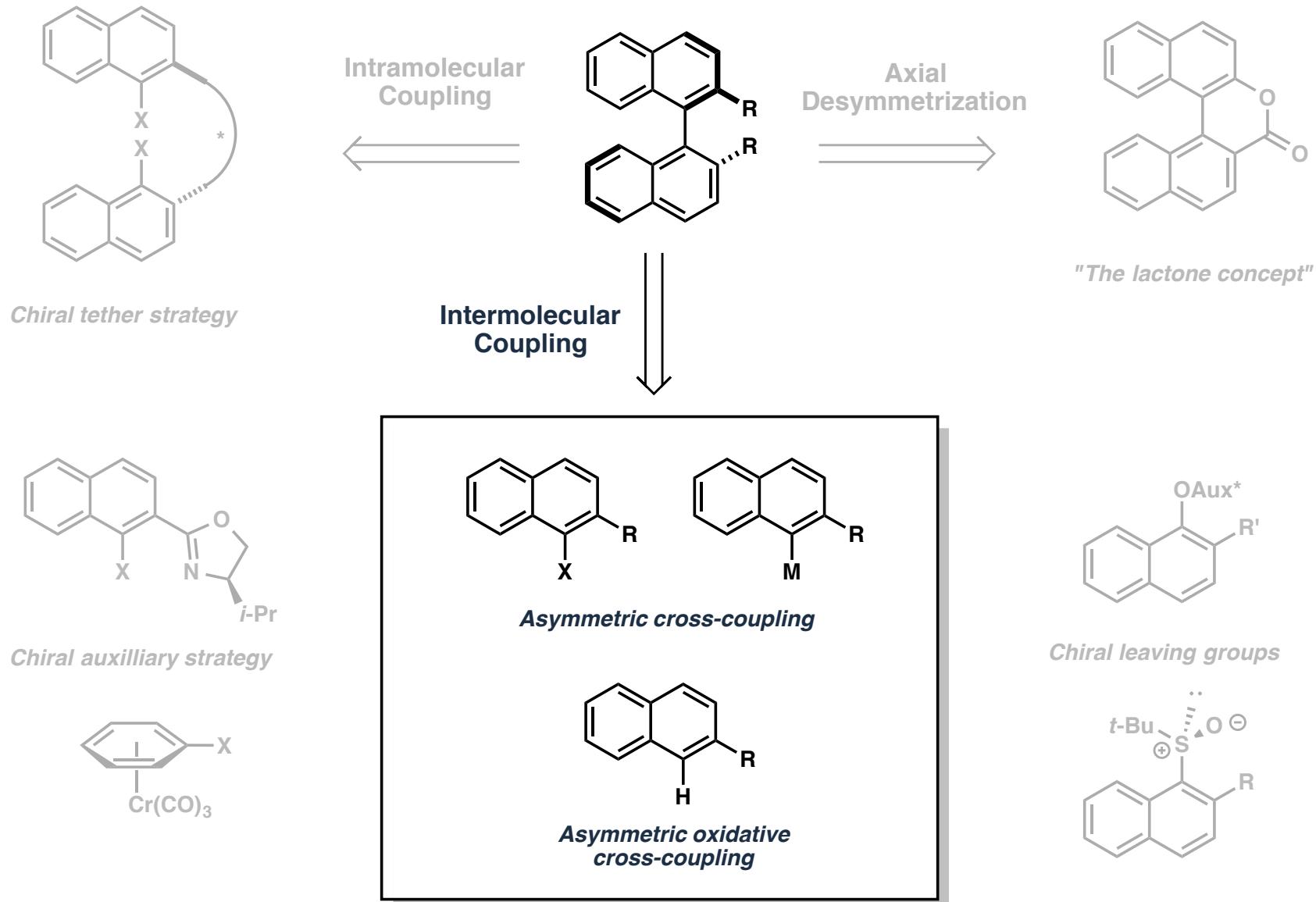
Methods and Approaches



Methods and Approaches



Methods and Approaches

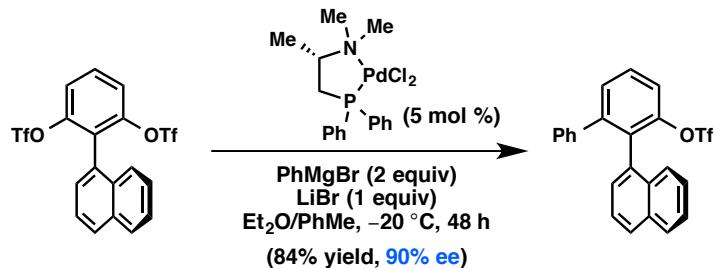


Redox-Neutral Couplings

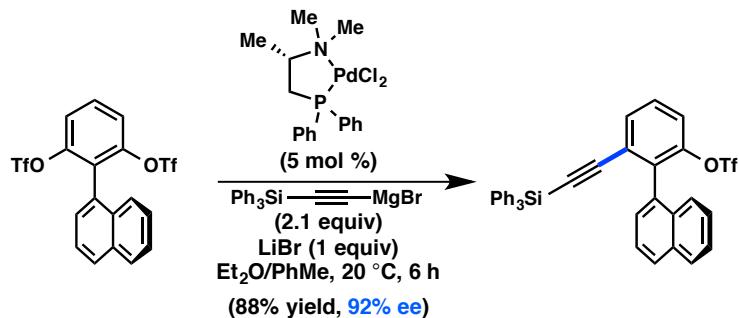
1. Enantioselective cross-coupling of a difunctionalized achiral biaryl substrate
2. Dynamic kinetic asymmetric transformations via cross-coupling of a racemic substrate
3. $\text{sp}^2\text{-sp}^2$ Cross-coupling producing axial chirality in the bond-forming event

Enantioposition-selective

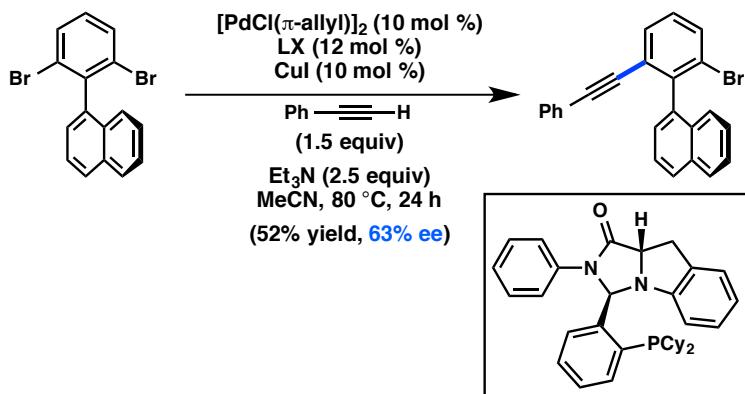
a) Kumada-Corriu cross-coupling of aryl Grignard reagent



b) Kumada-Corriu cross-coupling of alkynyl Grignard reagent



c) Sonogashira cross-coupling of alkynes



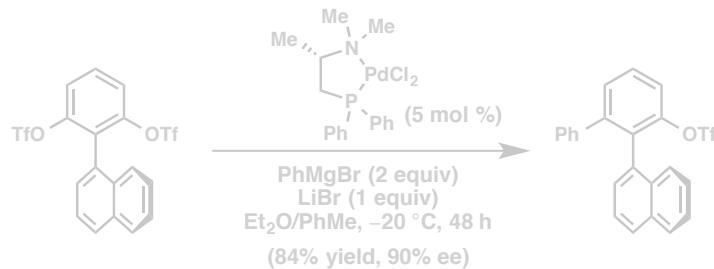
Uozumi, Y., et al. *J. Am. Chem. Soc.* **1995**, *117*, 9101

Kamikawa, T.; Hayashi, T. *Tetrahedron* **1999**, *55*, 3455.

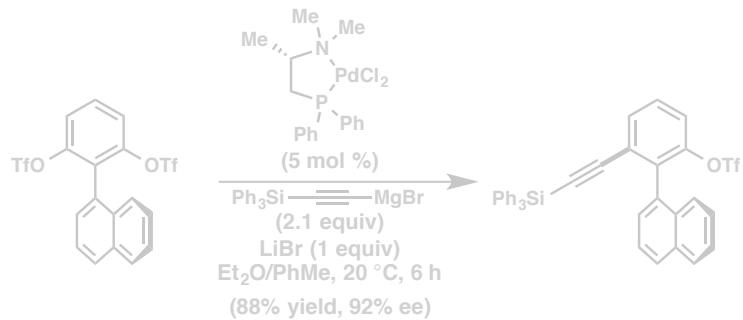
Kamikawa, T.; Uozumi, Y.; Hayashi, T. *Tetrahedron Letters* **1996**, *37*, 3161.

Enantioposition-selective

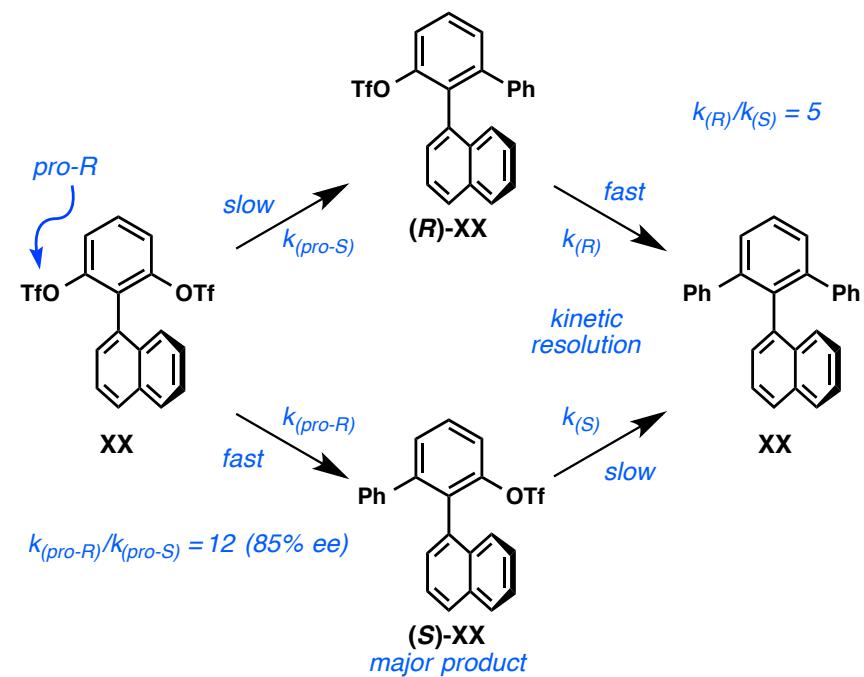
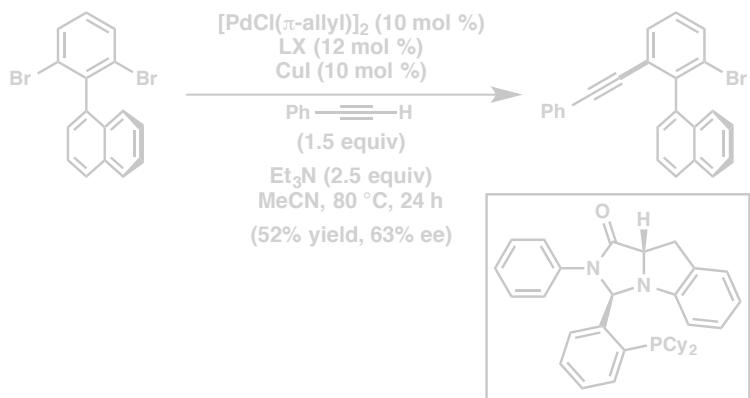
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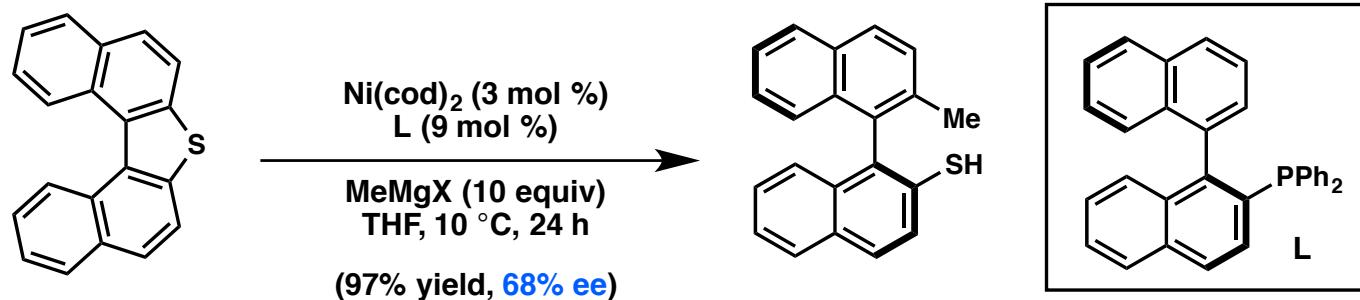
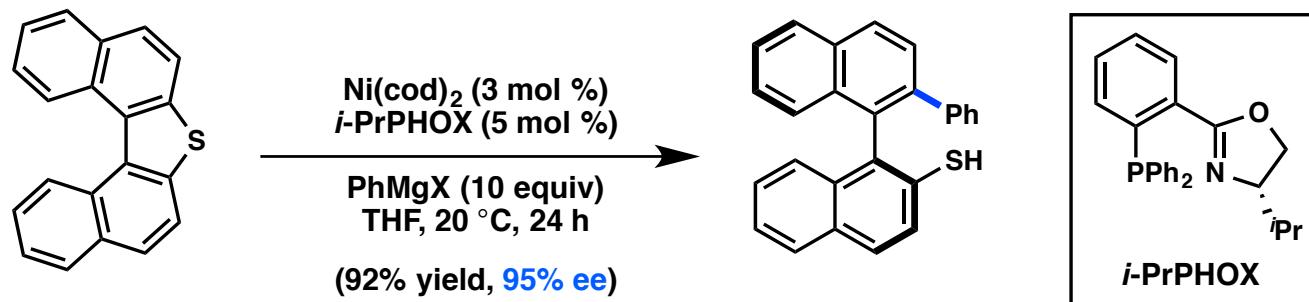
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Kamikawa, T.; Hayashi, T. *Tetrahedron* **1999**, *55*, 3455.

Kamikawa, T.; Uozumi, Y.; Hayashi, T. *Tetrahedron Letters* **1996**, *37*, 3161.

DYKAT

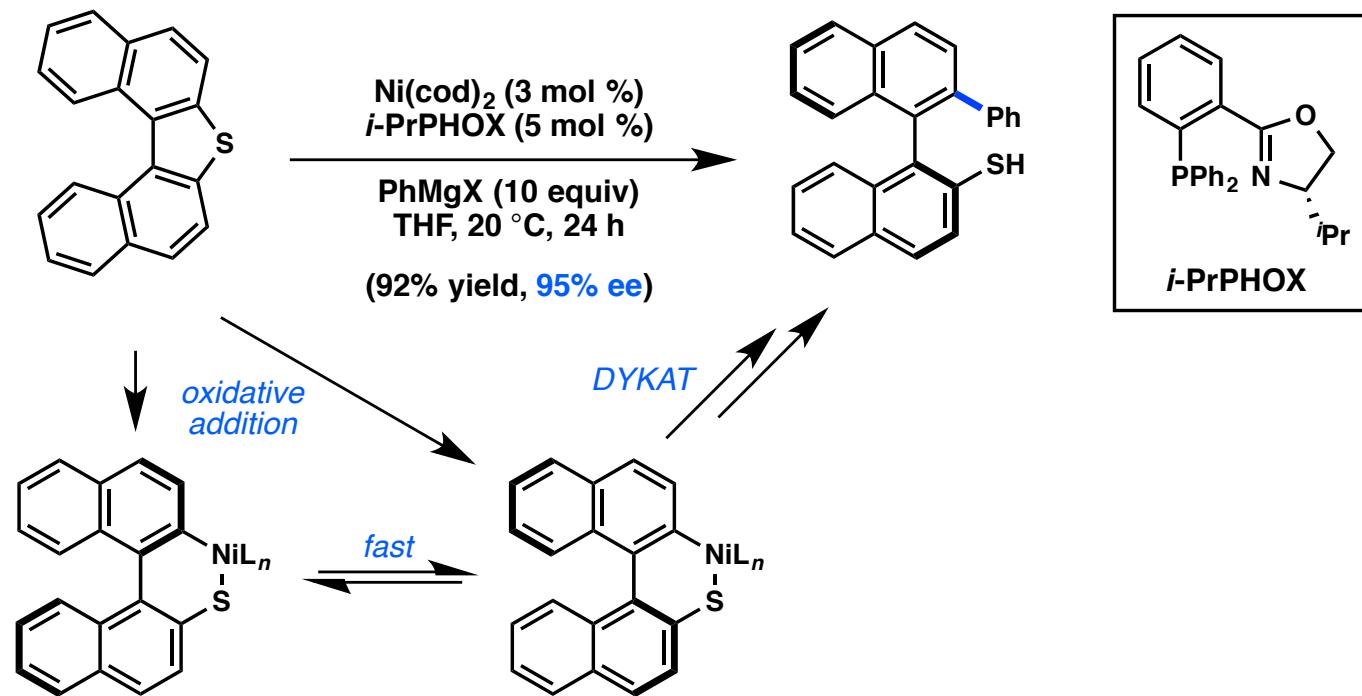
Hayashi, 2002 and 2004:



Shimada, T.; Cho, Y. H.; Hayashi, T. *J. Am. Chem. Soc.* **2002**, *124*, 13396.
Cho, Y. H.; Kina, A.; Shimada, T.; Hayashi, T. *J. Org. Chem.* **2004**, *69*, 3811.

DYKAT

Hayashi, 2002 and 2004:

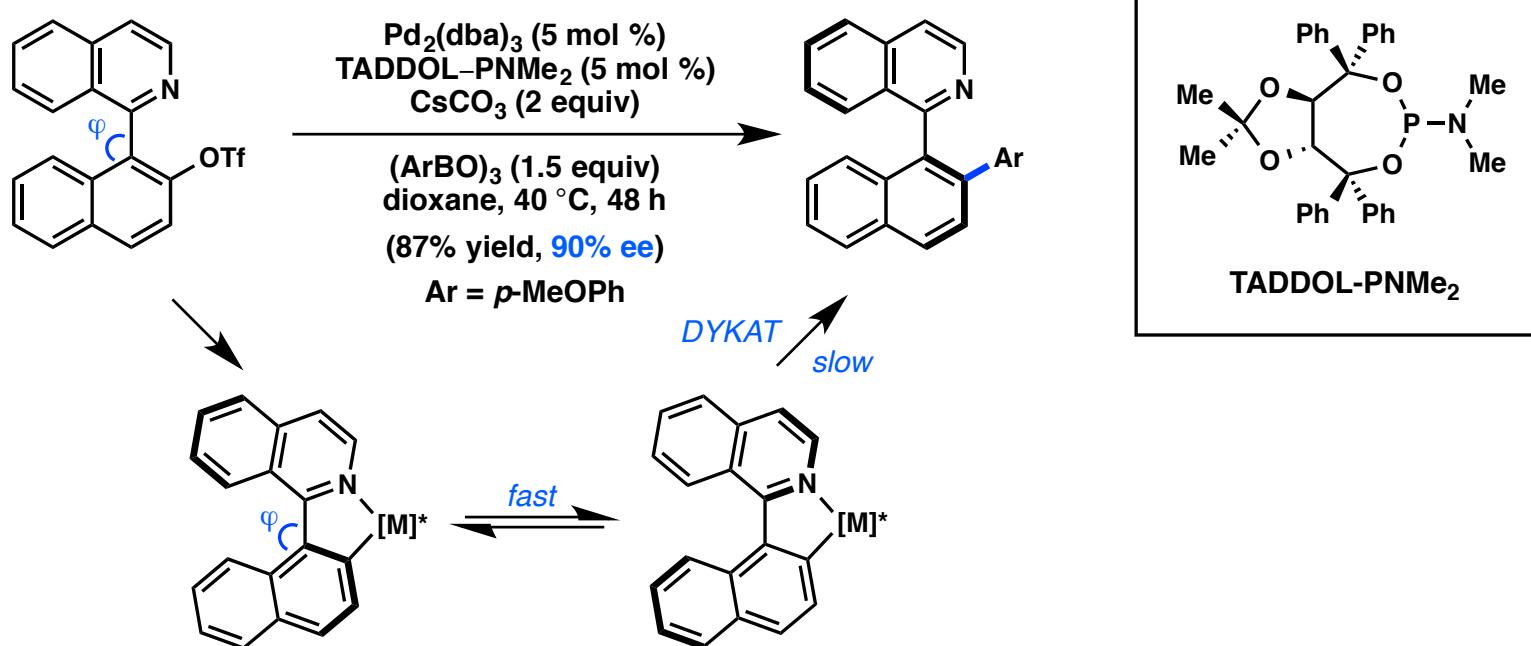


Shimada, T.; Cho, Y. H.; Hayashi, T. *J. Am. Chem. Soc.* **2002**, *124*, 13396.

Cho, Y. H.; Kina, A.; Shimada, T.; Hayashi, T. *J. Org. Chem.* **2004**, *69*, 3811.

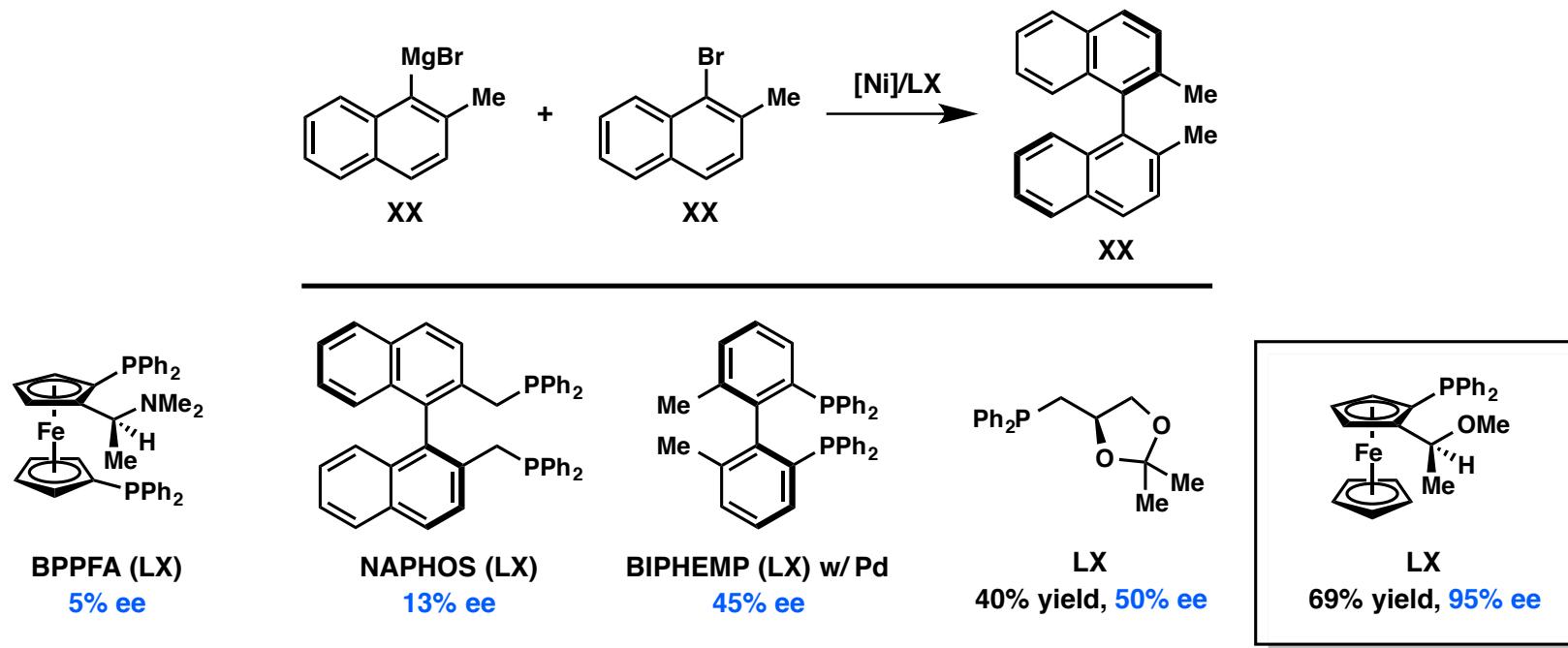
DYKAT

Lassaletta, 2013:



Cross-Coupling

-The 1980's: Asymmetric Kumada-Corriu coupling



Tamao, K.; Sumitani, K.; Kumada, M. *J. Am. Chem. Soc.* **1972**, *94*, 4374

Corriu, J. P.; Masse, J. P. *J. Chem. Soc., Chem. Commun.* **1972**, 144.

Kumada, M., et al. *Chemistry Letters* **1975**, 133.

Kumada, M. et al. *Tetrahedron Letters* **1977**, *18*, 1389.

Frejd, T.; Klingstedt, T. *Acta Chemica Scandinavica* **1989**, *43*, 670.

Terfort, A.; Brunner, H. *J. Chem. Soc., Perkin Trans. 1* **1996**, 1467.

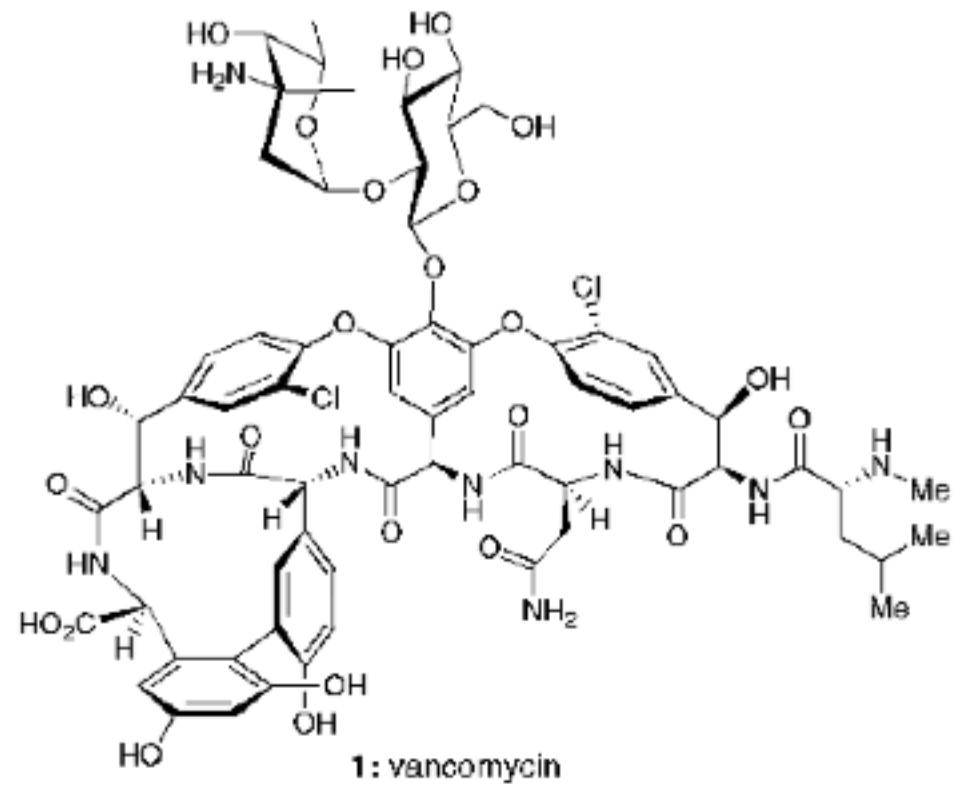
Kumada, M. et al. *Tetrahedron Letters* **1977**, *18*, 1389.

Ito, Y. et. al. *J. Am. Chem. Soc.* **1988**, *110*, 8153

Hayashi, T.; Hayashizaki, K.; Ito, Y. *Tetrahedron Letters* **1989**, *30*, 215.

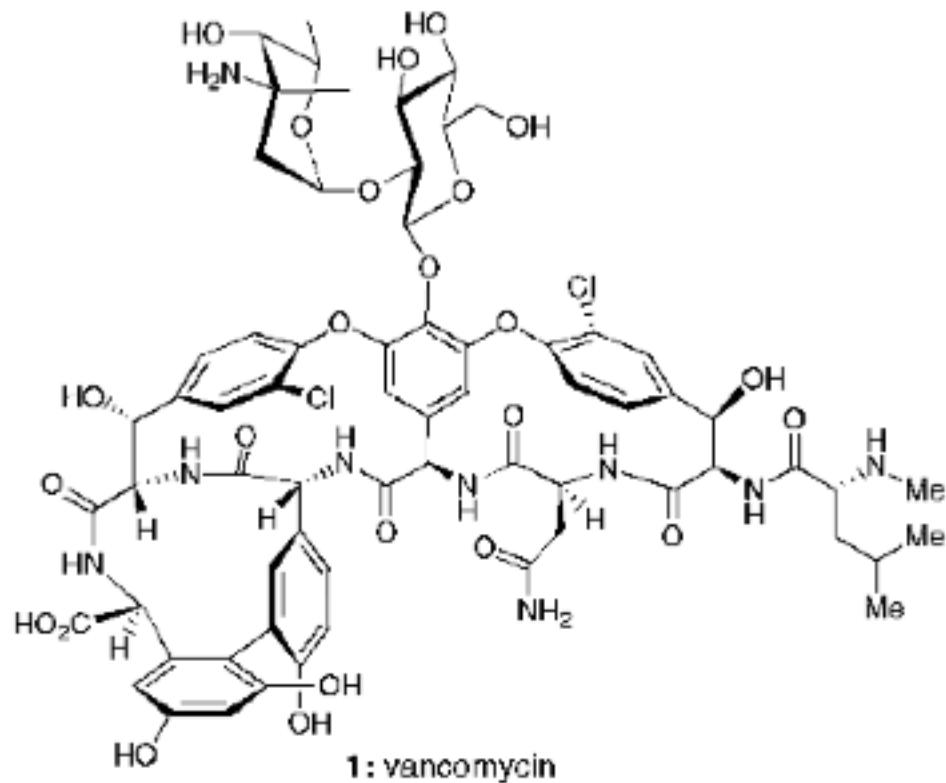
Cross-Coupling

-Skipping ahead to 1999.



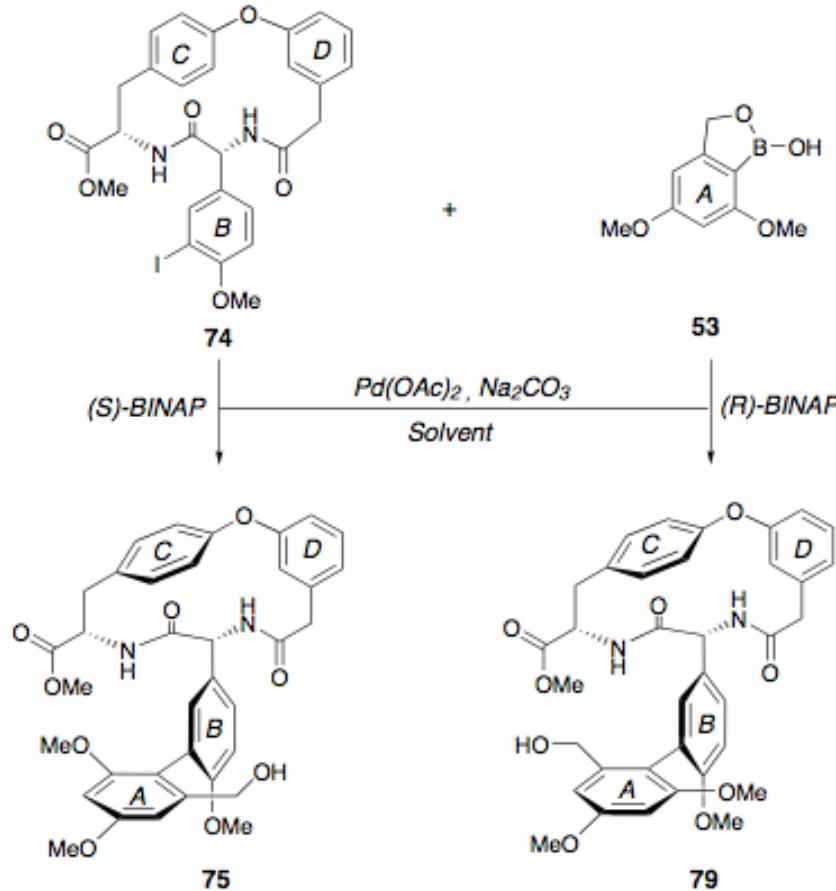
Vancomycin

-Skipping ahead to 1999.



- Isolated in 1953 by Eli Lilly from soil bacterium *Amycolatopsis orientalis*.
- Antibiotic used for infections by Gram-(+) bacteria, especially those resistant to more common drugs (e.g. MRSA).
- Inhibits biosynthesis of Gram-(+) bacterial cell wall.

Vancomycin



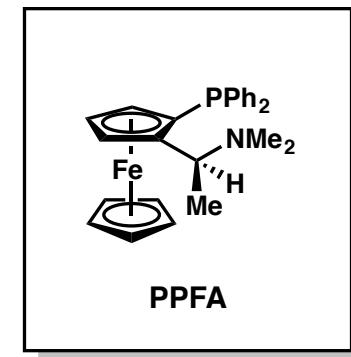
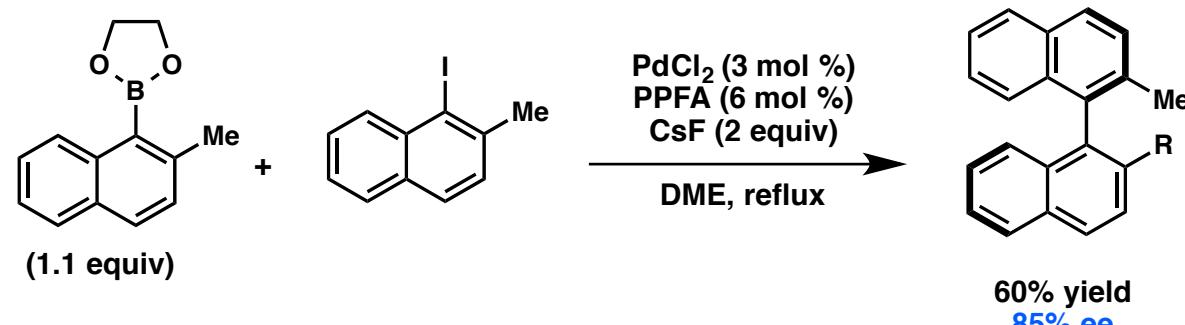
- No intrinsic substrate control in Suzuki reaction.
- Chiral ligand and condition screening identified BINAP as providing almost complete catalyst control.
- First catalyst-controlled enantioselective Suzuki coupling!

Entry	Ligand	Solvent	Temp (°C)	Time (h)	Yield (%)	Ratio (75:79)
1	Ph_3P	PhMe	90	2	80	1:1
2	BINAP	PhMe	90	12	trace	-
3	BINAP	THF	65	12	trace	-
4	(S)-BINAP	DMF	80	8	60	2.3:1
5	(S)-BINAP	PhMe:THF(1:1)	70	5	40(70 ^[a])	>95:5 ^[b]
6	(R)-BINAP	PhMe:THF(1:1)	70	5	40(70 ^[a])	<5:95 ^[b]

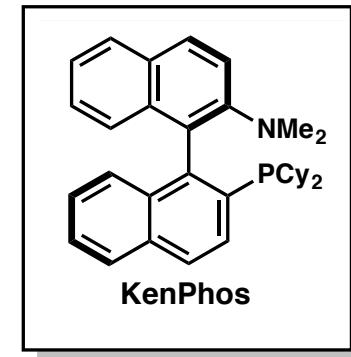
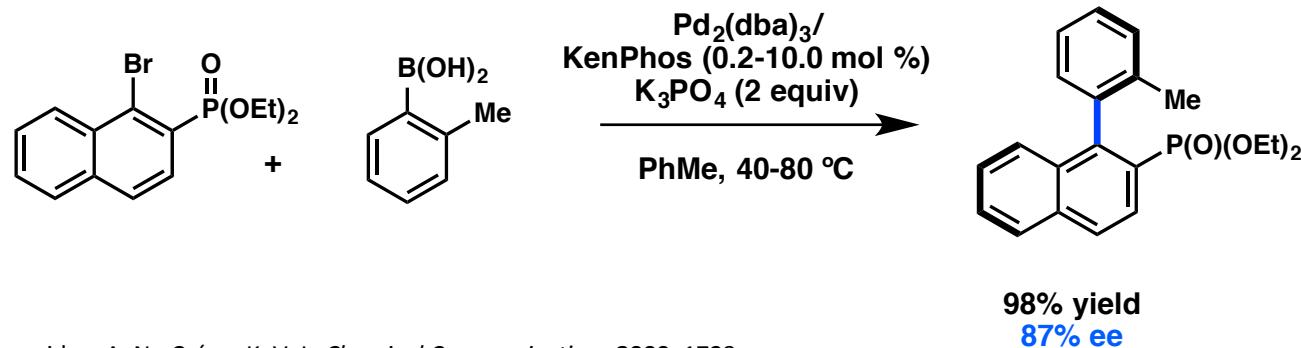
Suzuki Coupling

-The 2000's, the reign of Boron

Cammidge, 2000:



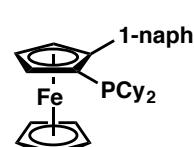
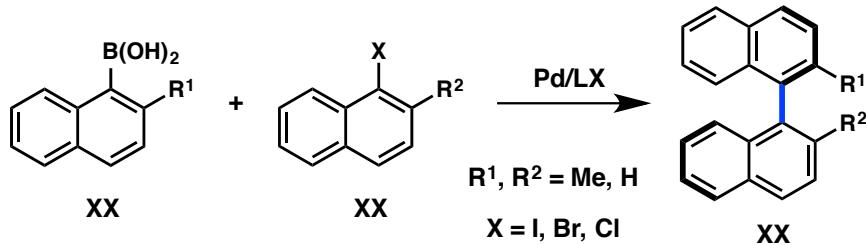
Buchwald, 2000:



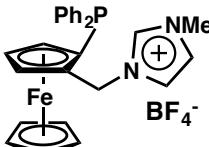
Cammidge, A. N.; Crépy, K. V. L. *Chemical Communications* **2000**, 1723.

Yin, J. J.; Buchwald, S. L. *J. Am. Chem. Soc.* **2000**, 122, 12051

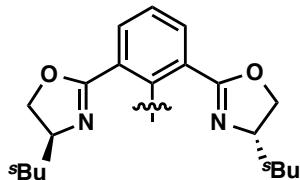
Suzuki Coupling



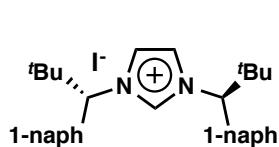
65% yield
54% ee
(Johannsen, 2003)



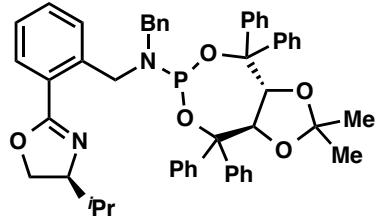
88% yield
42% ee
(Labande, 2010)



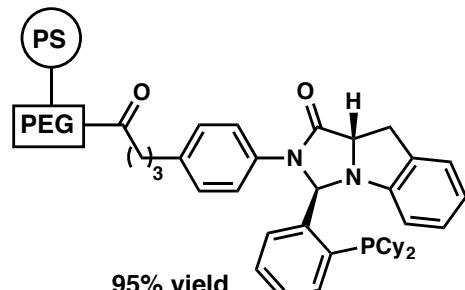
61% yield
49% ee
(Iwasa, 2007)



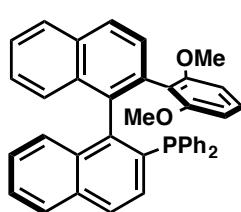
85% yield
80% ee
(Kündig, 2014)



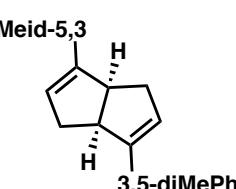
54% yield
46% ee
(Guiry, 2007)



95% yield
94% ee
(Uozumi, 2009)



73% yield
62% ee
(Putala, 2013)



78% yield
90% ee
(Lin, 2010)

Jensen, J. F.; Johannsen, *Org. Lett.* **2003**, 5, 3025.

Labande, A. et al. *New Journal of Chemistry* **2014**, 38, 338.

Uozumi, Y., et al. *Angew. Chem. Int. Ed.* **2009**, 48, 2708.

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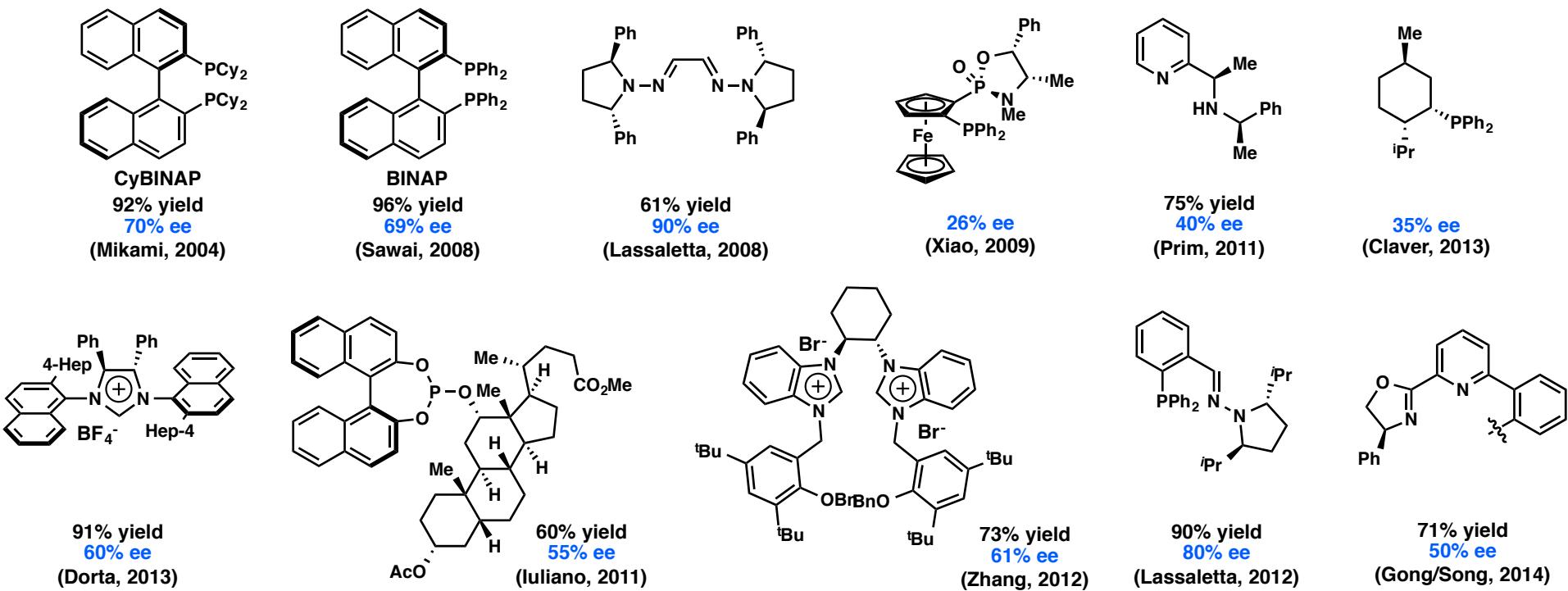
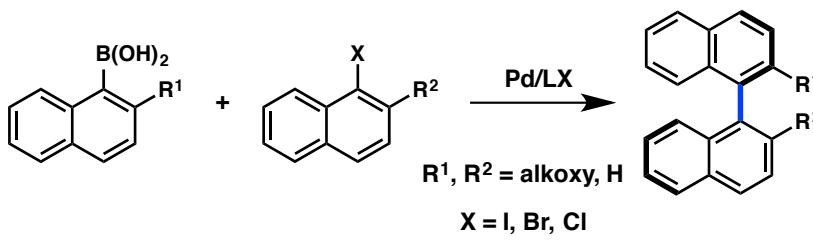
Benhamou, L.; Besnard, C.; Kundig, E. P. *Organometallics* **2014**, 33, 260.

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Broniger, R. P. J.; Guiry, P. J. *Tetrahedron-Asymmetry* **2007**, 18, 1094.

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Suzuki Coupling



Mikami, K.; Miyamoto, T.; Hatano, M. *Chem. Commun.* **2004**, 2082.

Sawai, K. *et al.* *Angew. Chem. Int. Ed.* **2008**, 47, 6917.

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Prim, D., *et al.* *Organometallics* **2011**, 30, 4074.

Dorta, R. *et al.* *Synlett* **2013**, 24, 1215.

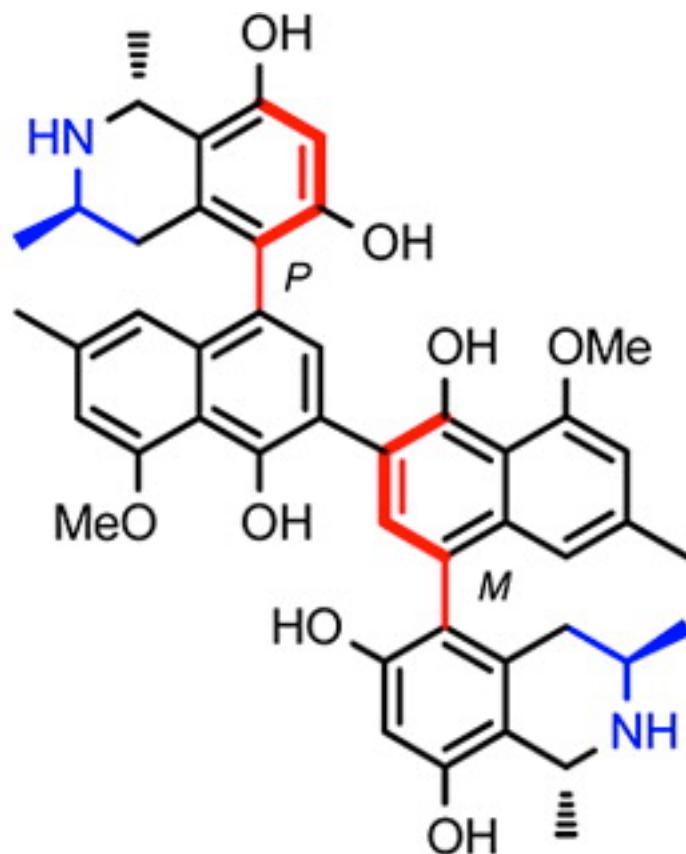
Zhang, D. *et al.* *Organometallics* **2014**, 33, 876.

Jumde, V. R.; Iuliano, A. *Tetrahedron-Asymmetry* **2011**, 22, 2151.

Claver, C. J. *et al.* *Organomet. Chem.* **2013**, 743, 31.

Gong, J. F.; Song, M. P., *et al.* *Organometallics* **2014**, 33, 194.

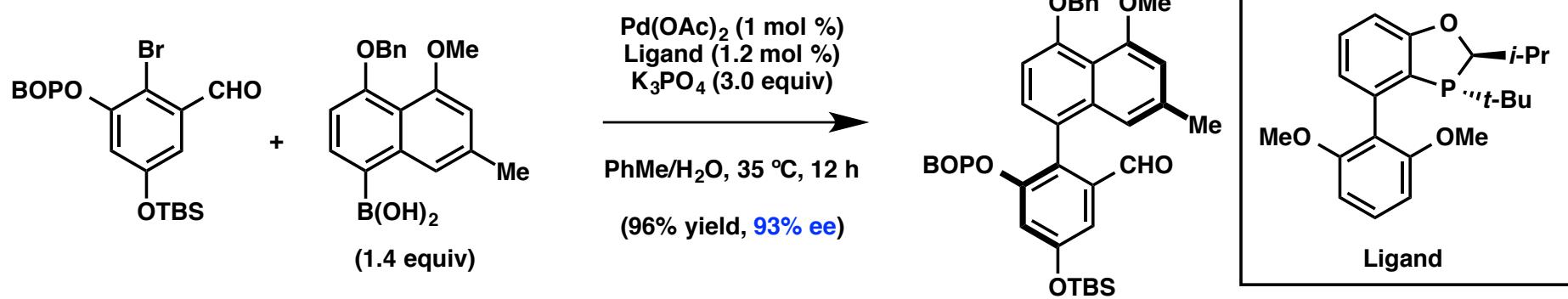
Michellamine B



- Isolated in 1991 by Boyd and coworkers
- Anti- HIV-1 (EC_{50} 10 μM) and HIV-2 (EC_{50} 2 μM) activity including resistant strains as well.
- Configurationally labile at binaphthyl junction.
- Significant activity dependence on stereochemistry of naphthylisoquinoline axes.
- Previous approaches: diastereoselective biaryl coupling; chiral Cr-complexes, and asymmetric lactone cleavage.

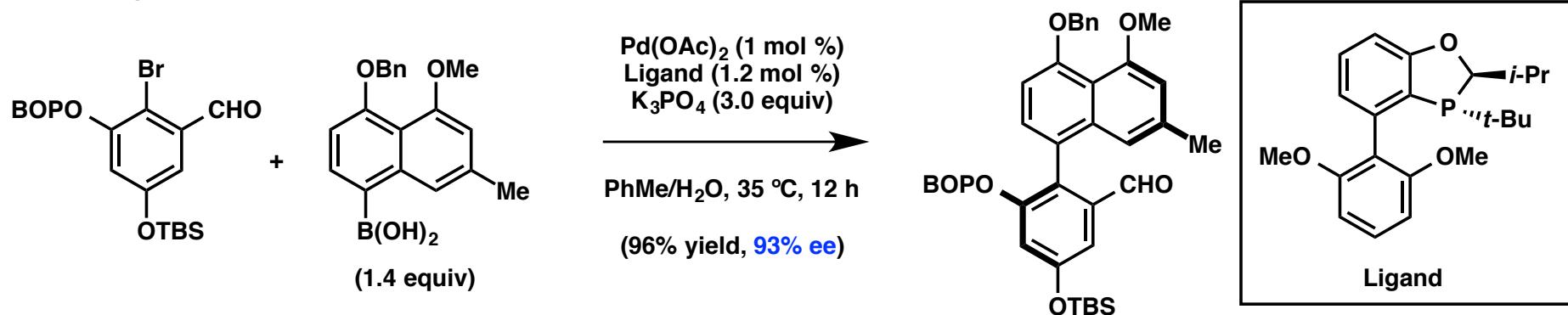
Michellamine B

Tang, 2014:

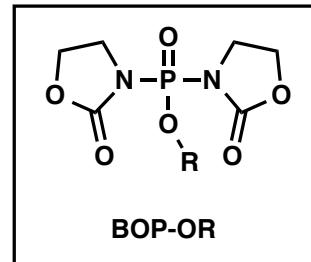


Michellamine B

Tang, 2014:

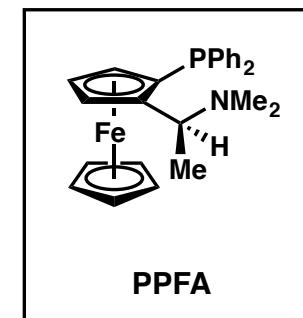
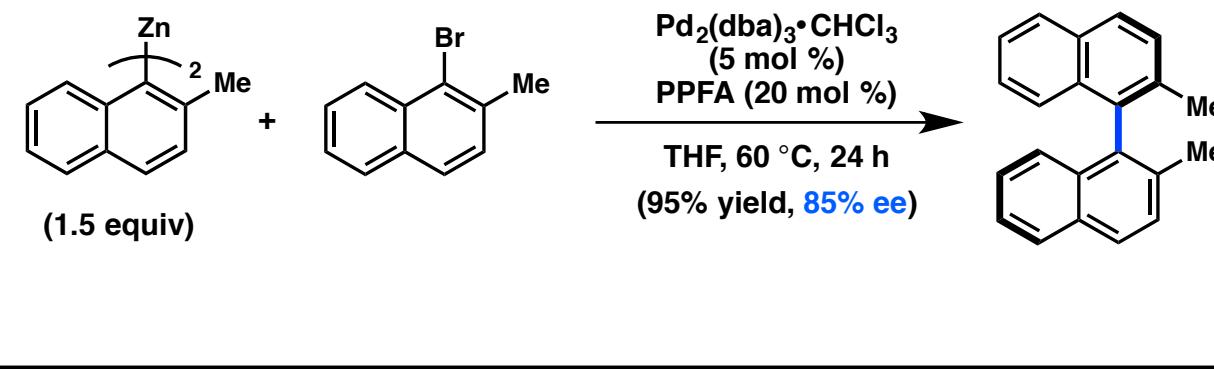


- Building from Buchwald's work, screened a variety of ligands against various *ortho*-directing groups, arrived at BOP:
- First catalytic asymmetric preparation of Michellamine B in 20+ years of efforts.
- Very mild conditions employed for hindered Suzuki coupling.

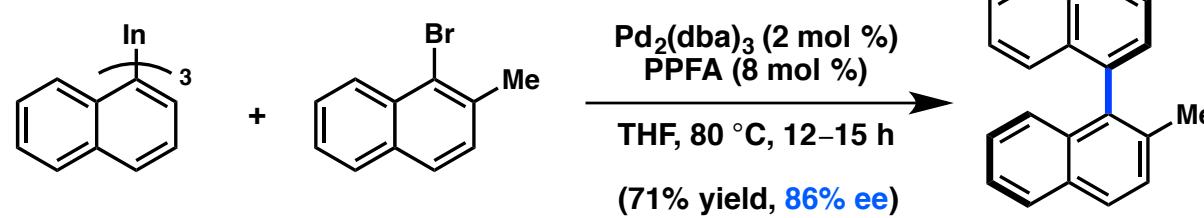


Cross-Coupling: Other Nucleophiles

Zinc: Espinet, 2006



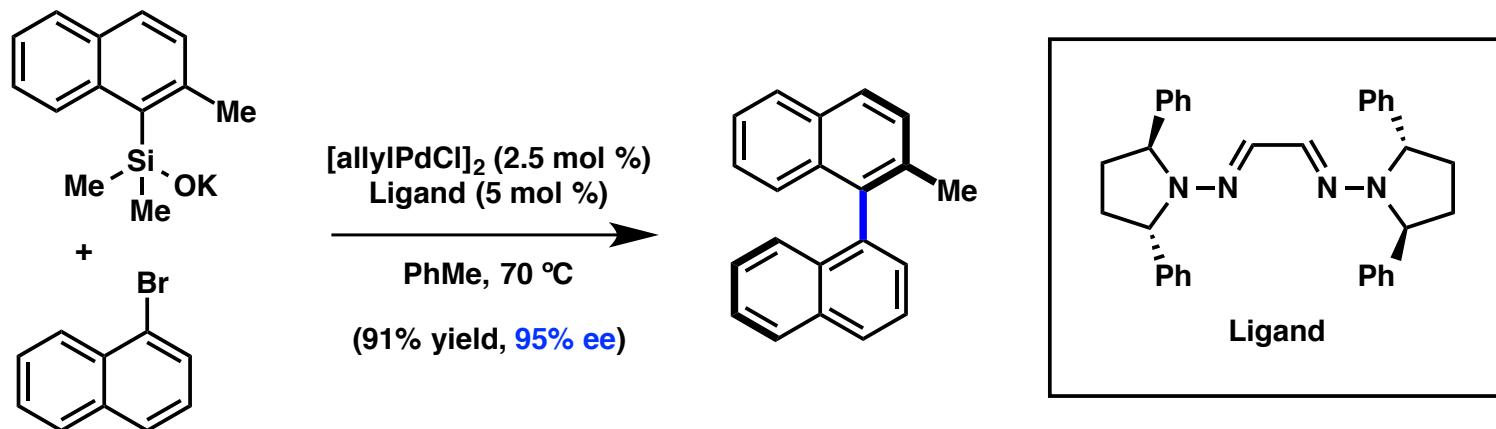
Indium: Sarandeses, 2013



Espinet, P. et al. *Tetrahedron-Asymmetry* **2006**, 17, 2593.
Sarandeses, L. A., et al. *Eur. J. Org. Chem.* **2013**, 2555.

Cross-Coupling: Other Nucleophiles

Silicon: Denmark, 2014

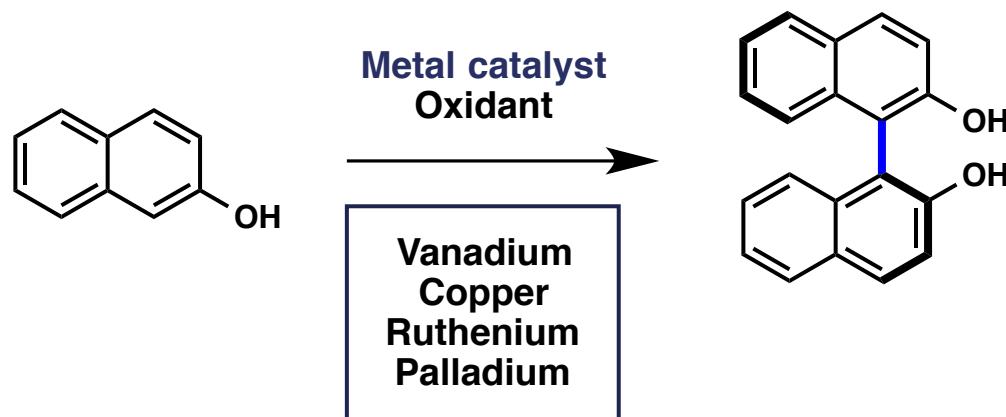


-Coordinating substituents at 2-position decrease ee by competitive coordination with Pd.

-Nucleophile/Electrophile swap gives identical ee. Combined with computational work indicates stereodetermining reductive elimination.

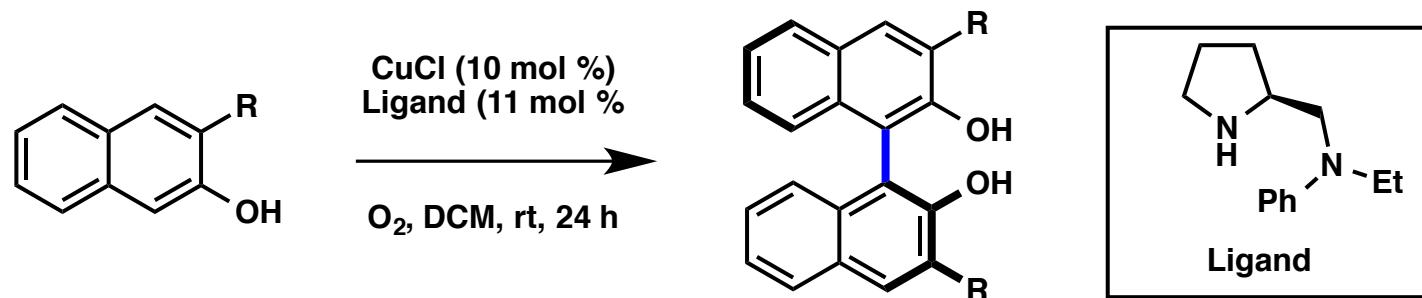
Oxidative Coupling- Intro

1. Catalytic asymmetric dimerization of activated phenols and naphthols
2. Oxidative cross-coupling of electronically differentiated arenes

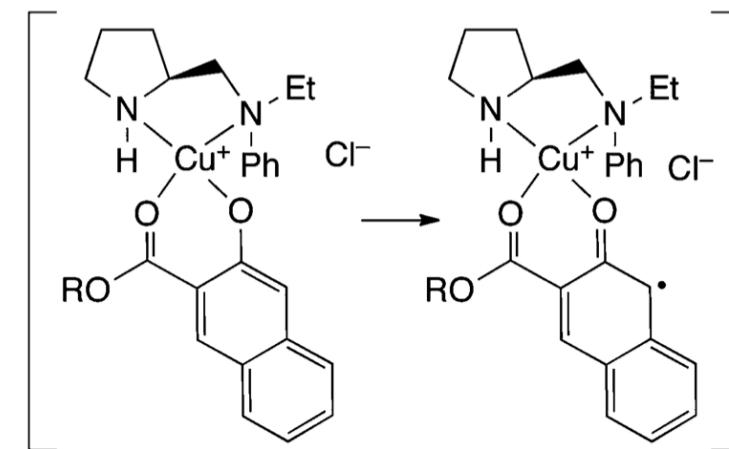


Oxidative Coupling- Nakajima

Nakajima, 1995

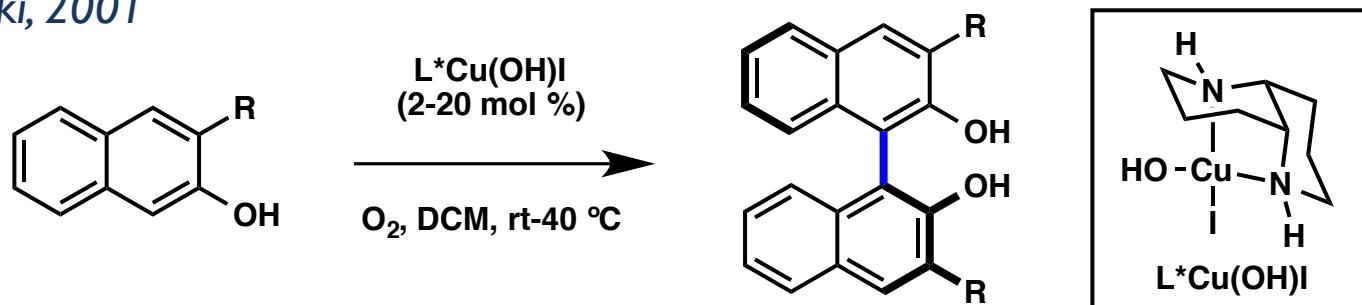


R	yield (%)	ee (%)
CO ₂ Me	85	78
CO ₂ Et	77	73
CO ₂ Bn	77	76
CO ₂ <i>t</i> -Bu	69	58
H	89	17
<i>i</i> -Pr	58	5
OBn	95	24



Oxidative Coupling- Kozlowski

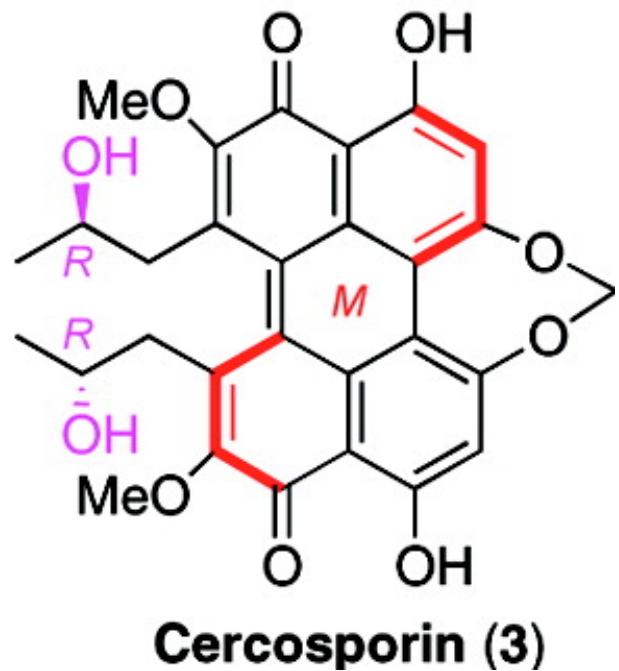
Kozlowski, 2001



Entry	R^1	Catalyst	Yield (%)	ee (%)
1	CO_2Me	(S,S)	85	93 (R)
2	CO_2Me	(R,R)	82	92 (S)
3	CO_2Bn	(S,S)	79	90 (R)
4	$\text{CO}_2n\text{-Hx}$	(S,S)	70	87 (R)
5	$\text{CON}(\text{CH}_2\text{CH}_2)_2\text{O}$	(S,S)	61	75 (R)
6	COPh	(S,S)	88	89 (R)
7	OBn	(R,R)- CuBr	74	46 (S)
8	$\text{P}(\text{O})(\text{OMe})_2$	(S,S)	76	92 (R)
9	$\text{SO}_2\text{C}_6\text{H}_4-p\text{-OMe}$	(S,S)	75	57 (R) (98, triturated)

Li, X.; Yang, J.; Kozlowski, M. C. *Org. Lett.* **2001**, 3, 1137.

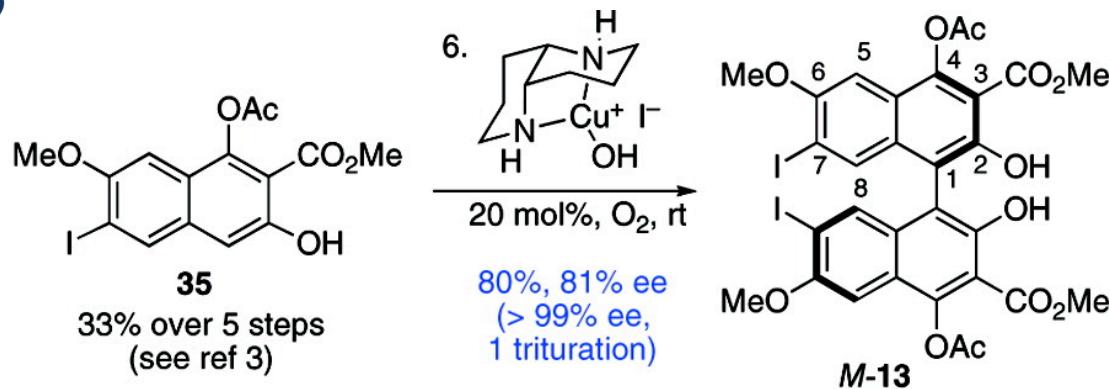
The Perylenequinones



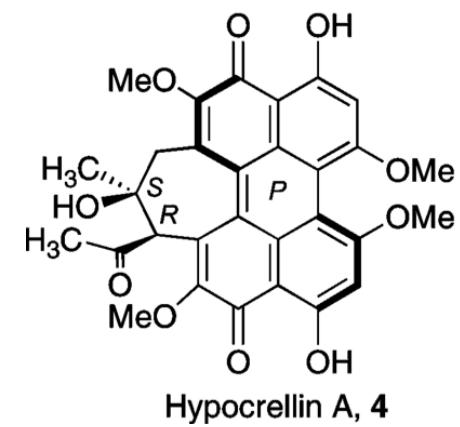
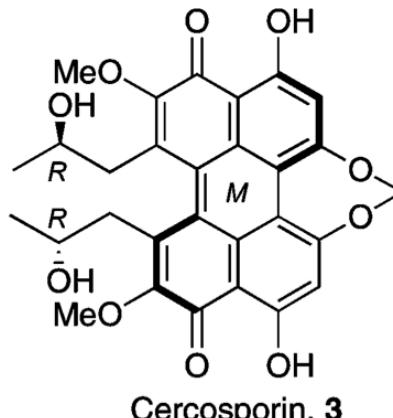
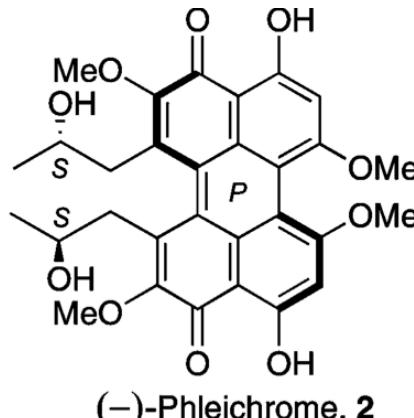
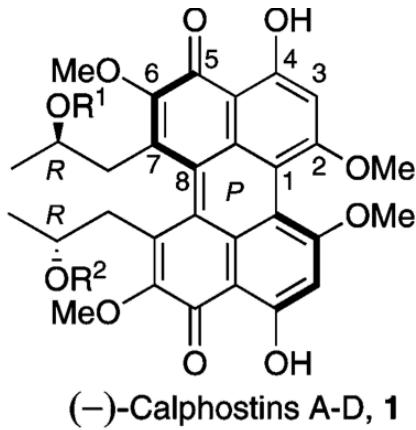
- Commonly isolated fungal natural products. Sources include *Cercospora kikuchii*, cause of soy bean “purple speck disease.”
- Possess helical chirality about the core pentacycle. Atrop-stability varies among members of the family.
- Light-induced biological activity (singlet oxygen generation, ROS) makes them potential photodynamic therapeutics.

Perylenequinone Syntheses

Kozlowski, 2009

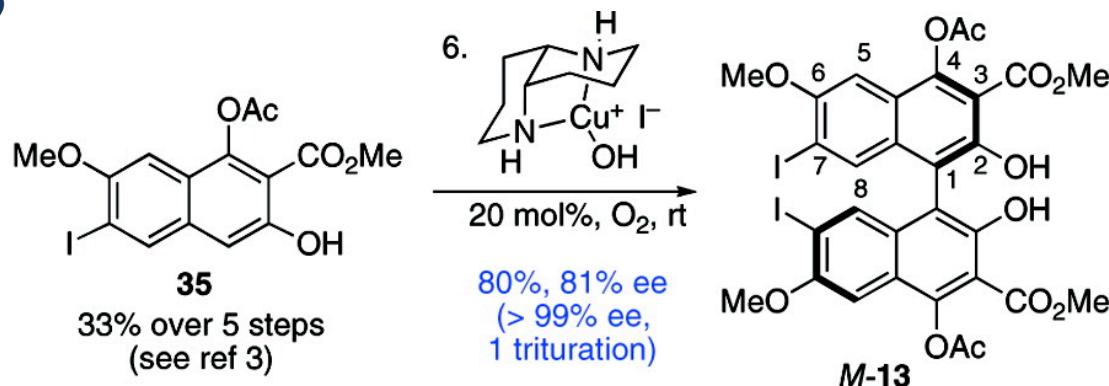


Prepared from common intermediate:

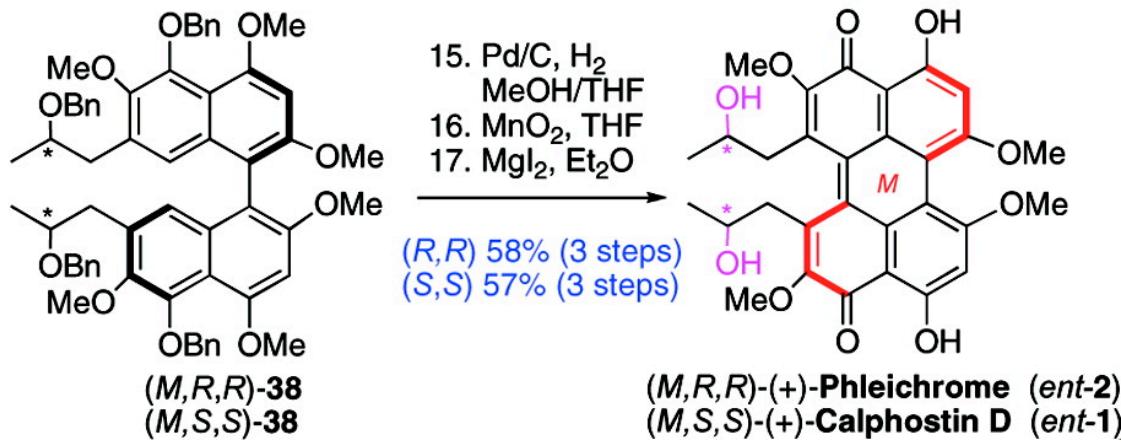


Perylenequinone Syntheses

Kozlowski, 2009

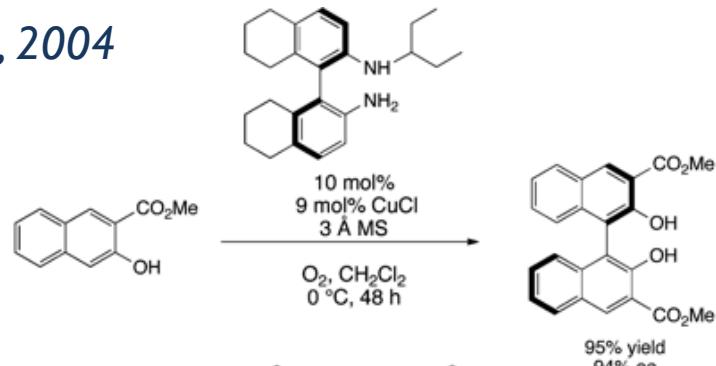


Interesting bisquinone closure with MnO₂:

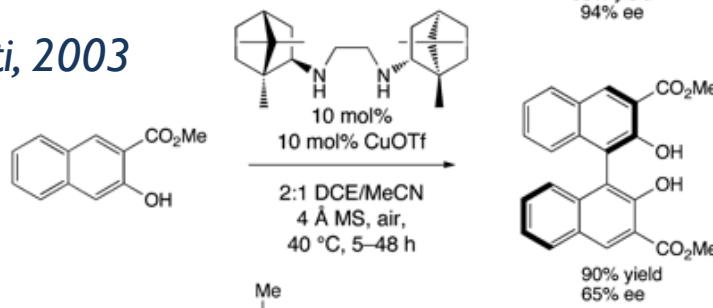


Oxidative Couplings

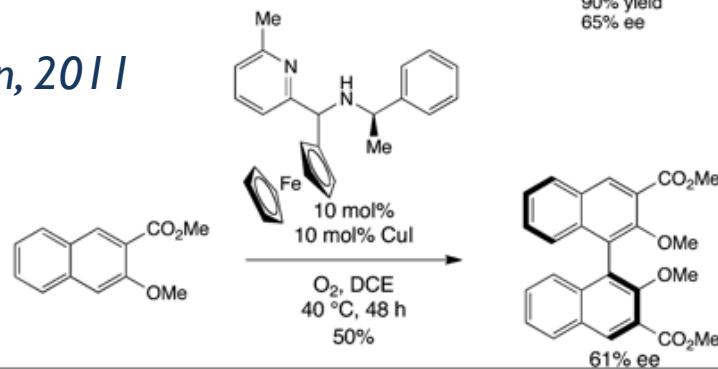
Ha, 2004



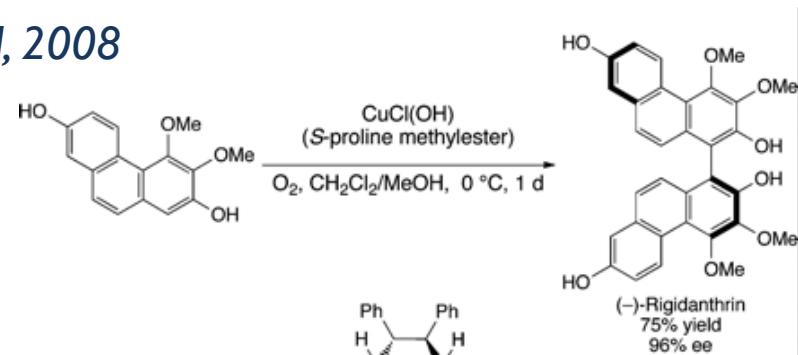
Pilati, 2003



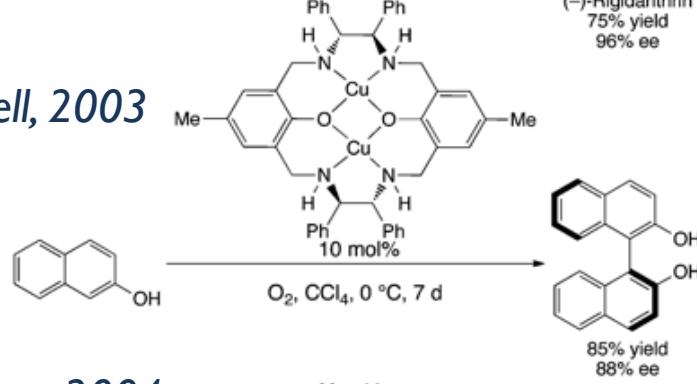
Troin, 2011



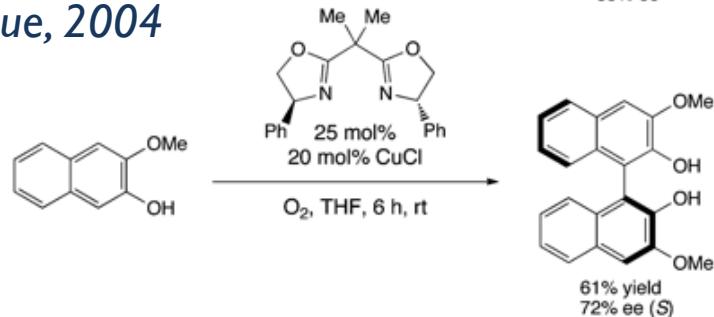
Pal, 2008



Martell, 2003



Habaue, 2004



Ha, D.-C., et al. *Tetrahedron* **2004**, 60, 9037.

Pilati, T., et al. *Tetrahedron: Asymmetry* **2003**, 14, 1451.

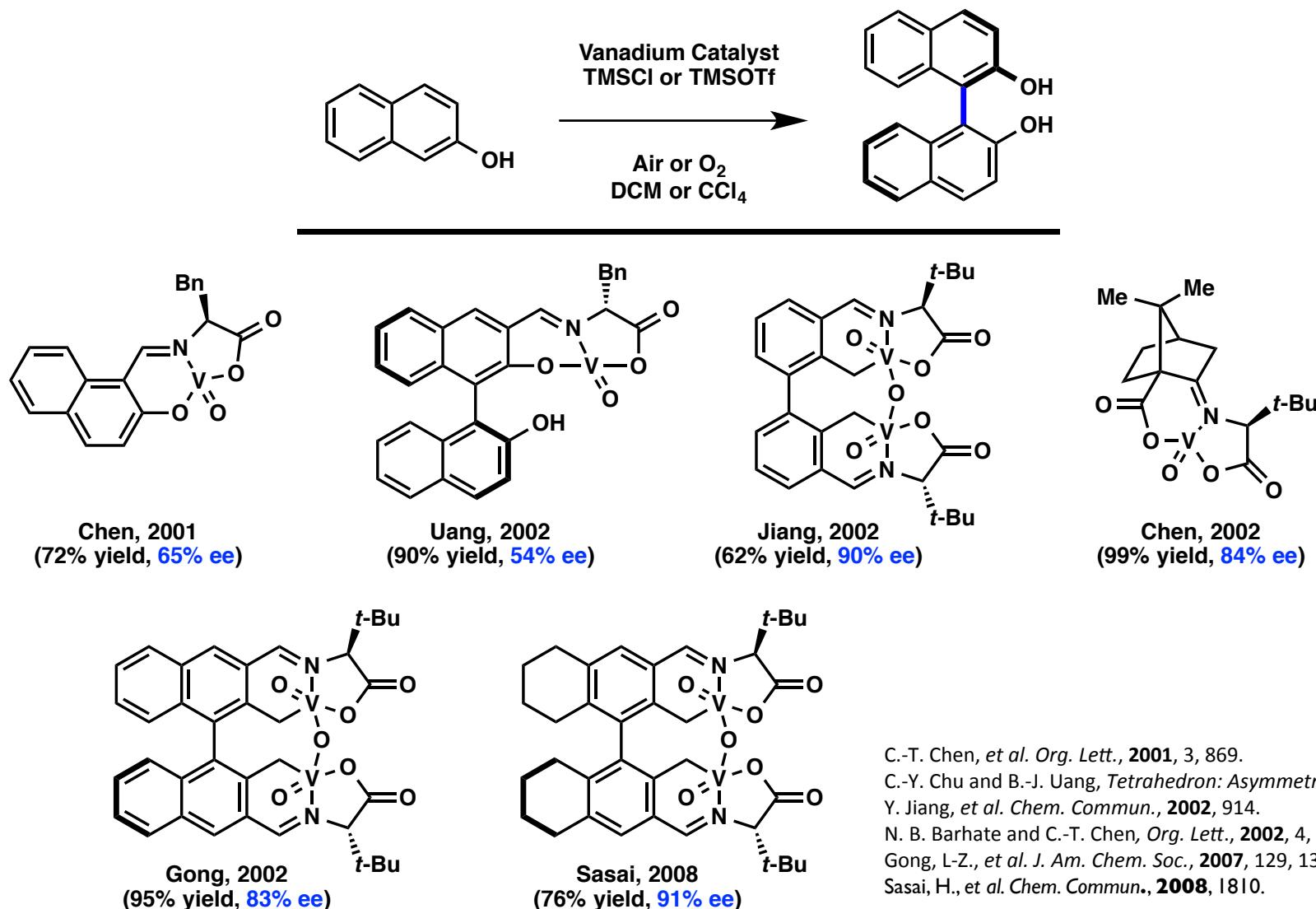
Troin, Y. et al. *Organometallics* **2011**, 30, 4047.

Pal, S., et al. *J. Indian Chem. Soc.* **2008**, 85, 1116.

Martell, A. E., et al. *Angew. Chem., Int. Ed.* **2003**, 42, 6008.

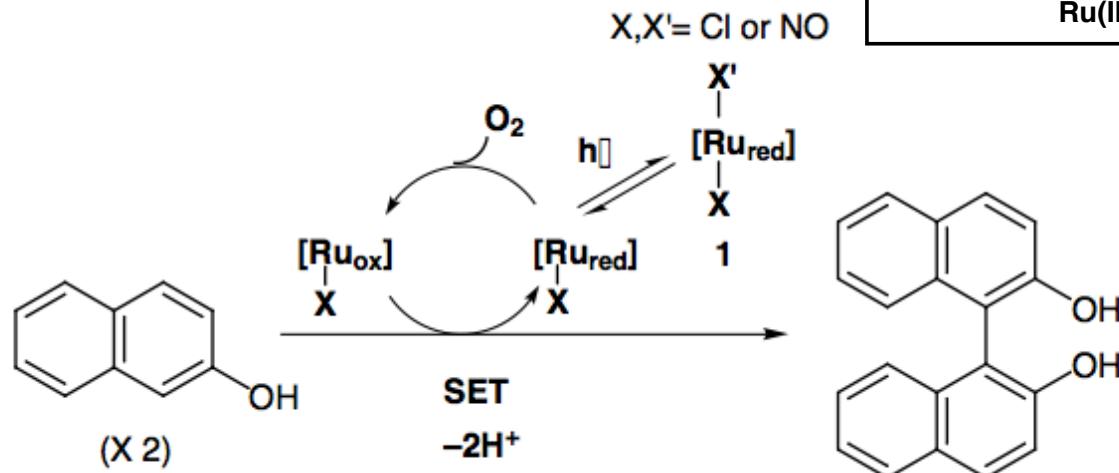
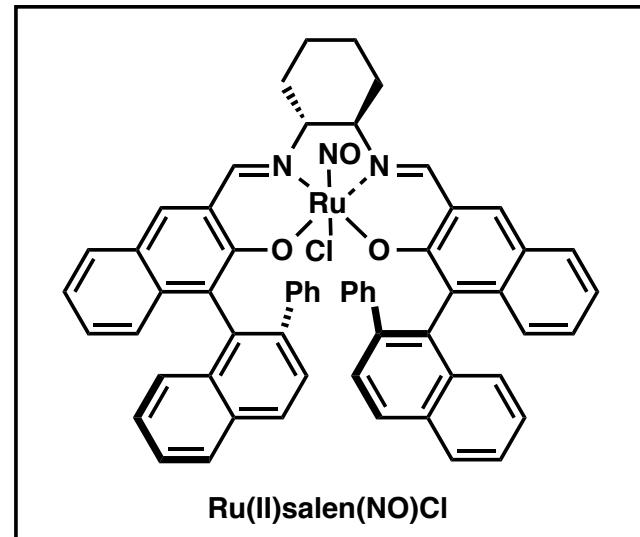
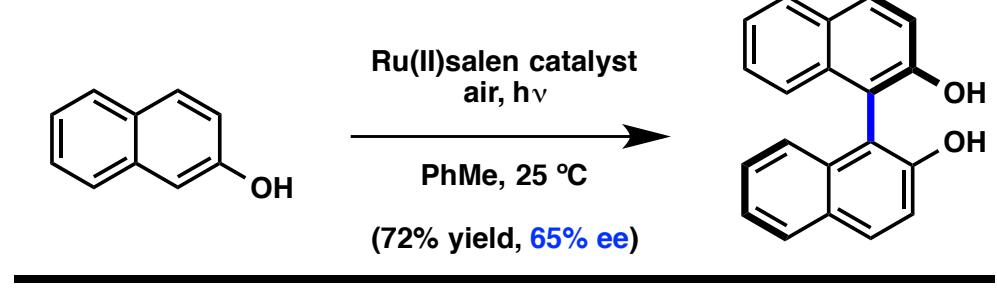
Habaue, S., et al. *Polym. Sci., Part A: Polym. Chem.* **2004**, 42, 4528.

Oxidative Coupling- Vanadium



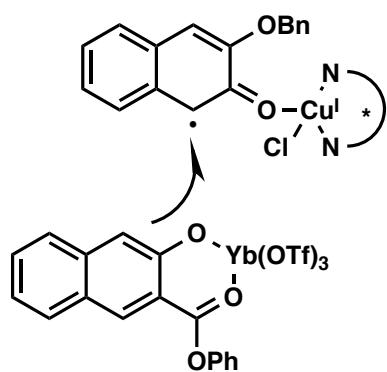
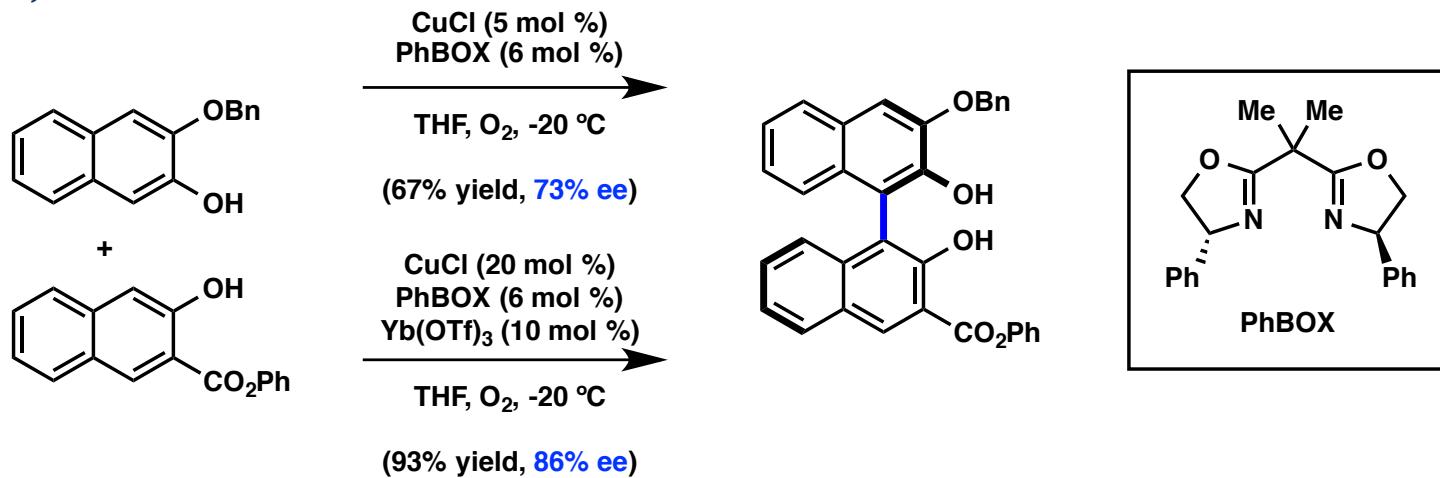
Oxidative Coupling- Ruthenium

Katsuki, 2000



Oxidative Heterocoupling

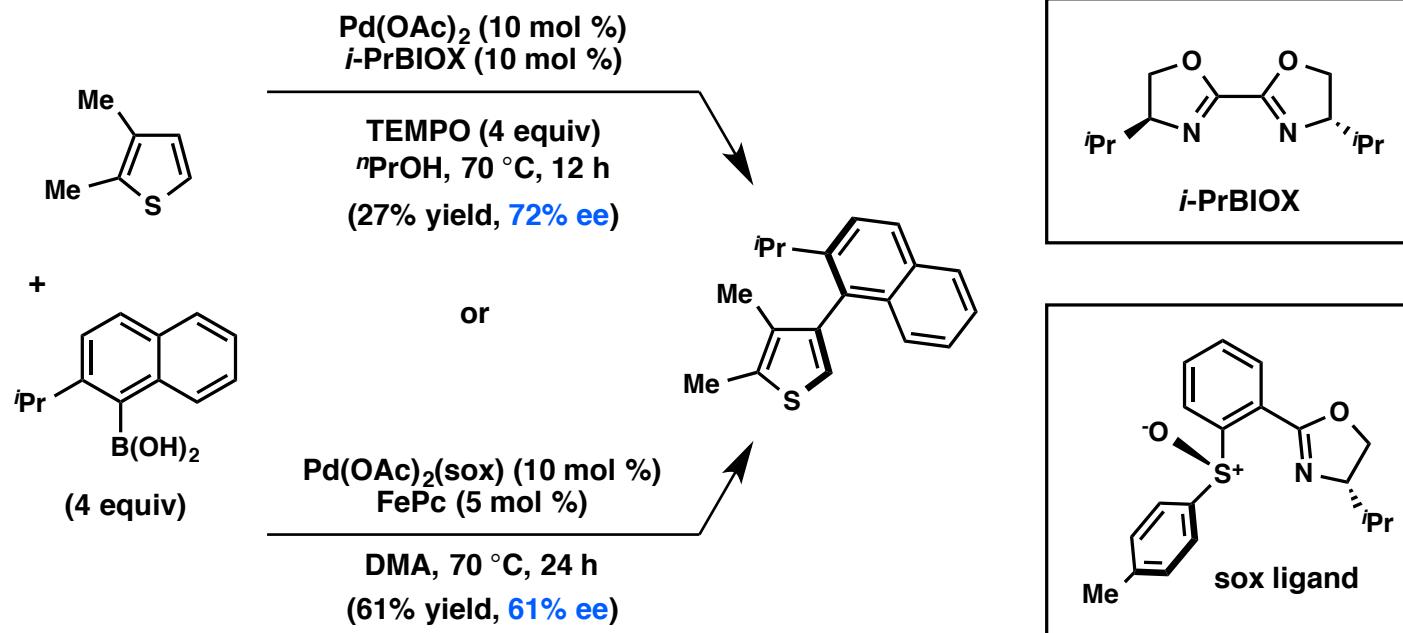
Habaue, 2007



- Selective oxidation of more electron-rich naphthol
- Cross-selectivity achieved by activating chelating substrate with Lewis acid
- No mechanistic details reported

Oxidative Heterocoupling

Itami, 2012



Yamaguchi, K.; Yamaguchi, J.; Studer, A.; Itami, K. *Chem. Sci.* **2012**, 3, 2165.

Yamaguchi, K.; Kondo, H.; Yamaguchi, J.; Itami, K. *Chem. Sci.* **2013**, 4, 3753.

Conclusion

- Metal-catalyzed asymmetric biaryl cross-coupling has developed into a rich and synthetically useful field.
- Both redox-neutral and oxidative methods have been optimized and employed in complex settings.
- Oxidative cross-coupling and non-Mg, non-B redox neutral cross-coupling are underdeveloped emerging fields with significant potential.

Resources

- Kevin Allen's group meeting, 2005:
 - Excellent coverage of diastereoselective methods and chiral leaving groups
http://stoltz.caltech.edu/seminars/2005_Allan.pdf
- B. Collins (Denmark) group meeting, 2004:
 - Detailed descriptions of Vanadium oxidative reactions and aryl-Pb couplings
http://www.scs.illinois.edu/denmark/presentations/2004/gm-2004-03_02.pdf
- M. Bruening review, 2011
 - Atropselective Total Synthesis of Axially Chiral Biaryl Natural Products
Chem. Rev., 2011, 111 (2), pp 563–639
- M. C. Kozlowski review, 2013
 - Aerobic Copper-Catalyzed Organic Reactions
Chem. Rev., 2013, 113 (8), pp 6234-6458